

## Supporting information

### Synthesis and crystal structure of a novel binuclear copper complex with the nalidixic acid derivative 1-ethyl-1,4-dihydro-7-methyl-4-oxo-N'-(1H-imidazol-4-ylmethylidene)-1,8-naphthyridine-3-carbohydrazide

Authors

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#### S1. Synthesis of the ligand and the Cu(II) complex

The first step, which involves the synthesis of the nalidixic acid hydrazide derivative (hzd), was performed according to a similar procedure described by Aboul-Fadl *et al.* 2010.

Nalidixic acid (2.0 mmol, 0.4465 g) was added to 10.0 mL of freshly distilled CH<sub>2</sub>Cl<sub>2</sub> followed by the addition of 0.50 mL of triethylamine. The solution was stirred for 15 min and then cooled to 0-5 °C followed by the slow addition of 0.2 mL of methyl chloroformate dissolved in 5.0 mL of dry CH<sub>2</sub>Cl<sub>2</sub>. The mixture was maintained under stirring for 30 min and then added dropwise to a dispersion of 0.50 mL of hydrazide monohydrate (N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O) in 1.0 mL of dry CH<sub>2</sub>Cl<sub>2</sub>. The white solid formed was maintained under stirring at room temperature for 5 h in order to assure the complete formation of the hydrazide. After this stage, 5.0 mL of water was added to the suspension and the volume of CH<sub>2</sub>Cl<sub>2</sub> was reduced in a rotavapor. The white solid obtained was separated by filtration, washed with several portions of 10 mL of distilled water and dried in a desiccator under P<sub>2</sub>O<sub>5</sub>.

The precursor hzd formed was used to obtain the ligand h4imi according to the following procedure: 1.5 mmol of 4-imidazolecarboxyaldehyde (4-imi, C<sub>4</sub>H<sub>4</sub>N<sub>2</sub>O, 96.09 g mol<sup>-1</sup>) was added to a suspension of 1.5 mmol of hzd (C<sub>12</sub>H<sub>14</sub>N<sub>4</sub>O<sub>2</sub>, 246.27 g mol<sup>-1</sup>) in 10.0 mL of methanol. Glacial acetic acid (4 drops) was added as a catalyst. The reaction was kept under stirring for 4 h at room temperature in order to

assure the complete formation of h4imi. The white-pearl solid formed was then separated by filtration, washed several times with cold methanol and dried in a desiccator over  $P_2O_5$  for 2 days. Yield: 43%. Elemental analysis: Calcd. for  $C_{16}H_{16}N_6O_2$  (%): C, 59.25; H, 4.97; N, 25.91. Found (%): C, 59.46; H, 5.20; N, 23.02. ESI-QTOF-MS (m/z): 325.1436 [ $C_{16}H_{16}N_6O_2+H^+$ ]. IR  $\nu(N-H)$  3235  $cm^{-1}$ , 3170  $cm^{-1}$  and 3126  $cm^{-1}$ ,  $\nu_{ar}(C-H)$  3065  $cm^{-1}$ ,  $\nu_{aly}(C-H)$  2968  $cm^{-1}$  and 2928  $cm^{-1}$ ,  $\nu(C=O)$  1671  $cm^{-1}$  and 1634  $cm^{-1}$ ,  $\nu(C=N)$  1610.3  $cm^{-1}$ , 1553  $cm^{-1}$  and 1532  $cm^{-1}$ ,  $\nu(C-N)$  1501  $cm^{-1}$  and  $\nu(N-N)$  1443  $cm^{-1}$ .

For  $[Cu_{1.5}(h4imi)Cl]Cl \cdot 2(CH_3OH)$ , a 1:1 mixture of  $CuCl_2 \cdot 2H_2O$  (0.50 mmol, 170.48  $gmol^{-1}$ ) in 10.0 mL of MeOH with h4imi (0.50 mmol, 324.34  $gmol^{-1}$ ) suspended in 10.0 mL of MeOH was maintained under stirring. The immediate formation of a green solid was observed but the reaction was left for 2 h under stirring in order to assure the complete formation of the complex. The solid was then separated by filtration and washed several times with  $CH_2Cl_2$ . Prior to the solid purification, the mother liquor was separated and left for crystallization. After two weeks, green crystals were formed and then evaluated as suitable crystals for X-ray diffraction. Elemental analysis: Calcd. For  $Cu_{1.5}C_{18}H_{23}N_6O_4$  (%): C, 39.05; H, 4.15; N, 15.18. Found (%): C, 40.07; H, 3.83; N, 17.39.

