



STRUCTURAL
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Supporting information for article:

Oxygen quenching of structurally characterized [5,10,15,20-tetrakis(4-fluoro-2,6-dimethylphenyl)porphyrinato]platinum(II)

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Supporting Information

Oxygen Quenching of Structurally Characterized 5,10,15,20-tetrakis-(2,6-dimethyl-4-fluorophenyl)porphyrinatoplatinum(II)

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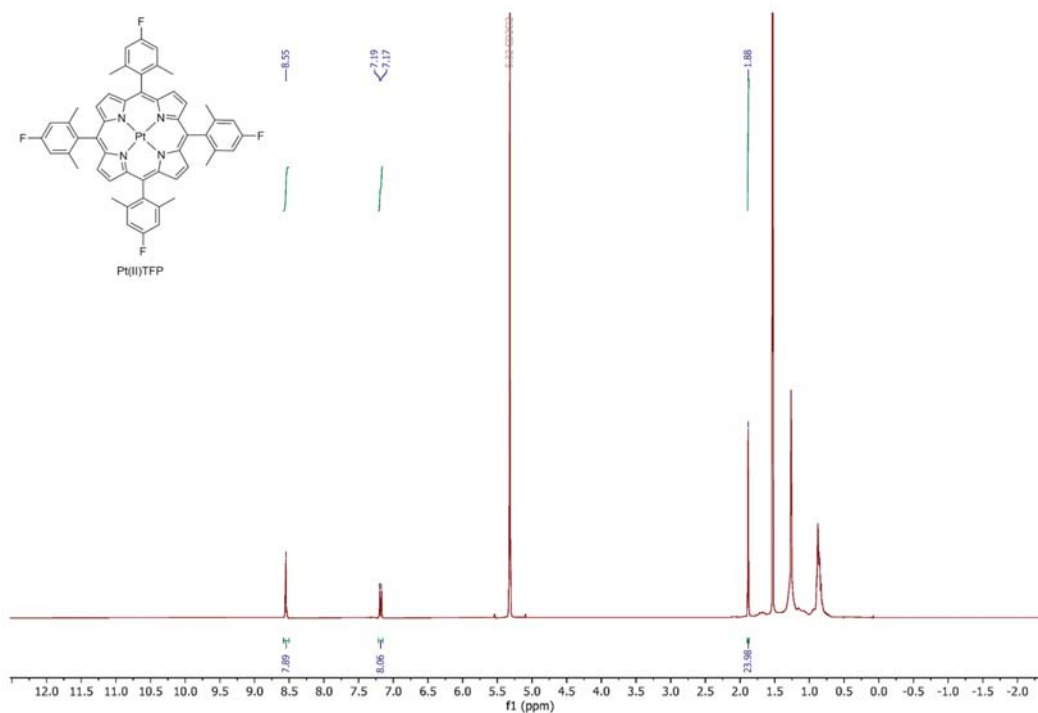


Figure S1. ^1H NMR spectrum of Pt(II)TFP in CD_2Cl_2 , 400 MHz.

