Influence of sublittoral microphytobenthos on the oxygen and nutrient flux between sediment and water: a laboratory continuous-flow study

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ABSTRACT: Influence of sublittoral microphytobenthos on the flux of oxygen and inorganic nutrients (N, P, Si) at the sediment-water interface was studied using undisturbed cores of sandy and muddy sediment incubated in a laboratory continuous-flow system, either in darkness or with a 16/8 h L/D cycle at in situ light level during summer. Sediment was collected in July at 15 m depth in a non-tidal, stratified bay in SE Kattegat. To test whether the higher content of inorganic nutrients below the halocline, compared with surface waters, could stimulate microphytobenthic growth, 2 levels of nutrient concentrations were used. Diel variations were found in L/D cores, but not in darkened cores, for oxygen, dissolved inorganic nitrogen and phosphorus content in the water overlying the sediment. The flux of NH_4^+ , NO_3^- and PO_4^{3-} out of the sediment decreased during light periods and occasionally a net uptake was recorded. Light-induced O_2 production, and correlations between Δ fluxes (differences between day and night fluxes of O2 and nutrients), chlorophyll a content and algal cell numbers in the sediment, indicate that the decreased outflux of IN and PO43- was mediated by photosynthetic organisms. Diel variations were not studied for silicon, but a significantly lower outflux, or even an uptake, of Si(OH)₄ from L/D cores supports this conclusion. This suggests that diatoms play a major role in the nutrient flux between sediment and water. Also, the differences in pore-water nutrient gradients between L/D and dark cores point to the importance of sediment-associated organisms. Daily (24 h) net fluxes of nutrients were primarily out of the sediment, but the magnitude depended on both light conditions and sediment type. Daily net outflux was significantly lower in L/D cores than in darkened cores for all nutrients except NO₃⁻ in muddy cores and NO₂⁻ in sandy cores. Net uptake in L/D cores was recorded for Si(OH)₄ and NO₃⁻ in sandy sediment. Outflux of nutrients was significantly higher from muddy sediments in comparison with sandy sediments (except NO₃⁻), especially in permanent darkness. No significant effect of nutrient enrichment on the abundance of sublittoral benthic microalgae could be shown. Results suggest that microphytobenthos can influence sediment-water exchange of inorganic nutrients even at sublittoral depths, and when measuring nutrient flux in permanently darkened cores from depths around 15 m in the Kattegat, summer flux rates will be overestimated by a factor varying between 2 and 6, depending on sediment type.

INTRODUCTION

Sediments in shallow coastal areas are important sites for mineralization and recirculation of nutrients from the particulate organic matter to the water mass (e.g. Nixon 1981, Balzer 1984, Jensen et al. 1990 and references therein). Until recently, the influence of sediment-associated microalgae on the nutrient flux between sediment and water had been largely overlooked, whereas other factors, such as heterotrophic microbial and faunal activity, have long been recognized (for references see Aller 1988, Henriksen & Kemp

1988). Nutrient exchange between sediment and water has traditionally been studied in darkened cores or chambers, also for sediments in shallow waters, thus excluding the influence of photosynthetic organisms. In coastal waters shallow and/or clear enough to enable light penetration to the sediment surface, sediment-associated microalgae (microphytobenthos) can be expected to influence exchange of nutrients at the sediment-water interface, and this possibility has also been discussed (Henriksen et al. 1980, Granéli & Sundbäck 1985, Nowicki & Nixon 1985, Asmus 1986, Granéli & Sundbäck 1986, Simon 1988, Keizer et al.

1989, Jensen et al. 1990). Thus, photosynthesis of benthic microalgae has been suggested as one of the mechanisms (besides faunal bioturbation) explaining why pore water nutrient profiles cannot be used to predict sediment-water flux rates during conditions when light reaches the sediment surface (Blackburn & Henriksen 1983, Ullman & Aller 1989). Consequently, some recent investigations, both in marine and fresh water areas, have aimed at demonstrating the effect of benthic microalgae on sediment-water nutrient flux by comparing flux rates under different light conditions (Andersen & Kristensen 1988, Carlton & Wetzel 1988, Kelderman et al. 1988, Sundbäck & Granéli 1988, Rizzo 1990) or by manipulating benthic microalgal abundance in situ (Hansson 1989). Microalgal influence on sediment-water nutrient flux has been shown to be due both to nutrient uptake and the oxygenation of the sediment/water interface by microalgal photosynthesis (Jensen et al. 1984, Andersen & Kristensen 1988, Carlton & Wetzel 1988).

The question remains whether microbenthic autotrophs play a significant role in regulating sedimentwater nutrient fluxes in sediments at depths greater than a few meters. Considerable microphytobenthic biomass and primary productivity have been documented for depths of 15 to 20 m in coastal temperate areas (Bodin et al. 1985, Herndl et al. 1989, Riaux-Gobin et al. 1989). In a stratified bay in SE Kattegat (Laholm Bay) some of the highest values of microphytobenthic biomass and potential productivity were in fact found at 14 to 16 m, i.e. at or just below the sharp halocline (Sundbäck & Jönsson 1988). Better nutrient availability in sediments in deeper waters and in below-halocline waters, in combination with algal adaptation to low light levels, are some of the factors suggested to explain this type of vertical distribution of benthic microalgae (Stevenson & Stoermer 1981, Sundbäck & Jönsson 1988)

Our aim was to study the influence of sublittoral microphytobenthos on the flux of O₂ and inorganic nutrients (N, P, Si) at the sediment-water interface in a non-tidal, stratified bay by using undisturbed sediment cores incubated in a laboratory continuous-flow system at simulated in situ conditions of light, temperature and nutrients. We also tested whether the higher content of inorganic nutrients below the halocline, compared with surface waters, could stimulate microphytobenthic growth.

MATERIALS AND METHODS

Sampling. Sediment samples were taken by divers at a water depth of 15 m in the middle of Laholm Bay (56°35' E, 12°50' N) on 31 July 1986 (Fig.1). For hydrography and sediment characteristics see Enoksson

LAHOLM BAY

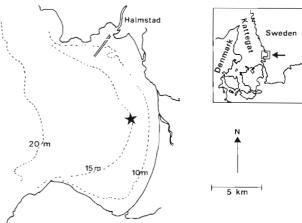


Fig. 1. Sampling site (star) in Laholm Bay, southeastern

et al. (1990) and references therein. Altogether 30 cores, around 15 cm deep, were taken with perspex tubes (8 cm ID) and placed in thermally insulated boxes. Surface water and water from ca 1 m above the bottom at the sampling site were collected with a 30 l van Dorn water sampler.

