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QUALITY CHARACTERISTICS OF QUERCUS MACEDONICA, Castanea sativa Mill. VS Quercus Alba IN ORDER TO PRODUCE AN INNOVATIVE BALSAMIC VINEGAR PRODUCT

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Article history:	ABSTRACT
Received:	Acetic fermentation is a vital function for microorganisms. Since antiquity,
5 May 2020	vinegar has been used both to dressing food, but also to characterize the
Accepted:	irritable people. The aim of this study was to investigate the quality
1 October 2020	characteristics of Greek oak (QuercusMacedonica - QM) and Greek
Keywords:	chestnut wood (Castanea sativa Mill CsM) in comparison with an
Castanea sativa Mill. (CsM);	American oak (Quercus Alba - QA) in order to produce vinegar from grape
Environmental indicators;	and apple. This study was conducted to make known the properties both of
Green product;	the Geek Macedonian Quercus and chestnut wood. All official OIV methods
Quercus Alba (QA);	were used to implement this study. Volatile acidity increased in the apple
Quercus Macedonica (QM);	vinegar more in QM and less in QA and CsM. In wine vinegar an increase
Vinegar.	in volatile acidity was observed at about the same level of 28 grams per liter
	expressed in acetic acid. It has therefore been observed that QM does not
	behave in the same way and rhythm as other forest species. We observe that
	QM does not release its ingredients quickly, so it is recommended for long
	aging of both wines and vinegars. QA and CsM release components from
	the first month. Polyphenols show a graduation from QM (about 1000 mg/lt)
	to CsM (approximately 15.500 mg/lt). All experimental data were confirmed
	by liquid chromatography. It was found that QM is best suited for aging
	apple vinegar. This study helps us to keep vinegar in every type of barrel, so
	to produce innovative balsamic vinegar.

1.Introduction

Acidic fermentation is responsible for sulfuric bacteria. These, through the enzymes contained in their cell, transfer oxygen to the alcohol to be oxidized until it is transformed into acetic acid. The history of vinegar is very great. The inhabitants of Attica in Greece called vinegar "idos" which mean pleasure or pleasant to taste. In ancient Lacedaemon (Greece), from the time of the king and legislator of Lykourgos in the 8th century BC century, people and army through the common soups eaten the famous «black broth» and gave them the reputation of prolific warriors. The Roman Empire set out an excellent cuisine in which the vinegar had a prominent place. Each house at the time of the Roman Empire had its own wine cellar (acetumlocum). The Romans used the vinegar to make "posca", the drink of the legionnaires. After they first mixed the vinegar with water, they drank it in order to maintain their strength in the constant campaigns sent by the Roman Empire. Vinegar participates in many metabolic pathways of our organism. Finally, it has a place in local customs, especially in the Holy Week, perhaps because it is connected with the remembrance of the crucifixion and the offering of vinegar by the Roman soldier to the crucified Christ. Acetic acid is the main component of vinegar that plays an important role in the release of energy from fats and carbohydrates. It also participates in the building of fats and amino acids. In the body it is transported with blood to the liver and tissues and undergoes complete oxidation with energy release (Kawa-Rygielska et al, 2018). In recent years there has been much talk about the balsamic vinegar of the Italian city of Modena, which is increasingly used in our kitchens. But long time is not the only thing that makes it so delicious and aromatic. It is aging in barrels made of wood of different species and in which vinegar is stored in succession. The maturation within them enriches it with a particular flavor. Polyphenols play an important role in giving flavour and special character to both wines and vinegars (Cerezo et al 2010, Del Alamo Sanza et al 2004, Sanzet al 2012a, Sanzet al2012b, Zhang et al Figueiredo-González et 2015. al 2014. Martínez-Gil et al 2019, Miriam Sanz et al 2011, Psarra et al 2015, Schwarz et al 2009, Tesfaye et al 2004, Bautista-Ortín et al 2008, Cerezo et al 2008, Cerezo et al 2010, Sarni et al 1990, Kanakaki et al 2015, De Rossoet al 2009, Sanz et al 2010, Alañón et al 2011, Karvela et al 2008). While acetic acid, in its pure form, is not so aromatic, it has a distinctly strong odor and irritates the olfactory centers when it comes into contact with them. Rapid induction of ageing character in brandy products is described many scientific projects (Van aarsveld et al 2009). Vinegars derived from sherry winesusually referred in bibliography (García Parrilla et al 1999, Alonso et al 2004). An innovative product of beer vinegar has been recently produced (Mudura et al 2018). In recent works, the chemical characteristics of the Greek CsM have identified (Kakavas et been al 2018. Chavenetidou et al 2019). In the present work we tried to produce an innovative vinegar product by dipping three different types of wood into wine and apple vinegar. It's the first attempt to use *Quercus Macedonica* for artificial aging of vinegars.

