



STRUCTURAL
CHEMISTRY

Volume 74 (2018)

Supporting information for article:

Topological identification of the first uninodal 8-connected Isz MOF built from 2,2'-difluorobiphenyl-4,4'-dicarboxylate pillars and cadmium(II)-triazolate layers

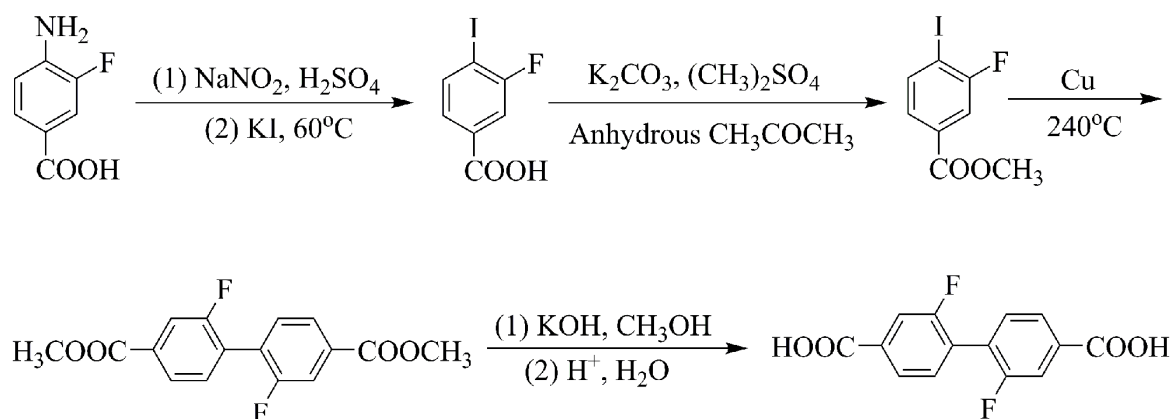
Yuchi Zhang, Yuanhua Wu, Xin He, Junhan Ma, Xuan Shen and Dunru Zhu

Topological identification of the first uninodal 8-connected Isz MOF built from 2,2'-difluoro-4,4'-biphenyldicarboxylate pillars and Cd(II)-triazolate layers

Yuchi Zhang^a, Yuanhua Wu^a, Xin He^a, Junhan Ma^a, Xuan Shen^a and Dunru Zhu^{a*}

^aCollege of Chemical Engineering, State Key Laboratory of Materials-Oriented Chemical Engineering, Nanjing Tech University, 5 XinMoFan Road, Nanjing, Jiangsu, 210009, People's Republic of China

Correspondence email: zhudr@njtech.edu.cn



Scheme S1 Synthetic route of 2,2'-difluoro-4,4'-biphenyldicarboxylic acid (H₂L).