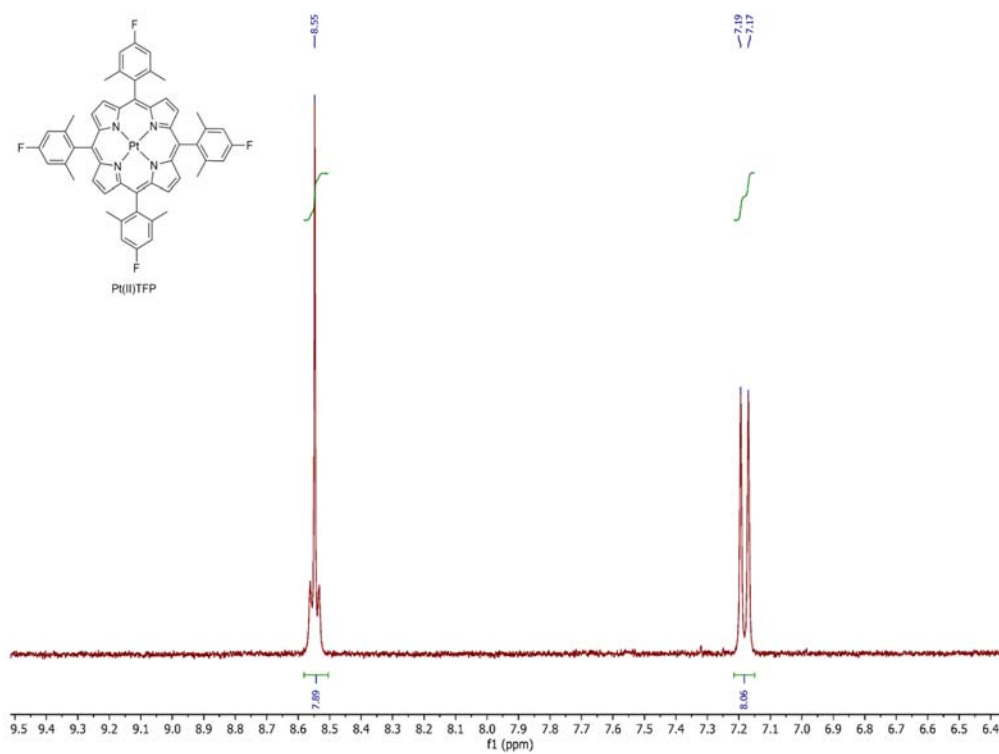


Figure S2: The aromatic region of Pt(II)TFP in CD_2Cl_2 , 400 MHz.

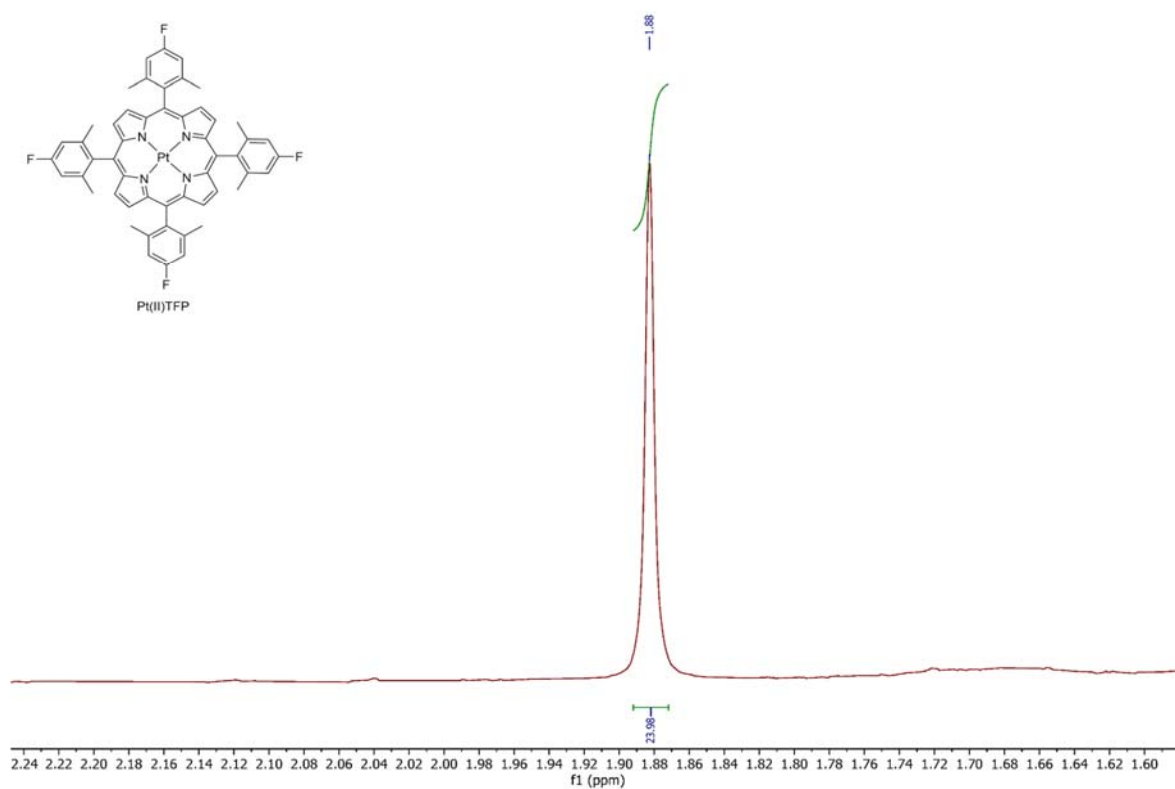


Figure S3. The aliphatic region of Pt(II)TFP in CD_2Cl_2 , 400 MHz.

