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Supporting information for article:

Ψ -Polyhedral symbols for coordination geometries of lead(II) with stereochemically active lone pair

Seik Weng Ng

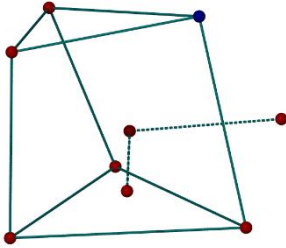
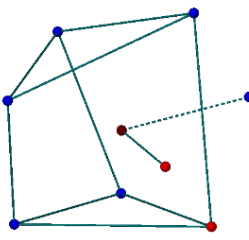
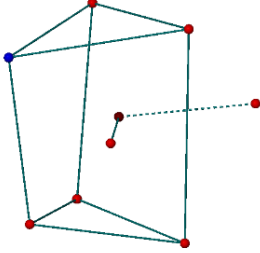
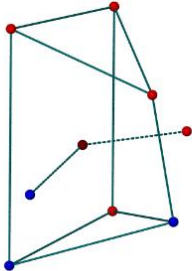
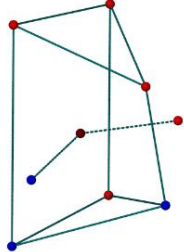
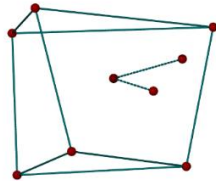
Table S1. CH&N % recalculated from published (Yin *et al.*, 2019) crystal structures

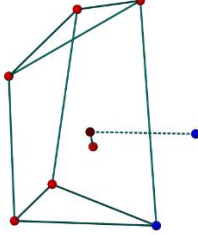
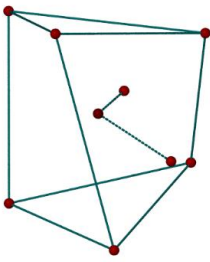
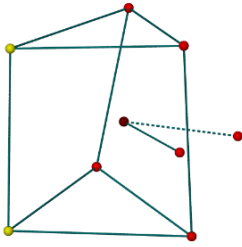
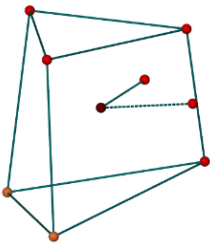
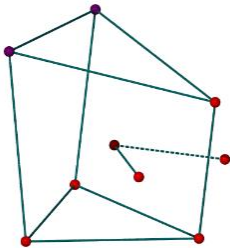
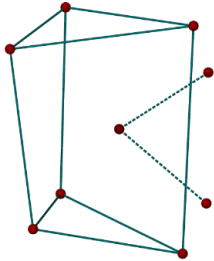
Compound	Formula	CH&N % (calculated)	CH&N % (found)
[Pb ₂ (H ₂ O) _{0.5} (C ₄ H ₉ NO) _{0.5} (C ₁₃ H ₇ NO ₄) ₂].1.5(C ₄ H ₉ NO)	C ₃₄ H ₃₃ N ₄ O _{10.5} Pb ₂	C 37.81 H 3.08 N 5.19	C 38.42 H 2.77 N 5.18
[Pb ₂ (H ₂ O) _{0.5} (C ₃ H ₇ NO) _{0.5} (C ₁₃ H ₇ NO ₄) ₂].1.5(C ₃ H ₇ NO)	C ₃₂ H ₂₉ N ₄ O _{10.5} Pb ₂	C 36.53 H 2.78 N 5.37	C 36.88 H 3.20 N 5.83
[Pb ₂ (H ₂ O) _{0.25} (C ₃ H ₇ NO) _{0.75} (C ₁₃ H ₇ NO ₄) ₂].C ₆ H ₁₅ N).0.5(C ₃ H ₇ NO)	C ₃₆ H _{38.5} N _{4.25} O _{9.5} Pb ₂	C 39.42 H 3.51 N 5.43	C 39.92 H 3.45 N 5.11

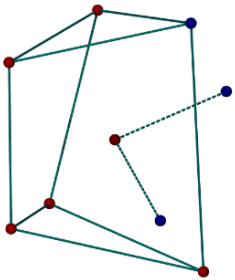
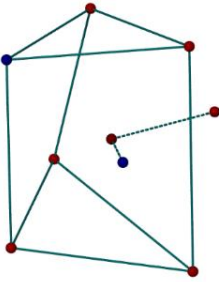
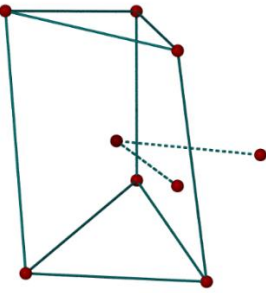
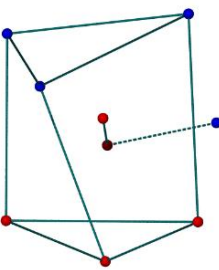
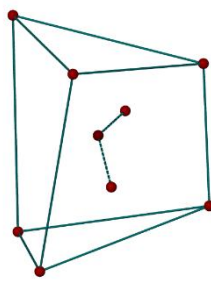
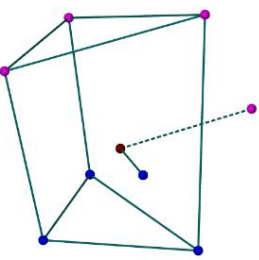
Table S2. Refinements and electron count in voids (voids on a center-of-symmetry) from published (Yin *et al.*, 2019) diffraction data

Structure	1	2	3
<i>R</i>	0.028	0.035	0.026
<i>S</i>	1.013	1.078	1.021
<i>w</i>	$[\sigma^2 + (0.035P)^2 + 0.774P]$ 1	$[\sigma^2 + (0.057P)^2 + 1.846P]$ 1	$[\sigma^2 + (0.031P)^2]$ 1
No. electrons	155	96	99
Volume of void, Å ³	534	526	485
Recovered electrons in void/unit cell	155	96	99
Solvent molecules squeezed/ unit cell	3 DMA	3 DMF	2 NEt ₃ + 1DMF
Number of electrons squeezed	144	120	156
Old CCDC Number	1943141	1943138	1943143
New CCDC Number	2069799	2069800	2069801

