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**Supporting information for article:**

**ACr(C<sub>2</sub>O<sub>4</sub>)<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub> (A = Li or Na): two new coordination polymers of low dimensionality with different hydrogen-bond networks**

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**Table S1**Selected geometric parameters (Å) for (**I**)

Cr1—O2	1.9792 (19)	Li1—O1	2.133 (2)
Cr1—O2 <sup>i</sup>	1.9792 (19)	Li1—O1 <sup>iv</sup>	2.133 (2)
Cr1—O2 <sup>ii</sup>	1.9792 (19)	Li1—O1 <sup>v</sup>	2.133 (2)
Cr1—O2 <sup>iii</sup>	1.9792 (19)	Li1—O1 <sup>i</sup>	2.133 (2)
Cr1—OW2	2.000 (2)	C1—O1	1.232 (2)
Cr1—OW2 <sup>iii</sup>	2.000 (2)	C1—O2	1.293 (2)
Li1—OW1B <sup>iv</sup>	2.066 (18)	C1—C1 <sup>i</sup>	1.569 (3)
Li1—OW1B	2.066 (18)	OW2—H2	0.835 (8)
Li1—OW1A	2.087 (9)	OW1A—H1	0.856 (9)
Li1—OW1A <sup>iv</sup>	2.087 (9)	OW1B—H1	0.855 (10)

Symmetry code(s): (i)  $x, -y, z$ ; (ii)  $-x+1, y, -z+1$ ; (iii)  $-x+1, -y, -z+1$ ; (iv)  $-x, -y, -z$ ; (v)  $-x, y, -z$ .**Table S2**Selected geometric parameters (°) for (**I**)

O2—Cr1—O2 <sup>i</sup>	83.43 (11)	O1—Li1—O1 <sup>iv</sup>	180.0
O2—Cr1—O2 <sup>ii</sup>	96.57 (11)	OW1B <sup>iv</sup> —Li1—O1 <sup>v</sup>	96.5 (11)
O2 <sup>i</sup> —Cr1—O2 <sup>ii</sup>	180.00 (9)	OW1B—Li1—O1 <sup>v</sup>	83.5 (11)
O2—Cr1—O2 <sup>iii</sup>	180.0	OW1A—Li1—O1 <sup>v</sup>	94.7 (3)
O2 <sup>i</sup> —Cr1—O2 <sup>iii</sup>	96.57 (11)	OW1A <sup>iv</sup> —Li1—O1 <sup>v</sup>	85.3 (3)
O2 <sup>ii</sup> —Cr1—O2 <sup>iii</sup>	83.43 (11)	O1—Li1—O1 <sup>v</sup>	98.38 (11)
O2—Cr1—OW2	89.44 (8)	O1 <sup>iv</sup> —Li1—O1 <sup>v</sup>	81.62 (11)
O2 <sup>i</sup> —Cr1—OW2	89.44 (8)	OW1B <sup>iv</sup> —Li1—O1 <sup>i</sup>	83.5 (11)
O2 <sup>ii</sup> —Cr1—OW2	90.56 (8)	OW1B—Li1—O1 <sup>i</sup>	96.5 (11)
O2 <sup>iii</sup> —Cr1—OW2	90.56 (8)	OW1A—Li1—O1 <sup>i</sup>	85.3 (3)
O2—Cr1—OW2 <sup>iii</sup>	90.56 (8)	OW1A <sup>iv</sup> —Li1—O1 <sup>i</sup>	94.7 (3)