Continuous-flow system. Sediment cores were stored in the dark at 15°C for 4 d prior to start of experiment. Surface water was filtered through 1.2 µm Millipore membrane filters and concentrations of inorganic nutrients in both surface and bottom waters were analysed (see below). Cores that did not contain macrofauna were selected for the experiment. However, a few cores that did contain specimens of Cyprina islandica (3.4 to 6.8 cm length), were also used, after the mussels had been gently removed from the cores. In one core a C. islandica was found at the end of the experiment. No other macrofauna (animals > 5 mm) was seen either at the start or at the termination of the experiment. The water in the tubes above the sediment (referred to as headspace water) was carefully replaced by 1.2 µm-filtered surface water, with or without nutrient enrichments. Surface water enriched with inorganic nitrogen (KNO₃), phosphorus (Na₂HPO₄), and silicon (Na₂SiO₃·9H₂O), corresponding to concentrations in the bottom water, was used instead of the original bottom water to avoid other differences (e.g. in salinity and content of organic substances). Nutrient concentrations of the inflowing surface water were: 0.1 to 0.8 μ M NH₄⁺, 0.1 to 0.5 μ M NO₃⁻, 0.05 to 0.17 μ M NO_2^- , 0.1 to 0.2 μ M PO_4^{3-} and 3 μ M $Si(OH)_4$. Corresponding values of the 'bottom water' were: 0.1 to 0.6 $\mu M NH_4^+$, 8 to 9 $\mu M NO_3^-$, 0.04 to 0.13 $\mu M NO_2^-$, 0.9 to 1.3 μ M PO₄³⁻ and 55 μ M Si(OH)₄. Mean areal loading rates of nutrients for enriched water were 57 µmol

 NO_3 , 7.3 μ mol PO_4 ³⁻ and 363 μ mol $Si(OH)_4$ m^{-2} h^{-1} (Si:N:P ratio 50:8:1).

The 24 cores used for the sediment-water flux determinations were kept in water baths at $15\,^{\circ}\text{C} \pm 0.2\,\text{C}^{\circ}$ (the temperature of bottom water on the sampling occasion) and the headspace water was connected to a flow through (Fig. 2). Water for refilling supply tanks during the experiment, was stored dark at $4\,^{\circ}\text{C}$. The headspace water (ca 500 ml) was gently stirred with a magnetic bar (60 rpm) and continuously exchanged (ca $33\,\text{ml h}^{-1}$) using multichannel peristaltic pumps, so that the total volume was renewed every 14 to 16 h with either surface water (12 cores) or 'bottom water' (12 cores) (Fig. 2). Twelve cores (6 with surface and 6 with

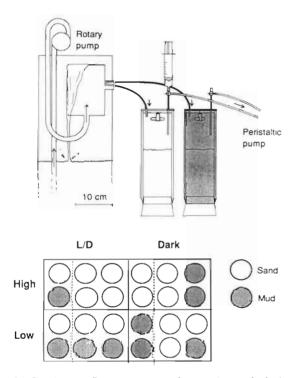


Fig. 2. Continuous-flow system and experimental design showing combinations of the 4 treatments: L/D = 16/8 light/dark cycle; Dark = permanent darkness; High = P, N and Si enriched water ('bottom water'); Low = nutrient-poor surface water. Circles left of dashed lines: cores sampled more frequently (see text)

'bottom' water) were exposed to a 16/8h light-dark cycle (referred to as L/D cores) at a photon flux density of 20 μE m $^{-2}$ s $^{-1}$ (= 1.2 E m $^{-2}$ d $^{-1}$) simulating in situ light conditions in July at 15 m in Laholm Bay (Sundbäck & Jönsson 1988). The other 12 cores (6 with surface and 6 with 'bottom' water) were kept dark (referred to as dark cores). Thus, the experiment comprised 4 treatments: dark cores with low or high nutrient concentration, and L/D cores with low or high nutrient concentration (Fig. 2). For practical reasons,

8 cores (2 for each treatment) were sampled more frequently. The curves for individual cores shown under 'Results' (Fig. 4a to e) are based on these cores. The other cores (4 for each treatment), were sampled less frequently. Of the less frequently sampled cores (4 for each treatments), 8 cores were used for analyses of sediment and microphytobenthos in the middle of the experiment. For statistical purposes, the calculation of 24 h net flux rates were based on 16 cores (see under 'Statistical analysis'). The experiment was run for 14 d (5 to 18 Aug 1986).

Analyses of headspace water. Samples of inflowing and of headspace water were taken with syringes through valves twice a day, at the end of the light and dark period, respectively. NH4+, NO3-, NO2- and PO₄³⁻ were analyzed using a Technicon Autoanalyzer as previously described by Enoksson & Samuelsson (1987) and Enoksson (1987). Because of analytical problems, NH₄⁺ was measured only up to Day 8. Silicon was analysed once a day according to the method described in Grasshoff et al. (1983). O2 concentration was measured with a Clark-type electrode (Hansatech, Kings Lynn, UK) using 2×5 ml subsamples. The electrode was calibrated by Winkler titrations. Flux rates were calculated according to the formula No. 6 in Propp et al. (1982), with the only difference being that we let outflux from the sediment be positive. The input concentration C_0 (supply vessel) varied slightly with time, and therefore the mean value of the initial and final value for each period was used. A correction for the extraction of water samples for analysis was made as follows: For a period between 2 measurements, the final concentration in the extracted water sample and the initial concentration $(C(t_0))$ was derived from the formula

$$C(t_0) = \frac{(\omega - E) \cdot C_b + E \cdot C_0}{\omega} \tag{1}$$

where ω = headspace volume; E = sample volume; C_b = concentration in the extracted water sample on the first occasion; C_0 = input concentration. For the conditions used, this approximation deviated $\ll 1$ % from a corresponding logarithmic formula. Differences in flux rates between light and dark periods in L/D cores were calculated and are referred to as Δ flux. Mean flux rates for 24 h periods (= 24 h net flux) were calculated as

$$\frac{\text{Flux}_{\text{dark period}} + 2 \cdot \text{Flux}_{\text{hght period}}}{3} \tag{2}$$

Analyses of sediment characteristics, pore water and microphytobenthos. Data on sediment variables rely on analyses of cores that were sliced before (2 cores), in the middle (8 cores on Days 6 and 7) and at the end of the experiment (16 cores, Day 13–14). The topmost 1.5 cm of each sediment core was sliced into 2

layers (0 to 0.5 cm and 0.5 to 1.5 cm) and subsamples for water content, particulate carbon (PC), nitrogen (PN), chlorophyll a content and number of algal cells were taken from the homogenized slices using cut-off 2 ml syringes. Pore water was obtained from the remainder of each slice by centrifugation at 2000 rpm and filtration through glass fibre filters. The reliability of the subsampling for chlorophyll and pheopigment determination was tested by analysing 15 replicate samples taken from the same slice. The coefficient of variation was 9.4 % for chlorophyll a and 15.9 % for pheopigment. Water content was measured by drying sediment at 105 °C. PC and PN were analysed using a Carlo Erba 1106 CHN analyser. Pore water samples were analysed for inorganic nutrients as described earlier.

Chlorophyll a content in the top 0.5 cm sediment was measured spectrophotometrically according to Lorenzen (1967) after extraction with 90 % acetone overnight followed by 5 min ultrasonication. The number of living autotrophic cells in the top 0.5 cm was counted in a Bürker counting chamber using epifluorescence microscopy (Sundbäck et al. 1990).

Statistical analysis. The fact that the sediment cores represented 2 clearly different types of sediment (see under 'Results') added an unexpected source of variation to the data sets. The 2 sediment types were not evenly distributed among the 4 treatments (Fig. 2), hampering the statistical analyses, and only 16 cores could be used in the Newman-Keuls multiple comparison test (NK). Average values given in the text comprise all available cores. To reduce non-normality of data and heterogeneity of variances, data were transformed as $\ln (x+1)$ or $\sqrt{x+1}$ (see further Green 1979)

before subjected to the NK-tests. In cases where the transformation was not successful, the non-parametric Wilcoxon signed rank test was applied. In some cases the non-parametric Mann-Whitney U-test (M-W U-test) was also used. Unless otherwise stated, differences are accepted as significant when p < 0.05.

