



STRUCTURAL SCIENCE  
CRYSTAL ENGINEERING  
MATERIALS

Volume 78 (2022)

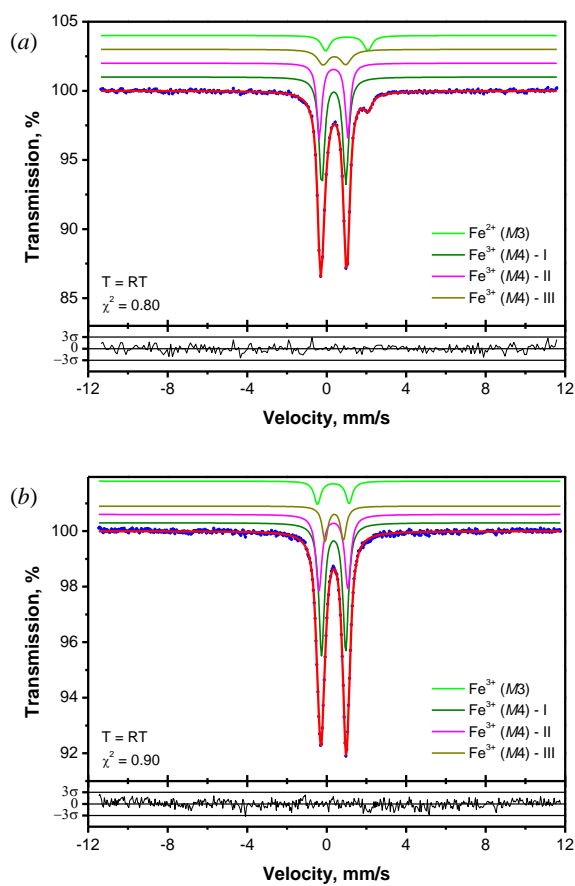
Supporting information for article:

**X-ray diffraction and Mössbauer spectroscopy study of oxoborate azoproite  $(\text{Mg}, \text{Fe}^{2+})_2(\text{Fe}^{3+}, \text{Ti}, \text{Mg}, \text{Al})\text{O}_2(\text{BO}_3)$ : an *in situ* temperature-dependent investigation ( $5 \leq T \leq 1650$  K)**

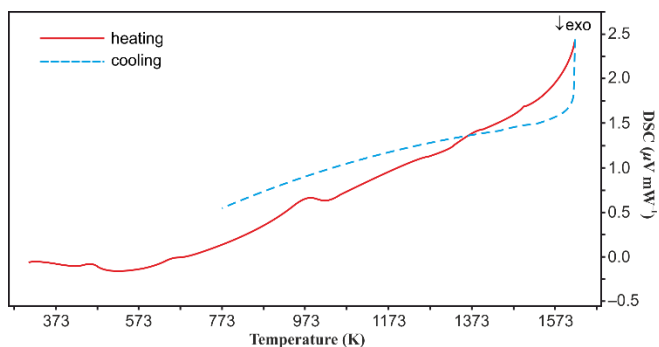
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**Figure S1** The investigated sample of azoprite shown by white arrow.



**Figure S2** The  $^{57}Fe$  Mössbauer spectra in a wide-velocity range of (a) initial azoprite and (b) azoprite calcined at 1023 K for 1 hour in air.



**Figure S3** DSC curves of azoprite in heating and cooling regimes.

**Table S1** Chemical composition (in wt. %) of azoprite.

Constituent	Mean	Range	Stand. Dev.	Probe Standard
[FeO <sub>total</sub> ]	21.32	20.63 – 22.00	0.97	FeS <sub>2</sub>
MgO	45.23	45.01 – 45.45	0.31	MgO
Al <sub>2</sub> O <sub>3</sub>	3.38	2.97 – 3.78	0.57	NaAlSi <sub>3</sub> O <sub>8</sub>
TiO <sub>2</sub>	11.22	11.20 – 11.24	0.03	Ti
B <sub>2</sub> O <sub>3</sub> *	18.93	18.78 – 19.07	0.20	
[FeO]**	4.00	3.00 – 5.00	1.41	
[Fe <sub>2</sub> O <sub>3</sub> ]**	23.69	23.34 – 24.04	0.49	
Total	100.07	99.03 – 101.10	1.46	

