

Electronic supplementary information (ESI)

Structural characterization of quaternary selenites of tungsten(VI), $A_2W_3SeO_{12}$ ($A = NH_4, Cs, Rb, K, Tl$)

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Details of other instruments employed

SEM and EDAX study

SEM images and EDAX data of single crystals of compounds **1-5** were recorded on FEI Quanta-200 and Quanta-450 scanning electron microscopes equipped with EDS detector.

Spectroscopic studies

The diffuse reflectance spectra were recorded on powder samples of compounds **1-5** at room temperature, on a JASCO UV-visible spectrophotometer with a diffuse reflectance accessory, over 200-900 nm range. BaSO₄ was used as a 100% reflectance standard. The absorption data was calculated (Kubelka & Munk 1931), using the Kubelka–Munk equation: $\alpha/S = (1 - R)^2/2R$, where R is the reflectance and α and S are the absorption and scattering coefficients, respectively.

Thermal studies

Thermogravimetric analytical (TGA) and Differential scanning calorimetry (DSC) data for compounds **3-5** were collected on 'Perkin-Elmer' Q-series TG and SDT Q600 V20.9 Build-20 instruments, respectively. The samples were heated to about 750 °C at a rate of 20 °C/min and cooled to about 120 °C at the rate of 10 °C/min under dynamic N₂ gas.

SUPPLEMENTARY TABLES

Table S1. Net dipole moments of WO_6 octahedra and SeO_3 pyramids and octahedral distortion of WO_6 octahedra of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (**1**), **Cs**(**2**), **Rb**(**3**) and **K**(**4a**)) compounds.

Table S2. Net dipole moment of WO_6 and TeO_3 polyhedra of $\text{Rb}_2\text{W}_3\text{SeO}_{12}$ (**3**) compound.

Table S3. Net dipole moment of WO_6 and TeO_3 polyhedra of $\text{K}_2\text{W}_3\text{SeO}_{12}$ (**4a**) compound.

Table S4. Quantities of reactants employed for the syntheses of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (**5**), **Cs**(**6**), **Rb**(**7**), **K**(**8a**) and **Tl**(**9**)) compounds.

SUPPLEMENTARY FIGURES

Figure S1a. Simulated (red) and observed (blue) powder XRD patterns of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (**1**) and **Cs**(**2**)) compounds.

Figure S1b. Simulated (red) and observed (blue) powder XRD patterns of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{Rb}$ (**3**), **K**(**4a**) and **Tl**(**5**)) compounds. Unidentified reflections are marked with asterisk.

Figure S2. ORTEP diagram of coordinations of A^+ ions of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (**1**), **Cs**(**2**), **Rb**(**3**) and **K**(**4a**)) compounds.

Figure S3. Thermal analysis of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{Rb}$ (**3**), **K**(**4a**) and **Tl**(**5**)) compounds.

Figure S4. Scanning electron microscope images of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (**1**), **Cs**(**2**), **Rb**(**3**), **K**(**4a**) and **Tl**(**5**)) compounds.

Table S1. Net dipole moments of WO_6 octahedra and SeO_3 pyramids and octahedral distortion of WO_6 octahedra of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (**1**), Cs (**2**), Rb (**3**) and K (**4a**)) compounds.

Compound	WO_6 octahedra	Dipole moment (D)	Δd	$\text{TeO}_3/ \text{SeO}_3$ polyhedra	Dipole moment (D)
$(\text{NH}_4)_2\text{W}_3\text{SeO}_{12}$ (1)	WO_6	1.52	0.73	SeO_3	6.95
$\text{Cs}_2\text{W}_3\text{SeO}_{12}$ (2)	WO_6	1.85	0.78	SeO_3	5.73
$\text{Rb}_2\text{W}_3\text{SeO}_{12}$ (3)	WO_6	1.28	0.77	SeO_3	6.40
$\text{K}_2\text{W}_3\text{SeO}_{12}$ (4a)	$\text{W}(1)\text{O}_6$	1.84	0.78	SeO_3	9.13
	$\text{W}(2)\text{O}_6$	1.30	0.71		
	$\text{W}(3)\text{O}_6$	0.79	0.86		

Unit Cell Parameters					
a (Å)	7.238	α (°)	90	V (Å ³)	549.5
b (Å)	7.238	β (°)	90		
c (Å)	12.1115	γ (°)	120		

Table S2. Net dipole moment of WO₆ and SeO₃ polyhedra of Rb₂W₃SeO₁₂(**3**) compound.