RESULTS

Sediment characteristics

Although all sediment cores were taken within a distance of less than 50 m in Laholm Bay, 2 clearly different types of sediment were obtained: sand with an average water content of 30 % (15 cores) and mud with a water content of 90 % (9 cores) (Table 1). The contents of particulate carbon and nitrogen were an order of magnitude higher in the muddy than in the sandy sediment, but the mean C/N mole ratio was about the same for both sediment types (9.5 to 9.7) (Table 1).

Pore water

The mean inorganic nutrient concentration in the pore water of the top 0.5 cm did not differ significantly between the 2 sediment types, except for PO_4^{3-} . The concentration of PO_4^{3-} was significantly higher (M-W U-test) in sandy than in muddy sediment (Table 1). This resulted in a lower mean N:P ratio (8:1) for the sandy than for the muddy sediment (22:1) (Table 1).

Values from 16 cores were used to illustrate profiles of inorganic nutrients, using mean concentrations for

Table 1. Contents of water (not corrected for salinity), carbon (PC), nitrogen (PN), chlorophyll a, pheopigment, number o
microalgal cells, and inorganic nutrient concentrations in pore water in the top 5 mm of 2 types of sediment used in the experimen

Characteristic	Sand			Mud		
	Mean	SD	(n)	Mean	SD	(n)
Water content %	32.1 ±	5.4	(17)	89.3	± 2.4	(9)
PC, µmol g ⁻¹ dry wt	253.6 ±	105.3	(7)	3675.1	± 127.4	(4)
PN, μmol g ⁻¹ dry wt	26.8 ±	11.1	(7)	379.7	± 22.0	(4)
CIN mole ratio	9.5 ±	0.3	(7)	9.7	± 0.3	(4)
Chl a, mg m ⁻²	31.2 ±	16.3	(17)	87.5	± 54.9	(9)
Pheopigment, mg m ⁻²	26.5 ±	10.9	(17)	37.5	± 15.0	(9)
Algal cells, \times 10 ⁶ cm ⁻²	0.66 ±	0.24	(15)	1.58	± 0.71	(7)
Conc. in pore water, µM:						
PO ₄ 3-	7.1 ±	3.6	(10)	2.9	± 0.9	(6)
NH ₄ ⁺	52.8 ±	21.7	(10)	58.6	± 30.2	(4)
NO ₃	2.9 ±	2.0	(10)	2.5	± 2.5	(6)
NO ₂	1.3 ±	1.3	(10)	1.8	± 1.4	(6)
Si(OH) ₄	233.1 ±	72.4	(10)	175.9	± 54.2	(6)
Si:N:P	32:8:1			60:22:1		

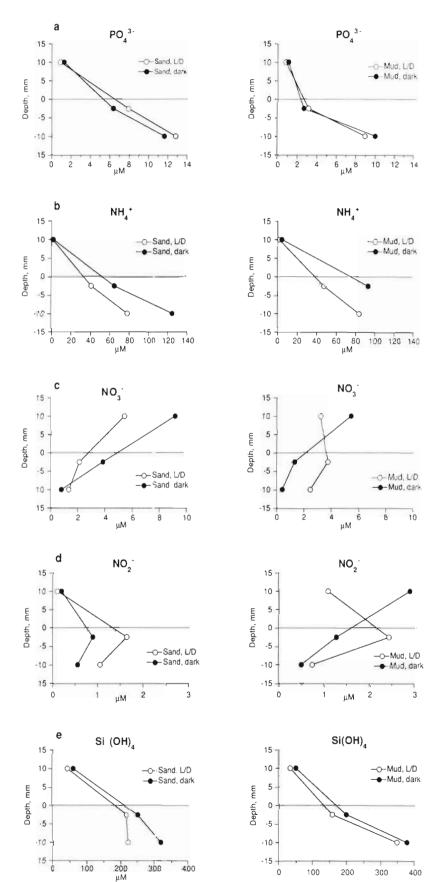


Fig. 3. Nutrient profiles for L/D and dark cores of sandy and muddy sediment. Values based on mean concentrations of headspace water (above 0-line) and pore water from the 0 to 0.5 cm and 0.5 to 1.5 cm sediment layers. Sand L/D, n=5; sand dark, n=5; mud L/D, n=3; mud dark, n=3. For treatments see Fig. 2

the headspace water, and for pore water from the 0 to 0.5 cm and 0.5 to 1.5 cm sediment layers (Fig. 3). For nitrate, ammonium (both sand and mud), and silica (only sand), there was a tendency for less steep gradients in L/D cores, when compared with dark cores (Fig. 3b, c, e). This tendency could, however, be statistically shown only for NH_4^+ in sandy sediment (M-W Utest, p = 0.05). No similar trend was seen for PO_4^{3-} , but the upper part of the gradient was less steep for muddy sediment, when compared with sandy sediment (Fig. 3a). A NO_2^- peak in the 0 to 0.5 cm layer was observed for both sandy and muddy light cores, as well as for sandy dark cores. The accumulation of nitrite in the headspace water of muddy dark cores (Fig. 5e) resulted in a steeply decreasing NO_2^- gradient.

Chlorophyll a and microphytobenthic abundance

At the termination of the experiment chlorophyll a content of the top 0.5 cm sediment was 3 times higher in the muddy than in the sandy sediment (significant difference with M-W U-test, p < 0.009) (Table 1). Nutrient additions did not significantly affect pigments or cell numbers. Although chlorophyll a values were lower in dark cores than in L/D cores (mean values 24 vs 33 mg m $^{-2}$ in sand and 65 vs 116 mg m $^{-2}$ in mud), the difference was not significant (M-W U-test). Pheopigment content was slightly higher in muddy sediment, but did not vary significantly (M-W U-test) between treatments (Table 1).

The number of viable (fluorescing) cells varied between 0.4 and 3×10^6 cm $^{-2}$. The number of algal cells was positively correlated with chlorophyll a content (r=0.95, n=20, p<0.005). Average number of viable cells was 2.5 times higher in the muddy sediment (1.58×10^6 cells cm $^{-2}$) than in the sandy sediment (0.66×10^6 cells cm $^{-2}$; M-W U-test, p<0.002). Significantly higher algal cell numbers (almost by a factor of 2) were also found in sandy L/D cores when compared with sandy dark cores (M-W U-test, p<0.05). In sandy sediment, the highest values for chlorophyll a (61~mg m $^{-2}$) and number of algal cells ($1.13\times 10^6~$ cm $^{-2}$) were found in a core that contained the mussel Cyprina~islandica.

Oxygen flux

There was diel variation in the O_2 concentrations of the headspace water in L/D cores, but not in dark cores. The minimum O_2 concentrations in the dark cores were around 150 µmol O_2 l⁻¹, while for the L/D cores the range of O_2 concentrations was 200 to 350 µmol O_2 l⁻¹ The O_2 saturation value at 15 °C and 20 % is 279 µmol O_2 l⁻¹ Thus, extreme values for the O_2 content above

the sediment were approximately 50 to $125\,\%$ of the saturation value.

