

**Supporting Information:**

**Weaving a two-dimensional Net of Hydrogen and  
Halogen Bonds: Cocrystal of a Pyrazolium  
Bromide with Tetrafluorodiodobenzene**

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## Additional figures

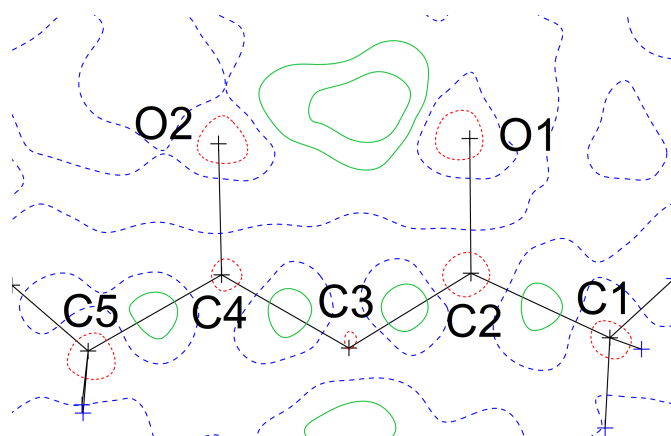


Figure S1: Difference Fourier synthesis map<sup>S1</sup> of **1** before including the enol hydrogen; contour lines drawn at  $0.2 e \text{ \AA}^{-3}$ .

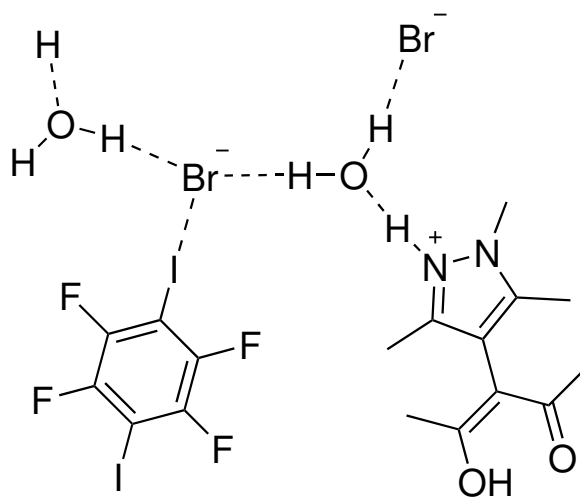


Figure S2: Fragment used for the single point calculation.

## Database searches

Database searches in the CCDC<sup>S2</sup> were conducted with ConQuest 2021.2.0, build 327809. The exact structures and constraints are compiled in Figure S3. Van der Waals radii ( $r_{\text{vdW}}$ ) were used as defined in ConQuest.

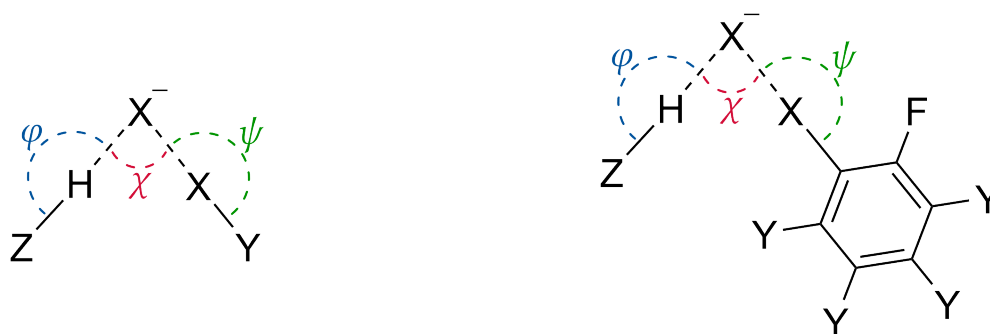


Figure S3: Left: More general query used to classify **3** in the range of related compounds. Right: Query limited to fluorinated halobenzenes as halogen bond donors with only twelve hits. Constraints:  $d_{\text{max}}(\text{H}\cdots\text{X}^-) = \sum(r_{\text{vdW}})$ ;  $d_{\text{max}}(\text{X}^- \cdots \text{X}) = \sum(r_{\text{vdW}}) - 0.2 \text{ \AA}$ ;  $\varphi = 150^\circ$  to  $180^\circ$ ;  $\chi = 60^\circ$  to  $120^\circ$ ;  $\psi = 150^\circ$  to  $180^\circ$ ; X = F, Cl, Br, I; Y = any atom; Z = O, N.

## Powder X-ray diffractograms

Experimental X-ray powder patterns depicted below were collected at room temperature while simulated patterns correspond to the single crystal data obtained at 100 K. This leads to a shift of the lines in the experimental pattern to lower  $2\theta$  values due to the lattice expansion at higher temperature. Reflection intensities of **1** only fit the simulated data if the microcrystalline powder is ground; otherwise the intensities suffer from preferred orientation.