Atom	Bond Valence Charge	Number of Protons	Number of Electrons	Fractional Co-ordinates			Cartesian Co-ordinates			W-O Distance (Å)	W-O Bond length (Å)
				xf	yf	zf	xc	yc	zc		
W	6.2	74	67.800	0.80780	0.66030	0.60560	3.45723070	4.13895312	7.33475774		
O1	-0.526	8	8.526	0.87260	0.75220	0.77670	3.59366700	4.71500915	9.40704481	2.15518656	2.155
O2	-1.685	8	9.685	0.79160	0.59450	0.46780	3.57810530	3.72649952	5.66578545	1.72342599	1.723
O3	-0.776	8	8.776	1.12150	0.87100	0.59280	4.96526800	5.45968222	7.17972983	2.01060575	2.011
O3	-1.255	8	9.255	0.74950	0.87850	0.59280	2.24558950	5.50669441	7.17972983	1.83380050	1.834
O4	-1.17	8	9.170	0.53830	0.45000	0.65120	2.26766540	2.82073134	7.88704465	1.85951475	1.86
O4	-0.788	8	8.788	0.91170	0.46170	0.65120	4.92799230	2.89407036	7.88704465	2.00446841	2.005

Unit Vector (W-O)			C Grav (Å)	C Charge (Å)	μ Debye	Dipole moment (W-O)		
UV xc	UV yf	UV zc				Dx	Dy	Dz
-0.06330603	-0.26728824	-0.96153489	0.21026210	0.24074523	11.16794546	-0.706998265	-2.9850605	-10.7383692
-0.07013623	0.23932191	0.96840381	0.16813912	0.21541435	17.58298042	-1.233203943	4.2079925	17.02742515
-0.75004127	-0.65688119	0.07710507	0.19615666	0.23042567	12.59608216	-9.447581502	-8.2741294	0.971221842
0.66072683	-0.74585065	0.08453913	0.17890737	0.22025597	15.29335945	10.10473298	-11.406562	1.292887351

Unit Cell Parameters											
a (Å)	7.231	α (°)	90	V (Å ³)	1050.56	0.0000000					
b (Å)	11.4863	β (°)	90.0959			0.7266933	1.71505456	-0.77695243	0.46562973	-0.42371439	1
c (Å)	12.6486	γ (°)	90								
O1	-1.296	8	9.296	0.02533300	1.55328273	0.72669330	1.71505456	0.79172338	0.44004567	-0.42371439	1
O1	-1.296	8	9.296	0.02533300	1.55328273	0.72669330	1.71505456	0.01477096	0.90567540	-0.42371439	1
centroid				0.00000000	0.00000000	0.72669330					
lone pair	-2	2	4	-7.45554E-16	0	1.22	1.22000000	0.00000000	0.00000000	1.00000000	1

C Grav (Å)	C Charge (Å)	μ Debye
0.32667706	0.40456626	14.73339671
0.32667706	0.40456626	14.73339671
0.32667706	0.40456626	14.73339671
0.06777778	0.14305816	12.32622933

Dipole moment (Se-O)			
Dx	Dy	Dz	
-11.4471483	6.860307468	-6.242752252	
11.6647747	6.483367497	-6.242752252	
-0.217626394	-13.34367497	-6.242752252	
-7.53268E-15	0	12.32622933	Magnitude
4.62427E-15	3.55271E-15	-6.402027422	6.40202742

Table S3. Net dipole moment of WO₆ and SeO₃ polyhedra of K₂W₃SeO₁₂(**4a**) compound.

Atom	Bond Valence Charge	Number of Protons	Number of Electrons	Fractional Co-ordinates			Cartesian Co-ordinates			W1-O Distance (Å)	W1-O Bond length (Å)
				xf	yf	zf	xc	yc	zc		
W1	6.161	74	67.839	0.22890	0.79650	0.08430	1.653391 20	9.1488379 5	1.0662767 0		
O1	-0.51	8	8.510	0.25360	0.60870	0.09130	1.831848 70	6.9917108 1	1.1548168 8	2.1663065 2	2.166
O2	-1.685	8	9.685	0.23870	0.94440	0.06190	1.724729 22	10.847661 72	0.7829481 3	1.7237651 8	1.724
O7	-1.205	8	9.205	0.06150	0.76250	-0.02190	0.445170 14	8.7583037 5	0.2770042 7	1.8484368 6	1.848
O9	-0.771	8	8.771	0.43780	0.76380	-0.01720	3.166095 94	8.7732359 4	0.2175558 6	2.0193015 0	2.019
O10	-0.788	8	8.788	0.43540	0.79500	0.19040	3.144346 47	9.1316085 0	2.4082928 1	2.0060537 7	2.006
O12	-1.202	8	9.202	0.05990	0.79240	0.19380	0.429033 99	9.1017441 2	2.4512980 4	1.8492032 1	1.849

