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Supporting information for article:

Structural and theoretical analysis of 2-chloro-4-nitroaniline and 2-methyl-6-nitroaniline salts

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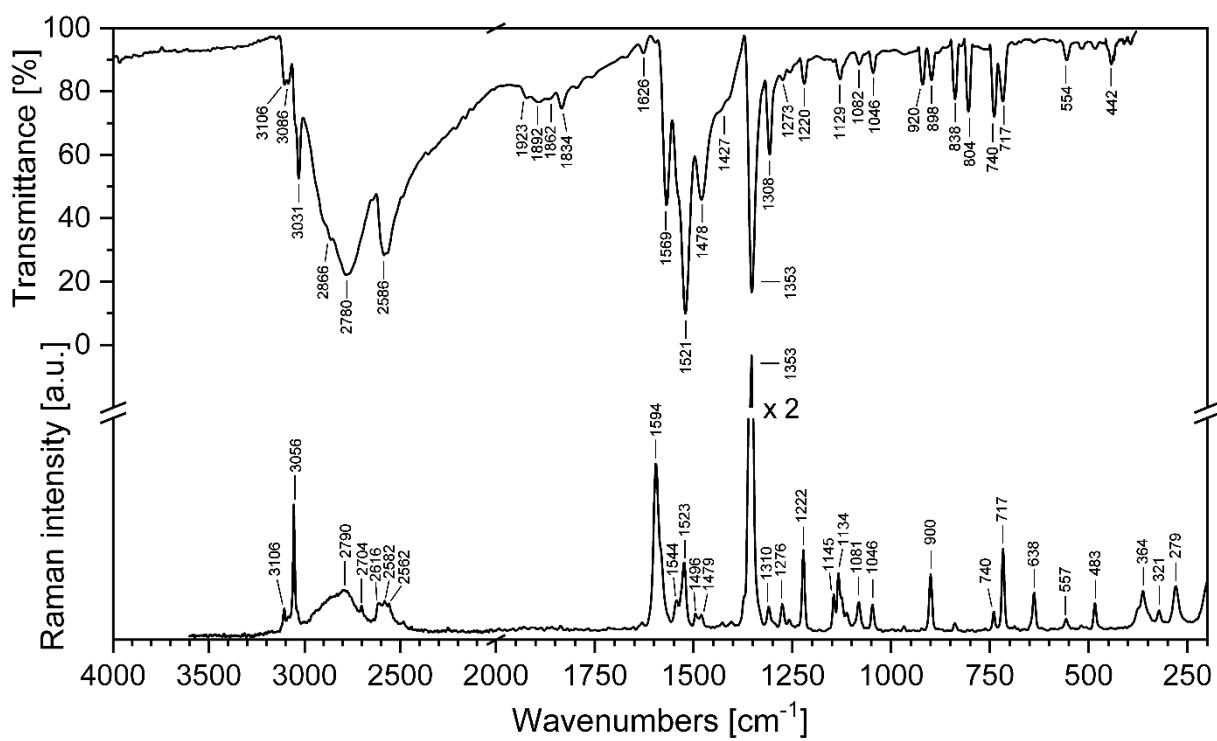


Figure S1. Room temperature FT-IR and FT-Raman spectra of (H₂Cl₄na)Br (**1**).

