Optimization of Ultrasound-Assisted Extraction of Polysaccharides from Mulberry Leaves by Response Surface Methodology

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Response surface methodology was applied to optimize the ultrasound-assisted extraction of polysaccharides from mulberry leaves. A central-composite rotatory design (CCRD) was used for experimental design and analysis of the results to obtain the optimum extraction conditions. Based on the response surface methodology analysis, optimum conditions were: ratio of water to raw material 15:1, power of ultrasound 60 W, extraction temperature 80 °C and extraction time 20 min. Under the optimized conditions, the experimental values were in close agreement with values predicted by the mathematical model. Compared with conventional solvent extraction, ultrasound-assisted extraction was a better method to extract polysaccharides. The crude polysaccharides prepared under optimum conditions contained 55.06 % total sugar, 14.02 % protein, 14.49 % moisture and 17.48 % ash.

Keywords: Mulberry leaves, Polysaccharides, Ultrasound-assisted extraction, Response surface methodology.

INTRODUCTION

Mulberry, a member of the genus *Morus* of the family Moraceae, is found from subtropical regions of the Northern Hemisphere to the tropics of the Southern Hemisphere¹. In most mulberry-growing countries, especially in China and India, mulberry is used to feed the silkworms (*Bombyx mori* L.) for its foliage. As a Chinese traditional herbal medicine, mulberry is also widely used to lower blood pressure^{2,3}, cure or prevent diabetes⁴, treat fever, protect the liver, improve eyesight, strengthen joints and facilitate discharge of urine.

Polysaccharide from plant, epiphyte and animal extracts is an interesting source of additives for several industries, especially food and drug industry. Polysaccharides from mulberry leaves have significant antihyperglycemic activity and lower blood lipid^{5,6}. Therefore, discovery and evaluation of polysaccharides from the mulberry leaves as new safe compounds for functional foods and medicines have become a hot research spot.

The conventional solvent extraction has been used as a classical extraction of polysaccharides. High temperature and long extraction time, the main disadvantages of conventional solvent extraction⁷, can cause the degradation of polysaccharides and the decrease of the pharmacological activity of polysaccharides. Compared with conventional solvent

extraction, ultrasound-assisted extraction has become a better extraction method due to its advantages, such as low energy and solvent consumption (no reflux or refrigeration are needed), high efficiency and high level of automation. Recently, the application of ultrasound-assisted extraction in analyzing trace organic compounds from soil, animal and plant tissues has been widely reported⁸⁻¹¹. High efficient extraction by ultrasound maybe ascribed to acoustic cavitation, which could cause the disruption of the cell walls, the reduction of the particle size and the enhancement on contact between solvent and targeted compounds¹².

Response surface methodology, a statistical-mathematical method, which was originally described by Box and Wilson in 1951, is an effective tool to optimize the process. Response surface methodology usually uses an experimental design such as central-composite rotatory design to fit an empirical, full second-order polynomial model. The main advantage of response surface methodology is to reduce the number of experimental trials needed to evaluate multiple variables, therefore, it is less laborious and time-consuming than other approaches¹³. Response surface methodology is widely applied in the food industry to determine the effects of several variables and optimize their conditions¹⁴⁻¹⁷.

The aim of the present work is to optimize the conditions for ultrasound-assisted extraction of polysaccharides from

mulberry leaves. The effects of extraction temperature, power of ultrasound, extraction time and ratio of water to raw material are also studied by response surface methodology on the extraction yield of polysaccharides.

EXPERIMENTAL

Mulberry leaves were provided by Huzhou Daybreak Import & Export Co., Ltd., in Huzhou, China. All samples (mulberry leaves) were ground with a disintegrator to obtain a particle size less than 0.4 mm, then defatted with petroleum ether and pretreated with 80 % ethanol twice to remove some colored materials, oligosaccharides and some small molecule materials before the experiment. All chemicals used were analytical grade unless otherwise specified.

Ultrasound-assisted extraction experiments were carried out in ultrasonic cleaning baths produced by Kunshan ultrasound instrument Co., Ltd., in Suzhou, China. It could operate at frequencies ranging from 20 to 99 kHz.