Unit Vector (W1-O)			C Grav (Å)	C Charge (Å)	μ Debye	Dipole moment (W1-O)			Magnitude
UV xc	UV yf	UV zc				Dx	Dy	Dz	
-	-	-	-	-	-	-	-	-	-
0.0823786 9	0.9957626 6	0.0408714 9	0.211346 98	0.2414605 1	11.035862 86	0.9091199 8	10.9891	0.4510521 66	
0.0413850 0	0.9855308 5	0.1643661 0	0.168172 21	0.2153483 5	17.554959 12	0.7265120 35	17.3009 54	2.8854402 25	
0.6536447 5	0.2112780 8	0.7267118 5	0.180335 30	0.2208460 3	14.981319 29	9.7924607 4	3.16522 44	10.887102 21	
0.7491227 7	0.1860059 1	0.6357805 2	0.197005 02	0.2311877 5	12.569944 99	9.4164320 52	2.33808 4	7.9917261 87	
0.7432279 7	0.0085887 3	0.6689831 2	0.195712 56	0.2300651 3	12.635203 04	9.3908363 35	0.10852 03	8.4527375 5	
0.6620998 7	0.0254670 9	0.7489827 7	0.180410 07	0.2208741 8	14.963499 78	9.9073312 96	0.38107 68	11.207403 48	Magnitude
						0.7431083 66	0.31894 81	1.6530754 31	1.8402707 15

Atom	Bond Valence Charge	Number of Protons	Number of Electrons	Fractional Co-ordinates			Cartesian Co-ordinates			W2-O Distance (Å)	W2-O Bond length (Å)
				xf	yf	zf	xc	yc	zc		
W2	6.079	74	67.921	0.01720	0.75070	0.82440	0.10691995	8.62276541	10.42750310		
O3	-0.509	8	8.509	-0.06340	0.57750	0.87420	-0.47695295	6.63333825	11.05740321	2.16691074	2.167
O4	-1.658	8	9.658	0.06800	0.89530	0.79940	0.47478403	10.28368439	10.11128818	1.73030856	1.73
O7	-0.855	8	8.855	0.06150	0.76250	0.97810	0.42399929	8.75830375	12.37159241	1.97443490	1.974
O8	-1.157	8	9.157	0.25490	0.69020	0.81020	1.82602928	7.92784426	10.24789303	1.86293104	1.863
O10	-1.135	8	9.135	-0.06460	0.70500	0.69030	-0.48173684	8.09784150	8.73132629	1.87058219	1.870
O11	-0.765	8	8.765	-0.24720	0.78550	0.86370	-1.80578846	9.02248865	10.92459295	2.01626647	2.016

Unit Vector (W2-O)			C Grav (Å)	C Charge (Å)	μ Debye	Dipole moment (W2-O)			Magnitude
UV xc	UV yf	UV zc				Dx	Dy	Dz	
0.26944945	0.91809373	0.29069038	0.21140593	0.24124354	10.94634515	2.949486687	10.049771	3.181997273	-
0.21260027	0.95989757	0.18275059	0.16881059	0.21541036	17.35278342	3.689206486	16.656895	3.171231478	-
0.16059245	0.06864665	0.98463074	0.19262780	0.22772248	12.93326135	2.07698414	0.8878251	12.73448674	-
0.92279815	0.37302570	0.09641263	0.18174937	0.22131944	14.63991173	13.50968353	5.4610633	1.411472328	-
0.31469175	0.28062061	0.90676412	0.18249582	0.22175779	14.52177662	4.56988336	4.0751098	13.16782604	-
0.94863871	0.19824921	0.24653976	0.19670892	0.23045374	12.42122432	11.78325418	2.4624979	3.062325654	-
						0.0267500	-	-	1.2987911

69	0.42127 37	1.2282798 27	53
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Atom	Bond Valence Charge	Number of Protons	Number of Electrons	Fractional Co-ordinates			Cartesian Co-ordinates			W3-O Distance (Å)	W3-O Bond length (Å)
				xf	yf	zf	xc	yc	zc		
W3	6.24	74	67.760	0.50660	0.75950	0.84420	3.6453521 7	8.723844 85	10.677945 31		
O5	-0.527	8	8.527	0.56760	0.57940	0.87640	4.0857614 7	6.655162 22	11.085230 13	2.1539009 0	2.154
O6	-1.676	8	9.676	0.43760	0.89560	0.80210	3.1473044 6	10.28713 028	10.145439 39	1.7249566 4	1.725
O8	-0.733	8	8.733	0.25490	0.69020	0.81020	1.8260292 8	7.927844 26	10.247893 03	2.0318704 9	2.032
O9	-1.282	8	9.282	0.43780	0.76680	0.98280	3.1449250 9	8.807694 84	12.431040 81	1.8250484 7	1.825
O11	-1.293	8	9.293	0.75280	0.78550	0.86370	5.4252115 4	9.022488 65	10.924592 95	1.8215165 5	1.822
O12	-0.729	8	8.729	0.55990	0.70760	0.69380	4.0339485 7	8.127705 88	8.7755963 7	2.0310884 6	2.031