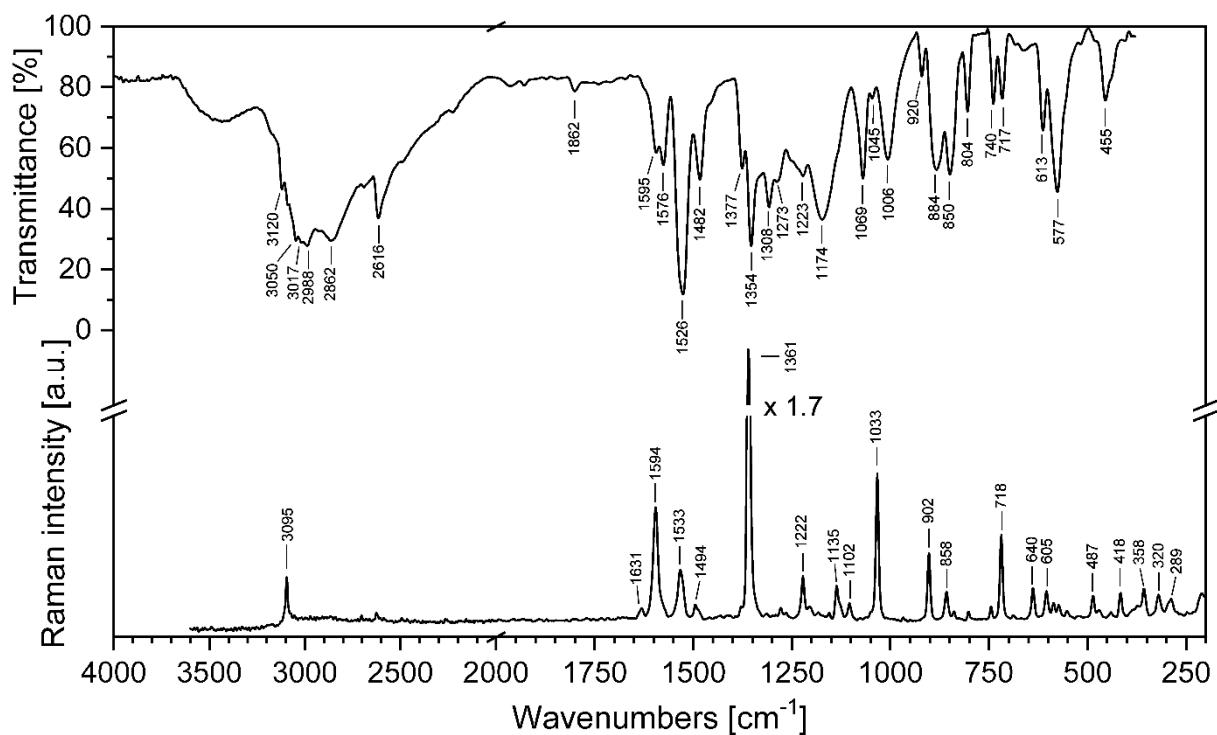


Figure S2. Room temperature FT-IR and FT-Raman spectra of (H₂Cl₄na)HSO₄ (**2**).