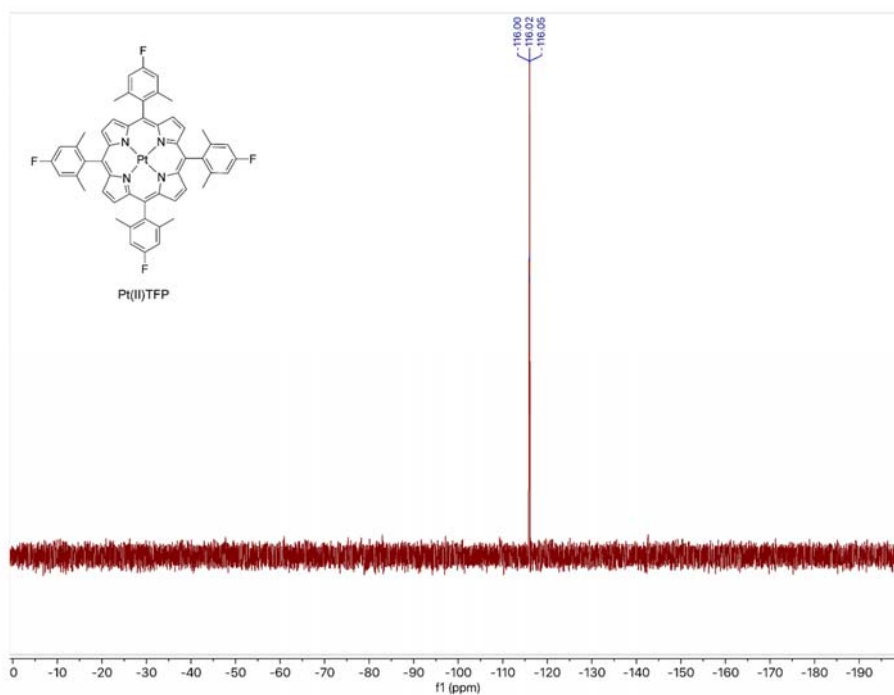


Figure S4. ^{19}F NMR spectrum of Pt(II)TFP in CD_2Cl_2 , 379 MHz.