Table S3. Lead(II) coordination assigned as Ψ -tricapped trigonal prism

Pb(II) compound		IUCr/IUPAC	Ref.
$[\text{Pb}_2(\text{H}_2\text{O})_{0.25}(\text{C}_3\text{H}_7\text{NO})_{0.75}(\text{C}_{13}\text{H}_7\text{NO}_4)_2] \cdot 1.5(\text{C}_3\text{H}_7\text{NO})$ Pb at (0.5087, 0.8373, 0.0453)		Ψ -TRPS-9	This work
$\text{Pb}_2(\text{NO}_3)_2(\text{C}_2\text{N}_3)_2(\text{C}_{16}\text{H}_{10}\text{N}_4\text{O}_2)_4$		Ψ -TRPS-9	Marandi <i>et al.</i> , 2019
$\text{Pb}(\text{C}_7\text{H}_3\text{NO}_4)$		Ψ -TRPS-9	Lush & Shen, 2011
$\text{Pb}_4(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{C}_9\text{H}_6\text{NO})_4(\text{C}_5\text{H}_2\text{ClO}_2\text{S})_4$ Pb at (0.5240, 0.3350, 0.2262)		Ψ -TRPS-9	Jennifer & Muthiah, 2014
$\text{Pb}(\text{ClO}_4)(\text{C}_{12}\text{H}_8\text{N}_2)(\text{C}_2\text{H}_3\text{O}_2)$		Ψ -TRPS-9	Hall <i>et al.</i> , 2000
$\text{Pb}_4(\text{H}_2\text{O})(\text{C}_7\text{H}_{12}\text{NO}_5\text{P})(\text{C}_8\text{H}_3\text{O}_5\text{S})_2 \cdot 2\text{H}_2\text{O}$ Pb at (0.8524, 0.7954, 0.3525)		Ψ -TRPS-9	Chen <i>et al.</i> , 2011
$\text{Pb}(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_7\text{H}_4\text{O}_6\text{S}) \cdot 0.5(\text{C}_{10}\text{H}_8\text{N}_2)$		Ψ -TRPS-9	Fan & Zhu, 2007

			
$\text{Pb}_2(\text{C}_3\text{H}_7\text{NO})(\text{C}_9\text{H}_3\text{O}_6)_2$		ψ -TRPS-9	Gao <i>et al.</i> , 2019
$2(\text{C}_3\text{H}_{10}\text{N})\text{Pb}_4\text{Cl}_2(\text{C}_{12}\text{H}_6\text{O}_4)_4$		ψ -TRPS-9	Lin <i>et al.</i> , 2020
$2(\text{C}_2\text{H}_8\text{N})\text{Pb}_4\text{Br}_2(\text{C}_{12}\text{H}_6\text{O}_4)_4$		ψ -TRPS-9	
$2(\text{C}_2\text{H}_8\text{N})\text{Pb}_4\text{I}_2(\text{C}_{12}\text{H}_6\text{O}_4)_4$		ψ -TRPS-9	
$\text{PbCu}_2(\text{NO}_3)_2(\text{C}_{10}\text{H}_{20}\text{N}_2\text{O}_2)$		ψ -TRPS-9	Chakraborty & Mohanta, 2017

$\text{Pb}(\text{H}_2\text{O})_3(\text{C}_{18}\text{H}_{14}\text{N}_8\text{O}_4)\cdot\text{H}_2\text{O}$		ψ -TRPS-9	Li <i>et al.</i> , 2021
$\text{Pb}(\text{C}_9\text{H}_7\text{N}_4\text{O}_2)(\text{C}_2\text{H}_3\text{O}_2)$		ψ -TRPS-9	
$\text{Pb}(\text{C}_{10}\text{H}_{14}\text{O}_4)$ Pb at (0.7934, 0.2692, 0.7062)		ψ -TRPS	Zhu <i>et al.</i> , 2016
$\text{Pb}(\text{C}_{24}\text{H}_{16}\text{N}_2)_2(\text{C}_{10}\text{H}_6\text{F}_3\text{O}_2)_2$		ψ -TRPS-9	Hosseini <i>et al.</i> , 2013
$\text{Pb}(\text{NO}_3)(\text{C}_6\text{H}_4\text{NO}_3)$		ψ -TRPS-9	Jin, 2020
$\text{Pb}(\text{SCN})_2(\text{C}_4\text{H}_6\text{N}_2)$		ψ -TRPS-9	Mohammadi <i>et al.</i> , 2019