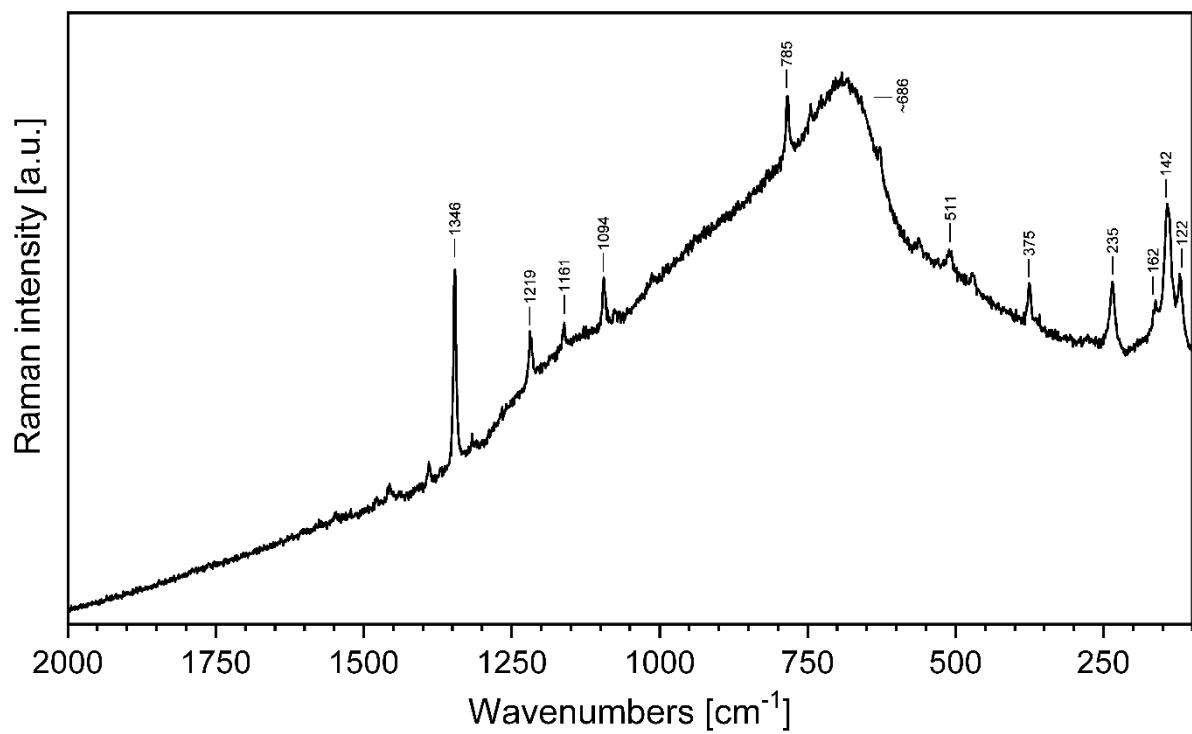


Figure S3. Room temperature FT-Raman spectra of (H₂m₆na)Br (**3**).

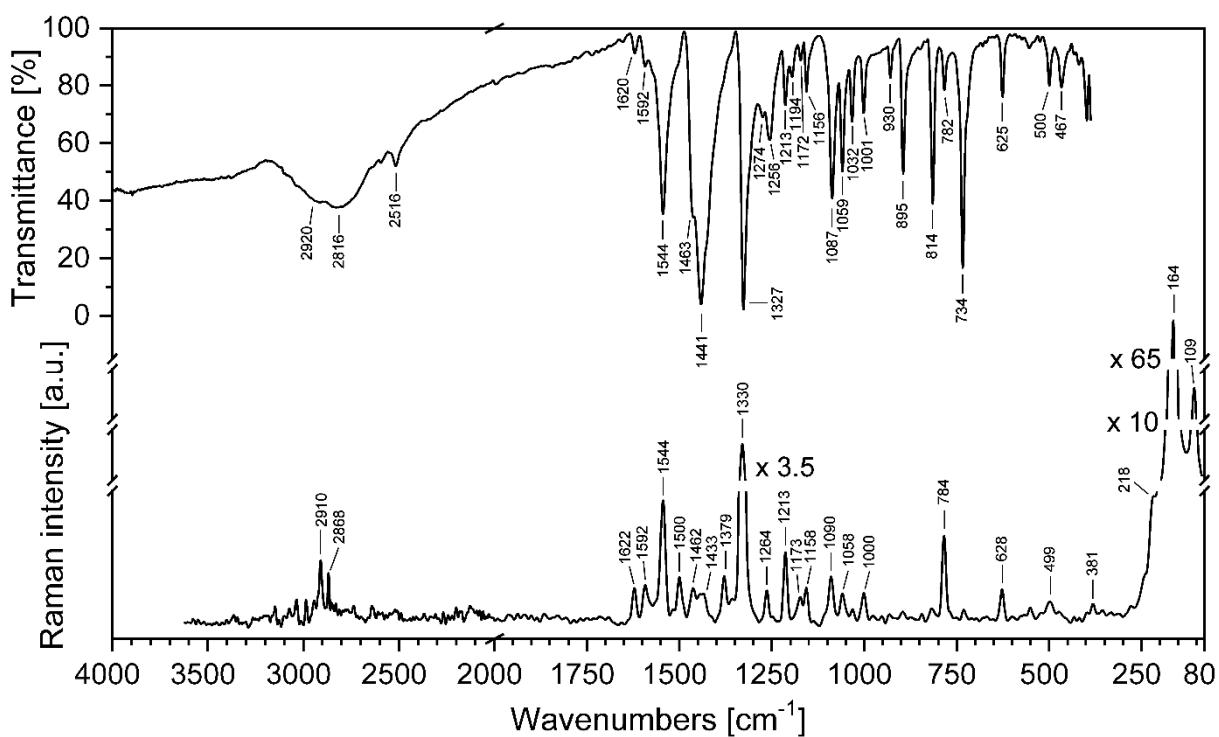


Figure S4. Room temperature FT-IR and FT-Raman spectra of $(\text{H}_2\text{m}6\text{na})\text{I}_3$ (4).