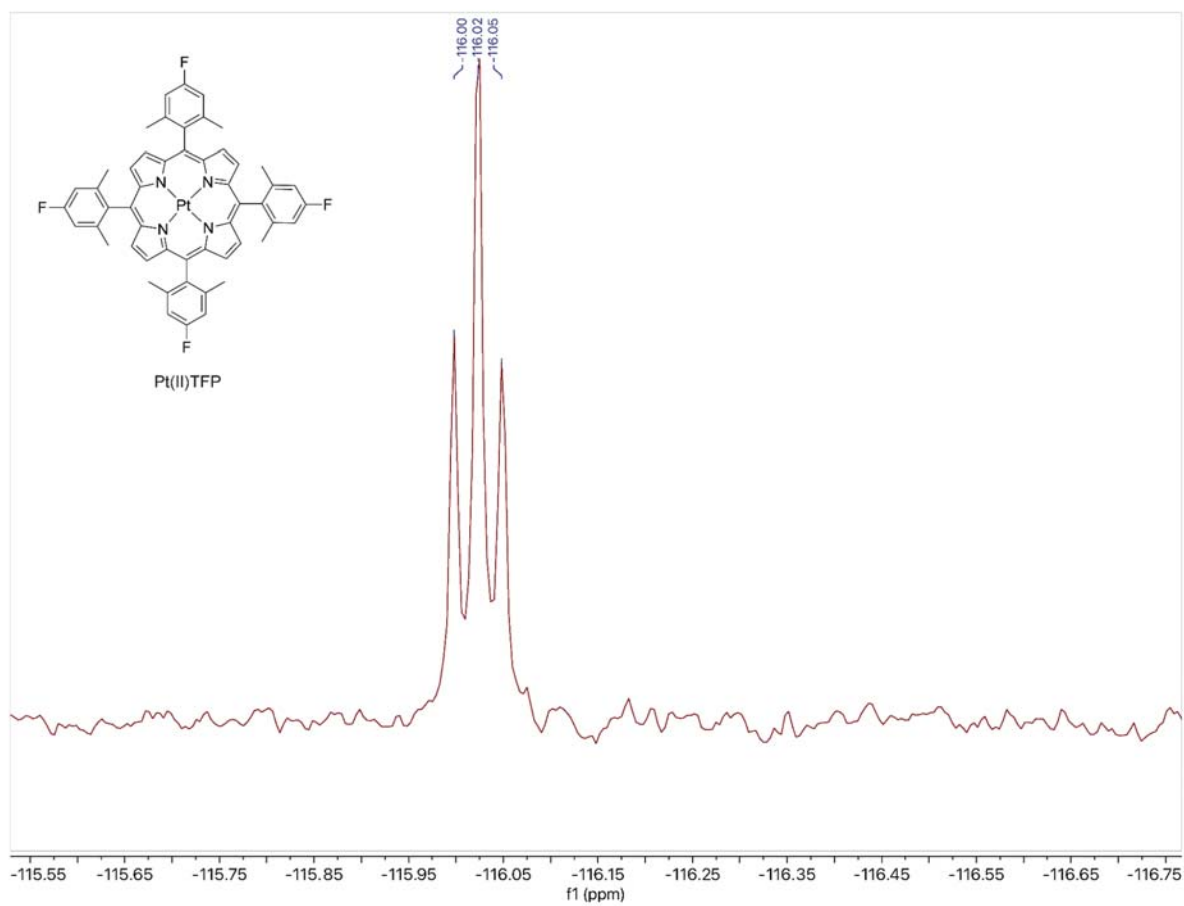


Figure S5. ^{19}F NMR spectrum of Pt(II)TFP in CD_2Cl_2 , zoomed in region, 379 MHz.