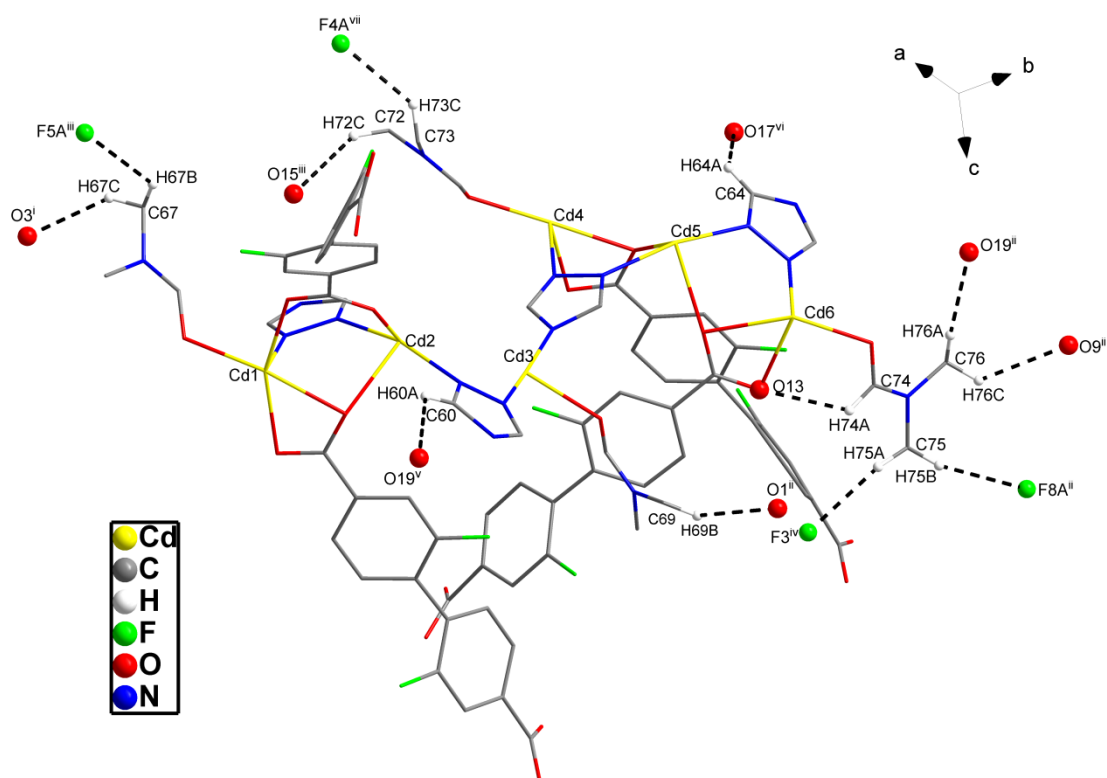


Figure S1 The network of (I) was stabilized by intermolecular C–H...F and C–H...O hydrogen bonds (black dotted lines).

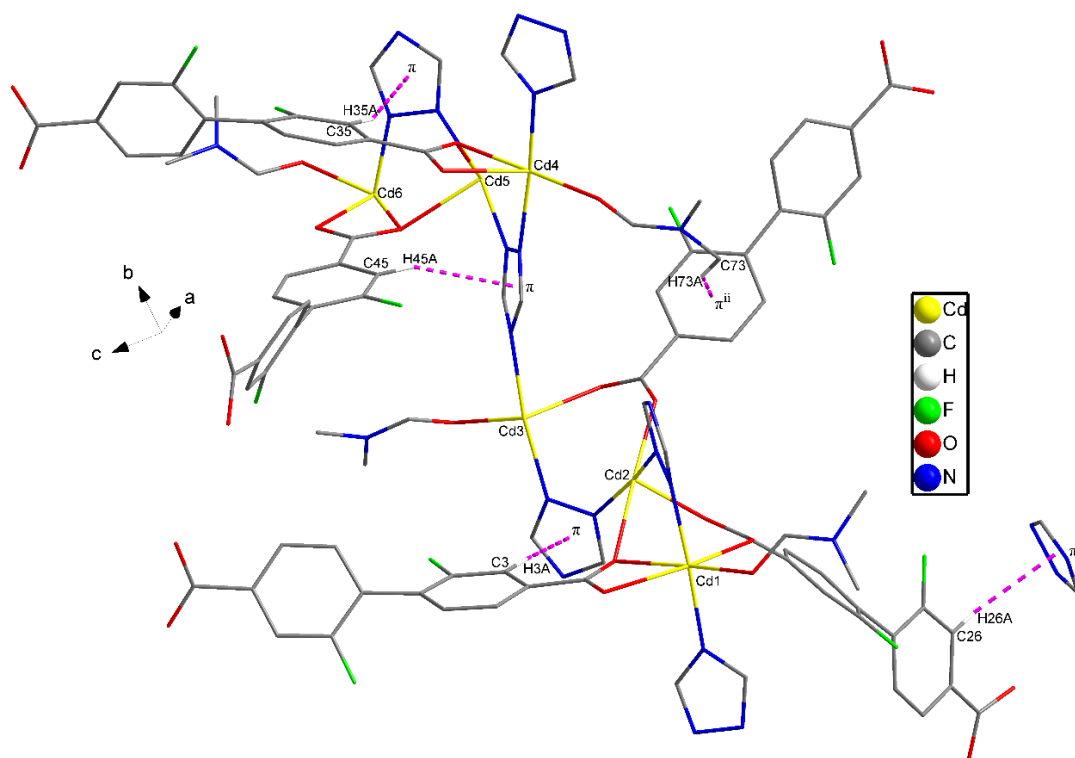


Figure S2 The intermolecular C–H... π interactions in (I) (pink dotted lines).