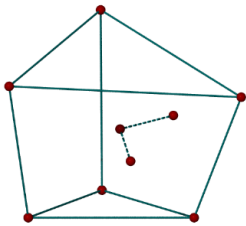
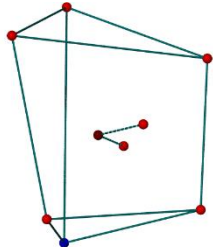
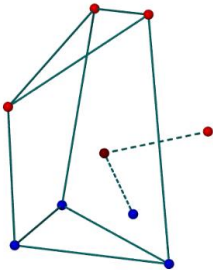
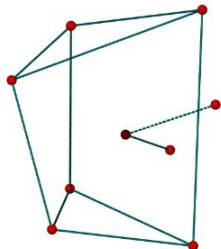
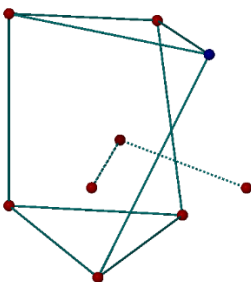
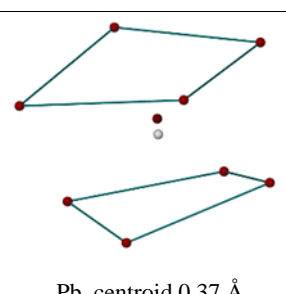
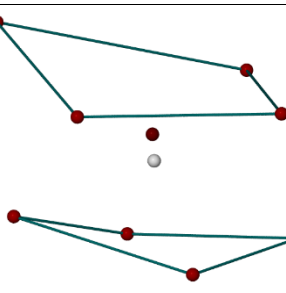
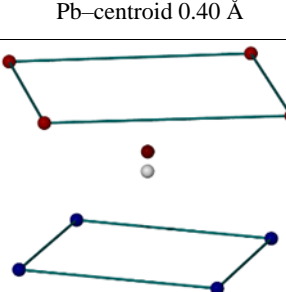
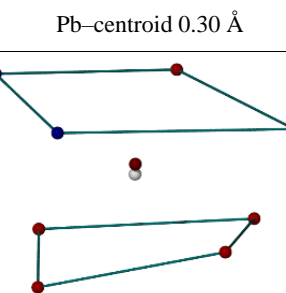
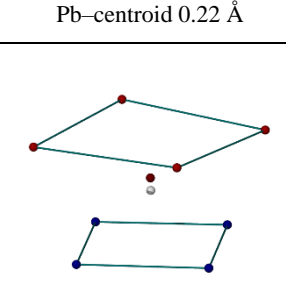
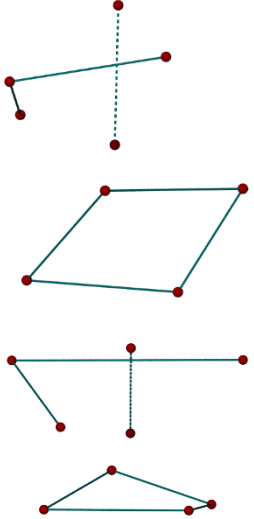
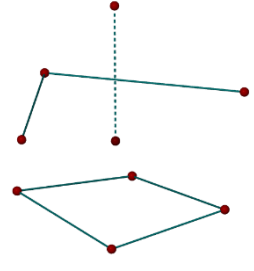
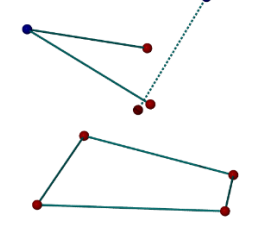
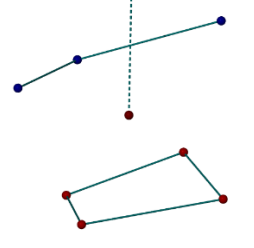
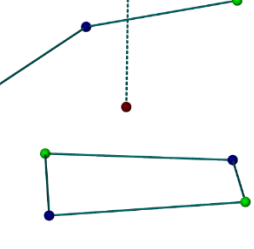
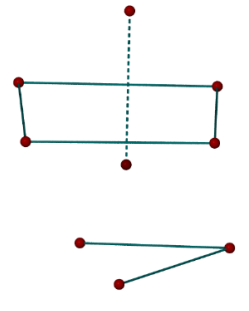
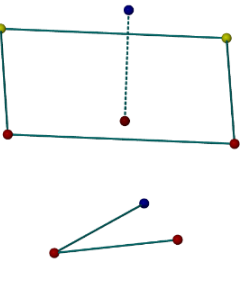
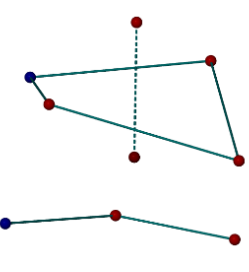
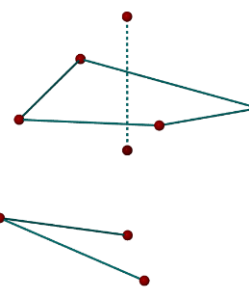
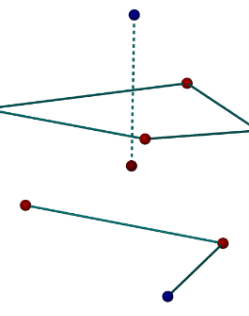
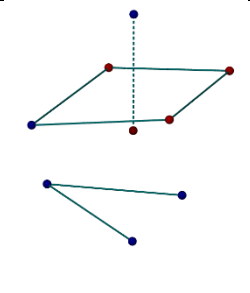
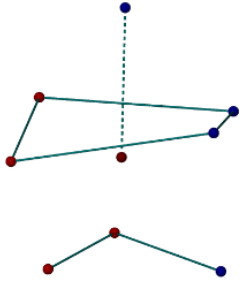
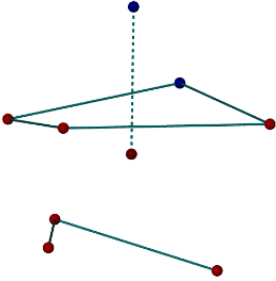
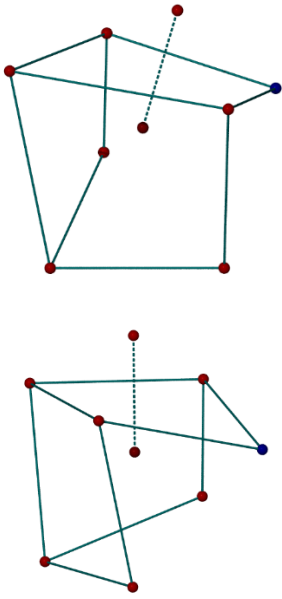
$\text{Pb}_4(\text{H}_2\text{O})(\text{C}_5\text{H}_4\text{NO}_2)_8$ Pb at (0.5133, 0.8733, 0.9954)		ψ -TRPS-9	Kowalik <i>et al.</i> , 2017
$\text{Pb}(\text{H}_2\text{O})_2(\text{C}_7\text{H}_3\text{NO}_4)(\text{C}_7\text{H}_4\text{NO}_4)\cdot\text{H}_2\text{O}$		ψ -TRPS-9	Harrowfield <i>et al.</i> , 2003
$\text{Pb}(\text{H}_2\text{O})(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{C}_6\text{H}_6\text{NO}_3\text{S})_2$		ψ -TRPS-9	Cheng <i>et al.</i> , 2014
$\text{Pb}_2(\text{C}_9\text{H}_3\text{O}_6)(\text{C}_9\text{H}_4\text{O}_6)\cdot 2\text{H}_2\text{O}$ Pb at (0.4259, 0.7468, 0.8557)		ψ -TRPS-9	Hayati <i>et al.</i> , 2018
$\text{Pb}(\text{H}_2\text{O})(\text{C}_6\text{H}_4\text{NO}_2)_2$		ψ -TRPS-9	Zhou <i>et al.</i> , 2007