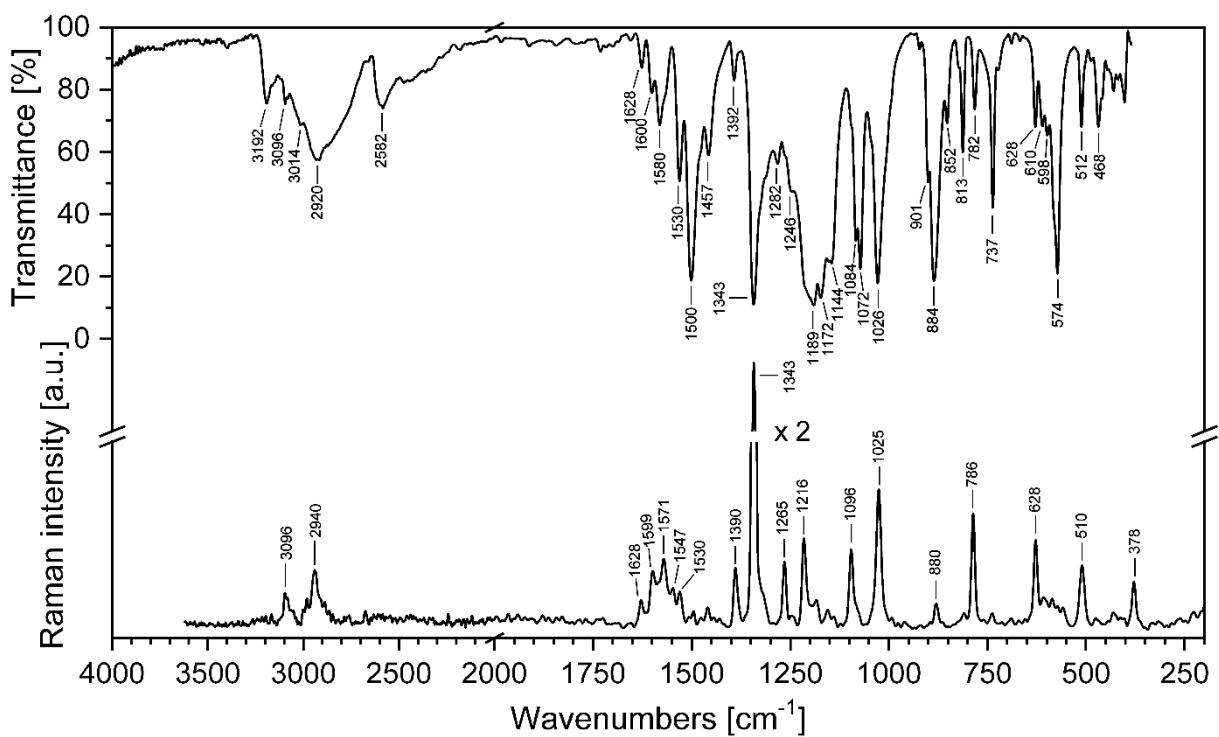


Figure S5. Room temperature FT-IR and FT-Raman spectra of $(\text{H}_2\text{m}6\text{na})\text{HSO}_4$ (5).

Table S1. XRD experimental details for studied compounds ($\text{H}_2\text{Cl4na}\text{Br}$ (**1**), ($\text{H}_2\text{Cl4na}\text{HSO}_4$ (**2**), ($\text{H}_2\text{m6na}\text{Br}$ (**3**), ($\text{H}_2\text{m6na}\text{I}_3$ (**4**) and ($\text{H}_2\text{m6na}\text{HSO}_4$ (**5**).

