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Supporting information for article:

Diversity of *N*-triphenylacetyl-L-tyrosine solvates with halogenated solvents

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Table S1 Experimental details

Experiments were carried out at 130 K with Cu $K\alpha$ radiation. H-atom parameters were constrained.

	(I)	(II)	(III)
Crystal data			
Chemical formula	C ₂₉ H ₂₅ NO ₄ ·CHCl ₃	C ₂₉ H ₂₅ NO ₄ ·CH ₂ Cl ₂	C ₂₉ H ₂₅ NO ₄ ·1.5(CHCl ₃)
M_r	570.87	536.42	570.87
Crystal system, space group	Monoclinic, $P2_1$	Monoclinic, $P2_1$	Orthorhombic, $P2_12_12$
a, b, c (Å)	10.0732 (3), 10.0051 (2), 14.7474 (4)	10.04475 (13), 9.82563 (12), 14.59465 (18)	9.27898 (12), 32.3574 (4), 9.96336 (12)
α, β, γ (°)	90, 109.945 (3), 90	90, 108.7422 (13), 90	90, 90, 90
V (Å ³)	1397.14 (6)	1364.05 (3)	2991.44 (6)
Z	2	2	4
D_x (Mg m ⁻³)	1.357	1.306	1.268
μ (mm ⁻¹)	3.27	2.43	3.05
Crystal size (mm)	0.22 × 0.05 × 0.03	0.30 × 0.06 × 0.03	0.35 × 0.02 × 0.01
Data collection			
Diffractometer	SuperNova, Single source at offset, Atlas	SuperNova, Single source at offset/far, Atlas	SuperNova, Single source at offset/far, Atlas
Absorption correction	Multi-scan	Gaussian	Multi-scan
T_{\min}, T_{\max}	0.408, 1.000	0.709, 1.000	0.713, 1.000
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	18782, 4905, 4842	20920, 5550, 5259	29945, 6241, 5987
R_{int}	0.050	0.034	0.040
$(\sin \theta/\lambda)_{\text{max}}$ (Å ⁻¹)	0.595	0.628	0.630
Refinement			
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.048, 0.129, 1.03	0.035, 0.095, 1.05	0.032, 0.084, 1.01
No. of reflections	4905	5550	6241
No. of parameters	399	336	344

No. of restraints	35	1	0
$\Delta\rho_{\max}, \Delta\rho_{\min}$ ($e \text{ \AA}^{-3}$)	0.59, -0.41	0.42, -0.49	0.21, -0.27
Absolute structure	Flack x determined using 2221 quotients [(I+)-(I-)]/[(I+)+(I-)]	Flack x determined using 2293 quotients [(I+)-(I-)]/[(I+)+(I-)]	Flack x determined using 2454 quotients [(I+)-(I-)]/[(I+)+(I-)]
Absolute structure parameter	0.003 (12)	-0.009 (9)	0.034 (6)

	(IV)	(V)
Crystal data		
Chemical formula	$C_{29}H_{25}NO_4 \cdot 0.1(CH_2Cl_2)$	$C_{29}H_{25}NO_4 \cdot 1.5(CHCl_3)$
M_r	459.99	630.55
Crystal system, space group	Orthorhombic, $P2_12_12_1$	Monoclinic, $P2_1/c$
a, b, c (\AA)	9.28862 (9), 10.34013 (9), 24.3525 (2)	10.00099 (16), 9.33129 (13), 31.8959 (5)
α, β, γ ($^\circ$)	90, 90, 90	90, 90.4958 (13), 90
V (\AA^3)	2338.95 (4)	2976.48 (8)
Z	4	4
D_x (Mg m^{-3})	1.306	1.407
μ (mm^{-1})	0.90	4.33
Crystal size (mm)	$0.20 \times 0.15 \times 0.03$	$0.35 \times 0.02 \times 0.02$
Data collection		
Diffractometer	SuperNova, Single source at offset/far, Atlas	SuperNova, Single source at offset/far, Atlas
Absorption correction	Multi-scan	Multi-scan
T_{\min}, T_{\max}	0.882, 1.000	0.773, 1.000
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	30278, 4276, 4225	20990, 6149, 4806
R_{int}	0.024	0.046
$(\sin \theta/\lambda)_{\max}$ (\AA^{-1})	0.602	0.630
Refinement		
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.031, 0.086, 1.05	0.058, 0.158, 1.04
No. of reflections	4276	6149
No. of parameters	494	372
No. of restraints	154	0

$\Delta\rho_{\max}, \Delta\rho_{\min}$ ($e \text{ \AA}^{-3}$) 0.13, -0.14 0.84, -0.84

Absolute structure Flack x determined using 1770
quotients $[(I^+)-(I^-)]/[(I^+)+(I^-)]$

Absolute structure
parameter 0.03 (3)

Computer programs: *CrysAlis PRO* 1.171.41.112a (Rigaku OD, 2021), *SHELXT* 2018/2 (Sheldrick, a2018),
SHELXL2018/3 (Sheldrick, 2018).