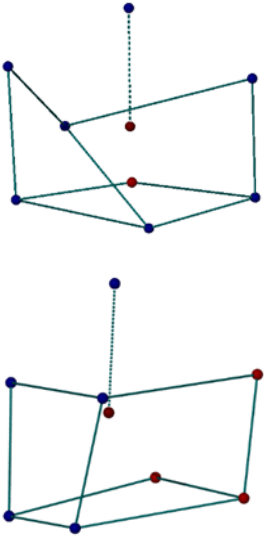
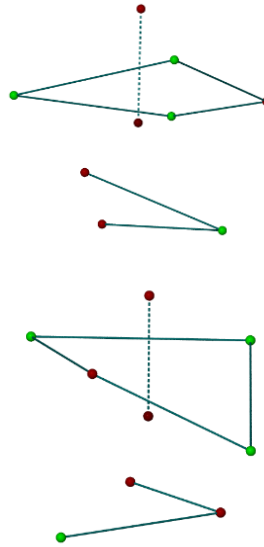
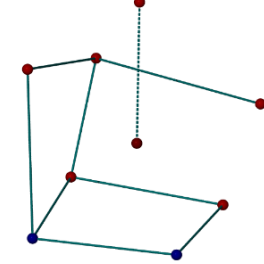
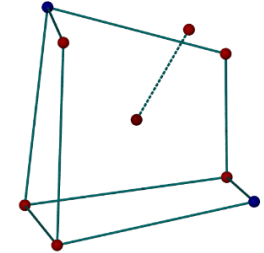
Table S4. Lead(II) coordination assigned as Ψ -monocapped square antiprism, Ψ -monocapped cube and Ψ -bicapped cube

Pb(II) compound		IUCr/IUPAC	Ref.
$\text{Pb}_3(\text{C}_7\text{H}_{12}\text{NO}_6\text{P})_2(\text{C}_7\text{H}_4\text{O}_6\text{S}) \cdot 2\text{H}_2\text{O}$ Pb at (1.2416, 0.0878, 0.4927)	 <p>Pb-centroid 0.37 Å</p>	Ψ -SAPRS-9	Chen <i>et al.</i> , 2011
$\text{Pb}(\text{C}_{10}\text{H}_{14}\text{O}_4)$ Pb at (0.9417, 0.7655, 0.6773)	 <p>Pb-centroid 0.40 Å</p>	Ψ -SAPRS-9	Zhu <i>et al.</i> , 2016
$\text{Pb}(\text{NO}_3)_2(\text{C}_{16}\text{H}_{32}\text{N}_8\text{O}_4) \cdot 3.5\text{H}_2\text{O}$	 <p>Pb-centroid 0.30 Å</p>	Ψ -SAPRS-9	Cuenot <i>et al.</i> , 2008
$\text{PbC}_{20}\text{H}_{14}\text{N}_4\text{O}_2(\text{C}_8\text{H}_4\text{O}_4)$	 <p>Pb-centroid 0.22 Å</p>	Ψ -SAPRS-9	Yan <i>et al.</i> , 2013
$\text{Pb}(\text{C}_{16}\text{H}_{32}\text{N}_8\text{O}_4)(\text{CF}_3\text{O}_3\text{S})_2$	 <p>Pb-centroid 0.29 Å</p>	Ψ -SAPRS-9	Lyczko <i>et al.</i> , 2020

<p>$\text{Pb}_4(\text{H}_2\text{O})(\text{C}_7\text{H}_{12}\text{NO}_5\text{P})(\text{C}_8\text{H}_3\text{O}_5\text{S})_2 \cdot 2\text{H}_2\text{O}$ Pb at (1.0282, -0.2667, -0.0103)</p> <p>$\text{Pb}_3(\text{C}_7\text{H}_{12}\text{NO}_6\text{P})_2(\text{C}_7\text{H}_4\text{O}_6\text{S}) \cdot 2\text{H}_2\text{O}$ Pb at (0.8665, 0.0909, 0.3887)</p>		Ψ -SAPRS-9	Chen <i>et al.</i> , 2011
<p>$\text{Pb}(\text{C}_4\text{H}_4\text{O}_6)$</p>		Ψ -SAPRS-9	Weil, 2014
<p>$\text{Pb}(\text{C}_6\text{H}_4\text{NO}_2)_2$</p>		Ψ -SAPRS-9	Shahverdizadeh <i>et al.</i> , 2008
<p>$\text{Pb}(\text{NO}_2)_2(\text{C}_{12}\text{H}_8\text{N}_2)_2$</p>		Ψ -SAPRS-9	Morsali, 2005
<p>$\text{Pb}(\text{C}_4\text{H}_3\text{N}_2\text{S})_2$</p>		Ψ -SAPRS	Schwade <i>et al.</i> , 2015

Pb ₂ (W ₁₂ SiO ₃₆)(C ₃ H ₇ NO) ₁₆		ψ -SAPRS-9	Pang <i>et al.</i> , 2019
PbI(C ₉ H ₇ N ₄ O ₂)		ψ -SAPRS-9	Li <i>et al.</i> , 2021
Pb(H ₂ O)(NO ₃) ₂ (C ₁₂ H ₈ N ₂)		ψ -SAPRS	Zhang & Lu, 2005
Pb ₂ (H ₂ O)(C ₁₀ H ₈ N ₂)(C ₈ H ₃ F ₂ O ₄)·3H ₂ O Pb at (0.1766, 0.5038, -0.0816)		ψ -SAPRS-9	Song <i>et al.</i> , 2014
Pb(NO ₃)(C ₅ H ₆ N ₂ O ₃) ₂ ·2NO ₃		ψ -SAPRS-9	Masternak <i>et al.</i> , 2005
Pb(H ₂ O) ₂ [B(CN) ₄] ₂		ψ -SAPRS-9	Siegesmund <i>et al.</i> , 2021

Pb[B(CN) ₄](C ₂ H ₃ O ₂) ₂ C ₂ H ₆ O		ψ-SAPRS-9	
Pb ₂ (C ₁₂ H ₈ N ₂) ₂ (C ₇ H ₄ ClO ₃) ₄ ·2H ₂ O		ψ-SAPRS-9	Tang & Wen, 2021
<p>[Pb₂(H₂O)_{0.5}(C₄H₉NO)_{0.5}(C₁₃H₇NO₄)₂]·1.5(C₄H₉NO)</p> <p>Pb at (0.4855, 0.4368, 0.1638)</p> <p>Pb₂(H₂O)_{0.5}(C₃H₇NO)_{0.5}(C₁₃H₇NO₄)₂]·1.5(C₃H₇NO)</p> <p>Pb at (0.4870, 0.9424, 0.664)</p>		<p>ψ-CUS-9</p> <p>ψ-CUS-9</p>	This work