Unit Vector (W3-O)			C Grav (Å)	C Charge (Å)	μ Debye	Dipole moment (W3-O)		
UV xc	UV yf	UV zc				Dx	Dy	Dz
-	-	-				-		-
0.2044705 5	0.9604353 8	0.1890917 1	0.210136 67	0.2407528 5	11.210959 63	2.2923110 65	10.7674 02	2.1198995 71
0.2887305 7	- 0.9062752	0.3087068 4	0.168288 45	0.2155416 1	17.563659 97	5.0711654 87	- 15.9175	5.4220220 34

-0.01474824	-0.75411368	0.65657829	0.32290829	0.40164969	14.95091222	-0.220499707	-11.2746874	9.816444334	
0.77700586	0.49477783	0.38917451	0.33120161	0.40667817	14.27412702	11.09108034	7.06252155	5.555126332	
-0.75928230	0.50849412	0.40610852	0.33044268	0.40618468	14.32905214	-10.87979563	7.28623871	5.819150104	Magnitude
						-0.009214996	3.07407282	21.19072077	21.41253501

lone pair

Atom	Bond Valence Charge	Number of Protons	Number of Electrons	Fractional Co-ordinates			Se-O distance (Å)	Unit Vector (Se-O)			UV
				xc	yc	zc		UV xc	UV yf	UV zc	
Se	3.815	34	30.185	0.00000000	0.00000000	0.00000000					
O1	-1.372	8	9.372	0.02500223	1.27842519	1.11307651	1.69526853	0.01474824	0.75411368	0.65657829	1
O3	-1.215	8	9.215	1.35106436	0.86032387	0.67669992	1.73880846	0.77700586	0.49477783	0.38917451	1
O5	-1.228	8	9.228	1.31722122	0.88214784	0.70452683	1.73482408	0.75928230	0.50849412	0.40610852	1
centroid				0.00294697	0.15468217	0.83143442					
lone pair	-2	2	4	0.004251245	0.223141701	1.199412234	1.22000000	0.00348463	0.18290303	0.98312478	1

C Grav (Å)	C Charge (Å)	μ debye
0.32290829	0.40164969	14.95091222
0.33120161	0.40667817	14.27412702

Dipole moment (Se-O)		
Dx	Dy	Dz
0.220499707	11.27468744	-9.816444334
11.09108034	7.062521552	5.555126332

0.3304426 8	0.4061846 8	14.329052 14
0.0677777 8	0.1427526 7	12.302480 00

10.879795 63	- 7.2862387 08	- 5.8191501 04	
0.0428695 55	2.2501609 14	12.094872 97	Magnitud e
0.0520845 51	- 0.8239119 11	- 9.0958477 98	9.133235 51

	Compound	Reactants (g, mmol)				Amount of water	Temperature (°C) (duration (h))		Yield (%)
							Preheating in open air	Heating in evacuated and sealed quartz tube	
Single crystal growth	(NH ₄) ₂ W ₃ SeO ₁₂ (1)	NH ₄ OH (6ml, 150.60)	H ₂ WO ₄ (0.4360, 1.74)	SeO ₂ (1.2, 10.80)	NH ₄ Cl (1, 18.35)	5ml	-	230 °C (96h)	~93 %
	CS ₂ W ₃ SeO ₁₂ (2)	CS ₂ CO ₃ (0.1495, 0.46)	H ₂ WO ₄ (0.3443, 1.38)	SeO ₂ (0.2548, 2.30)	-	-	100 °C for 12 h	150°C(6h) + 600°C(24h)	-
	Rb ₂ W ₃ SeO ₁₂ (3)	Rb ₂ CO ₃ (0.1156, 0.50)	WO ₃ (0.3501, 1.50)	SeO ₂ (0.1126, 1.00)	-	-	100 °C for 12 h	100 °C (2h) + 200 °C (24h) + 400 °C (24h) + 500 °C (24h)	-
	K ₂ W ₃ SeO ₁₂ (4a)	(COOK) ₂ ·H ₂ O (0.1057, 0.57)	H ₂ WO ₄ (0.4310, 1.72)	@SeO ₂ (0.0643, 0.57)	-	-	500 °C for 16 h	200 °C (24h) + 400 °C (24h) + 500 °C (24h)	-
	Tl ₂ W ₃ SeO ₁₂ (5)	Tl ₂ WO ₄ (0.2666, 0.41)	WO ₃ (0.1882, 0.81)	SeO ₂ (0.0901, 0.81)	-	-	100 °C for 12 h	200 °C (20h) + 250 °C (16h) + 350 °C (16h) + 400 °C (48h) + 500 °C (24h) + 600 °C (18h)	~98 %