O2 <sup>i</sup> —Cr1—OW2 <sup>iii</sup>	90.56 (8)	O1—Li1—O1 <sup>i</sup>	81.62 (11)
O2 <sup>ii</sup> —Cr1—OW2 <sup>iii</sup>	89.44 (8)	O1 <sup>iv</sup> —Li1—O1 <sup>i</sup>	98.38 (11)
O2 <sup>iii</sup> —Cr1—OW2 <sup>iii</sup>	89.44 (8)	O1 <sup>v</sup> —Li1—O1 <sup>i</sup>	180.00 (10)
OW2—Cr1—OW2 <sup>iii</sup>	180.0	O1—C1—O2	126.01 (16)
OW1B <sup>iv</sup> —Li1—OW1B	180.0	O1—C1—C1 <sup>i</sup>	119.66 (9)
OW1A—Li1—OW1A <sup>iv</sup>	180.0	O2—C1—C1 <sup>i</sup>	114.33 (9)
OW1B <sup>iv</sup> —Li1—O1	83.5 (11)	C1—O1—Li1	109.33 (13)
OW1B—Li1—O1	96.5 (11)	C1—O2—Cr1	113.86 (12)
OW1A—Li1—O1	85.3 (3)	Cr1—OW2—H2	117.9 (13)
OW1A <sup>iv</sup> —Li1—O1	94.7 (3)	Li1—OW1A—H1	116.4 (14)
OW1B <sup>iv</sup> —Li1—O1 <sup>iv</sup>	96.5 (11)	Li1—OW1B—H1	118 (3)
OW1B—Li1—O1 <sup>iv</sup>	83.5 (11)	H1—OW1A—H1 <sup>i</sup>	106.8 (9)
OW1A—Li1—O1 <sup>iv</sup>	94.7 (3)	H1—OW1B—H1 <sup>i</sup>	106.9 (9)
OW1A <sup>iv</sup> —Li1—O1 <sup>iv</sup>	85.3 (3)	H2—OW2—H2 <sup>i</sup>	113.7 (8)
O2—C1—O1—Li1	176.45 (12)	O1—C1—O2—Cr1	-177.87 (13)
C1 <sup>i</sup> —C1—O1—Li1	-4.52 (11)	C1 <sup>i</sup> —C1—O2—Cr1	3.07 (10)

Symmetry code(s): (i)  $x, -y, z$ ; (ii)  $-x+1, y, -z+1$ ; (iii)  $-x+1, -y, -z+1$ ; (iv)  $-x, -y, -z$ ; (v)  $-x, y, -z$ .

### Table S3

Selected geometric parameters (Å) for (II)

Cr—O1	1.9581 (9)	Na—C1 <sup>ii</sup>	3.1007 (13)
Cr—O1 <sup>i</sup>	1.9581 (9)	Na—C2 <sup>ii</sup>	3.1187 (13)
Cr—O3	1.9751 (10)	Na—C2	3.1187 (13)
Cr—O3 <sup>i</sup>	1.9751 (10)	O1—C1	1.2744 (16)
Cr—OW2 <sup>i</sup>	1.9956 (10)	O2—C1	1.2286 (15)
Cr—OW2	1.9956 (10)	O3—C2	1.2832 (16)
Na—O2 <sup>ii</sup>	2.3525 (11)	O4—C2	1.2191 (17)
Na—O2	2.3525 (11)	C1—C2	1.5571 (19)
Na—O4	2.4263 (10)	OW1—H1	0.840 (9)
Na—O4 <sup>ii</sup>	2.4263 (10)	OW1—H2	0.848 (9)
Na—OW1 <sup>ii</sup>	2.4357 (11)	OW2—H3	0.848 (9)
Na—OW1	2.4357 (11)	OW2—H4	0.831 (9)

Na—C1 3.1007 (13)

Symmetry code(s): (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x+2, -y+2, -z$ .**Table S4**Selected geometric parameters (°) for (**II**)