The maximum Δ O_2 flux (light minus dark flux rate in L/D cores) was as high as 1400 μmol O_2 m^{-2} h^{-1} , a value that was obtained in a muddy core. The flux in this core varied between a net O_2 production of 800 μmol O_2 m^{-2} h^{-1} and a net consumption of 1080 μmol O_2 m^{-2} h^{-1} and a net consumption of 1080 μmol O_2 m^{-2} h^{-1} (Fig. 4a). Δ O_2 flux was also high in the core containing a mussel (960 μmol m^{-2} h^{-1}) (Fig. 4a). The average Δ O_2 flux was 290 \pm 220 (SD) μmol m^{-2} h^{-1} if these 2 cores were excluded. The O_2 consumption during night in the L/D cores (mean 560 μmol m^{-2} h^{-1}) was not significantly different (M-W U-test) from the O_2 consumption in the dark cores (mean 610 μmol m^{-2} h^{-1}). Extreme recordings of O_2 consumption for dark cores were 240 and 1480 μmol m^{-2} h^{-1}).

The mean 24 h net O_2 uptake was always lower for the L/D cores (290 µmol m⁻² h⁻¹) than for dark cores (600 µmol m⁻² h⁻¹) (Fig. 5a). The difference was significant from Day 9 and onwards for muddy cores (NK) but never for sandy cores (NK). Muddy dark cores consumed, on an average, more O_2 (670 µmol m⁻² h⁻¹) than sandy dark cores (550 µmol m⁻² h⁻¹), mainly due to a significant (nearly 2-fold) difference during the latter part of the experiment (NK). Higher nutrient levels in the 'bottom' water did not affect O_2 exchange, neither in dark nor in L/D cores (NK).

Flux of inorganic phosphorus

Fluxes of PO_4^{3-} showed diel variations in L/D cores, with lower flux out of the sediment, or uptake by the sediment, during the light period (Fig. 4b). No diel patterns were observed in the dark cores. The highest Δ PO_4^{3-} flux rates of 3 to 4 µmol m⁻² h⁻¹ (dark minus light flux rates in L/D cores) were measured in the one extreme muddy core and in the sandy core with a mussel. The average Δ PO_4^{3-} flux for the rest of the L/D cores was 0.6 ± 0.8 (SD) µmol m⁻² h⁻¹. The L/D treatment also affected the night values of the efflux; during the latter part of the experiment the dark cores released 2.5 µmol m⁻² h⁻¹ more PO_4^{3-} (i.e. 2 to 3 times more) than the L/D cores during night.

The 24 h net outflow of PO_4^{3-} in muddy dark cores increased significantly with time (NK), being nearly 3 times higher (3 to 7 µmol m⁻² h⁻¹) than the net outflow from L/D cores (0 to 3 µmol m⁻² h⁻¹) during the latter part of the experiment (NK) (Fig. 5b). A significant difference between L/D and dark cores was also found for sandy sediment (NK), but mean flux rates were, both over time and for single cores, less than 2.2 µmol m⁻² h⁻¹ (Fig. 5b). Muddy dark cores released about 3 times (NK) more PO_4^{3-} than sandy dark cores during the latter part of the experiment (Fig. 5b).

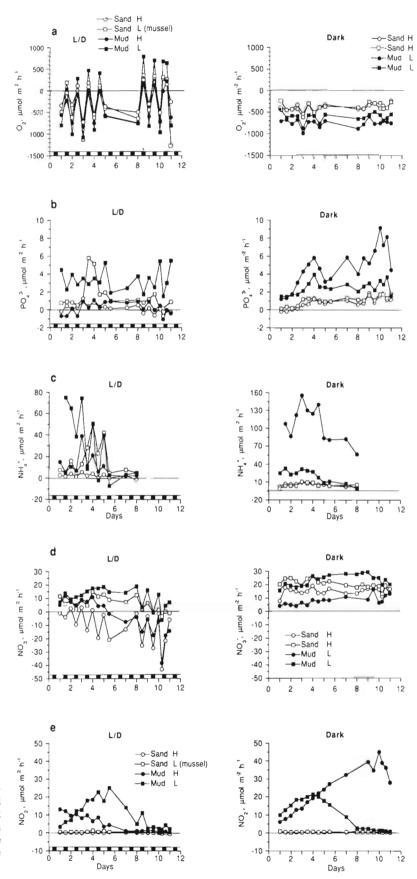


Fig. 4. Flux of oxygen (a), phosphate (b), ammonium, note different scales (c), nitrate (d) and nitrite (e) shown for individual L/D and dark cores measured twice a day. Dark squares on horizontal axes denote dark periods during the L/D cycle. H = high, L = low. For treatments see Fig. 2

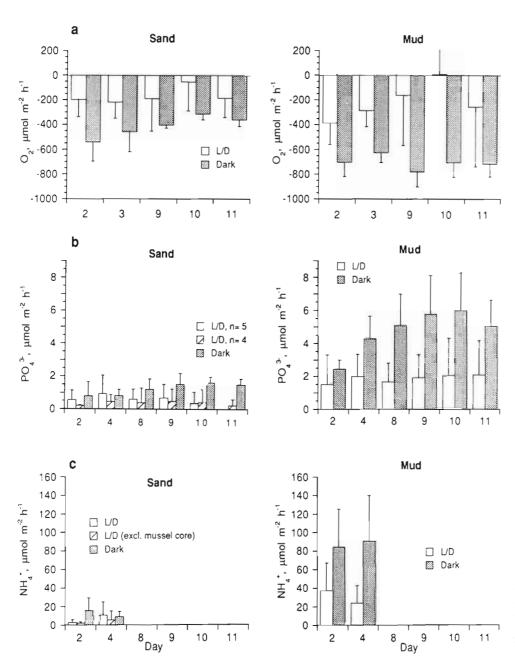


Fig. 5. Mean 24 h net flux of oxygen (a), phosphate (b), and ammonium (c), in sandy and muddy sediment. Each bar shows mean of 4 cores + SD. For treatments see Fig. 2

Flux of inorganic nitrogen

Ammonium

Fluxes of NH_4^+ exhibited strong diel variations in muddy L/D cores, with a mean Δ NH_4^+ -flux of 22 ± 14 $\mu mol\ m^{-2}\ h^{-1}$ and an extreme Δ flux of $65\ \mu mol\ m^{-2}\ h^{-1}$ (Fig. 4c; data on NH_4^+ missing after Day 8). In sandy sediment, Δ NH_4^+ flux rate averaged 1.0 \pm 2.4 (SD) $\mu mol\ m^{-2}\ h^{-1}$, excluding the core containing a mussel, for which Δ flux was up to 40 $\mu mol\ m^{-2}\ h^{-1}$ (Fig. 4c). The direction of the flux was, for all cores, mainly out of the sediment, although uptake of less than 7 $\mu mol\ m^{-2}$

 h^{-1} was also noted during light periods. The highest rate of NH_4^+ efflux, 160 $\mu mol\ m^{-2}\ h^{-1}$, was measured in a dark muddy core, the maximum value being only 54 $\mu mol\ m^{-2}\ h^{-1}$ for sandy cores.