Pow-der	Rb ₂ W ₃ SeO ₁₂ (3)	Rb ₂ CO ₃ (0.1156, 0.50)	H ₂ WO ₄ (0.3773, 1.51)	@SeO ₂ (0.2815, 2.51)	-	-	300 °C for 24 h	300 °C (24h) + 400 °C (12h) + 500 °C (12h)	~99 %
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Table S4. Quantities of reactants employed for the syntheses of A₂W₃SeO₁₂ (A = NH₄(**1**), Cs(**2**), Rb(**3**), K(**4a**) and Tl(**5**) compounds.

@SeO₂ was added after initial preheating.

SUPPLEMENTARY FIGURES

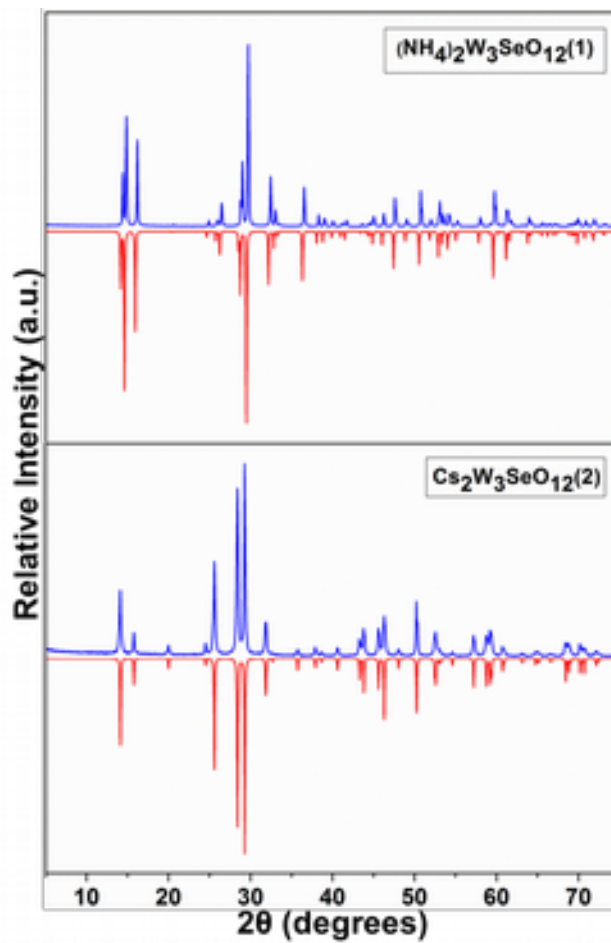


Figure S1a. Simulated (red) and observed (blue) powder XRD patterns of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{NH}_4$ (1) and Cs (2)) compounds.