O1—Cr—O1 <sup>i</sup>	180.0	OW1—Na—C1 <sup>ii</sup>	86.03 (4)
O1—Cr—O3	82.62 (4)	C1—Na—C1 <sup>ii</sup>	180.0
O1 <sup>i</sup> —Cr—O3	97.38 (4)	O2 <sup>ii</sup> —Na—C2 <sup>ii</sup>	49.78 (3)
O1—Cr—O3 <sup>i</sup>	97.38 (4)	O2—Na—C2 <sup>ii</sup>	130.22 (3)
O1 <sup>i</sup> —Cr—O3 <sup>i</sup>	82.62 (4)	O4—Na—C2 <sup>ii</sup>	158.98 (4)
O3—Cr—O3 <sup>i</sup>	180.0	O4 <sup>ii</sup> —Na—C2 <sup>ii</sup>	21.02 (4)
O1—Cr—OW2 <sup>i</sup>	87.71 (4)	OW1 <sup>ii</sup> —Na—C2 <sup>ii</sup>	96.39 (4)
O1 <sup>i</sup> —Cr—OW2 <sup>i</sup>	92.30 (4)	OW1—Na—C2 <sup>ii</sup>	83.61 (4)
O3—Cr—OW2 <sup>i</sup>	89.31 (5)	C1—Na—C2 <sup>ii</sup>	151.00 (4)
O3 <sup>i</sup> —Cr—OW2 <sup>i</sup>	90.69 (5)	C1 <sup>ii</sup> —Na—C2 <sup>ii</sup>	29.00 (4)
O1—Cr—OW2	92.30 (4)	O2 <sup>ii</sup> —Na—C2	130.22 (3)
O1 <sup>i</sup> —Cr—OW2	87.70 (4)	O2—Na—C2	49.78 (3)
O3—Cr—OW2	90.69 (5)	O4—Na—C2	21.02 (4)
O3 <sup>i</sup> —Cr—OW2	89.31 (5)	O4 <sup>ii</sup> —Na—C2	158.98 (4)
OW2 <sup>i</sup> —Cr—OW2	180.0	OW1 <sup>ii</sup> —Na—C2	83.61 (4)
O2 <sup>ii</sup> —Na—O2	180.0	OW1—Na—C2	96.39 (4)
O2 <sup>ii</sup> —Na—O4	109.21 (3)	C1—Na—C2	29.00 (4)
O2—Na—O4	70.80 (3)	C1 <sup>ii</sup> —Na—C2	151.00 (4)
O2 <sup>ii</sup> —Na—O4 <sup>ii</sup>	70.79 (3)	C2 <sup>ii</sup> —Na—C2	180.0
O2—Na—O4 <sup>ii</sup>	109.21 (3)	C1—O1—Cr	114.76 (8)
O4—Na—O4 <sup>ii</sup>	180.0	C1—O2—Na	116.40 (9)
O2 <sup>ii</sup> —Na—OW1 <sup>ii</sup>	91.17 (4)	C2—O3—Cr	114.55 (8)
O2—Na—OW1 <sup>ii</sup>	88.83 (4)	C2—O4—Na	113.44 (9)
O4—Na—OW1 <sup>ii</sup>	83.06 (4)	O2—C1—O1	126.48 (12)
O4 <sup>ii</sup> —Na—OW1 <sup>ii</sup>	96.94 (4)	O2—C1—C2	118.94 (12)
O2 <sup>ii</sup> —Na—OW1	88.83 (4)	O1—C1—C2	114.57 (10)
O2—Na—OW1	91.17 (4)	O2—C1—Na	42.81 (7)

O4—Na—OW1	96.94 (4)	O1—C1—Na	169.28 (9)
O4 <sup>ii</sup> —Na—OW1	83.06 (4)	C2—C1—Na	76.14 (7)
OW1 <sup>ii</sup> —Na—OW1	180.0	O4—C2—O3	126.21 (12)
O2 <sup>ii</sup> —Na—C1	159.21 (3)	O4—C2—C1	120.40 (12)
O2—Na—C1	20.79 (3)	O3—C2—C1	113.38 (11)
O4—Na—C1	50.01 (3)	O4—C2—Na	45.55 (7)
O4 <sup>ii</sup> —Na—C1	129.99 (3)	O3—C2—Na	171.75 (9)
OW1 <sup>ii</sup> —Na—C1	86.03 (4)	C1—C2—Na	74.86 (7)
OW1—Na—C1	93.97 (4)	Na—OW1—H1	104.8 (16)
O2 <sup>ii</sup> —Na—C1 <sup>ii</sup>	20.79 (3)	Na—OW1—H2	101.9 (15)
O2—Na—C1 <sup>ii</sup>	159.21 (3)	H1—OW1—H2	108.7 (17)
O4—Na—C1 <sup>ii</sup>	129.99 (3)	Cr—OW2—H3	115.4 (14)
O4 <sup>ii</sup> —Na—C1 <sup>ii</sup>	50.01 (3)	Cr—OW2—H4	115.1 (14)
OW1 <sup>ii</sup> —Na—C1 <sup>ii</sup>	93.97 (4)	H3—OW2—H4	110.9 (17)
Na—O2—C1—O1	-179.22 (10)	O2—C1—C2—O4	-0.4 (2)
Na—O2—C1—C2	1.27 (15)	O1—C1—C2—O4	-179.91 (12)
Cr—O1—C1—O2	178.24 (11)	Na—C1—C2—O4	0.54 (12)
Cr—O1—C1—C2	-2.24 (14)	O2—C1—C2—O3	179.41 (11)
Cr—O1—C1—Na	175.4 (4)	O1—C1—C2—O3	-0.16 (17)
Na—O4—C2—O3	179.55 (11)	Na—C1—C2—O3	-179.70 (11)
Na—O4—C2—C1	-0.73 (16)	O2—C1—C2—Na	-0.89 (11)
Cr—O3—C2—O4	-177.81 (12)	O1—C1—C2—Na	179.55 (11)
Cr—O3—C2—C1	2.45 (14)		