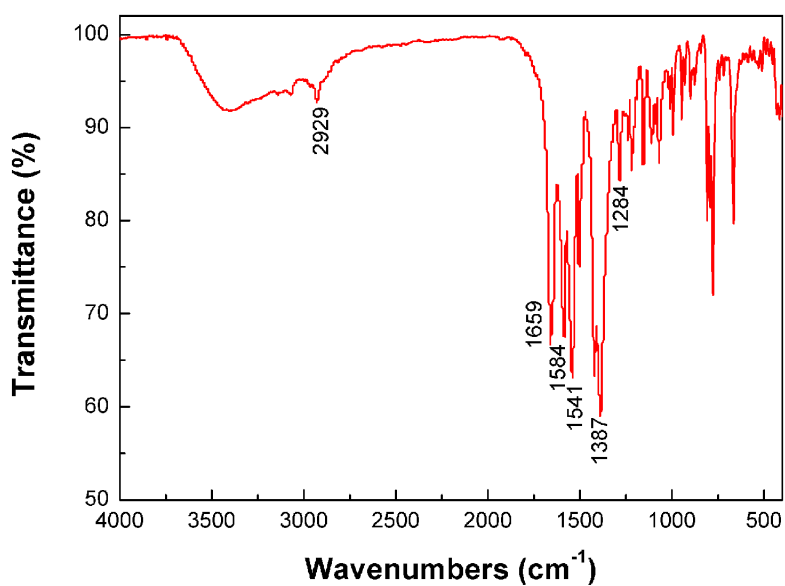


Figure S3 The IR spectrum of (I).

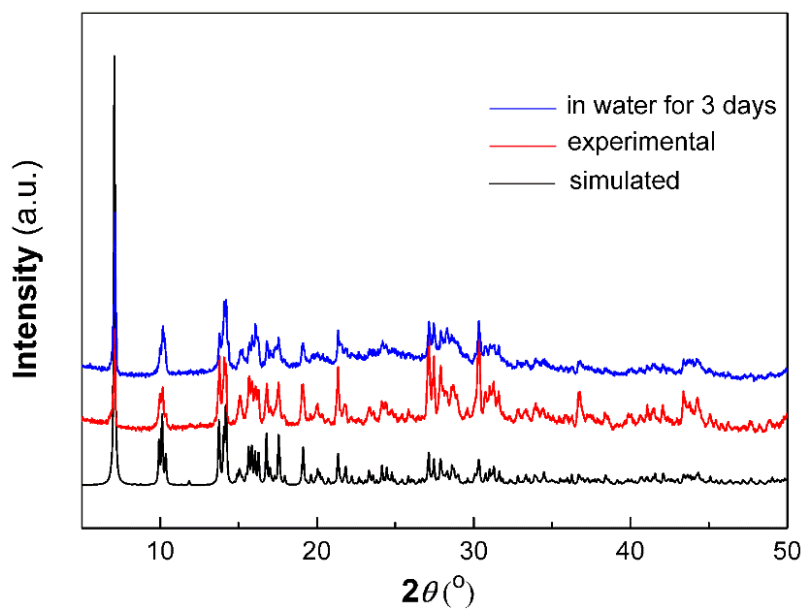


Figure S4 Experimental and simulated powder X-ray diffraction patterns for (I) and the water-treated sample.

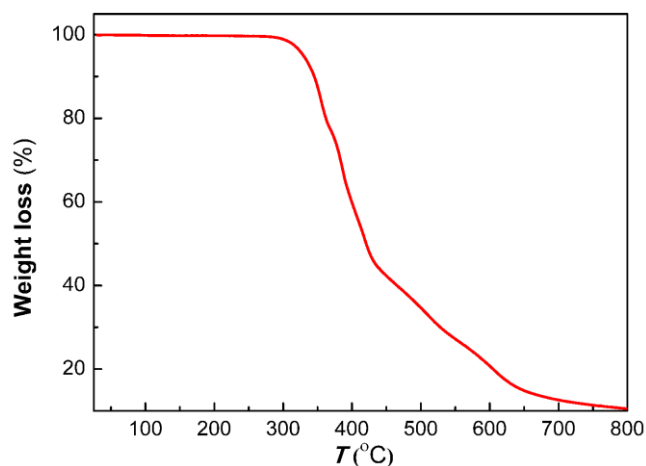


Figure S5 TGA curve for (I).

