Initial data for determining the thermodynamic functions of $Ca_2P_2O_7$, $Sr_2P_2O_7$, $Ba_2P_2O_7$, and $Ra_2P_2O_7$ by the Neumann–Kopp and the Kellogg–Kubaschewski methods

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Table 1

Heat capacity of the initial solid compounds under standard conditions

Compound	C_p, $J \cdot mol^{-1} \cdot K^{-1}$	Compound	C_p, $J \cdot mol^{-1} \cdot K^{-1}$	Compound	C_p, $J \cdot mol^{-1} \cdot K^{-1}$
CaO	42.122 [1]	P_4O_{10}	211.710 [1]	$Ca_2P_2O_7$	187.763 [1, 4]
SrO	45.15 [2]				
BaO	47.278 [1]				
RaO	46.4 [3]				

Table 2

Melting point of selected $Ca_2(Sr_2, Ba_2, Ra_2)P_2O_7$ pyrophosphates

Compound	$T_{m.p.}$, K	Source
$Ca_2P_2O_7$	1631	[1]
$Sr_2P_2O_7$	1648	[5]
$Ba_2P_2O_7$	1703	[5]
$Ra_2P_2O_7$	1676	[result of these authors]

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