Symmetry code(s): (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x+2, -y+2, -z$ .

**Table S5**

Selected hydrogen-bond parameters (Å, °) for (**I**)

$D-H\cdots A$	$D-H$ (Å)	$H\cdots A$ (Å)	$D\cdots A$ (Å)	$D-H\cdots A$ (°)
OW2—H2 $\cdots$ O1 <sup>i</sup>	0.835 (8)	1.899 (8)	2.716 (3)	165.4 (16)
OW1A—H1 $\cdots$ O2 <sup>ii</sup>	0.856 (9)	2.017 (8)	2.830 (4)	158.5 (19)
OW1B—H1 $\cdots$ O2 <sup>ii</sup>	0.855 (10)	2.017 (8)	2.828 (5)	158 (2)

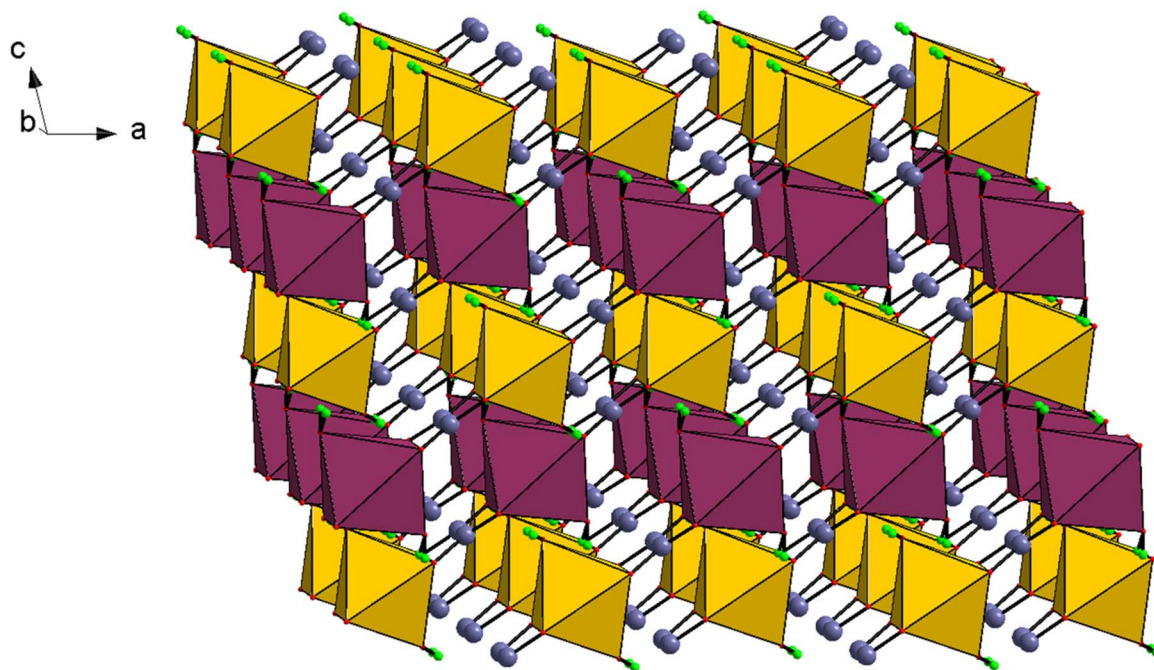
Symmetry code(s): (i)  $x+1/2, y-1/2, z$ ; (ii)  $-x+1/2, -y+1/2, -z$ .