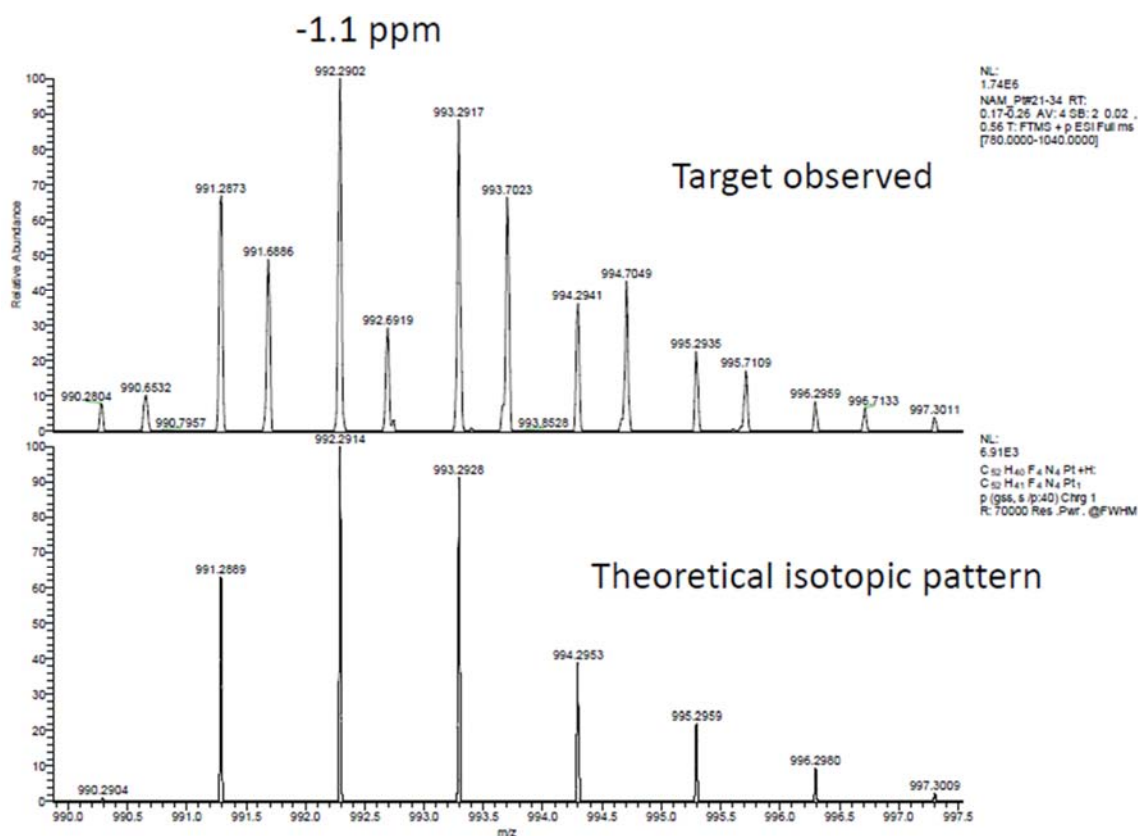


Figure S6. High resolution mass spectrometry of Pt(II)TFP.

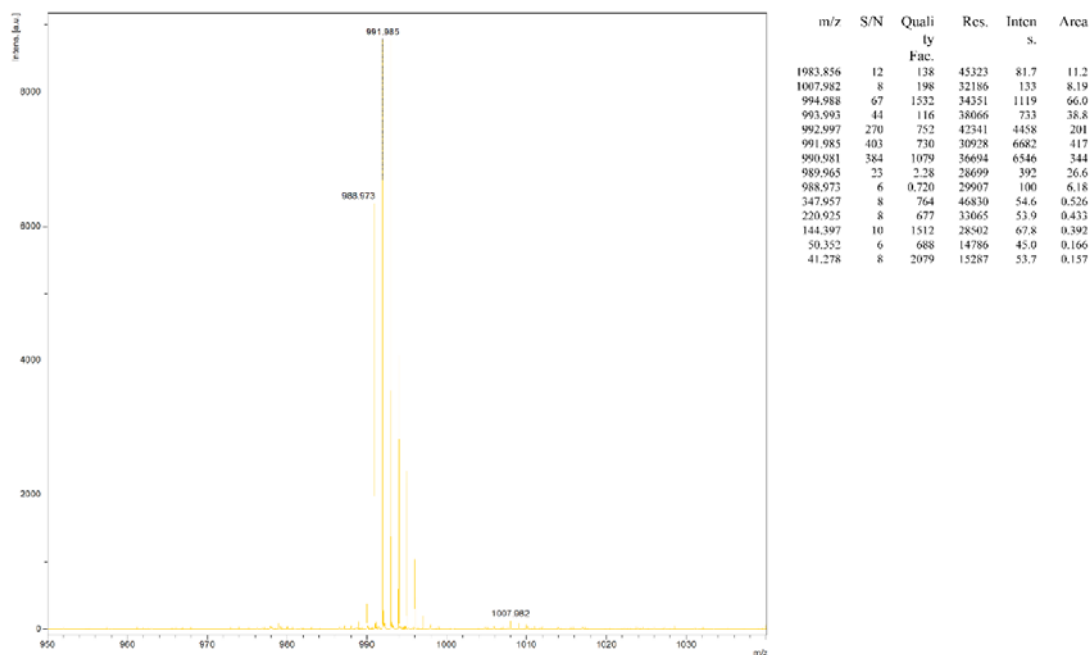
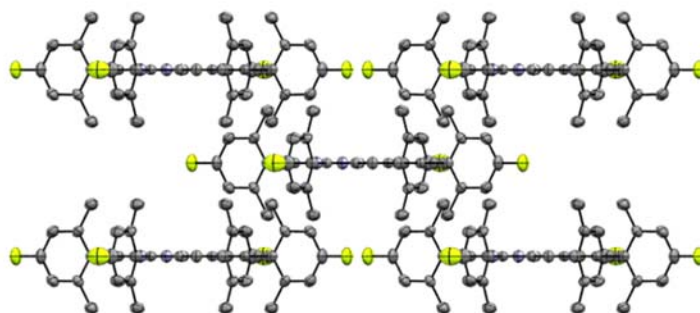


Figure S7. MALDI-TOF of Pt(II)TFP.