Ultrasound-assisted extraction of polysaccharides from mulberry: The ultrasound-assisted extraction procedure used in the experiment was developed by Wu *et al.*¹⁸ with a little modification. The pretreated dry powder (10 g) was extracted with distilled water [ratio of water to raw material (mL/g) ranging from 5:1 to 25:1] at the power of ultrasound ranging from 40 to 80 W, while the temperature of the water bath ranged from 40 to 80 °C. The water-raw slurry in a stainless steel boiler in the water bath was stirred with an electric mixing paddle for a given time (extraction time ranging from 15 to 30 min) during the entire extraction process. Debris fragments of polysaccharides extracts were removed by centrifugation. Protein was removed with Sevag method. The polysaccharides solution was concentrated with a rotary evaporator and precipitated with four volumes of 95 % ethanol for 48 h at 4 °C. The precipitates were collected by centrifugation (6000 rpm, 0.5 h), washed with acetone and then dried under reduced pressure at -40 °C to get crude polysaccharides. All experiments were performed at least in duplicate.

Conventional solvent extraction of polysaccharides from mulberry: 10 g of ground powder was mixed with 250 mL of distilled water in a 500 mL round bottom flask fitted with a cooling system. Extractions were carried out at 80 °C for 30, 60, 90 min, respectively. The following processes were similar to the ultrasound-assisted extraction of polysaccharides.

Analytical methods: The yield of crude polysaccharides was calculated as a percentage of the weight of the pretreated dry mulberry powder. The total sugar content was determined by the phenol-sulfuric acid method using D-glucose as the standard¹⁹. The protein content was determined using Kjeldahl Nitrogen Determination method (Kjeldahl, Foss. Tecator) with a conversion factor of 6.25. Moisture content was determined by drying at 105 °C to constant mass. Ash content was determined by weight difference after calcining samples. Fat content was determined using a Soxhlet apparatus with the official method of AOAC.

The treated samples in conventional solvent extraction or ultrasound-assisted extraction were placed into a preparation chamber, which was attached to a scanning electron microscope (Philips XL 30 ESEM, England). The samples were

sputter coated with a thin layer of conductive gold. The shape and the surface characteristics of the samples were magnified and digitally recorded. The crude polysaccharides were purified by a DEAE-52 cellulose column (16 × 200 mm) and characterized by FT-IR spectrum (Thermo Nicolet nexus, USA) to detect functional groups.

Response surface methodology: The interaction effects of extraction temperature, power of ultrasound, extraction time and ratio of water to raw material were studied by response surface methodology. A central-composite rotatory design (CCRD) with four variables at 5 levels²⁰ was designed by SAS (SAS version 9.1, SAS Institute Inc., USA). The variables were coded according to eqn. 1:

$$x_i = \frac{X_i - X_0}{\Delta X_i} \quad (1)$$

where x_i was the independent variable coded value, X_i was the independent variable real value, X_0 was the independent variable real value on the centre point and ΔX_i was the step change value. Based on the preliminary single-factor experiment results, the selected independent variables and their levels were presented in Table-1. The whole design (Table-2) consisted of 31 experimental points carried out in random order, which included 16 factorial points, 7 centre points and 8 axial points.

Statistical analysis: Experimental data were fitted to a second-order polynomial model and regression coefficients were obtained. The second-order polynomial model for predicting the optimum point was expressed according to the eqn. 2:

$$Y = \beta_0 + \sum_{i=1}^4 \beta_i X_i + \sum_{i=1}^4 \beta_{ii} X_i^2 + \sum_{i < j=1}^4 \sum_{j=1}^4 \beta_{ij} X_i X_j \quad (2)$$

where Y was the response function, β_0 was the centre point of the system, β_i , β_{ii} and β_{ij} represented the coefficients of the linear, quadratic and interactive terms, respectively; X_i , X_{ii} and $X_i X_j$ represented the linear, quadratic and interactive terms of the coded independent variables, respectively. The fitted polynomial equation was expressed as response surface and contour plots in order to visualize the relationship between the response and experimental levels of each factor and to deduce the optimum conditions²¹.

Regression analysis of the data and estimation of the coefficients of the regression equation were analyzed by SAS (SAS version 9.1, SAS Institute Inc., USA). The Design Expert package (Version 7.1.1, 2007, Stat-Ease, Minneapolis, MN, USA) was used to generate three-dimensional response surface and two-dimensional contour plots.