**Table S6**

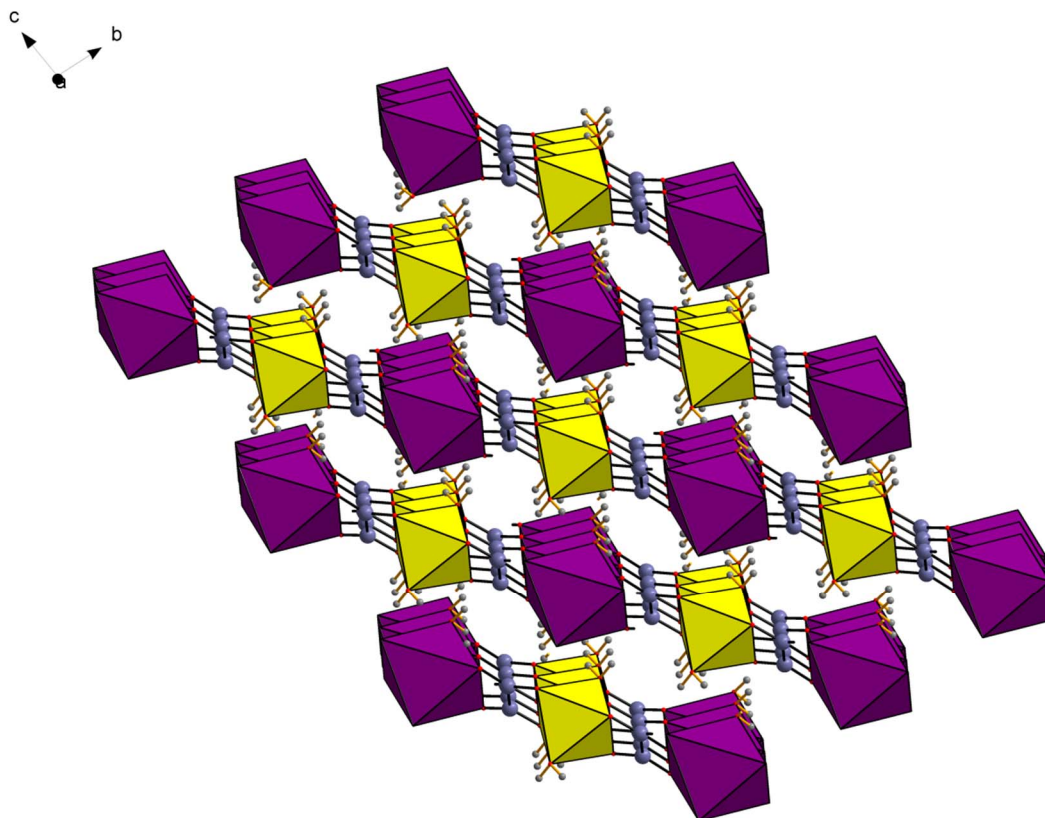
Selected hydrogen-bond parameters(Å, °) for **(II)**

$D-H\cdots A$	$D-H$ (Å)	$H\cdots A$ (Å)	$D\cdots A$ (Å)	$D-H\cdots A$ (°)
OW1—H1 $\cdots$ O4 <sup>i</sup>	0.840 (9)	1.985 (12)	2.7731 (16)	156 (2)
OW1—H2 $\cdots$ O3 <sup>ii</sup>	0.848 (9)	1.973 (12)	2.7612 (14)	154 (2)
OW2—H3 $\cdots$ O2 <sup>iii</sup>	0.848 (9)	1.843 (10)	2.6849 (14)	172 (2)
OW2—H4 $\cdots$ OW1 <sup>iv</sup>	0.831 (9)	1.914 (11)	2.7336 (16)	168 (2)