<p>$\text{Pb}_2(\text{C}_{51}\text{H}_{47}\text{N}_{15}\text{O}_2)(\text{CH}_4\text{O})(\text{CF}_3\text{O}_3\text{S})\cdot 3\text{CF}_3\text{O}_3\text{S}\cdot 2\text{CH}_4\text{O}\cdot 2\text{H}_2\text{O}$ Pb at (0.1421, 0.3699, 0.6108)</p> <p>$[\text{Pb}_2(\text{C}_{51}\text{H}_{47}\text{N}_{15}\text{O}_2)(\text{CH}_4\text{O})(\text{CF}_3\text{O}_3\text{S})]\cdot 3\text{CF}_3\text{O}_3\text{S}\cdot 2\text{CH}_4\text{O}\cdot 2\text{H}_2\text{O}$ Pb at (0.7551, 0.3882, 0.9049)</p>		<p>ψ-CUS-9</p> <p>ψ-CUS-9</p>	<p>Lobo <i>et al.</i>, 2019</p>
<p>$\text{Pb}(\text{NO}_3)_2(\text{C}_7\text{H}_6\text{N}_2\text{S})_2$</p> <p>$\text{Pb}(\text{NO}_3)_2(\text{C}_3\text{H}_5\text{N}_3\text{S})_4$</p>		<p>ψ-CUS-9</p> <p>ψ-CUS-9</p>	<p>Fan & Zhu, 2006</p>
<p>$[\text{Pb}_3(\text{C}_{12}\text{H}_8\text{N}_2)_3(\text{C}_7\text{H}_3\text{O}_6\text{S})_2]$ Pb at (0.8519, 0.3547, 0.2850)</p>		<p>ψ-CUS-9</p>	<p>Imran <i>et al.</i>, 2015</p>
<p>$\text{Pb}(\text{C}_6\text{H}_4\text{NO}_2)_2$</p>		<p>ψ-CUS-9</p>	<p>Etaiw <i>et al.</i>, 2018</p>

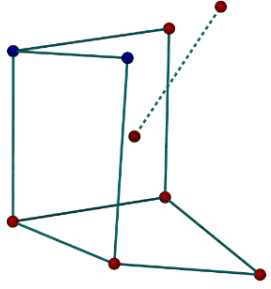
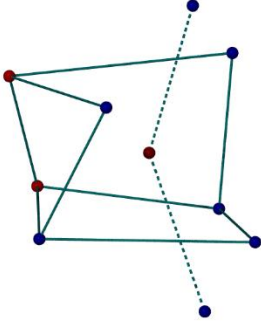
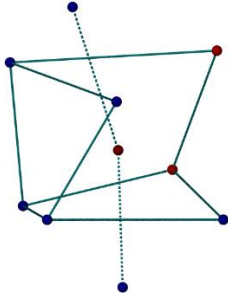
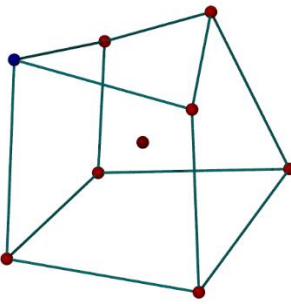
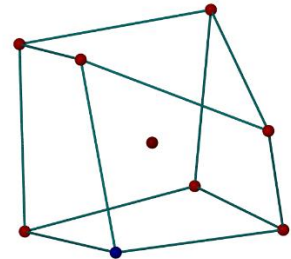
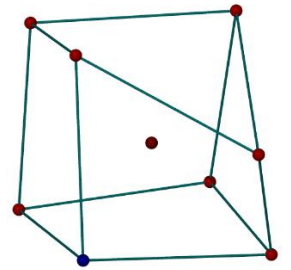
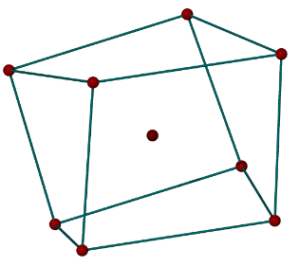
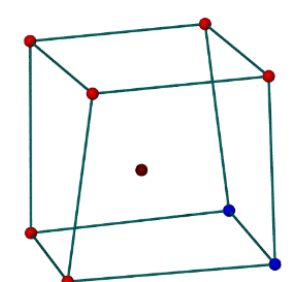
$\text{Pb}(\text{NO}_3)_2(\text{C}_{14}\text{H}_{12}\text{N}_2)$		ψ -CUS-9	Morsali, 2004
$[\text{Pb}_2(\text{C}_{51}\text{H}_{47}\text{N}_{15}\text{O}_2)(\text{CF}_3\text{O}_3\text{S})_2] \cdot 3\text{CF}_3\text{O}_3\text{S} \cdot \text{C}_2\text{H}_3\text{N}$ Pb at (0.6666, 0.2320, 1.0034)		ψ -CUS-10	Lobo <i>et al.</i> , 2019
$[\text{Pb}_2(\text{C}_{51}\text{H}_{47}\text{N}_{15}\text{O}_2)(\text{CF}_3\text{O}_3\text{S})_2] \cdot 3\text{CF}_3\text{O}_3\text{S} \cdot \text{C}_2\text{H}_3\text{N}$ Pb at (0.7551, 0.3882, 0.9049)		ψ -CUS-10	

Table S5. Lead(II) coordination assigned as 8-vertex polyhedral (stereochemically-inert lone pair)

Pb(II) compound		IUCr/IUPAC	Ref.
<p>[Pb₂(H₂O)_{0.5}(C₄H₉NO)_{0.5}(C₁₃H₇NO₄)₂]·1.5(C₄H₉NO)</p> <p>)</p> <p>Pb at (-0.1472, -0.4080, 0.0488)</p>		CU-8	This work
<p>[Pb₂(H₂O)_{0.5}(C₃H₇NO)_{0.5}(C₁₃H₇NO₄)₂]·1.5(C₃H₇NO)</p> <p>)</p> <p>Pb at (-0.1460, 0.0933, 0.5471)</p>		CU-8	
<p>Pb₂(H₂O)_{0.25}(C₃H₇NO)_{0.75}(C₁₃H₇NO₄)₂·1.5(C₃H₇NO)</p> <p>)</p> <p>Pb at (1.1443, 0.9460, 0.9024)</p>		CU-8	
<p>Pb₅O(C₁₂H₆O₄)₄</p> <p>Pb at (0, 1.1331, ¼)</p>		CU-8	Zhou, 2020
<p>Pb₂(C₁₄H₁₂N₂)₂(C₁₁H₈F₃O₃)₄</p>		CU-8	Marandi & Krautscheid, 2009