TABLE-1
RANGE OF INDEPENDENT VARIABLES
AND THEIR CORRESPONDING LEVELS

Independent variable	Symbol		Coded factor level					
	Uncoded	Coded	-2	-1	0	+1	+2	
Extraction temp. (°C)	X ₁	x ₁	40	50	60	70	80	
Power of ultrasound (W)	X ₂	x ₂	40	50	60	70	80	
Extraction time (min)	X ₃	x ₃	10	15	20	25	30	
Ratio of water to raw material (mL/g)	X ₄	x ₄	5:1	10:1	15:1	20:1	25:1	

TABLE-2
CENTRAL COMPOSITE ROTATABLE
DESIGN (CCRD) AND RESPONSE^A

Run	Independent variables ^b				Response ^c (%) Y
	x ₁	x ₂	x ₃	x ₄	
1	-1	-1	-1	-1	3.06
2	-1	-1	-1	1	7.39
3	-1	-1	1	-1	4.60
4	-1	-1	1	1	4.59
5	-1	1	-1	-1	3.72
6	-1	1	-1	1	3.65
7	-1	1	1	-1	5.87
8	-1	1	1	1	3.30
9	1	-1	-1	-1	7.49
10	1	-1	-1	1	9.90
11	1	-1	1	-1	7.37
12	1	-1	1	1	7.45
13	1	1	-1	-1	7.55
14	1	1	-1	1	7.31
15	1	1	1	-1	9.40
16	1	1	1	1	7.03
17	-2	0	0	0	3.29
18	2	0	0	0	10.99
19	0	-2	0	0	6.51
20	0	2	0	0	5.94
21	0	0	-2	0	3.79
22	0	0	2	0	4.52
23	0	0	0	-2	5.96
24	0	0	0	2	6.79
25	0	0	0	0	7.46
26	0	0	0	0	8.33
27	0	0	0	0	8.13
28	0	0	0	0	8.18
29	0	0	0	0	7.81
30	0	0	0	0	7.90
31	0	0	0	0	7.93

^aNon-randomized; ^bCoded symbols and levels of independent variables refer to Table-1; ^cAverage of duplicated determination from different experiments.

RESULTS AND DISCUSSION

Fitting the models: Regression analysis was carried out to match the mathematical model with the experimental data, aiming at an optimum region for the responses. Some insignificant terms (x₃, x₄, x₁x₂, x₁x₃, x₁x₄) were neglected and the predicted model was not refitted. The regression model for polysaccharides extraction (Y) was predicted by eqn. 3 as follows:

$$Y = 7.962857 + 1.780375x_1 - 0.215458x_2 - 0.163121x_1^2 - 0.391871x_2^2 - 0.909371x_3^2 - 0.354371x_4^2 + 0.450563x_2x_3 - 0.754437x_2x_4 - 0.705688x_3x_4 \quad (3)$$

The results of analysis of variance (ANOVA) for the CCRD were shown in Table-3. It was obvious that the linear terms except β₃ and β₄, all quadratic terms and the crossproduct terms except β₁₂, β₁₃ and β₁₄ were significant (p < 0.01 or p < 0.05). The results demonstrated that the effects of extraction temperature and power of ultrasound were the major contributing factors to the yield of polysaccharides.

Table-4 showed the results of analysis of variance (ANOVA) for the CCRD. From Table-4, we found that the coefficient of determination (R²) was 0.9804. This implied that 98.04 % of the variations could be explained by the fitted model. Joglekar and May²² suggested that, for a good fit of a model, R² should

TABLE-3
SIGNIFICANCE OF REGRESSION COEFFICIENTS
OF THE FITTED SECOND-ORDER POLYNOMIAL
MODEL FOR RESPONSE (Y)

Term	Regression coefficients	Standard error	t-Value	p-Value
β ₀	7.962857	0.150288	52.98	< 0.0001***
Linear				
β ₁	1.780375	0.081165	21.94	< 0.0001***
β ₂	-0.215458	0.081165	-2.65	0.0173*
β ₃	0.042042	0.081165	0.52	0.6116
β ₄	0.133708	0.081165	1.65	0.1190
Quadratic				
β ₁₁	-0.163121	0.074357	-2.19	0.0434*
β ₂₂	-0.391871	0.074357	-5.27	< 0.0001***
β ₃₃	-0.909371	0.074357	-12.23	< 0.0001***
β ₄₄	-0.354371	0.074357	-4.77	0.0002**
Cross product				
β ₁₂	0.136812	0.099406	1.38	0.1877
β ₁₃	-0.096937	0.099406	-0.98	0.3440
β ₁₄	-0.111938	0.099406	-1.13	0.2767
β ₂₃	0.450563	0.099406	4.53	0.0003**
β ₂₄	-0.754437	0.099406	-7.59	< 0.0001***
β ₃₄	-0.705688	0.099406	-7.10	< 0.0001***

*Significant at 0.05 level; **Significant at 0.01 level; ***Significant at 0.001 level