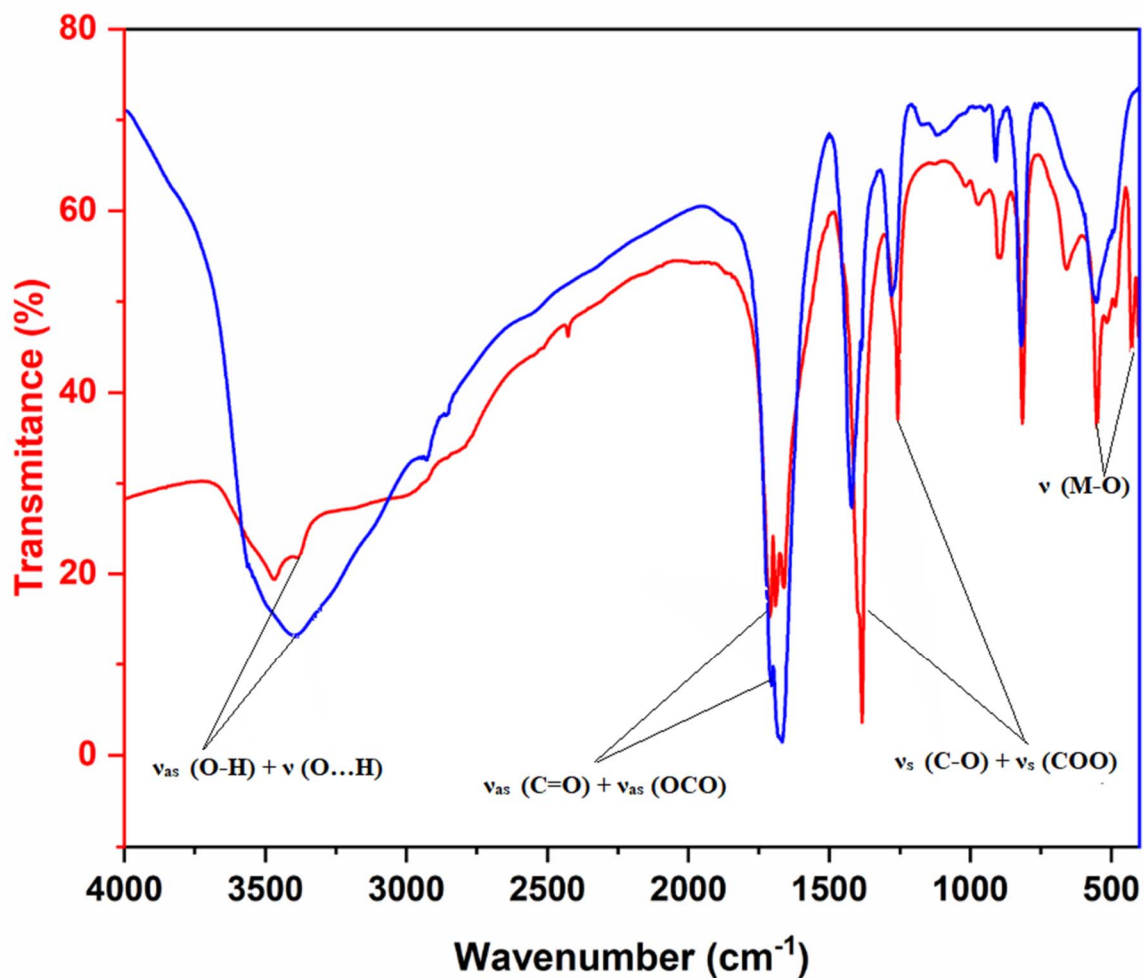
Symmetry code(s): (i)  $x+1, y, z$ ; (ii)  $-x+1, -y+1, -z$ ; (iii)  $x-1, y-1, z$ ; (iv)  $-x+2, -y+1, -z$ .