(a)



(b)

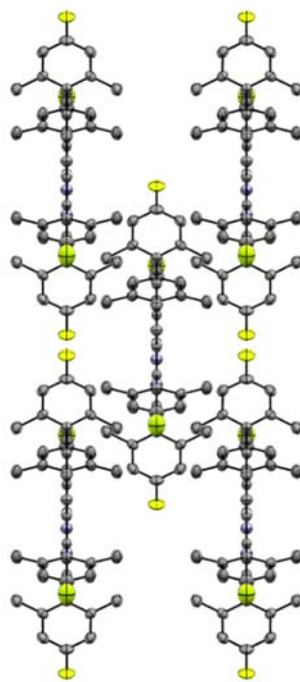


Figure S8. Crystal packing of the **Pt(II)TFP** along the (a) a-axis and (b) b-axis. Hydrogen atoms are omitted for clarity. Color scheme: C (gray), N (light purple), F (yellow-green), Pt (white).

Table S1. Selected geometric parameters for **Pt(II)TFP***Bond Distances (Å)*

Pt1–N1 ⁱ	2.014 (4)
Pt1–N1 ⁱⁱ	2.014 (4)
Pt1–N1 ⁱⁱⁱ	2.014 (4)
Pt1–N1	2.014 (4)
F1–C9	1.358 (7)
N1–C4	1.362 (8)
N1–C1	1.383 (7)
C1–C5 ⁱⁱ	1.391 (8)
C1–C2	1.430 (8)
C2–C3	1.352 (8)
C2–H2	0.9496
C3–C4	1.437 (8)

C3–H3	0.9500
C4–C5	1.400 (8)
C5–C6	1.494 (7)
C6–C7 ^{iv}	1.403 (5)
C6–C7	1.403 (5)
C7–C8	1.389 (6)
C7–C10	1.506 (6)
C8–C9	1.367 (6)
C8–H8	0.9517
C10–H10A	0.9800
C10–H10B	0.9801
C10–H10C	0.9800

Bond Angles (°)

N1 ⁱ –Pt1–N1 ⁱⁱ	180.0
N1 ⁱ –Pt1–N1 ⁱⁱⁱ	90.0
N1 ⁱⁱ –Pt1–N1 ⁱⁱⁱ	90.0
N1 ⁱ –Pt1–N1	89.999 (1)
N1 ⁱⁱ –Pt1–N1	90.0
N1 ⁱⁱⁱ –Pt1–N1	180.0
C4–N1–C1	106.4 (4)
C4–N1–Pt1	127.2 (4)
C1–N1–Pt1	126.3 (4)
N1–C1–C5 ⁱⁱ	126.7 (5)
N1–C1–C2	109.1 (5)

C1 ⁱ –C5–C4	123.3 (5)
C1 ⁱ –C5–C6	118.8 (5)
C4–C5–C6	117.9 (5)
C7 ^{iv} –C6–C7	120.6 (5)
C7 ^{iv} –C6–C5	119.7 (2)
C7–C6–C5	119.7 (2)
C8–C7–C6	118.9 (4)
C8–C7–C10	119.7 (4)
C6–C7–C10	121.3 (3)
C9–C8–C7	119.3 (4)
C9–C8–H8	119.7

C5 ⁱⁱ –C1–C2	124.1 (5)
C3–C2–C1	107.7 (5)
C3–C2–H2	125.9
C1–C2–H2	126.4
C2–C3–C4	106.7 (6)
C2–C3–H3	126.7
C4–C3–H3	126.7
N1–C4–C5	126.4 (5)
N1–C4–C3	110.1 (5)
C5–C4–C3	123.5 (5)
C4–N1–C1–C5 ⁱⁱ	180.0
Pt1–N1–C1–C5 ⁱⁱ	0.0
C4–N1–C1–C2	0.0
Pt1–N1–C1–C2	180.0
N1–C1–C2–C3	0.0
C5 ⁱⁱ –C1–C2–C3	180.0
C1–C2–C3–C4	0.0
C1–N1–C4–C5	180.0
Pt1–N1–C4–C5	0.0
C1–N1–C4–C3	0.0
Pt1–N1–C4–C3	180.0
C2–C3–C4–N1	0.0
C2–C3–C4–C5	180.0
N1–C4–C5–C1 ⁱ	0.0
C3–C4–C5–C1 ⁱ	180.0