2. Materials and methods

2.1.Materials

2.2.1.Samples

QM was given to us by the forestry authority of the Prefecture of Kozani and comes from Pindosmountain in Greece. The other types of wood and vinegars were purchased from the Greek market.We used glass bottles with a capacity of 300ml, in which 250ml of apple or wine (white) vinegar and 35gr of cubed wood were placed. Monthly analyzes were conducted for seven months. Total acidity (gr/lt tartaric acid), volatile acidity by Jaulmes (gr/lt acetic acid) and dry mass were (%) determined based on Hitos P. et al. Polyphenols were determined by the Folin-Ciocalteumethod: initially, the standard curve (Y=0,0014X+0,0015 R²=0,993) was determined with various concentrations of gallic acid. Absorption was set at 765nm.Chromatograms data was performed with HPLC Agilent 1200.All official OIV methods were used to implement this study.

3.Results and discussions

3.1.Chemical Properties of Vinegar Samples *3.1.1. Total acidity*

Total acidity has a declining trend for QM, with an increasing trend in QA and CsM before and after wood dipping in apple and wine vinegar (Figure 1). Total acidity was expressed in tartaric acid.





Figure 1. Total acidity of apple vinegar (left) and white vinegar (right).

3.1.2. Volatile acidity

Volatile acidity increased in the apple vinegar more in QM (28.7 gr/lt) and less in QA and CsM. In wine vinegar an increase in volatile acidity was observed more in CsM(29.1 gr/lt), (Figure 2). Volatile acidity expressed in acetic acid.



Figure 2.Volatile acidity of apple vinegar (left) and white vinegar (right).

3.1.3. Total dry mass

Total dry mass in both apple and wine vinegar is doubled in QA and CsM in relation to QM. It has therefore been observed that QM does not behave in thesame way and rhythm as other forest species (Figure 3).





Figure 3. Total dry mass of apple vinegar (left) and white vinegar (right).

3.2.Biochemical Properties of Vinegar samples

3.2.1. Total polyphenols

It is observed that CsM yields about 15,730 mg / lt, QA 9,990 mg / lt and QM 1020 mg / lt total polyphenols for seven months period, representing the average in the case of apple vinegar and grape vinegar (Figures 4, 5). We observe that among the three forest species, CsM yields most polyphenols in both apple vinegar and grape vinegar. In contrast, QM

yields fewer polyphenols. Polyphenols was expressed in gallic acid.



Figure 4. Variation of polyphenols during aging in apple vinegar



Figure 5. Variation of polyphenols during aging in grape vinegar

3.2.2.HPLC chromatograms

If we compare the results of blank (wine vinegar) chromatogram with QM, QA and CsM samples (Figure 6 a - d) we see that, 6a chromatogram shows a limited number of peaks. 6b chromatogram shows all the peaks of the blind (6a) but in addition we can see peaks with deferent retention times. From 15 to 20, 22 to 29

and 53 to 56 minutes we can see many peaks. 6c and 6d chromatograms have a variety of peaks and verify that both QA and CsM have a large amount of polyphenols (see also Figure 4 and 5).



Figure 6.a) blank chromatogram (vinegar from wine), b) QM chromatogram, c) QA chromatogram, d) CsM chromatogram.

4. Conclusions

QM can create excellent balsamic vinegar from apples because the increase in volatile acidity has reached high levels(28.7 gr/lt). In contrast CsMcan create excellent balsamic vinegar from wine because the increase in volatile acidity has reached also very high levels(29.1 gr/lt). Total dry mass reached 2.52% in CsM and over doubled from its initial number (1.1%), in the case of apple vinegar. Total dry mass reached 2.04% toQA (2.02% to CsM) in the case of wine vinegar. CsMattributed more polyphenols to both apple vinegar (16.626 mg/lt)and wine vinegar (14.840 mg/lt). For this reason, CsM and QM can be used additionally in aging of wines and balsamic vinegars. This study helps us to keep vinegar the proper time period in every type of barrel, so to produce innovative balsamic vinegar.Our work revealed that Greek chestnut (CsM) showed much better quality characteristics than QA, in contrast to QM which did not yield rich quality characteristics.

5.References

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