

Volume 74 (2018)

Supporting information for article:

Two new cobalt(II) rhodamine 6G hydrazone complexes: structure, fluorescence and magnetism

Juan Yuan, Xia Wang, Nan Zhang, Mei-Jiao Liu and Hui-Zhong Kou

D-HA	D-H	НА	DA	<(DHA)
complex (1)		11 11	DA	
N7-H7O14	0.86	2.33	2.951(5)	129.0
N8-H8O6	0.86	2.14	2.913(5)	150.0
N10-H10O5	0.86	2.08	2.877(5)	153.1
O5–H5O11 <sup>i</sup>	0.82	2.39	3.103(8)	146.3
O5-H5O12 <sup>i</sup>	0.82	2.50	3.277(9)	157.5
O6-H6AO7	0.82	2.30	2.918(7)	132.0
C6-H6O14 <sup>ii</sup>	0.93	2.54	3.373(5)	148.7
C4-H4O13 <sup>ii</sup>	0.93	2.49	3.323(6)	149.6
C19–H19 O10 <sup>iii</sup>	0.93	2.58	3.509(6)	178.6
C28–H28A O8 <sup>iii</sup>	0.97	2.54		
C2-H2O8 <sup>iv</sup>	0.93	2.41	3.107(6)	131.5
C44-H44O13 <sup>v</sup>	0.93	2.50	3.239(6)	137.0
C12-H12O3 <sup>vi</sup>	0.93	2.54	3.438(5)	162.1
C67–H67AN6 complex (2)	0.97	2.36	3.281(7)	159.0
N8-H8O6	0.88	2.10	2.900(4)	150.6
N7-H7F3	0.88	2.25	2.903(4)	130.4
N10-H10O5	0.88	2.07	2.882(4)	152.7
O5-H5F1A <sup>vii</sup>	0.84	1.99	2.762(10)	152.5
O5-H5F4 <sup>vii</sup>	0.84	2.22	2.897(11)	137.4
O6–H6AF6	0.84	2.06	2.809(4)	148.7
C2-H2F7 <sup>viii</sup>	0.95	2.43	3.142(4)	132.0
C4-H4F2 <sup>ix</sup>	0.95	2.40	3.275(6)	152.9
C6-H6F3 <sup>ix</sup>	0.95	2.50	3.323(4)	145.6
C12-H12O3 <sup>x</sup>	0.95	2.46	3.379(4)	161.5
C22-H22F8 <sup>xi</sup>	0.95	2.52	3.466(4)	179.0
C31–H31B F7 <sup>xi</sup>	0.99	2.43	3.302(4)	147.1
C44-H44F2 <sup>xii</sup>	0.95	2.46	3.177(6)	132.7
C67-H67BN6	0.99	2.34	3.277(6)	157.8

**Table S1**Hydrogen bonds in complexes (1) and (2) (with esds except fixed and riding H).

Symmetry codes: (i) x, y, z+1; (ii) -x, 1-y, 1-z; (iii) 1-x, -y, 1-z; (iv) x, 1+y, z; (v) 1+x, y, 1+z; (vi) -x, -y, 1-z;

(vii) x, y, z-1; (viii) x, y-1, z; (ix) 2-x, 1-y, 1-z; (x) 2-x, 2-y, 1-z; (xi) 1-x, 2-y, 1-z; (xii) -1+x, y, -1+z



**Figure S1** Hydrogen bonding interactions in complex (1).



Figure S2 3D supramolecular structure of complex (1).



Figure S3 The IR spectra of complexes (1) and (2).



**Figure S4** Fluorescence spectra for complexes (1), (2) and Ligand (L) in ethanol ( $\lambda_{ex} = 350$  nm, [1] = [2] = 10 µmol/L, [L] = 20 µmol/L). Inset: Enlarged view of the spectra for L.