**Figure S1** Perspective view of the structure of  $\text{LiCr}(\text{C}_2\text{O}_4)_2(\text{H}_2\text{O})_4$ , **(I)** showing infinite polyhedra chains running along  $[101]$  direction (Li violet, Cr yellow)

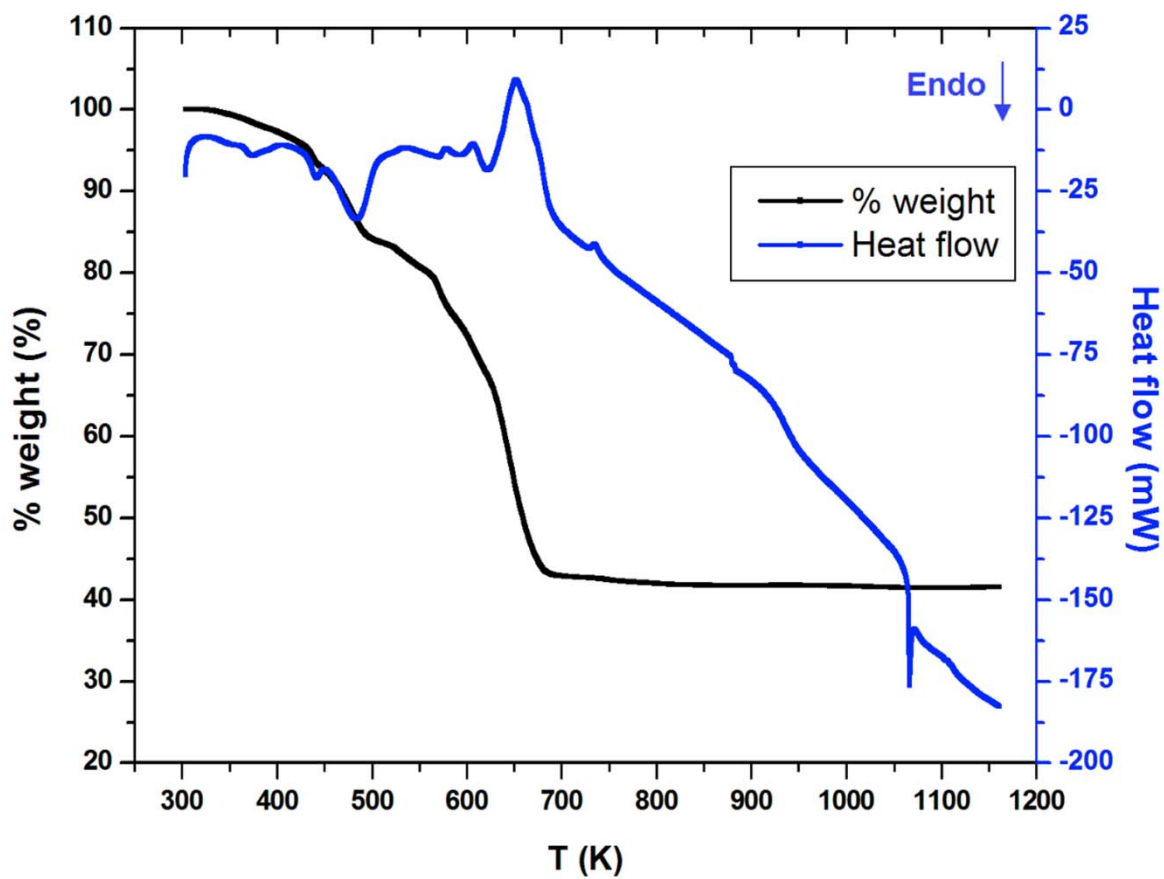


**Figure S2** Perspective view of the structure of  $\text{NaCr}(\text{C}_2\text{O}_4)_2(\text{H}_2\text{O})_4(\text{II})$  showing infinite polyhedra chains running along  $[101]$  direction (Na violet, Cr yellow)



**Figure S3** The FT-IR spectra of (I) (blue) and (II) (red), obtained in pure KBr. Absorption bands are indexed for comparison.





**Figure S4** Thermogravimetric (black) and differential thermal analysis (blue) curves for  $\text{NaCr}(\text{C}_2\text{O}_4)_2(\text{H}_2\text{O})_4$ , (**II**)