TABLE-4
ANALYSIS OF VARIANCE (ANOVA) OF THE REGRESSION
PARAMETERS FOR THE RESPONSE SURFACE MODEL

Regression	Degree of freedom	Sum of squares	R ²	F-Value	p-Value
Total model ^a	14	126.342726	0.9804	57.08	< 0.0001
Linear	4	77.659268	0.6026	122.80	< 0.0001
Quadratic	4	27.710307	0.2150	43.82	< 0.0001
Crossproduct	6	20.973150	0.1627	22.11	< 0.0001
Error	16	2.529699	0.1581	—	—

^aThe coefficient of determination (R²) of the predicted model was 0.9804

be at least 0.80. The significance of each coefficient was also determined by the F-value and p-value²². The corresponding variables would be more significant if the absolute F-value became greater and p-value became smaller²³. The p-value of the regression model was less than 0.0001. Therefore, the developed model could adequately represent the real relationships among the chosen parameters.

Effects of independent variables on responses: The main goal of response surface methodology was to obtain the optimum values of the independent variables efficiently so that the responses were maximized. Response surface methodology could also identify interactions of the chosen variables by three-dimensional response surface and two-dimensional contour plots. Two kinds of figures were graphical representations of the regression model eqn. 3. In one three-dimensional response surface or two-dimensional contour plots, two variables were depicted while the other two were kept constant.

Figs. 1 to 3 showed the interaction effects of extraction temperature and three other variables (power of ultrasound, extraction time and ratio of water to raw material) on the yield of polysaccharides. Extraction temperature displayed a significant linear effect on the response and the yield of

polysaccharides increased with increasing extraction temperature. All other variables of the three figures exerted quadratic effects on the response. When extraction temperature was fixed at 60 °C in Fig. 1, the yield of polysaccharides increased with increasing power of ultrasound. It reached the peak near the central condition of power of ultrasound and then declined. The effects of extraction time (Fig. 2) and ratio of water to raw material (Fig. 3) on the yield of polysaccharides were similar to power of ultrasound in Fig. 1. The general form of three-dimensional relationship demonstrated that the changes of extraction temperature had more significant effect on the yield of polysaccharides than those of three other variables (power of ultrasound, extraction time and ratio of water to raw material). The results were in accordance with Table-3.

Figs. 4 and 5 showed the interaction effects of extraction time and two other variables (power of ultrasound and ratio of water to raw material) on the yield of polysaccharides. Quadratic effects for all variables on the response could be observed in two figures. The optimum yield of polysaccharides was obtained when the levels of the three variables were approximately at coded zero level.

Fig. 1. Response surface (3D) and contour plots for the effects of extraction temperature and power of ultrasound on the yield of polysaccharides

Fig. 2. Response surface (3D) and contour plots for the effects of extraction temperature and extraction time on the yield of polysaccharides

Fig. 3. Response surface (3D) and contour plots for the effects of extraction temperature and ratio of water to raw material on the yield of polysaccharides

When extraction time was fixed at 20 min in Fig. 4, the yield of polysaccharides increased obviously, reached the optimum value and then decreased slowly with increasing power of ultrasound. It was well known that the powerful ultrasound could induce chemical effect²⁴. The cavitation which occurred in water yielded hydroxyl radicals, leading to chemical decomposition²⁵. More chemical decomposition occurred with stronger power of ultrasound. So we believed that more chemical decomposition was responsible for the decreasing yield of polysaccharides. Under power of ultrasound 60 W, the yield of polysaccharides also increased with increasing extraction time and then decreased after it reached the peak. More chemical decomposition of polysaccharides occurred as the extraction time prolonged which caused the decrease of the yield. Similar results were obtained in Fig. 5.

Fig. 6 showed response surface and contour plots of the interaction of extraction time and power of ultrasound on the yield of polysaccharides. The power of ultrasound displayed a linear increase on the response, demonstrating that the yield of the polysaccharides increased with increasing power of ultrasound at fixed ratio of water to raw material. The effect of ratio of water to raw material on the yield of polysaccharides was similar to that of power of ultrasound. Fig. 6 showed that larger ratio of water to raw material and weaker power of ultrasound led to larger response. The optimum response was obtained when the ratio of water to raw material was 25:1 and power of ultrasound was 40 W.