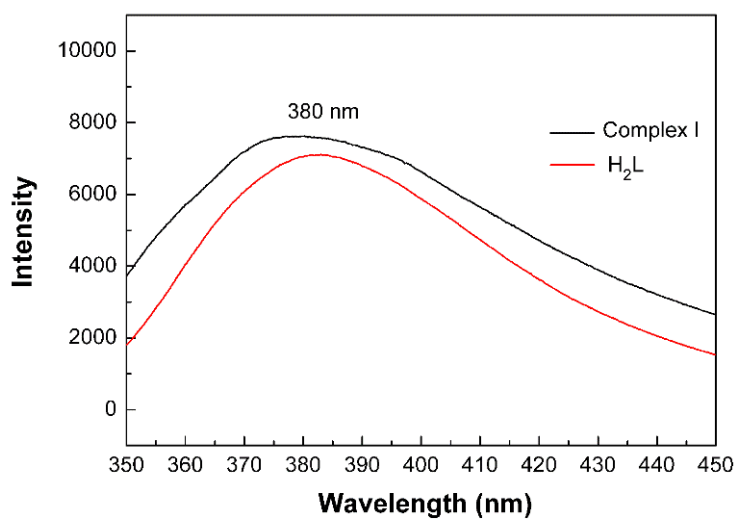


Figure S6 Solid state photoluminescence spectra for (I) and the free H₂L ($\lambda_{\text{ex}} = 320$ nm).

Table S1 Selected geometric parameters (\AA , $^\circ$) of (I)

Cd1—N1	2.214 (8)	Cd4—N6 ^{iv}	2.246 (7)
Cd1—N12 ⁱ	2.242 (7)	Cd4—N7	2.286 (8)
Cd1—O5	2.277 (6)	Cd4—O11 ^v	2.266 (7)
Cd1—O17	2.285 (6)	Cd4—O19	2.288 (7)
Cd1—O1	2.367 (7)	Cd4—O9	2.367 (7)
Cd1—O2	2.391 (7)	Cd4—O10	2.410 (7)
Cd2—N2	2.243 (8)	Cd5—N8	2.248 (7)
Cd2—N4	2.255 (7)	Cd5—N10	2.258 (7)
Cd2—O6	2.233 (7)	Cd5—O12 ^v	2.235 (7)