In muddy sediments, 24 h net NH_4^+ flux rates were on average 3 times higher in dark cores (20 to 120 μ mol m⁻² h⁻¹) than in L/D cores (10 to 50 μ mol m⁻² h⁻¹) (significant difference, Wilcoxon-test, p < 0.001) (Fig. 5c). In sandy cores, a significant difference between L/D (0 to 12 μ mol m⁻² h⁻¹) and dark cores (6 to 18 μ mol m⁻² h⁻¹) (NK) was found when the mussel core was excluded (Fig. 5c). The NH_4^+ flux from muddy dark sediment was 20 to 120 μ mol m⁻² h⁻¹, while the flux

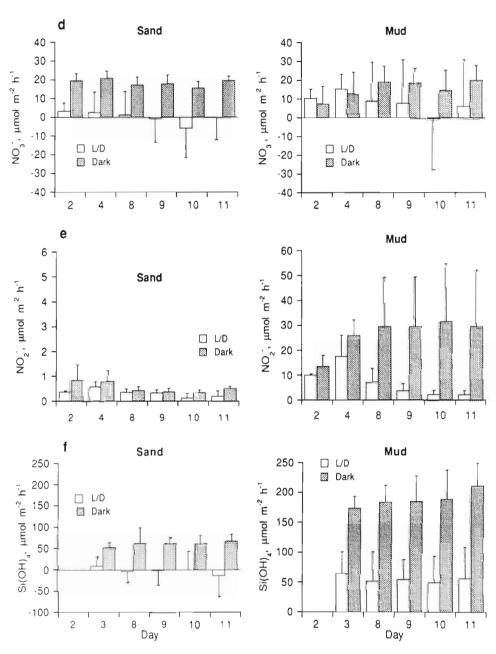


Fig. 5 (continued). Mean 24 h net flux of nitrate (d), nitrite (e) and silica (f) in sandy and muddy sediment

from sandy dark sediment generally was 5 to 20 μ mol m⁻² h⁻¹ (Fig. 5c).

Nitrate

The NO_3^- flux in the L/D cores exhibited diel variation with an amplitude that increased with time. Maximum Δ NO_3^- flux rate corresponded to an uptake of 40 μ mol m⁻² h⁻¹ during day and an insignificant flux rate during night (Fig. 4d). Mean Δ flux rates were similar in sandy and muddy cores, averaging 9±9 (SD) μ mol m⁻² h⁻¹ during the latter part of the experiment. In the

dark cores, no diel flux variations, and no changes in flux rates over the experimental period were observed. NO_3^- flux out of the sediment varied between 4 to 29 μ mol m⁻² h⁻¹ among cores.

Although the 24 h net fluxes generally were higher for the dark cores than the L/D cores, this difference was only significant in sandy cores (Wilcoxon test, p < 0.001), the dark ones releasing at least 4 times as much NO_3^- as the L/D cores (Fig 5d). A significant difference between sand and mud was only recorded on one occasion when sandy dark cores released NO_3^- at a significantly higher rate than the muddy dark cores (NK).

Nitrite

Rates of NO_2^- efflux were lower than 1.5 µmol m⁻² h⁻¹ in sandy cores, but in muddy cores they were on an average 20 times higher (5 to 31 µmol m⁻² h⁻¹), both for L/D and dark treatments (Fig. 4e). In muddy cores NO_2^- fluxes were similar to NO_3^- fluxes. Small diel flux variations were observed in both sandy and muddy L/D cores, most often showing the largest outflux during the light period, unlike NO_3^- , NH_4^+ , and PO_4^{-3-} fluxes that were usually highest during the dark period.

For 24 h net fluxes, there was a significant difference (NK) between L/D and dark treatments in muddy sediment during the latter part of the experiment, but not in sandy sediment (Fig. 5e). The difference found in the muddy cores was a consequence of a significantly (NK) increased efflux from the dark cores (from Day 2) and a decreased efflux from L/D cores (after Day 4).

In muddy, but not in sandy cores, there was a strong correlation between the efflux of NO_3^- and NO_2^- out of the dark cores on Day 2 ($r=0.99,\ p<0.001,\ n=5$). However, NO_2^- efflux did not correlate with NO_3^- concentration in the headspace waters. In many cores there was a tendency for the peak of NO_2^- efflux to occur after the peak of NH_4^+ efflux and therefore, on no single occasion was there a clear relation between these 2 nutrients. Instead, a significant correlation was obtained between the maximum efflux rate of NO_2^- for each core and the early (Day 2) efflux rate for NH_4^+ ($r=0.94,\ p<0.001,\ n=24$). As NH_4^+ release ceased in some cores (mainly L/D cores), NO_2^- release also ceased.

Flux of inorganic silicon

Concentrations of $Si(OH)_4$ were measured only at the end of the light period, and diel variations were not studied. Significantly lower flux rates (NK) out of the sediment were found for L/D than dark cores (Fig. 5f), although for sandy cores the difference was significant (NK) only on the last day. Uptake of Si was observed only in the sandy cores. In the dark the outflux was 2 to 3 times higher (significant difference, NK) from the muddy than from the sandy sediment (ca 150 to 250 and 50 to 100 μ mol m⁻² h⁻¹, respectively) (Fig. 5f), whereas no significant difference between mud and sand was found for the L/D cores. A net uptake (max. 77 μ mol m⁻² h⁻¹) was observed for some of the sandy L/D cores, whereas there was always a net release of Si from the muddy sediment (Fig. 5f).

Summary of differences between 24 h net fluxes

For muddy sediment, significant differences in 24 h net fluxes between L/D and dark cores were found for

 O_2 and all measured nutrients except for NO_3^- (Table 2). In sandy sediment a significant difference was found for NO_3^- but not for NO_2^- or O_2 . Significant differences between L/D and dark treatments usually occurred only in the latter part of the experiment. Even

Table 2. Summary of statistical tests on the effect of treatment and sediment type on mean 24 h flux of oxygen and inorganic nutrients in sediment cores

	L/D cycle	e vs dark	Sand vs mud		
Variable	Sand	Mud	L/D	Dark	
O ₂	NS		NS		
PO ₄ 3-			NS		
NH ₄ +		• d			
NO_3^-	• ल	NS	NS		
NO ₂ -	NS				
Si(OH) ₄			NS		

A significant difference (Newman-Keuls multiple comparison test, p < 0.05) found at least for 1 d

^a Wilcoxon signed rank test

in the cases where significant differences could not be statistically supported, mean fluxes were generally lower for L/D than for dark cores. Significant differences between 24 h net fluxes from muddy and sandy cores were observed for O_2 and all nutrients in permanently darkened cores, but only for NH_4^+ and nitrite in cores exposed to a L/D cycle (Table 2). Statistical tests on the effect of 'bottom' and surface water were not possible because of the small number of replicates.

Correlations between chlorophyll a, oxygen and nutrient fluxes

The influence of algae and light on nutrient fluxes can be studied by plotting Δ nutrient flux rates (the differences in flux rates between light and dark periods in the L/D cores) against chlorophyll a and Δ O_2 fluxes (Fig. 6). The best correlations were obtained when the most complete data sets from the first part of the experiment were used (Days 2 to 4). However, for Δ PO₄ flux, significant correlations were obtained when data from the whole experimental period were used.