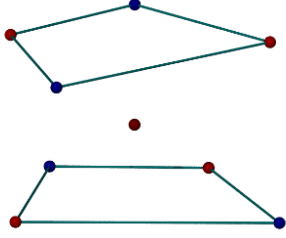
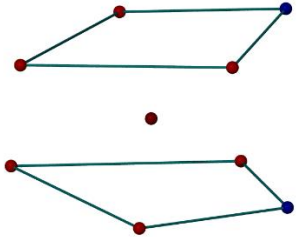
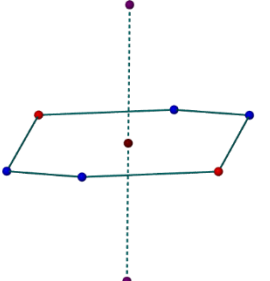
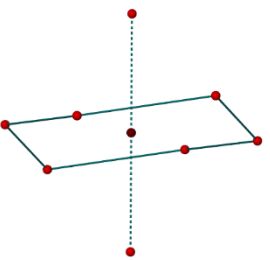
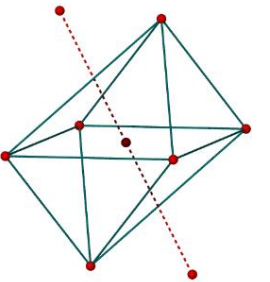
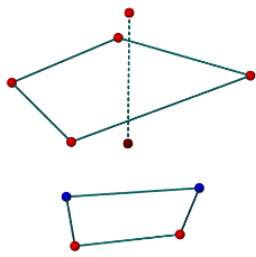
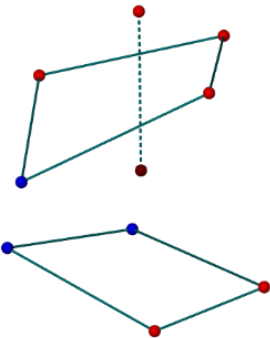
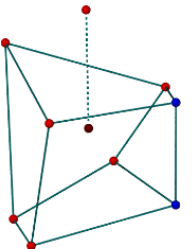
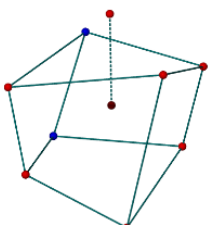
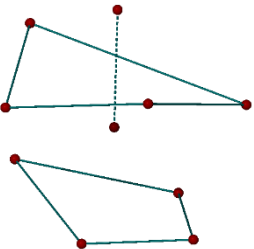
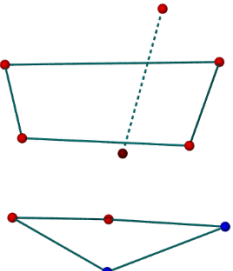
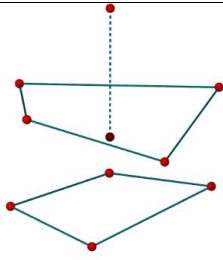
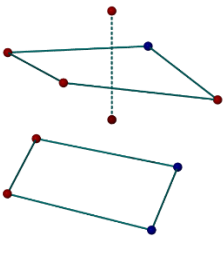
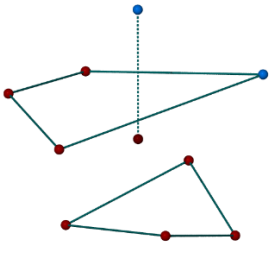
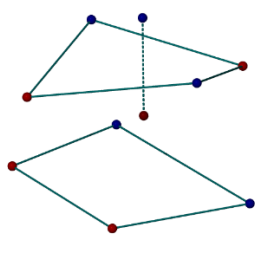
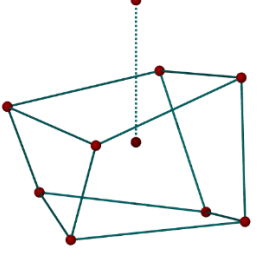
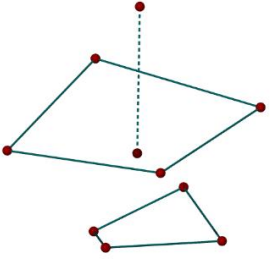
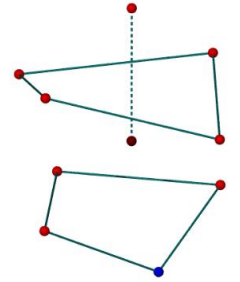
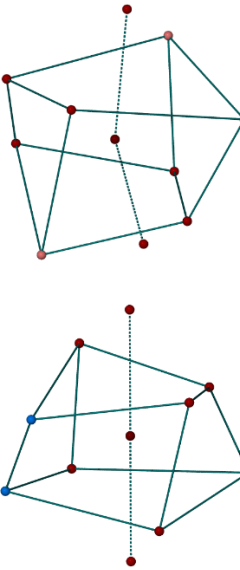
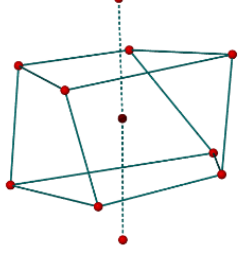
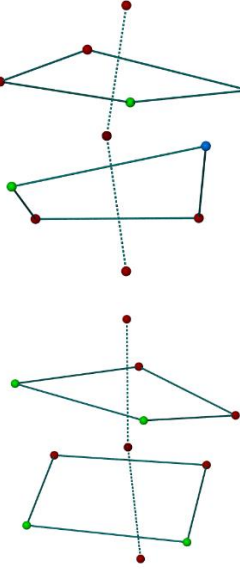
$\text{Pb}(\text{C}_5\text{H}_3\text{N}_2\text{O}_2)_2$		SAPR-8	Echeverría <i>et al.</i> , 2017
$\text{Pb}(\text{C}_4\text{H}_4\text{NO}_4\text{S})_2$		SAPR-8	Starosta & Leciejewicz, 2010
$\text{PbI}_2(\text{C}_{18}\text{H}_{14}\text{N}_4\text{O})_2$		HBPY-8	Mahmoudi, <i>et al.</i> , 2019
$\text{Pb}_{2.5}(\text{H}_2\text{O})_2(\text{C}_8\text{H}_6\text{O}_4)_2$ Pb at (0, 0, 0)		HBPY-8	Ghosh & Sanguramath, 2008
$\text{Pb}(\text{C}_4\text{H}_2\text{O}_4\text{S})$ Pb on two-fold rotation axis		OCT-8	He <i>et al.</i> , 2018