Cd2—O4 ⁱⁱ	2.263 (7)	Cd5—O16 ^{vi}	2.249 (7)
Cd2—O2	2.436 (7)	Cd5—O10	2.401 (6)
Cd2—O8 ⁱⁱⁱ	2.444 (7)	Cd5—O14	2.456 (7)
Cd3—N5	2.239 (8)	Cd6—N11	2.212 (7)
Cd3—N9	2.244 (8)	Cd6—N3 ^{vii}	2.223 (7)
Cd3—O18	2.294 (8)	Cd6—O15 ^{vi}	2.293 (7)
Cd3—O3 ⁱⁱ	2.300 (7)	Cd6—O20	2.300 (7)
Cd3—O8 ⁱⁱⁱ	2.378 (6)	Cd6—O13	2.354 (6)
Cd3—O7 ⁱⁱⁱ	2.410 (7)	Cd6—O14	2.410 (6)
Cd1...Cd2	3.670(8)	Cd4...Cd5	3.675(10)
Cd2...Cd3	3.652(9)	Cd5...Cd6	3.666(10)
N1—Cd1—N12 ⁱ	177.1 (3)	N6 ^{iv} —Cd4—O11 ^v	92.1 (3)
N1—Cd1—O5	92.8 (3)	N6 ^{iv} —Cd4—N7	174.8 (3)
N12 ⁱ —Cd1—O5	89.7 (3)	O11 ^v —Cd4—N7	93.2 (3)
N1—Cd1—O17	92.7 (3)	N6 ^{iv} —Cd4—O19	88.6 (3)
N12 ⁱ —Cd1—O17	88.9 (3)	O11 ^v —Cd4—O19	83.8 (3)
O5—Cd1—O17	86.9 (3)	N7—Cd4—O19	92.2 (3)
N1—Cd1—O1	90.3 (3)	N6 ^{iv} —Cd4—O9	83.8 (3)
N12 ⁱ —Cd1—O1	86.7 (3)	O11 ^v —Cd4—O9	152.0 (2)
O5—Cd1—O1	155.5 (2)	N7—Cd4—O9	91.5 (3)
O17—Cd1—O1	117.3 (3)	O19—Cd4—O9	123.6 (2)
N1—Cd1—O2	86.6 (3)	N6 ^{iv} —Cd4—O10	91.7 (3)
N12 ⁱ —Cd1—O2	91.5 (3)	O11 ^v —Cd4—O10	97.3 (2)
O5—Cd1—O2	100.7 (3)	N7—Cd4—O10	87.4 (3)
O17—Cd1—O2	172.4 (3)	O19—Cd4—O9	123.6 (2)
O1—Cd1—O2	55.2 (2)	O9—Cd4—O10	55.3 (2)
O6—Cd2—N2	102.8 (3)	O12 ^v —Cd5—N8	102.9 (3)
O6—Cd2—N4	86.5 (3)	O12 ^v —Cd5—O16 ^{vi}	94.6 (3)
N2—Cd2—N4	165.6 (3)	N8—Cd5—O16 ^{vi}	89.2 (3)
O6—Cd2—O4 ⁱⁱ	94.8 (3)	O12 ^v —Cd5—N10	87.5 (3)
N2—Cd2—O4 ⁱⁱ	90.4 (3)	N8—Cd5—N10	166.7 (3)
N4—Cd2—O4 ⁱⁱ	99.8 (3)	O16 ^{vi} —Cd5—N10	98.3 (3)
O6—Cd2—O2	85.6 (3)	O12 ^v —Cd5—O10	88.0 (3)
N2—Cd2—O2	84.4 (3)	N8—Cd5—O10	85.3 (2)
N4—Cd2—O2	85.4 (3)	O16 ^{vi} —Cd5—O10	174.3 (2)
O4 ⁱⁱ —Cd2—O2	174.7 (3)	N10—Cd5—O10	86.9 (2)
O6—Cd2—O8 ⁱⁱⁱ	172.9 (2)	O12 ^v —Cd5—O14	174.0 (2)

N2—Cd2—O8 ⁱⁱⁱ	84.3 (2)	N8—Cd5—O14	82.8 (3)
N4—Cd2—O8 ⁱⁱⁱ	86.5 (2)	O16 ^{vi} —Cd5—O14	83.7 (3)
O4 ⁱⁱ —Cd2—O8 ⁱⁱⁱ	85.5 (3)	N10—Cd5—O14	87.1 (2)
O2—Cd2—O8 ⁱⁱⁱ	94.7 (2)	O10—Cd5—O14	94.3 (2)
N5—Cd3—N9	170.3 (3)	N11—Cd6—N3 ^{vii}	167.1 (3)
N5—Cd3—O18	90.0 (3)	N11—Cd6—O15 ^{vi}	87.2 (2)
N9—Cd3—O18	91.2 (3)	N3 ^{vii} —Cd6—O15 ^{vi}	80.2 (3)
N5—Cd3—O3 ⁱⁱ	89.4 (3)	N11—Cd6—O20	89.2 (3)
N9—Cd3—O3 ⁱⁱ	81.4 (3)	N3 ^{vii} —Cd6—O20	93.8 (3)
O18—Cd3—O3 ⁱⁱ	117.0 (3)	O15 ^{vi} —Cd6—O20	115.9 (3)
N5—Cd3—O8 ⁱⁱⁱ	87.5 (3)	N11—Cd6—O13	100.8 (3)
N9—Cd3—O8 ⁱⁱⁱ	97.4 (3)	N3 ^{vii} —Cd6—O13	91.9 (3)
O18—Cd3—O8 ⁱⁱⁱ	141.4 (3)	O15 ^{vi} —Cd6—O13	156.5 (2)
O3 ⁱⁱ —Cd3—O8 ⁱⁱⁱ	101.4 (2)	O20—Cd6—O13	86.5 (3)
N5—Cd3—O7 ⁱⁱⁱ	104.0 (3)	N11—Cd6—O14	87.6 (2)
N9—Cd3—O7 ⁱⁱⁱ	85.7 (3)	N3 ^{vii} —Cd6—O14	97.9 (2)
O18—Cd3—O7 ⁱⁱⁱ	88.5 (3)	O15 ^{vi} —Cd6—O14	103.2 (2)
O3 ⁱⁱ —Cd3—O7 ⁱⁱⁱ	151.4 (3)	O20—Cd6—O14	140.5 (3)
O8 ⁱⁱⁱ —Cd3—O7 ⁱⁱⁱ	55.1 (2)	O13—Cd6—O14	55.7 (2)