\* – calculated from crystal structure

\*\* – calculated from charge balance

**Table S2** Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ ) for azoproteite (293 K).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$ ( $\text{\AA}^2$ )	Occupancy
Mg1	0	0	0	0.0030 (4)	1
Mg2	0.5	0	0.5	0.0137 (5)	1
Mg3	0.00090 (12)	0.28056 (11)	0	0.0162 (4)	0.95
Fe3	0.00090 (12)	0.28056 (11)	0	0.0162 (4)	0.05
Fe4	0.73794 (7)	0.38401 (6)	0.5	0.0090 (2)	0.4
Ti4	0.73794 (7)	0.38401 (6)	0.5	0.0090 (2)	0.3
Mg4	0.73794 (7)	0.38401 (6)	0.5	0.0090 (2)	0.2
Al4	0.73794 (7)	0.38401 (6)	0.5	0.0090 (2)	0.1
O1	0.8503 (3)	0.0423 (2)	0.5	0.0090 (7)	1
O2	0.3832 (3)	0.0757 (2)	0	0.0109 (7)	1
O3	0.6248 (3)	0.1420 (2)	0.5	0.0097 (7)	1
O4	0.1089 (3)	0.1432 (2)	0	0.0101 (7)	1
O5	0.3495 (3)	0.2628 (2)	0.5	0.0093 (7)	1
B1	0.2735 (4)	0.3608 (3)	0.5	0.0082 (9)	1

**Table S3** Refined site-scattering factors and assignment for cation sites in the structure of azoproteite (293 K).

Site	SC	SOF	$SSF_{\text{exp}}[e^-]$	Assigned occupancy	$SSF_{\text{calc}}[e^-]$
M4	Ti	0.99308	21.85	$\text{Fe}_{0.4}\text{Ti}_{0.3}\text{Mg}_{0.2}\text{Al}_{0.1}$	20.70

\*SC – scattering curve used to refine site occupancy; SOF – refined site-occupation factor;  $SSF_{\text{exp}}$  and  $SSF_{\text{calc}}$  – experimental and calculated site-scattering factors.

**Table S4** Anisotropic atomic displacement parameters ( $\text{\AA}^2$ ) of azoproite (293 K).

Atom	$U_{11}$	$U_{22}$	$U_{33}$	$U_{12}$	$U_{13}$	$U_{23}$
Mg1	0.0044 (8)	0.0020 (7)	0.0026 (7)	-0.0001 (4)	0	0
Mg2	0.0124 (9)	0.0141 (9)	0.0146 (10)	0.0013 (6)	0	0
Mg3	0.0157 (7)	0.0155 (7)	0.0174 (7)	-0.0008 (4)	0	0
Fe3	0.0157 (7)	0.0155 (7)	0.0174 (7)	-0.0008 (4)	0	0
Fe4	0.0087 (3)	0.0092 (3)	0.0091 (3)	0.0010 (2)	0	0
Ti4	0.0087 (3)	0.0092 (3)	0.0091 (3)	0.0010 (2)	0	0
Mg4	0.0087 (3)	0.0092 (3)	0.0091 (3)	0.0010 (2)	0	0
Al4	0.0087 (3)	0.0092 (3)	0.0091 (3)	0.0010 (2)	0	0
O1	0.0083 (11)	0.0096 (12)	0.0090 (11)	-0.0002 (8)	0	0
O2	0.0079 (11)	0.0103 (12)	0.0146 (12)	0.0005 (9)	0	0
O3	0.0085 (11)	0.0098 (12)	0.0107 (12)	0.0009 (8)	0	0
O4	0.0094 (12)	0.0067 (11)	0.0141 (13)	0.0007 (8)	0	0
O5	0.0088 (11)	0.0077 (11)	0.0114 (12)	-0.0005 (8)	0	0
B1	0.0070 (16)	0.0102 (17)	0.0072 (16)	0.0024 (12)	0	0