 Δ O₂ flux was strongly correlated with chlorophyll a content, i.e. the higher chlorophyll a content, the larger the Δ O₂ flux (Fig. 6a). A similar correlation was found with algal cell number (r = 0.82, p < 0.001, n = 12). Δ PO₄³⁻ flux was better correlated with Δ O₂ flux (Fig. 6d) than with chlorophyll a content (Fig. 6c). For inorganic nitrogen (IN; NH₄"+NO₂"+NO₃"), significant correlations were found with both chlorophyll a and

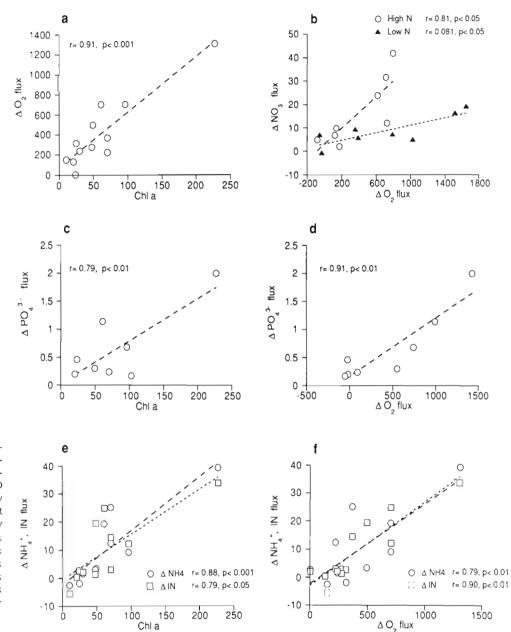


Fig. 6. Plots showing correlations between chlorophyll a and Δ flux for oxygen and nutrients in L/D cores. Δ oxygen flux = day minus night flux, Δ nutrient fluxes = night minus day flux. Oxygen vs Chl a (a); nitrate vs oxygen (b); phosphate vs Chl a (c); phosphate vs oxygen (d); ammonium vs Chl a (e); ammonium and IN vs oxygen (f)

 Δ O_2 flux, although Δ NH_4^+ flux was more strongly correlated with chlorophyll $\it a$ (Fig. 6e), whereas Δ IN correlated better with Δ O_2 flux (Fig. 6f). Furthermore, these plots revealed a higher Δ IN flux/chlorophyll $\it a$ ratio in sand than in mud. During the latter part of the experiment 2 separate regressions were found for cores exposed to low and high nutrient water when Δ NO_3^- flux was plotted against chlorophyll and Δ O_2 flux, the correlation being stronger with Δ O_2 (Fig. 6b) than with chlorophyll a (not shown). No such difference between high and low nutrient treatments was found for Δ $PO_4^{~3-}$ flux (Fig. 6c, d). No significant correlation was found between Δ $PO_4^{~3-}$ and Δ IN flux (Fig. 7).

DISCUSSION

Light-induced effects

The diel variations in sediment-water flux of nutrients, i.e. a decreased outflux of $\mathrm{NH_4}^+$, $\mathrm{NO_3}^-$ and $\mathrm{PO_4}^{3-}$ during the light period, were shown to be mediated by photosynthetic benthic organisms. This conclusion is based on the fact that there was a light-induced benthic oxygen production as well as significant correlations between Δ O₂ flux (day minus night flux rate), the corresponding Δ nutrient fluxes (night minus day flux), the chlorophyll a content and the algal cell num-

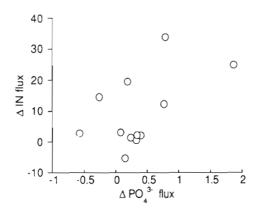


Fig. 7. Δ flux of inorganic nitrogen vs Δ flux of phosphate. For explanation of Δ flux see Fig. 6

bers in the sediment (Fig. 6). The significantly lower release of silicon from illuminated sediment also supports this conclusion and suggests that diatoms play a major role. Also, the differences in pore-water nutrient gradients between L/D and dark cores point to the importance of sediment-associated organisms. Our results agree with those of Carlton & Wetzel (1988), who showed, by using ³²P, that the efflux of P from the sediment was inversely related to the magnitude of microalgal photosynthesis. Similarly, Andersen & Kristensen (1988) found an inverse relationship between the release of dissolved inorganic nitrogen and benthic primary production. The only unequivocal evidence of a light-induced effect on the algal biomass during our incubation was the higher cell numbers found in the sandy L/D cores compared with the sandy dark cores.

The diel variation in nutrient fluxes is coupled to the light-dark periods with at least 2 mechanisms: (1) algal uptake, (2) photosynthetic oxygen production. The microalgae are expected to assimilate more inorganic nutrients during the light period (Syrett 1981). Nutrient levels have been observed to be much lower in the biologically active thin top-most layer of the sediment than in both the overlying water and the sediment below, suggesting efficient removal mechanisms of nutrients from the interstitial water (Simon 1988 and references therein).

The oxygen produced by the benthic microalgae influences a number of microbial processes as well as purely chemical redox-reactions. The high-resolution microelectrode technique has enabled analysis of the variation in the sediment oxygen profile on microscales and the oxic/anoxic interface has been shown to rapidly move up and down with changed light conditions (Revsbech & Jørgensen 1983, Lindeboom et al. 1985, Wit et al. 1989), affecting redox-sensitive flux processes (for further discussion see below).

Could biological processes, other than those related to the sediment organisms, have influenced the flux of

nutrients and oxygen? Even if picoplankton and bacteria could have passed through the 1.2 µm filter, the interference by biological processes in the very shallow column of headspace water was not likely to be important. The density of bacteria in sediments is generally ca 1000 times higher than that in the water phase (Meyer-Reil 1984), and this is also true for sedimentassociated microalgae. The primary productivity in the top 5 mm of illuminated sediment is, when expressed per unit area, within the same range as the phytoplankton productivity in a water column of several metres (e.g. Charpy-Roubaud & Sournia 1990). Thus, it is unlikely that the microorganisms in the headspace water could have increased to densities high enough to overshadow, or even compete with, the effect of the sediment organisms. However, the growth of bacteria and benthic microalgae, mainly diatoms, on the walls of the perspex tubes might have interfered. This interference was, however, likely to be important only during the latter part of the experiment, whereas the diel fluctuations were observed already during the initial phase of the experiment. A brownish colouring, restricted to the stirring bar and a narrow zone in the top part of the tube, became visible only late in the experiment. This is consistent with observations that the growth rate of benthic diatoms (which dominated the algal biomass) is much lower (0.06 to 0.27 d^{-1} ; Gould & Gallagher 1990) than that of phytoplankton species, and consequently, the colonization rate can be expected to be slow. Thus, the variations related to the L/D cycle appear to be mediated mainly by organisms within the sediment.

Several in situ investigations in intertidal and other shallow areas point to the fact that the microphytobenthos undoubtedly regulates nutrient flux between sediment and water (Vries & Hopstaken 1984, Nowicki & Nixon 1985, Asmus 1986, Kelderman et al. 1988, Keizer et al. 1989, Rizzo 1990). In our experiment, however, we dealt with sublittoral sediments from 15 m depth and the question arises as to the importance of microphytobenthos for total nutrient fluxes at this depth. Riaux-Gobin et al. (1989), who investigated 2 subtidal sediments at 10 and 20 m depth on the coast of Bretagne (France), found that the decrease of inorganic nitrogen in interstitial water in surface sediments was related to microphytobenthic spring bloom. Henriksen et al. (1981) interpreted the zero-flux of NO₃⁻ at 14 m depth in Kattegat as a result of uptake by a layer of benthic diatoms, and Jørgensen & Revsbech (1989) recorded benthic photosynthesis at this depth. According to previous calculations by Granéli & Sundbäck (1986), oxygen production by microphytobenthos (calculated from ¹⁴C uptake values) may play a considerable role in Laholm Bay. The latter 3 studies represent a summer situation with adequate light conditions for

microphytobenthic productivity at 15 m depth. At this depth, where usually less than 5% of surface irradiance remains, benthic primary productivity is negligible during October to April (Sundbäck & Jönsson 1988). However, in lower latitudes, where light penetrates to greater depths during most of the year, the regulating mechanism of microphytobenthos can be expected to play a more important role than in northern temperate areas (cf. Plante-Cuny 1984, Hansen et al. 1987, Herndl et al. 1989).