Fig. 4. Response surface (3D) and contour plots for the effects of extraction time and power of ultrasound on the yield of polysaccharides

Fig. 5. Response surface (3D) and contour plots for the effects of extraction time and ratio of water to raw material on the yield of polysaccharides

Fig. 6. Response surface (3D) and contour plots for the effects of power of ultrasound and ratio of water to raw material on the yield of polysaccharides

Verification of results: Based on all the 31 experimental data for extraction of polysaccharides from mulberry leave with ultrasound-assisted extraction, the optimal process was found to be: $x_1 = 79.92$ °C (extraction temperature), $x_2 = 61.12$ W (power of ultrasound), $x_3 = 20$ min (extraction time), $x_4 = 14.31:1$ (ratio of water to raw material). The model predicted the yield of polysaccharides could be 10.88 %. For convenience, the optimum parameters were taken as follows: $x_1 = 80$ °C (extraction temperature), $x_2 = 60$ W (power of ultrasound), $x_3 = 20$ min (extraction time), $x_4 = 15:1$ (ratio of water to raw material). In order to confirm the fit of the predicted and actual data, three parallel experiments were carried out under the condition above and showed an average yield of 10.79 %, which was reasonably close to the predicted value and was regarded as satisfactory under the current constraints of experiments. The crude polysaccharides extracted under the optimum conditions were further analyzed for chemical compositions. We found the crude polysaccharides contained 55.06 % total sugar, 14.02 % protein, 14.49 % moisture and 17.48 % ash.

Comparison of ultrasound-assisted extraction to conventional solvent extraction of polysaccharides from mulberry leaves: Under the same extraction temperature, particle size and ratio of water to raw material, the yield of polysaccharides from ultrasound-assisted extraction was compared with that obtained from conventional solvent extraction. The results were presented in Table-5. From Table-5, we found that ultrasound-assisted extraction showed larger yield of polysaccharides and shorter extraction time than conventional solvent extraction. It was well known that ultrasound in ambient fluids could induce the expansion and collapse of cavities near the cell walls, the turbulent vibration on the solvent-solid interfaces and the furious agitation of extraction solvent. These physical phenomenons facilitated the disruption of cell walls, the release of the targeted components and the diffusion of the targeted components into extraction solvent. Therefore, the yield of polysaccharides was improved significantly²⁶. Similar result was reported by researchers, while studying the effect of ultrasound on extraction from vegetable tissues²⁷.

TABLE 5
COMPARISON OF OPTIMUM BETWEEN ULTRASOUND-
ASSISTED EXTRACTION AND CONVENTIONAL
SOLVENT EXTRACTION

Method	Conventional solvent extraction	Ultrasound-assisted extraction
Extraction temp. (°C)	80	80
Power of ultrasound (W)	–	60
Extraction time (min)	90	20
Ratio of water to raw material (mL/g)	15:1	15:1
Yield (%)	4.76 ± 0.36	10.88 ± 0.28

FT-IR spectra: The polysaccharides were analyzed using by FT-IR to detect functional groups (Fig. 7). The relatively strong absorption peak at around 1650 cm^{-1} , the weak one at around 1250 cm^{-1} and the other two between 1100 and 1010 cm^{-1} indicated the characteristic bands of polysaccharides. The intense peak at 3425 cm^{-1} was -OH stretching peak and the band at around 2930 cm^{-1} was C-H stretching vibration^{28,29}.

The weak peak at 2368 cm^{-1} was C-H transiting angle peak. The carbonyl (C=O) stretching was at 1741 and 1614 cm^{-1} . The bands at 1325 and 1242 cm^{-1} were indication of ring breathing with C-O stretching³⁰. The peak at 1242 cm^{-1} was unsymmetrical carbonyl stretching peak. The 1200-1000 cm^{-1} region, referred to as the "finger print" region, was unique to a sample. The two obvious absorptions between 1100 and 1010 cm^{-1} were originated from the glycosidic linkage $\nu(\text{C-O-C})$, indicating the stretching vibration of furanose. A characteristic absorption at 920 cm^{-1} was observed, indicating the existence of α -D-glucan³¹.

Fig. 7. FT-IR spectrum of the polysaccharides from mulberry leaves

Conclusion

Response surface methodology was effective for estimating the effects of four independent variables on the yield of polysaccharides. Using response surface methodology and contour plots, the optimum set of the independent variables was obtained graphically to get the desired levels of ultrasound-assisted extraction of polysaccharides. Optimum conditions (extraction temperature 80 °C, power of ultrasound 60 W, extraction time 20 min and ratio of water to raw material 15:1) for the extraction procedure of polysaccharides from mulberry leaves were identified. Under the optimum conditions, the yield of polysaccharides was 10.79 %. Compared to conventional solvent extraction, ultrasound-assisted extraction reduced extraction time and gave a larger yield of polysaccharides due to the acoustic cavitation of ultrasound-assisted extraction.

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