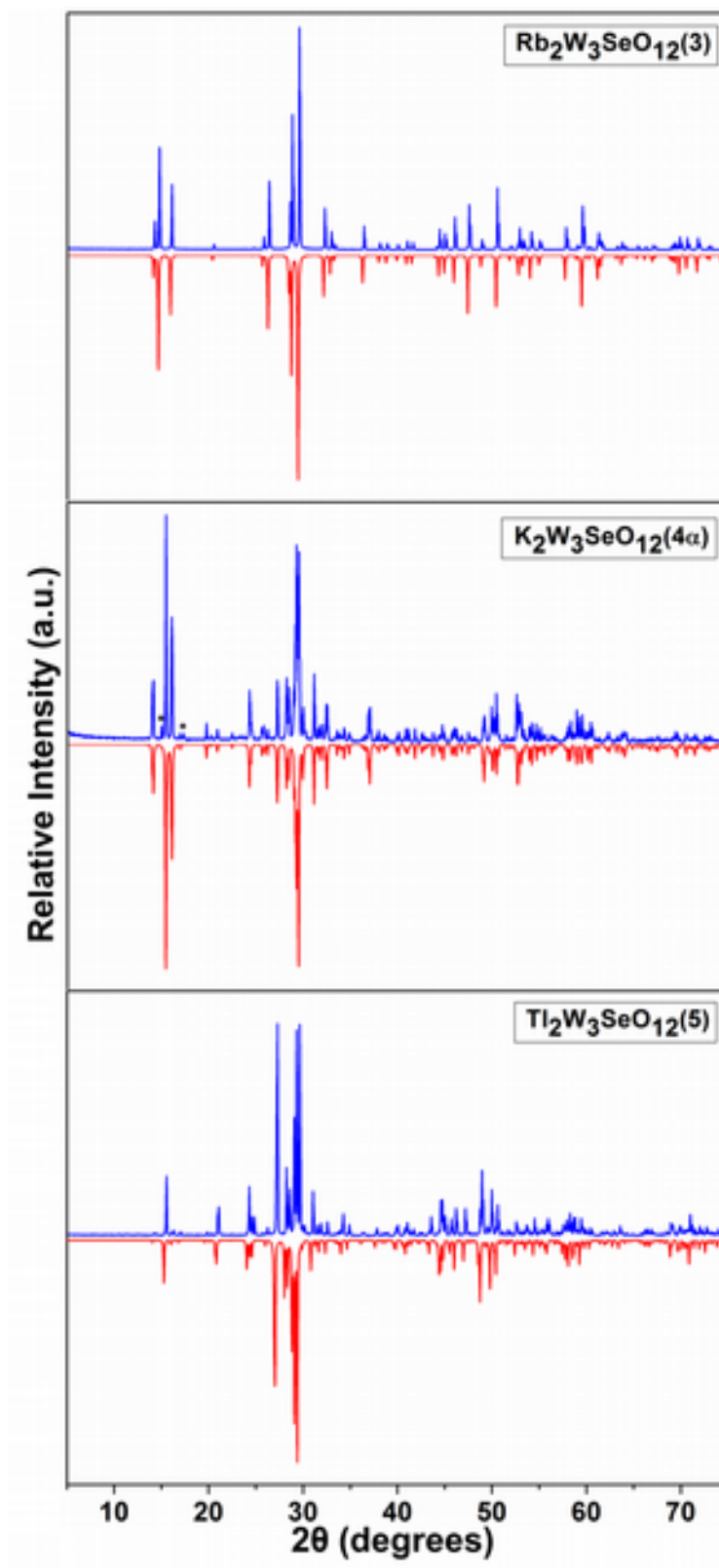


Figure S1b. Simulated (red) and observed (blue) powder XRD patterns of $A_2\text{W}_3\text{SeO}_{12}$ ($A = \text{Rb}(3)$, $\text{K}(4\alpha)$ and $\text{Tl}(5)$) compounds. Unidentified reflections are marked with asterisk.