C7–C8–H8	121.0
F1–C9–C8 ^{iv}	118.6 (3)
F1–C9–C8	118.6 (3)
C8 ^{iv} –C9–C8	122.9 (6)
C7–C10–H10A	109.6
C7–C10–H10B	109.4
H10A–C10–H10B	109.5
C7–C10–H10C	109.3
H10A–C10–H10C	109.4
H10B–C10–H10C	109.6
N1–C4–C5–C6	180.0
C3–C4–C5–C6	0.0
C1 ⁱ –C5–C6–C7 ^{iv}	-90.8 (5)
C4–C5–C6–C7 ^{iv}	89.2 (5)
C1 ⁱ –C5–C6–C7	90.8 (5)
C4–C5–C6–C7	-89.2 (5)
C7 ^{iv} –C6–C7–C8	0.9 (9)
C5–C6–C7–C8	179.4 (5)
C7 ^{iv} –C6–C7–C10	-179.3 (4)
C5–C6–C7–C10	-0.8 (8)
C6–C7–C8–C9	-0.7 (8)
C10–C7–C8–C9	179.5 (5)
C7–C8–C9–F1	-179.8 (6)
C7–C8–C9–C8 ^{iv}	0.6 (11)

Symmetry code(s): (i) $y, -x + 1, -z + 1$; (ii) $-y + 1, x, z$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $x, y, -z + 1$.

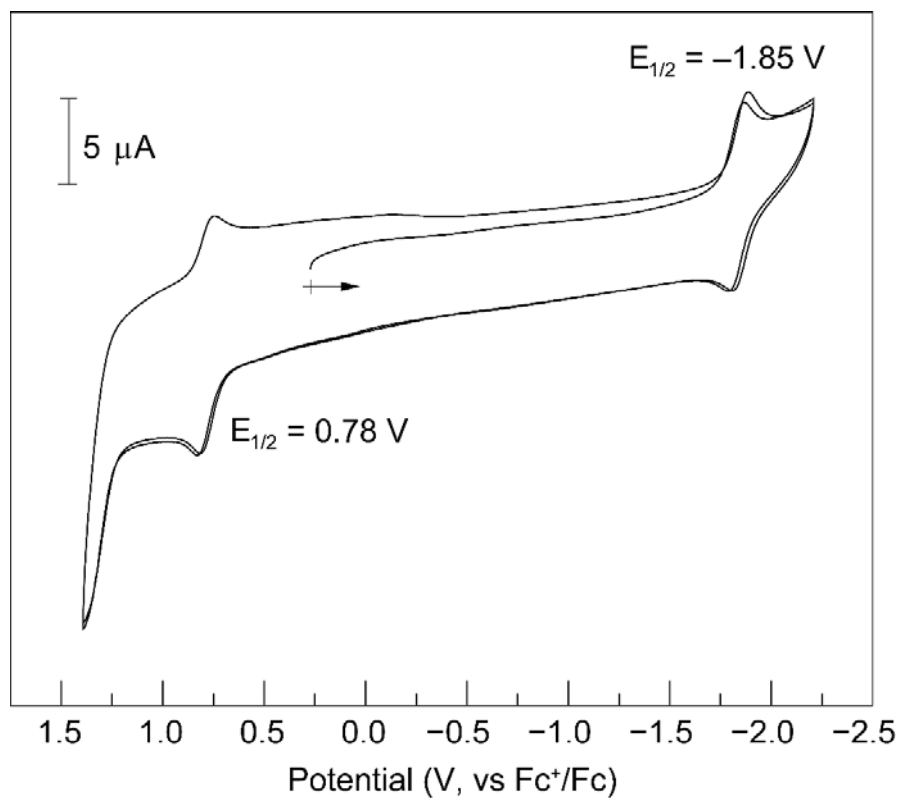


Figure S9. Cyclic voltammogram of Pt(II)TFP, 0.1 M TBAPF₆ in CH₂Cl₂, 100 mV/s, in N₂ filled glovebox.