	1	2	3	4	5
Chemical formula	$\text{C}_6\text{H}_6\text{ClN}_2\text{O}_2 \cdot \text{Br}$	$\text{C}_6\text{H}_6\text{ClN}_2\text{O}_2 \cdot \text{HO}_4\text{S}$	$\text{C}_7\text{H}_9\text{N}_2\text{O}_2 \cdot \text{Br}$	$\text{C}_7\text{H}_9\text{N}_2\text{O}_2 \cdot \text{I}_3$	$\text{C}_7\text{H}_9\text{N}_2\text{O}_2 \cdot \text{HO}_4\text{S}$
M_f	253.49	270.65	233.07	533.86	250.23
Crystal system, space group	Orthorhombic, $P2_12_12_1$	Monoclinic, $P2_1/c$	Triclinic, $P\bar{1}$	Monoclinic, $P2_1/n$	Monoclinic, $P2_1/n$
a, b, c (Å)	5.72152 (9), 7.90017 (17), 20.2729 (4)	16.1771 (6), 4.95335 (16), 12.8431 (5)	6.8920 (2), 7.7281 (3), 8.9210 (2)	7.0840 (2), 11.5247 (4), 16.3921 (6)	7.0966 (5), 8.3785 (8), 17.5115 (18)
α, β, γ (°)	90, 90, 90	90, 101.584 (4), 90	76.646 (3), 73.897 (2), 76.045 (2)	90, 98.738 (3), 90	90, 100.185 (9), 90
V (Å ³)	916.35 (3)	1008.17 (7)	436.12 (2)	1322.73 (8)	1024.81 (16)
Z	4	4	2	4	4
$F(000)$	496	552	232	960	520
D_x (Mg m ⁻³)	1.837	1.783	1.775	2.681	1.622
No. of reflections for cell measurement	7548	1898	6602	6876	5045
θ range (°) for cell measurement	2.8–27.4	3.2–31.7	2.8–29.2	3.0–26.6	3.0–30.2
μ (mm ⁻¹)	4.74	0.60	4.67	7.07	0.33
Crystal shape	Block	Plate	Block	Block	Plate
Colour	Colourless	Colourless	Colourless	Black	Colourless
Crystal size (mm)	0.36 × 0.26 × 0.22	0.56 × 0.46 × 0.06	0.46 × 0.33 × 0.18	0.21 × 0.11 × 0.09	0.42 × 0.32 × 0.09
Absorption correction	Analytical	Multi-scan	Gaussian	Gaussian	Multi-scan
T_{\min}, T_{\max}	0.300, 0.446	0.967, 1.000	0.947, 0.973	0.966, 0.988	0.939, 1.000

No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	18773, 2669, 2340	6028, 2923, 2351	15285, 2299, 1986	5822, 5822, 3181	4495, 4495, 3463
R_{int}	0.035	0.022	0.027	twin	twin
θ values (°)	$\theta_{\text{max}} = 30.0, \theta_{\text{min}} = 2.8$	$\theta_{\text{max}} = 30.0, \theta_{\text{min}} = 2.6$	$\theta_{\text{max}} = 29.0, \theta_{\text{min}} = 3.1$	$\theta_{\text{max}} = 29.7, \theta_{\text{min}} = 3.1$	$\theta_{\text{max}} = 27.5, \theta_{\text{min}} = 2.7$
$(\sin \theta / \lambda)_{\text{max}} (\text{\AA}^{-1})$	0.703	0.703	0.682	0.697	0.650
Range of h, k, l	$h = -8 \rightarrow 8, k = -11 \rightarrow 11, l = -28 \rightarrow 28$	$h = -22 \rightarrow 22, k = 6 \rightarrow 6, l = -9 \rightarrow 18$	$h = -9 \rightarrow 9, k = -10 \rightarrow 10, l = -12 \rightarrow 12$	$h = -9 \rightarrow 9, k = -15 \rightarrow 15, l = -22 \rightarrow 22$	$h = -9 \rightarrow 9, k = -10 \rightarrow 10, l = -22 \rightarrow 22$
$R[F^2 > 2\sigma(F^2)],$ $wR(F^2), S$	0.028, 0.062, 1.13	0.036, 0.099, 1.07	0.026, 0.061, 1.11	0.031, 0.059, 0.80	0.044, 0.136, 1.07
No. of reflections	2669	2923	2299	5822	4495
No. of parameters	113	161	122	130	149
$(\Delta/\sigma)_{\text{max}}$	0.002	0.001	0.001	0.001	0.001
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}} (\text{e \AA}^{-3})$	0.33, -0.70	0.52, -0.55	0.33, -0.39	0.70, -0.78	0.66, -0.42
Absolute structure parameter	-0.008 (4)	—	—	—	—