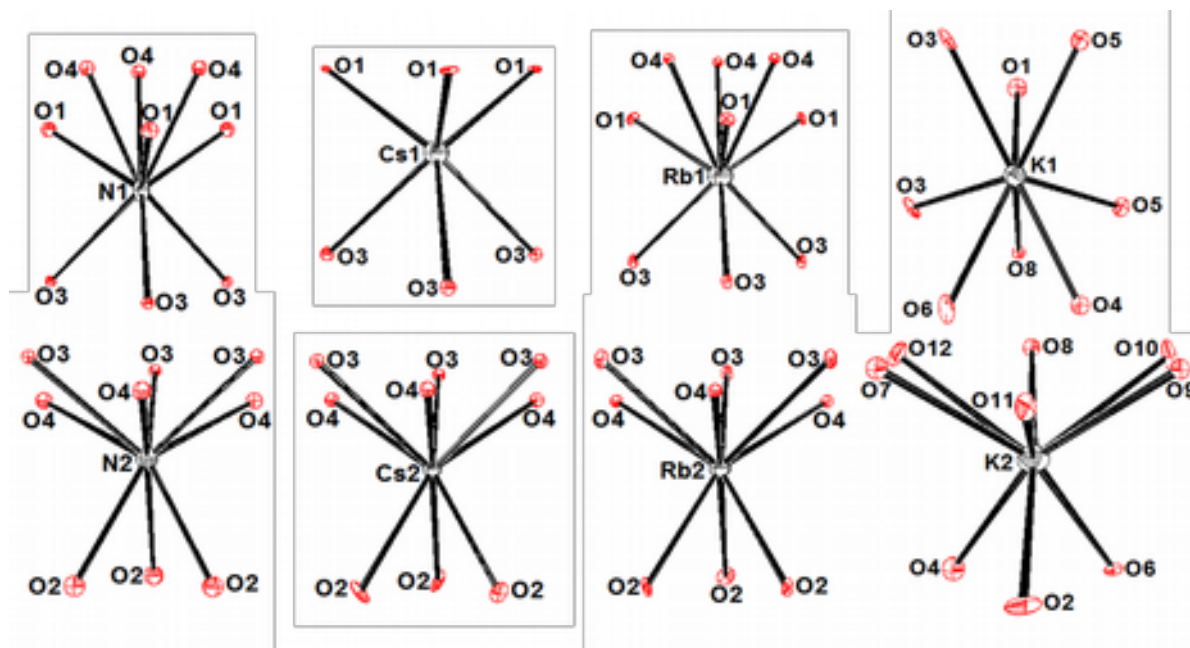


Figure S2. ORTEP diagram of coordinations of A^+ ions of $A_2W_3SeO_{12}$ ($A = NH_4$ (**1**), Cs(**2**), Rb(**3**) and K(**4a**)) compounds.

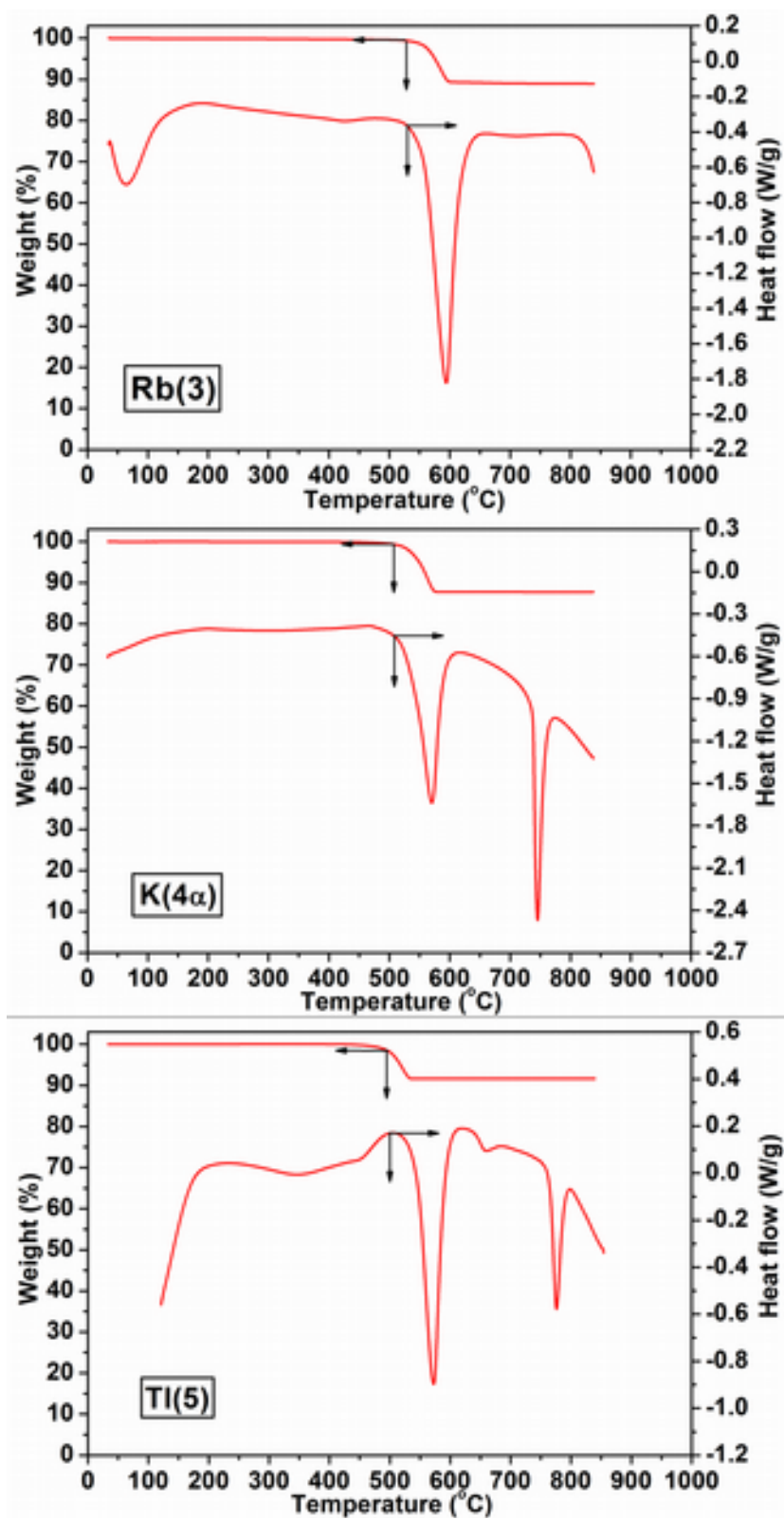


Figure S3. Thermal analysis of $A_2W_3SeO_{12}$ ($A = \text{Rb(3)}$, $\text{K(4}\alpha\text{)}$ and Tl(5)) compounds.