**Table S1** Atomic coordinates output for h4imi DFT optimized structure

Atom	X	y	z
C	4.29908492925532	3.10919222023542	-1.49140209129262
N	2.48482754458878	2.58523460573466	-0.21687067470614
H	1.85673920878275	2.50365899091285	-1.02518999215764
O	2.65292519496002	2.44882615280949	2.06052021538461
O	-0.03069280818113	2.07053601195793	-1.20286985575754
N	-1.09930353819085	1.04651566446929	2.60924919115669
N	-3.21851082603991	0.41073790593288	1.91897149026274
N	3.78206390766014	2.91526884267078	-0.33630011425517
C	2.00040181270191	2.33989871371184	1.04134248578708
C	0.13049624475764	1.51734297846901	2.32495507030917
H	0.81749288763598	1.61793870012022	3.15811501338823
C	-1.65392567773408	1.26006459099197	0.29029464832141
C	-2.02054335550967	0.90025195996109	1.59125022848199
C	-2.62258823685082	1.10647655597375	-0.69987467806207

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H	-2.35480540679011	1.38479568044798	-1.71261502920917
C	-0.31723465088609	1.77237397381276	-0.04594360088204
C	4.35299165502656	3.23796013045195	-4.01362296049513
N	2.36977348066687	2.84237286049544	-3.03863960804831
C	-4.12448595945281	0.27149175067851	0.96327141797550
C	0.57612765859633	1.87857022183093	1.08699623026490
C	3.70448068859654	3.06184858756041	-2.81078414148027
C	-1.44224021557152	0.65146994938051	3.97485357072492
H	-2.44608776131338	1.01896113981498	4.18749356166393
H	-0.74432869168952	1.17095944789856	4.63292330232160
C	-1.37478806625063	-0.84656542534945	4.18385960621713
H	-2.07850036412660	-1.35523664266965	3.52634276256498
H	-1.63923118875317	-1.08966095049946	5.21582756014556
H	-0.37044822603440	-1.22572264821820	3.98213052346847
C	-3.86137260411546	0.61762713111957	-0.37011423072391
H	-4.63165688605937	0.48393820702617	-1.12045180708595
C	2.20819548965944	2.88980720295532	-4.32987261549675
C	-5.46144471856641	-0.28462029742921	1.34209838739040
H	-5.47139623426898	-0.58210746361133	2.38967081863569
H	-5.70781032891093	-1.15175657474581	0.72435318221666
H	-6.24751767670307	0.45757119084580	1.17819108606811
N	3.38371936424393	3.12519437731751	-4.96613260203776
H	3.51878169580344	3.20696744605624	-5.95867302583803
H	1.27431528266462	2.76128641330775	-4.85607752426869
H	5.38280712754908	3.43477069578023	-4.26287523114046
H	5.35757924884954	3.35833970179129	-1.46330056981211

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**Table S2** Selected distances of h4imi optimized structure and h4imi coordinated structure

Bond	Distance/ Å	
	crystal	h4imi DFT
O1-C7	1.262(2)	1.2286
C5-N4	1.343(2)	1.3704
N3-N4	1.400(2)	1.3439
N3-C4	1.284(2)	1.2803
C5-C6	1.471(2)	1.4978
C5-O2	1.272(2)	1.2151

**Table S3** Selected angles of h4imi optimized structure and h4imi coordinated structure

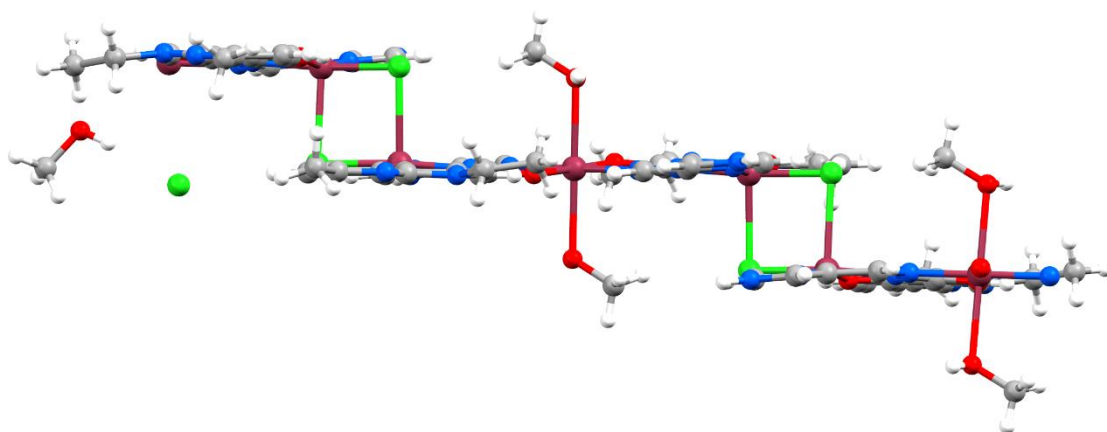
Angle	crystal	h4imi DFT
O1-C7-C6	125.5(2)	125.14
C7-C6-C5	125.4(2)	125.87
N4-C5-O2	123.3(2)	124.35
C5-N4-N3	107.9(1)	117.83
N4-N3-C4	123.1(2)	120.48
N3-C4-C3	128.5(2)	130.64