Table S2. Selected hydrogen-bond parameters for studied compounds **1–5**.

$D-H \cdots A$	$D-H (\text{\AA})$	$H \cdots A (\text{\AA})$	$D \cdots A (\text{\AA})$	$D-H \cdots A (^{\circ})$
(H₂Cl₄na)Br (1)				
N1—H1A···Br1	0.89	2.47	3.329 (2)	162.7
N1—H1B···Br1 ⁱ	0.89	2.45	3.295 (2)	159.3
N1—H1C···Br1 ⁱⁱ	0.89	2.41	3.266 (3)	162.5
C3—H3···Br1 ⁱⁱⁱ	0.93	2.79	3.712 (3)	172.0
C5—H5···O2 ^{iv}	0.93	2.56	3.199 (4)	126.2
(H₂Cl₄na)HSO₄ (2)				
N1—H1A···O3	0.93 (2)	1.96 (2)	2.854 (2)	161 (2)
N1—H1B···O6 ^v	0.90 (3)	1.92 (3)	2.813 (2)	171 (2)
N1—H1C···O3 ^{vi}	0.87 (3)	2.14 (3)	2.952 (2)	156 (2)
N1—H1C···O5 ^{vii}	0.87 (3)	2.39 (3)	2.911 (2)	119 (2)
C3—H3···O1 ^{viii}	0.93	2.41	3.326 (2)	167.6
C6—H6···O3	0.93	2.47	3.184 (2)	134.0
O4—H4···O5 ^{ix}	0.79 (3)	1.85 (3)	2.6350 (18)	167 (3)
(H₂m₆na)Br (3)				
N1—H1A···O1	0.90 (3)	1.98 (3)	2.693 (2)	135 (2)
N1—H1B···Br1 ^x	0.86 (3)	2.47 (3)	3.2717 (17)	155 (2)
N1—H1C···Br1	0.92 (3)	2.34 (3)	3.2650 (18)	174 (2)
(H₂m₆na)I₃ (4)				
N1—H1A···O1	0.89	1.86	2.591 (6)	138.1
N1—H1B···I1	0.89	2.82	3.598 (4)	146.4
N1—H1C···I1 ^{xi}	0.89	2.82	3.566 (4)	143.0
C5—H5···O1 ^{xii}	0.93	2.54	3.304 (6)	140.1
(H₂m₆na)HSO₄ (5)				
O6—H6···O5 ^{xiii}	0.82	1.75	2.568 (3)	172.9
N1—H1A···O5 ^{xiv}	0.89	2.39	2.993 (3)	125.6
N1—H1A···O1	0.89	2.01	2.650 (3)	127.5
N1—H1B···O4 ^{xi}	0.89	1.92	2.802 (3)	168.5
N1—H1C···O3	0.89	1.86	2.731 (3)	167.3

Symmetry codes: (i) $x-1, y, z$; (ii) $x-1/2, -y+3/2, -z+1$; (iii) $x-1, y-1, z$; (iv) $-x+1, y+1/2, -z+3/2$; (v) $x, -y+1/2, z-1/2$; (vi) $x, y+1, z$; (vii) $-x, -y+1, -z$; (viii) $-x+1, -y+2, -z$; (ix) $-x, y-1/2, -z+1/2$; (x) $-x+1, -y, -z+2$; (xi) $x+1, y, z$; (xii) $-x+3/2, y+1/2, -z+1/2$; (xiii) $-x+1/2, y+1/2, -z+3/2$; (xiv) $-x+3/2, y+1/2, -z+3/2$.