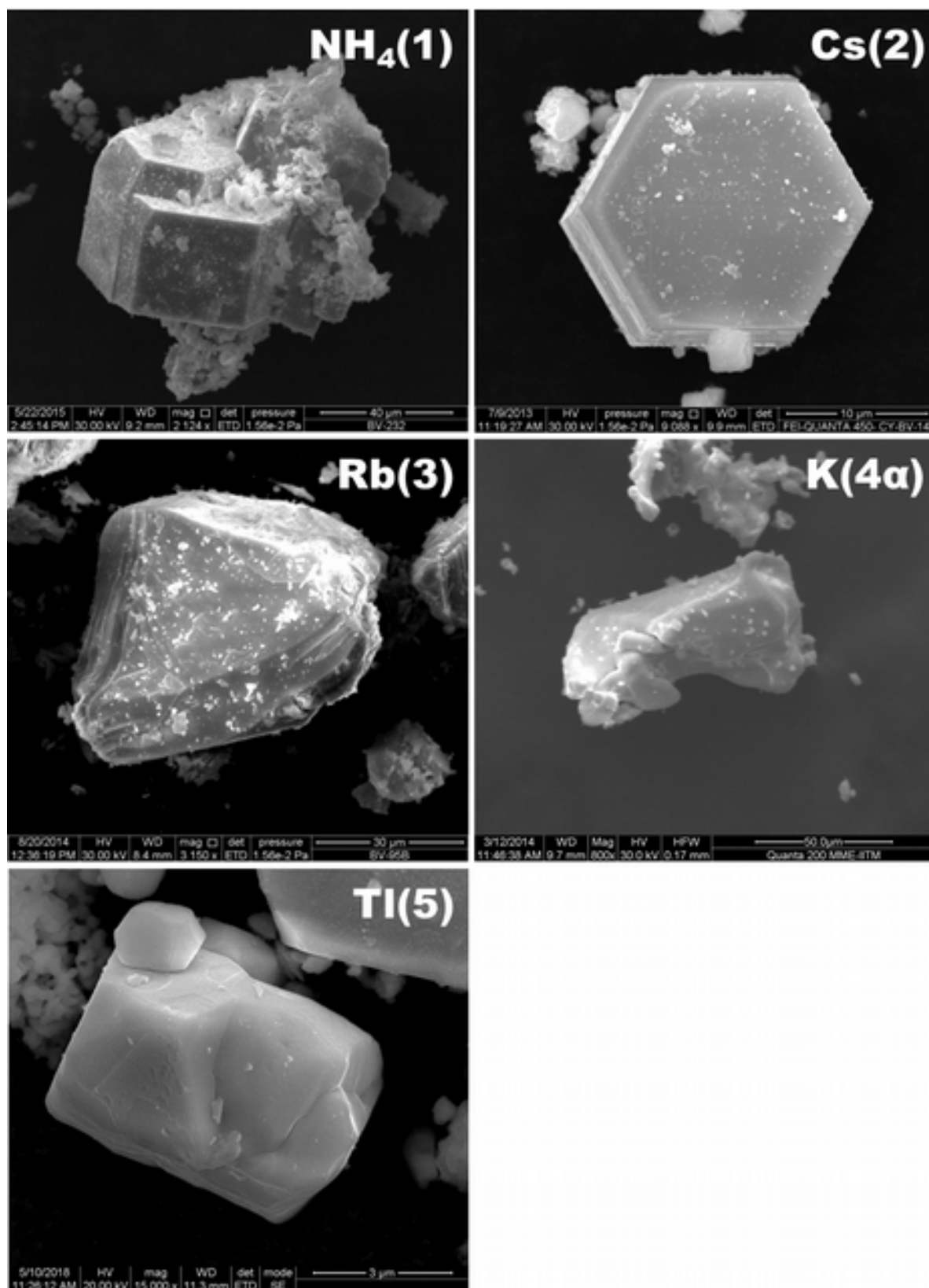


Figure S4. Scanning electron microscope images of $A_2W_3SeO_{12}$ ($A = NH_4$ (1), Cs(2), Rb(3), K(4 α) and Tl(5)) compounds.