Symmetry codes: (i) $x, y-1, z$; (ii) $x, 2-y, z-1/2$; (iii) $x+1/2, 3/2-y, z+1/2$; (iv) $x+1/2, y+1/2, z$; (v) $x, 3-y, z-1/2$; (vi) $x-1/2, 5/2-y, z-1/2$; (vii) $x-1/2, y+1/2, z$.

Table S2 Hydrogen-bond geometry (Å, °) for (I)

D—H...A	d(D—H)	d(H...A)	d(D...A)	∠D—H...A
C60—H60A...O19 ^v	0.95	2.67	3.185(4)	115
C64—H64A...O17 ^{vi}	0.95	2.65	3.174(4)	115
C67—H67B...F5A ⁱⁱⁱ	0.98	2.63	3.265(4)	123
C67—H67C...O3 ⁱ	0.98	2.57	3.402(2)	143
C69—H69B...O1 ⁱⁱ	0.98	2.62	3.255(9)	122
C72—H72C...O15 ⁱⁱⁱ	0.98	2.54	3.389(3)	145
C73—H73C...F4A ^{vii}	0.98	2.50	3.300(2)	138
C74—H74A...O13	0.95	2.41	3.015(5)	121
C75—H75A...F3 ^{iv}	0.98	2.51	3.431(7)	157
C75—H75B...F8A ⁱⁱ	0.98	2.65	3.428(6)	137
C76—H76A...O19 ⁱⁱ	0.98	2.54	3.495(4)	165
C76—H76C...O9 ⁱⁱ	0.98	2.53	3.232(3)	129

Symmetry codes: (i) $x+1/2, 3/2-y, z-1/2$; (ii) $x-1/2, y+1/2, z$; (iii) $x, 2-y, z-1/2$; (iv) $x, 2-y, z+1/2$; (v) $x-1/2, y-1/2, z$; (vi) $x, y+1, z$; (vii) $x+1/2, y+1/2, z$.

Table S3 C–H... π interactions (Å, °) for (I)

D–H...A	d(D–H)	d(H...A)	d(D...A)	\angle D–H...A
C3–H3A... π (trz)	0.95	2.70	3.625(11)	165
C26–H26A... π (trz) ⁱ	0.78	2.57	3.349(11)	140
C35–H35A... π (trz)	0.95	2.54	3.438(11)	157
C45–H45A... π (trz)	0.95	2.62	3.508(11)	157
C73–H73A... π (Ph) ⁱⁱ	0.98	2.70	3.607(16)	155

Symmetry codes: (i) $x-1/2, 3/2-y, z-1/2$; (ii) $x, 2-y, z-1/2$.**Table S4** The dihedral angles for L²⁻ ligand in (I)

L ²⁻	Dihedral angles (°)	
	F-Ph/F-Ph ring	F-Ph ring/CO ₂ ²⁻
L (F ₁ /F ₂)	54.5(3)	(F1) 3.4(3); (F2) 35.5(3)
L (F ₃ /F ₄)	57.5(3)	(F3) 16.9(2); (F4) 15.0(3)
L (F ₅ /F ₆)	63.3(3)	(F5) 7.2(3); (F6) 16.4(3)
L (F ₇ /F ₈)	55.2(3)	(F7) 2.7(3); (F8) 36.1(3)