Table S2 Selected hydrogen-bond parameters

$D-H\cdots A$	$D-H$ (Å)	$H\cdots A$ (Å)	$D\cdots A$ (Å)	$D-H\cdots A$ (°)
(I)				
O1—H1 \cdots O3 ⁱⁱ	0.84	1.81	2.642 (4)	173.7
O4—H4 \cdots O2 ⁱⁱⁱ	0.84	1.93	2.762 (4)	174.3
C8—H8 \cdots O3 ⁱ	0.95	2.53	3.427 (4)	156.7
C37 ^a —H37 ^a \cdots O4	1.00	2.32	3.165 (7)	141.8
C26—H26 \cdots O3	0.95	2.27	2.946 (5)	127.0
(II)				
O1—H1 \cdots O4 ⁱⁱ	0.84	1.80	2.641 (3)	175.4
O3—H3 \cdots O2 ⁱⁱⁱ	0.84	1.93	2.772 (3)	175.7
C26—H26 \cdots O4	0.95	2.27	2.940 (3)	126.6
C6—H6 \cdots O2 ⁱⁱⁱ	0.95	2.60	3.288 (3)	129.2
C6—H6 \cdots O4 ⁱ	0.95	2.53	3.409 (3)	153.8
(III)				
O1—H1 \cdots O1 ^{iv}	0.84	1.83	2.635 (4)	160.3
O2—H2 \cdots O2 ^{iv}	0.84	1.80	2.627 (3)	168.7
O4—H4 \cdots O3 ^v	0.84	1.80	2.642 (3)	175.9
C36—H36 \cdots O3	0.95	2.32	2.825 (3)	112.8
C25—H25 \cdots O4 ^{vi}	0.95	2.41	3.292 (4)	153.9
(IV)				
O4—H4 \cdots O3 ^{vii}	0.84	1.91	2.729 (2)	166.0
O1—H1 \cdots O4 ⁱⁱⁱ	0.84	1.89	2.700 (2)	161.4
C16A ^b —H16A ^b \cdots O3	0.95	2.04	2.708 (10)	126.2
(V)				

O1—H1 \cdots O2 ^{viii}	0.84	1.81	2.643 (3)	174.6
O4—H4 \cdots O3 ⁱⁱⁱ	0.84	1.79	2.628 (3)	173.0
C22—H22 \cdots O3	0.95	2.38	2.822 (4)	107.8

Symmetry code(s): (i) $-x+2, y-1/2, -z+2$; (ii) $-x+2, y+1/2, -z+2$; (iii) $x, y-1, z$; (iv) $-x+2, -y+1, z$; (v) $x-1, y, z$; (vi) $x+1, y, z+1$; (vii) $x-1/2, -y+3/2, -z+1$; (viii) $-x+1, -y+2, -z+1$.

Table S3 Intermolecular hydrogen bonds in the known **TrCOTyr** polymorphs, solvates and cocrystals

refcode	composition	1	2	3	4	5	6	7	8
EGUVUA	L-TrCOTyr	-	-	√	-	-	√	√	2D
EGUVUA01	L-TrCOTyr	-	-	-	√	√	-	-	1D
NOQSAQ	L-TrCOTyr·MeOH	√	-	-	√	-	√	√	2D*
NOQSEU	L-TrCOTyr·EtOH	√	-	-	√	-	√	√	2D*
NOQSIY	<i>rac</i> -TrCOTyr	-	-	-	√	√	-	-	1D
NOQSOE	<i>rac</i> -TrCOTyr·MeOH	√	-	-	√	-	√	√	2D*
NOQSUK	<i>rac</i> -TrCOTyr·EtOH	√	-	-	√	-	√	√	2D*
NOQTAR	L-TrCOTyr·NPRD	-	-	-	-	-	√	-	2D**
NOQTEV	L-TrCOTyr·QX	-	-	-	-	-	√	-	1D
NOQTIZ	(L-TrCOTyr) ₂ ·QX	-	-	-	√	-	√	-	1D**
NOQTOF	(TrCOTyr) ₂ ·BIPY	-	-	-	√	-	-	-	1D**
NOQTUL	(TrCOTyr) ₂ ·BIPY	-	-	-	√	-	-	-	1D**
NOQVAT	L-TrCOTyr·DABCO	-	√	-	√	-	-	-	1D

1 – $R_2^2(8)$ carboxylic supramolecular synthon; 2 – O=CO-H \cdots O=COH; 3 – O=CO-H \cdots OH; 4 – O-H \cdots O=COH; 5 – C=OO-H \cdots O=CNH; 6 – O-H \cdots O=CNH; 7 – O=CNH \cdots O; 8 – dimensionality of the supramolecular structure by hydrogen bonds.

* alcohol molecules (methanol or ethanol) take part in the expansion of the supramolecular motif by O-H \cdots O bonds

** diazaheterocyclic amine molecules interact with the **TrCOTyr** molecule through O-H \cdots N hydrogen bonds and participate in the development of the supramolecular motif.