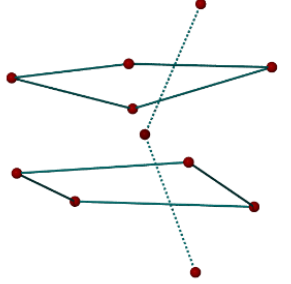
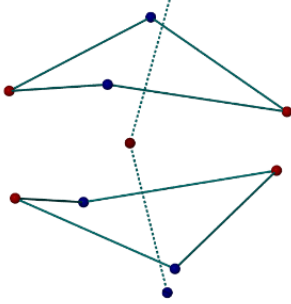
Table S6. Lead(II) coordination assigned as 9- and 10-vertex polyhedral (stereochemically-inert lone pair)

Pb(II) compound		IUPAC	Ref.
$\text{Pb}_2(\text{NO}_3)_2(\text{C}_{12}\text{H}_{10}\text{N}_4\text{O})_2\text{ClO}_4$		TRPS-9	Mahmoudi <i>et al.</i> , 2019
$\text{Pb}_2(\text{H}_2\text{O})(\text{C}_{26}\text{H}_{20}\text{N}_6\text{O}_2)_4\text{ClO}_4$		TRPS-9	
$[\text{Pb}_2(\text{C}_{21}\text{H}_{23}\text{N}_7\text{O}_2)_2(\text{CF}_3\text{O}_3\text{S})_3] \cdot (\text{CF}_3\text{O}_3\text{S}) \cdot \text{C}_2\text{H}_3\text{N}$ Pb at (0.3822, 0.6267, 0.8662)		TRPS-9	Lobo <i>et al.</i> , 2019
$\text{Pb}(\text{H}_2\text{O})(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_7\text{H}_4\text{O}_6\text{S})$		SAPRS-9	Imran <i>et al.</i> , 2015
$\text{Pb}_5(\text{C}_8\text{H}_3\text{F}_2\text{O}_4)_4(\text{C}_7\text{H}_4\text{FO}_2)_2$ Pb at (1.0696, 0.8235, -0.0581)		SAPRS-9	Song <i>et al.</i> , 2014
$\text{Pb}_2(\text{H}_2\text{O})(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_8\text{H}_3\text{F}_2\text{O}_4) \cdot 3\text{H}_2\text{O}$ Pb at (0.4827, 0.5741, 0.1937)		SAPRS-9	

			
$\text{Pb}(\text{H}_2\text{O})(\text{NO}_3)_2(\text{C}_{10}\text{H}_9\text{ClON}_6\text{O}_2)\cdot\text{H}_2\text{O}$		SAPRS-9	Deng <i>et al.</i> , 2013
$\text{Pb}_2\text{Cd}(\text{NO}_3)_6(\text{C}_{10}\text{H}_8\text{N}_2)_2$ Pb at (0.5362, 0.3732, 0.4164)		SAPRS-9	Najafi <i>et al.</i> , 2018
$\text{Pb}_2\text{Cd}(\text{NO}_3)_6(\text{C}_{10}\text{H}_8\text{N}_2)_2$ Pb at (0.8543, 0.1384, 0.0811)		SAPRS-9	
$\text{Pb}_4(\text{H}_2\text{O})(\text{C}_5\text{H}_4\text{NO}_2)_8$ Pb at (0.3051, 0.6932, 0.6134)		SAPRS-9	Kowalik <i>et al.</i> , 2017
$\text{Pb}(\text{H}_2\text{O})(\text{NO}_3)_2(\text{C}_{12}\text{H}_8\text{N}_2)$		SAPRS-9	Zhao <i>et al.</i> , 2007

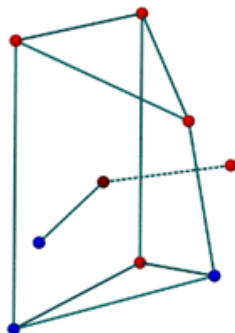
$\text{Pb}_2(\text{C}_9\text{H}_3\text{O}_6)(\text{C}_9\text{H}_4\text{O}_6) \cdot 2\text{H}_2\text{O}$ Pb at (0.0864, 0.0872, 0.6115)		SAPRS-9	Hayati <i>et al.</i> , 2018
$\text{Pb}(\text{NO}_3)_2(\text{H}_2\text{O})_2(\text{C}_{19}\text{H}_{13}\text{N}_5) \cdot \text{C}_{19}\text{H}_{13}\text{N}_5 \cdot \text{H}_2\text{O}$		SAPRS-9	Morsali & Mahjoub, 2004
$[\text{Pb}(\text{H}_2\text{O})(\text{BF}_4)(\text{C}_{12}\text{H}_{24}\text{O}_6)]_2 \cdot 2\text{BF}_4$		SAPRS-9	Farin <i>et al.</i> , 2013
$[\text{Pb}_2(\text{C}_{37}\text{H}_{51}\text{N}_{15}\text{O}_4)(\text{CF}_3\text{O}_3\text{S})_4] \cdot \text{CF}_3\text{O}_3\text{S} \cdot \text{C}_4\text{H}_{10}\text{O} \cdot 2\text{H}_2\text{O}$ O Pb at (0.4049, -0.1301, 0.1738)		SAPRS-9	Lobo <i>et al.</i> , 2019
$\text{Pb}(\text{NO}_3)_2(\text{C}_{10}\text{H}_{20}\text{O}_5)$		CUS-9	Farin <i>et al.</i> , 2013
$\text{Pb}(\text{H}_2\text{O})(\text{C}_{16}\text{H}_{10}\text{O}_7) \cdot \text{H}_2\text{O}$		CUS-9	Yang <i>et al.</i> , 2019