**Table S5** Selected bond lengths (Å) of of azoproite (293 K).

Bond	Distance (Å)	BVS (v.u.)	Bond	Distance (Å)	BVS (v.u.)
Mg1—O1 <sup>ix</sup>	2.109 (2)	0.32	M4—O5 <sup>i</sup>	2.076 (3)	0.42
Mg1—O1 <sup>x</sup>	2.109 (2)	0.32	M4—O1 <sup>ii</sup>	2.107 (3)	0.38
Mg1—O1 <sup>xi</sup>	2.109 (2)	0.32	M4—O4 <sup>iii</sup>	1.949 (2)	0.59
Mg1—O1 <sup>iv</sup>	2.109 (2)	0.32	M4—O4 <sup>i</sup>	1.949 (2)	0.59
Mg1—O4	2.026 (3)	0.41	M4—O2 <sup>iii</sup>	2.077 (2)	0.42
Mg1—O4 <sup>xii</sup>	2.026 (3)	0.41	M4—O2 <sup>i</sup>	2.077 (2)	0.42
<M1—O> <sub>6</sub>	2.08	2.11	<M4—O> <sub>6</sub>	2.04	2.81
Mg2—O3	2.090 (3)	0.34	B1—O5	1.393 (5)	0.94
Mg2—O3 <sup>iv</sup>	2.090 (3)	0.34	B1—O3 <sup>viii</sup>	1.375 (5)	0.99
Mg2—O2	2.072 (2)	0.36	B1—O1 <sup>viii</sup>	1.385 (5)	0.96
Mg2—O2 <sup>v</sup>	2.072 (2)	0.36	<B1—O> <sub>3</sub>	1.38	2.89
Mg2—O2 <sup>iv</sup>	2.072 (2)	0.36			
Mg2—O2 <sup>vi</sup>	2.072 (2)	0.36			
<M2—O> <sub>6</sub>	2.08	2.12			
Mg3—O5 <sup>vii</sup>	2.123 (2)	0.31			
Mg3—O5 <sup>viii</sup>	2.123 (2)	0.31			
Mg3—O3 <sup>vii</sup>	2.117 (2)	0.32			
Mg3—O3 <sup>viii</sup>	2.117 (2)	0.32			
Mg3—O4	1.959 (3)	0.49			
Mg3—O2 <sup>vii</sup>	2.073 (3)	0.36			
<M3—O> <sub>6</sub>	2.09	2.12			

BVS – bond valence sum

**Table S6** Equations of approximation of temperature dependencies of unit cell parameters of azoproite.

<i>T</i> (K)	Equation $l(t) = l_0 + l_1t + l_2t^2$			
	<i>a</i> ( <i>t</i> ) (Å)	<i>b</i> ( <i>t</i> ) (Å)	<i>c</i> ( <i>t</i> ) (Å)	<i>V</i> ( <i>t</i> ) (Å <sup>3</sup> )
<b>93–293</b>	9.2551(1) +	12.2814(2) +	3.0108(3) +	342.23(1) +
	0.000057(3)× <i>t</i> +	0.000061(6)× <i>t</i> +	0.000027(1)× <i>t</i> +	0.0069(3)× <i>t</i> +
	0.000000048(1)× <i>t</i> <sup>2</sup>	0.000000015(3)× <i>t</i> <sup>2</sup>	0.000000041(5)× <i>t</i> <sup>2</sup>	0.0000069(2)× <i>t</i> <sup>2</sup>
<b>293–1373</b>	9.2534(3) +	12.2765(1) +	3.01591(4) +	343.83(1) +
	0.000087(1)× <i>t</i> +	0.000068(1)× <i>t</i> +	0.000039(1)× <i>t</i> +	0.0094(7)× <i>t</i> +
	0.000000016(1)× <i>t</i> <sup>2</sup>	0.000000043(1)× <i>t</i> <sup>2</sup>	0.000000085(1)× <i>t</i> <sup>2</sup>	0.0000029(6)× <i>t</i> <sup>2</sup>