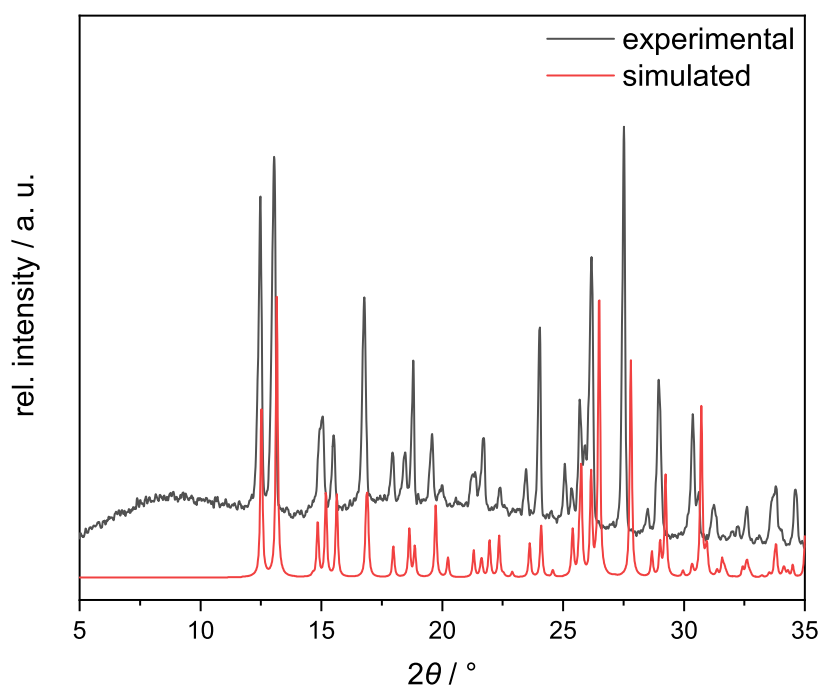


Figure S4: Experimental and simulated powder X-ray diffractogram of **1**.

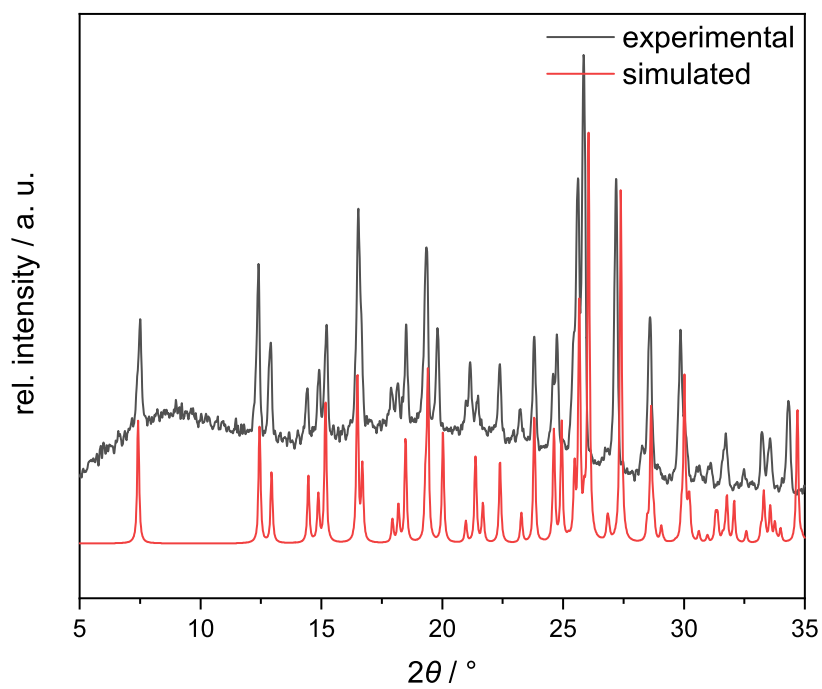


Figure S5: Experimental and simulated powder X-ray diffractogram of **2**.

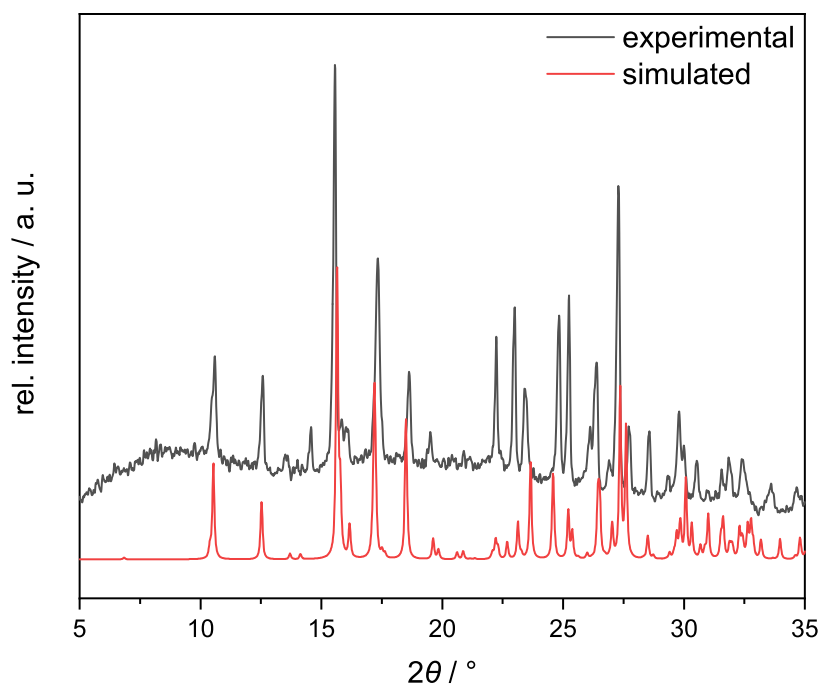


Figure S6: Experimental and simulated powder X-ray diffractogram of **3**.

## References

- (S1) Spek, A. L. Structure validation in chemical crystallography. *Acta Crystallogr.* **2009**, *D65*, 148–155.
- (S2) Groom, C. R.; Bruno, I. J.; Lightfoot, M. P.; Ward, S. C. The Cambridge Structural Database. *Acta Crystallogr.* **2016**, *B72*, 171–179.