$\text{Pb}(\text{C}_{21}\text{H}_{23}\text{NO}_6)_2 \cdot 2\text{ClO}_4 \cdot \text{C}_2\text{H}_3\text{N} \cdot \text{H}_2\text{O}$		CUS-9	Németh <i>et al.</i> , 2015
$\text{Pb}(\text{NO}_3)_2(\text{C}_{12}\text{H}_{24}\text{O}_4\text{Se}_2)$ $\text{Pb}(\text{H}_2\text{O})_{1.6}(\text{C}_{12}\text{H}_{24}\text{O}_6)] \cdot 2\text{PF}_6 \cdot \text{H}_2\text{O}$		CUS-10 CUS-10	Farin <i>et al.</i> , 2013
$[\text{Pb}(\text{C}_{10}\text{H}_{20}\text{O}_5)_2[\text{SbCl}_6]][\text{SbCl}_3(\text{C}_{10}\text{H}_{20}\text{O}_5)]$		CUS-10	von Armin <i>et al.</i> , 1993
$\text{Pb}(\text{H}_2\text{O})_2(\text{BF}_4)(\text{C}_{12}\text{H}_{24}\text{O}_4\text{S}_2)\text{BF}_4$ $\text{Pb}(\text{C}_{10}\text{H}_{20}\text{O}_3\text{S}_2)_2 \cdot 2\text{BF}_4 \cdot \text{CH}_2\text{Cl}_2 \cdot 1.22$		SAPRS-10 SAPRS-10	Farin <i>et al.</i> , 2013

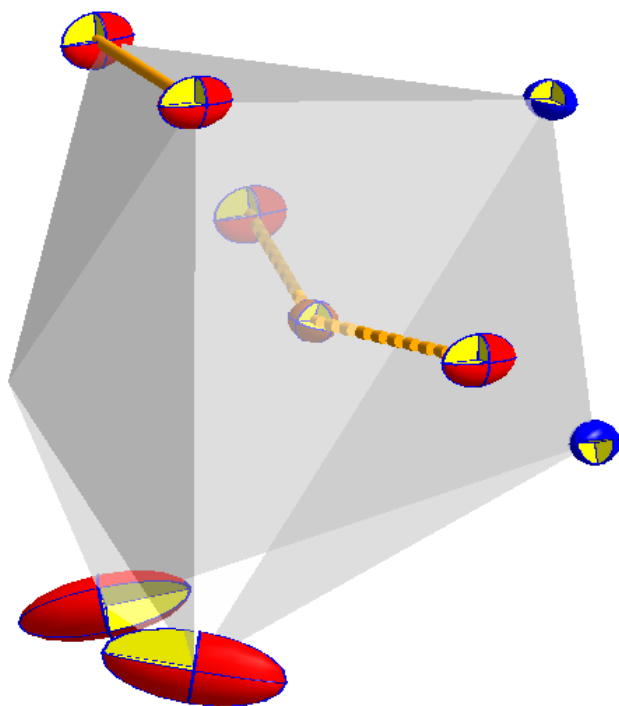
$[(C_4H_9)_4N]_3[Pb(NO_3)_5]$		SAPRS-10	Asanbaeva <i>et al.</i> , 2021
$[Pb(ClO_4)(NO_3)(C_{24}H_{16}N_6)]$		SAPRS-10	Morsali & Ramazani, 2005

Appendix 1. Divining the position of the lone-pair electron occupying a capping site of a Ψ -tricapped trigonal prism

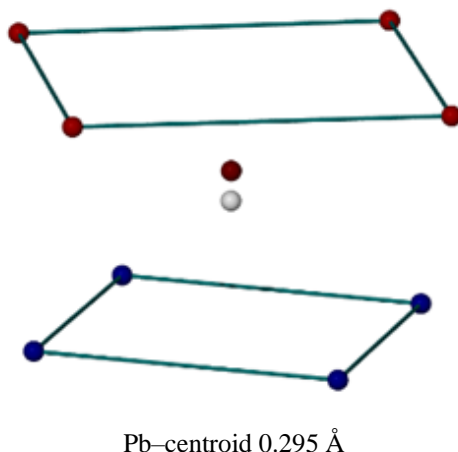
Example $\text{Pb}(\text{ClO}_4)(\text{C}_{12}\text{H}_8\text{N}_2)(\text{C}_2\text{H}_3\text{O}_2)$ (Hall *et al.*, 2000) from Table S4



The lone pair is presumed to be located over a near flat O_3N rectangle. The centroid of the O_6N_2 polyhedron is the site of Pb itself, and is 1.364 Å from the O_3N rectangle. The lone pair is then estimated at 1.236 Å above this rectangle so that the $\text{Pb}-\text{E}_{\text{lone pair}}$ distance would be 2.60 Å; its position is (-0.2679, 0.2439, -0.1372). The centroid of the nonahedron is also coincident with Pb.

**Appendix 2. Divining the position of the lone-pair electron occupying the capping site of a Ψ -monocapped square antiprism**

Example $\text{Pb}(\text{NO}_3)_2(\text{C}_{16}\text{H}_{32}\text{N}_8\text{O}_4) \cdot 3.5\text{H}_2\text{O}$ (Cuenot *et al.*, 2008) from Table S4



The coordinates of the centroid of the 4 oxygen—4 nitrogen polyhedron as calculated by using the *XP* graphics software of Bruker's APEX package is (0.1797, 0.0993, -0.1947). The mid-point of the larger square that made by the 4 oxygen atoms is at (0.1839, 0.0518, -0.1953). The distance from Pb to this mid-point square is 0.964 Å. A carbon atom is set at this site.

Assume the lone pair to be 2.60 Å from Pb, and on the opposite of this O_4 square.

In a zero-cycle SHELXL refinement run, the centroid (atom) is connected to Pb, so that an **AFIX 163 1.635** command written in the **ins** input after this arbitrary C atom generates a hydrogen atom at a distance of 2.60 Å from Pb. The lone pair is then assumed as the atom generated this way at (0.1908, -0.0100, -0.1954).

With 8 real atoms and one imaginary lone pair, the new centroid is coincident with Pb; other Pb— $E_{\text{lone pair}}$ distances lead to some divergence of the centroid from Pb.