**Table S4** Selected torsions of h4imi optimized structure and h4imi coordinated structure

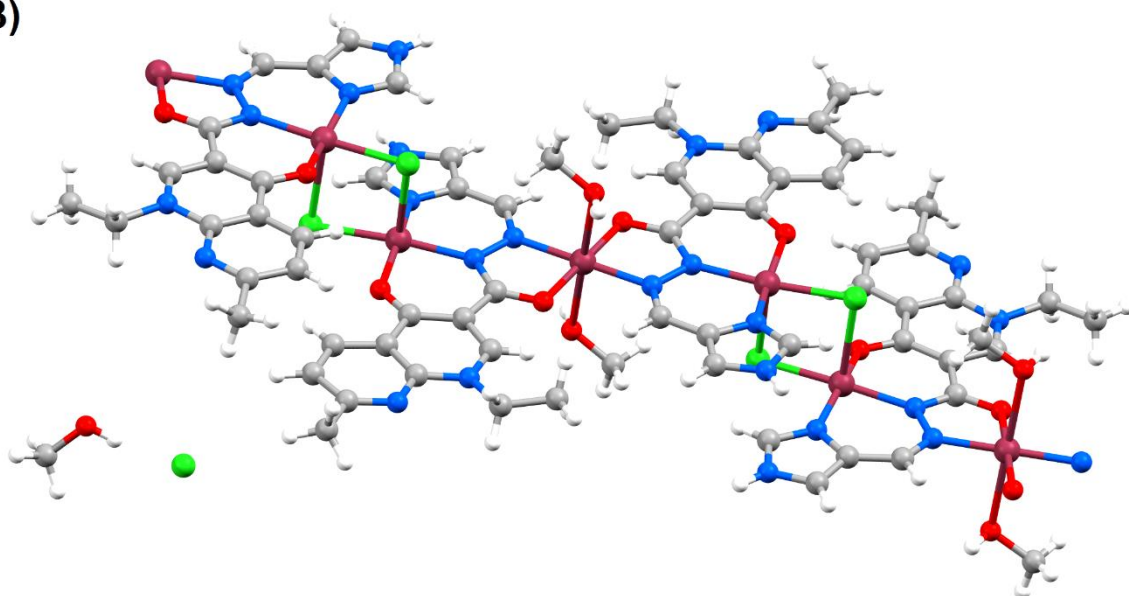
Torsions	crystal	h4imi DFT
O1-C7-C6-C5	1.8(3)	-1.31
O2-C5-N4-N3	-2.3(2)	2.84
C5-N4-N3-C4	176.8(2)	177.54
N4-N3-C4-C3	-2.3(3)	1.34
N3-C4-C3-N2	2.9(3)	3.34

**Figure S1** Extended structure of the  $\{[\text{Cu}_{1.5}(\text{h4imi})\text{Cl}]\text{Cl}\cdot 2(\text{CH}_3\text{OH})\}_n$  polymer. The counter ions are intentionally not represented.

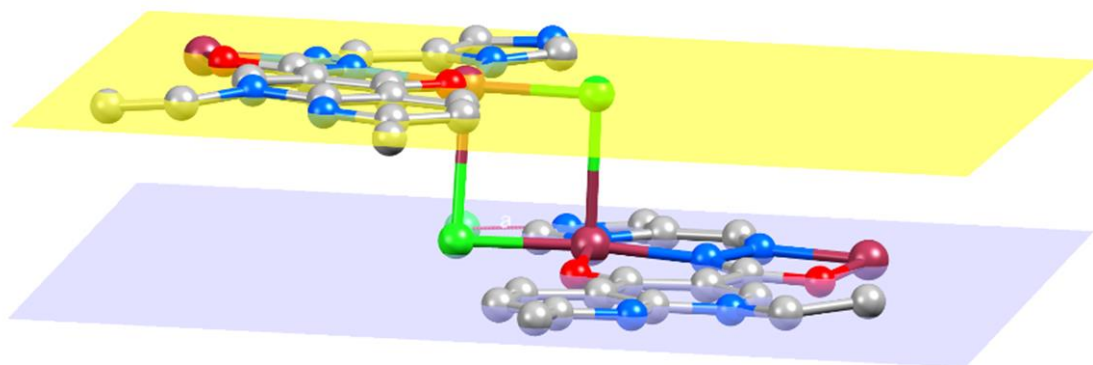
(A)



(B)



**Figure S2** Minimum square planes representation of almost parallel h4imi (22 atoms) molecules. Due to the disorder, the ethyl groups of h4imi were intentionally not picked for the calculation.



**Figure S3** (A) PBE0/def2-TZVP optimized structure of h4imi and (B) comparison between coordinated (X ray data in purple) and non-coordinated h4imi (blue).