Oxygen flux

Oxygen fluxes in this study were similar to rates reported for laboratory-incubated cores simulating conditions in Laholm Bay during summer (Granéli & Sundbäck 1986). Benthic respiration rates for muddy cores were also similar to rates measured in situ for silty-sandy sediment from the northern Adriatic Sea sampled at depths of 15 and 22 m (Herndl et al. 1989). The mean gross primary production (here approximated by Δ O₂ flux) during the 16 h light period for Laholm Bay sediment was 420 μ mol O_2 m⁻² h⁻¹ and net production was 290 μ mol O₂ m⁻¹ h⁻¹, which is similar to or somewhat lower than values for March to September at comparable or deeper sites in the Northern Adriatic Sea with noon light levels of 5 to 50 $\mu E m^{-2} s^{-1}$ (Herndl et al. 1989). The microphytobenthic oxygen production in our laboratory study is thus not unrealistically high compared with in situ conditions at a depth of 15 m. As the Kattegat has a mean depth of only 23 m, the microphytobenthic algae may play a significant role in the regulation of oxygen conditions at the sedimentwater interface, as well as the nutrient exchange processes. The effect of microphytobenthic algae on oxygen conditions may be more dramatic than our measurements indicate, because diel changes in the interstitial water in the top few mm of the sediment may be much larger due to altered light conditions than the

changes in the overlying water measured in this study (cf. Revsbech & Jørgensen 1986, Jensen et al. 1984).

Phosphorus flux

The PO_4^{3-} fluxes in this study agree with other measurements in the Kattegat, at 20 m and deeper (Enoksson et al. 1990), where both mud and especially sand released PO_4^{3-} at rates that were low in proportion to oxygen consumption. This phenomenon has also been observed in Kiel Bight at 20 m depth (Balzer 1984) and in Chesapeake Bay (Boynton & Kemp 1985).

The suppression of PO₄³⁻ release in light occurred in both sandy and muddy samples, and in mud the light effect clearly continued also during the dark periods. As stated also by Carlton & Wetzel (1988), the light-suppression of PO43- release may be due to oxygeninduced redox reactions and/or microphytobenthic uptake. As there was no correlation between oxygen concentration in the headspace water and PO₄³⁻ flux during night in our experiment, oxygen may not have been the most important regulator of P-flux. On the other hand, if oxygen regulates the P-flux, it is probably the concentration gradient in the topmost few mm of the sediment that is important, not oxygen concentration in the bulk water above the sediment. Algal uptake of PO_4^{3-} , if taking place only during the day, does not fully fit our data, because the $\Delta O_2/\Delta PO_4^{3-}$ flux ratio (cf. Fig. 6d) was much higher than both the Redfield ratio of 106 and the ratio 160 found by Brzezinski (1985), which is to be expected assuming the photosynthetic quotient to be 1. However, a PO₄³⁻uptake may have occurred also during night. Both PO₄3- uptake (Sundberg & Nilshammer-Holmval 1975) and increase in cell phosphorus in the dark (Eppley et al. 1967) have been observed for microalgae, and therefore the net flux of PO₄³⁻ over the whole light-dark cycle (24 h net flux) (Table 3) would compare better with the light induced changes in oxy-

Table 3. Mean O_2 , IN and PO_4^{3-} fluxes for Day 2 and 4 given as net sediment water exchange rates for dark and L/D cores, differences in exchange rates between dark and L/D cores and atomic flux ratios calculated from these net fluxes and differences in net fluxes

Sediment	Treatment	O ₂	IN — μmol m ⁻² h ⁻¹ —	PO ₄ ³⁻	O ₂ /IN	O ₂ /PO ₄ ³⁻
Mud	Dark	621	108	3.1	5.7	199
Mud	L/D	328	51	1.6	6.4	210
Mud	Dark minus L/D	293	57	1.6	5.2	188
Sand	Dark	561	37	1.6	15.0	351
Sand	L/D	303	10	0.6	30.0	534
Sand	Dark minus L/D	258	27	1.0	9.4	250

gen fluxes. In addition, heterotrophic activity that is stimulated by the organic carbon input from algal photosynthesis (Hall & Fisher 1985, Andersen & Kristensen 1988) would probably affect both day and night fluxes of PO_4^{3-} . Indeed, the differences in 24 h net PO_4^{3-} fluxes between dark and L/D cores were similar to what would be predicted from the corresponding differences in oxygen fluxes, giving O_2/PO_4^{3-} ratios of 188 and 250 for mud and sand, respectively (Table 3). Therefore, there is no indication of a secondary, O2-mediated light effect on PO₄³⁻ fluxes. Carlton & Wetzel (1988) found that photosynthesis-mediated oxygen production was the main controlling factor for PO₄³⁻ flux in carbonate-rich lake sediments, but there are important differences that prevent a direct application of their results on marine sediments. According to Caraco et al. (1990), in most salt-water systems PO₄³⁻ is released from sediments and behaves essentially as a conservative tracer of benthic decomposition, while in freshwater systems PO₄³⁻ is strongly immobilized in sediments under oxic conditions.

Nitrogen fluxes

The dark sandy sediment released IN at rates that were within the range observed for the same depths in July in the Kattegat (Blackburn & Henriksen 1983) and in Aarhus Bight (Jensen et al. 1990). The high rates of IN release that we found for muddy cores has only been occasionally observed in Laholm Bay (Enoksson 1987, Enoksson et al. 1990).

Light efficiently decreased the release of $\mathrm{NH_4}^+$, causing strong diel fluctuations in all samples where there was a high $\mathrm{NH_4}^+$ concentration in the overlying water, regardless if the sediment was sand or mud. Concurrent with a decreasing $\mathrm{NH_4}^+$ release, the $\mathrm{NO_3}^-$ flux exhibited enhanced diel fluctuations (Fig. 4c, d). $\mathrm{NH_4}^+$ was probably the nitrogen source that was preferred by the benthic microalgae, but when $\mathrm{NH_4}^+$ became less abundant, the algae started to utilize $\mathrm{NO_3}^-$, which was available in adequate amounts in the headspace water of all cores.

As was found for the Δ PO₄³ fluxes, the Δ IN fluxes in L/D cores were markedly lower than what would be expected from Δ O₂ fluxes (Fig. 6f). This might either indicate algal uptake of IN during night (Pettersson & Sahlsten 1990) and/or that other microbial processes, indirectly affected by light, might have dampened the diel variations in the IN fluxes. The latter would include a stimulation of the strictly aerobic nitrifying bacteria via algal oxygen production, although there are also reports on nitrification being inhibited in algal mats (Henriksen & Kemp 1988). Denitrification would also be affected in various ways, including (1) competition for NO₃⁻ by the algae, (2) changes in NO₃⁻ production, (3) stimulation

by algal excretion of labile carbon compounds, (4) inhibition via algal oxygen production during daytime. The latter has been shown to be true for a stream sediment by Nielsen et al. (1990). If there was a similar inhibition of denitrification in our experiment, this would partly explain the low Δ IN fluxes.

The 24 h net fluxes of IN and O_2 (Table 3) were used to calculate if algal uptake (incl. night uptake) could account for light effects on IN fluxes. O2/IN ratios of 5.2 and 9.4 were obtained using the dark-L/D differences for mud and sand, respectively. The value for sand is close to 10, which would be expected for benthic microalgal activity (C/N atomic ratio of 10, Brzezinski 1985). Therefore, we have no indication that light dependent mechanisms other than algal uptake were important in the sandy sediment. In mud, the relatively large light effect on IN flux (Dark-L/D; Table 3) may, however, point to a stimulation of denitrifiers during night in the L/D cores, or to an increase in bacterial biomass, bacteria being much richer in protein than algae. NO₂⁻ was released slightly faster during the light period but, as NO_2^- is an intermediate product in a number of nitrogen transforming processes, we do not know the underlying mechanism. However, light had only a weak influence on the efflux of NO2-.

Silicon flux

Flux of silicon is not considered to be directly influenced by redox-conditions (Balzer 1984). Therefore silicon has often been used as a conservative tracer to evaluate the magnitude of sedimentary fluxes (e.g. Ullman & Aller 1989). However, the potential uptake of silicon by diatoms at the sediment-water interface has not been considered. Our flux measurements revealed a markedly lower Si outflux in L/D than in dark cores, thus giving us a measure of algal activity not masked by either oxygen conditions or heterotrophic processes. The silicon fluxes show that diatoms play an important role in the nutrient flux between sediment and water. Kelderman et al. (1988) reported results indicating that the growth of benthic diatoms can be limited by silica availability.

Nutrient fluxes and sediment type

In the dark, the outflow of inorganic nutrients was significantly higher from the muddy than from the sandy sediment. This agrees with Enoksson's (1987) measurements of nutrient fluxes for sediments from Laholm Bay and with other results (e.g. Nowicki & Nixon 1985, Ullman & Sandström 1987). In the L/D cores, on the other hand, the mean 24 h net fluxes were only occasionally higher in mud (Table 2). This may be

a result of the higher microalgal activity in muddy than in sandy sediment during the light period. This conclusion is corroborated by the higher chlorophyll a content and algal cell numbers in the mud (Table 1) in combination with higher Δ $\rm O_2$ fluxes.

Muddy cores released exceptionally large amounts of NO₂⁻, the source of which was most probably NH₄⁺. There appears to be a mechanism operating in the muddy sediment that blocks the further conversion of NO_2^- , either to nitrogen gas or to NO_3^- (or NH_4^+). Muddy sediment was characterized by a rather high organic content (around 4% org. C), a high oxygen consumption, probably an ephemeral nature of the muddy layer (Flodérus & Håkanson 1989), and absence of bioturbation. In darkness, such a sediment would only be oxic in the top few millimeters (Jørgensen & Revsbech 1985), and only there would NH₄⁺ oxidation take place. If a close coupling between NH₄⁺ oxidizing and NO₂⁻ oxidizing bacteria is lacking, a large fraction of the produced NO₂⁻ would diffuse out into the water, where the abundance of NO_2^- oxidizers is lower than in the sediment. In addition, the low release, or even uptake, of NO₃⁻ by the sediment, suggests that the NO₂⁻ oxidizers were not fully established or that they were inhibited by reduced sulphur compounds (Bremner & Bundy 1974). NO₃⁻ reducers, especially those that are able to produce NH₄⁻ and that accumulate NO₂ (Samuelsson et al. 1988), are believed to thrive in highly organic sediments and therefore cannot be ruled out as the responsible group. However, our failure to show any dependence of NO₂⁻ on NO₃⁻ concentrations strongly suggests that NO₂⁻ is produced directly by NH₄⁺ oxidizers.

Surface vs bottom water

No stimulation of microbenthic algae from higher nutrient concentrations in the simulated below-halocline water was found. This agrees with previous experiments with muddy sublittoral sediment from Laholm Bay (Sundbäck 1986). For sandy sediment, however, nutrient limitation of microphytobenthos has been shown both for shallow water (Nilsson et al. 1991) and for sediment from 15 m depth in Laholm Bay (Granéli & Sundbäck 1985, Sundbäck 1986, Sundbäck & Granéli 1988). The lack of response of the microflora to the nutrient enrichment of the overlying water may have several explanations: (1) Nutrient release by the sediment itself may have overshadowed a possible stimulatory effect on algal growth by the enriched overlying water. The fact that the release of both IN and PO_4^{3-} from muddy sediment during the dark period was within the same range as the input rate of nutrients in the inflowing water, supports this idea. Adding much

higher amounts of nutrients to the overlying water did, however, stimulate the algal population of sandy sublittoral sediment (Sundbäck 1986). It appears that the high abundance of microalgae observed below the halocline in Laholm Bay (Sundbäck & Jönsson 1988) is instead related to the nutrient supply from the sediment itself rather than the concentration in the overlying water. (2) Because of the low light quantity used in the experiment, the algae may have been light rather than nutrient limited. Production vs light (P/I) curves for sublittoral microbenthic algae from Laholm Bay, showed that algae were light limited below depths of 5 m (Sundbäck & Jönsson 1988). (3) The duration of the experiment (14 d) may have been too short to reveal a stimulatory effect. The response of sediment-associated microalgae to changed nutritional conditions appears to exhibit a minimum lag time of ca 2 wk, suggesting that the sediment system is fairly well buffered against shortterm environmental changes (Levinton 1985, Sundbäck et al. 1990, Nilsson et al. 1991, Lindström Swanberg in press). Significant differences in nutrient fluxes between treatments were usually recorded only during the latter part of the experiment, which supports this conclusion. (4) Finally, a possible stimulating effect of the enriched bottom water could also have been masked because of the high individual variation among the cores and because we could not follow the development of the microphytobenthos in the individual cores during the experiment.

Faunal activity

We used cores without macrofauna. The accidental inclusion of one core containing a mussel demonstrated the effect of animal activity (in this case a suspension-feeder) by increased release of $\mathrm{NH_4}^+$ (Fig. 4c) and $\mathrm{PO_4}^{3-}$ (Fig. 4b). The question is, whether animal excretion and bioturbation can overshadow the influence of microalgae in situ. Kelderman et al. (1988) found no differences in P and Si fluxes between dark and light bell jars when placed over a dense cockle field with low primary productivity. On the other hand, Henriksen et al. (1980), Asmus (1986), Andersen & Kristensen (1988), and Lindström Swanberg (in press) all observed significant effects of microphytobenthos on nutrient fluxes in sediments with natural densities of macrofauna.

CONCLUSIONS

(1) Our results suggest that diel oscillations of nutrient fluxes related to the light/dark cycle can occur even at sublittoral depths (15 m). (2) These diel oscillations, with decreased outflux of both IN and PO_4^{3-} during the light period, were shown to be mediated by

photosynthetic organisms. (3) Significantly lower outflux, or even uptake of silicon from cores exposed to a L/D cycle also supports this conclusion and indicates that diatoms play a major role for the nutrient flux between sediment and water. (4) In the sublittoral sediment used in this experiment, IN flux mainly appeared to be directly influenced by algal uptake. Furthermore, PO_4^{3-} flux appeared to a higher degree to depend on algal uptake than on changes in the redox conditions mediated by photosynthesis. (5) Although microphytobenthos appears to regulate nutrient flux even at sublittoral depths, this effect is probably limited to the growth season, when enough light penetrates to the sediment surface. (6) Our results suggest, that when studying nutrient flux in permanently darkened cores, the flux rates will, in the summer, be overestimated by a factor of 2 to 6, depending on sediment type. (7) In shallow water, where light reaches the bottom, flux measurements should be made under light conditions corresponding to those in situ.

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