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**Supporting information for article:**

**A general method for searching homometric structures**

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## S1. Supporting text

### S1.1. The distance homometric structure search method

For a given structure, the structure factor is given by:

$$F(hkl) = \sum_n (f_n \exp[2\pi i(hx_n + ky_n + lz_n)])$$

$$= \sum_n f_n \{\sin[2\pi(hx_n + ky_n + lz_n)] + i\cos[2\pi(hx_n + ky_n + lz_n)]\} \quad (1)$$

The intensity is proportional to  $F(hkl)F^*(hkl)$ , which is expressed as:

$$F(hkl)F^*(hkl) =$$

$$\sum_{m,n} f_m f_n^* \{\sin[2\pi(hx_m + ky_m + lz_m)]\sin[2\pi(hx_n + ky_n + lz_n)]$$

$$+ \cos[2\pi(hx_m + ky_m + lz_m)]\cos[2\pi(hx_n + ky_n + lz_n)]\}$$

$$= \sum_{m,n} f_m f_n^* \cos[2\pi(hx_m + ky_m + lz_m - hx_n - ky_n - lz_n)] \quad (2)$$

Divide the crystallographic orbits into 3 unions P, Q, and R. For simplicity, we use the matrix forms below for further discussion:

$$\mathbf{H} = \begin{bmatrix} h \\ k \\ l \end{bmatrix}, \mathbf{X}_m = \begin{bmatrix} x_m \\ y_m \\ z_m \end{bmatrix}$$

Suppose there are p atoms locate at the crystallographic orbits from P, q atoms for Q and r atoms for R, then total intensity can be divided into:

$$I(\mathbf{H}) = CF(\mathbf{H})F^*(\mathbf{H})$$

$$= \frac{1}{2} [I_{PP}(\mathbf{H}) + I_{QQ}(\mathbf{H}) + I_{RR}(\mathbf{H})] + I_{PQ}(\mathbf{H}) + I_{PR}(\mathbf{H}) + I_{QR}(\mathbf{H}) \quad (3)$$

And for any pair of (P, Q),  $I_{PQ}(\mathbf{H})$  is defined as:

$$I_{PQ}(\mathbf{H}) = C \sum_{p \text{ in } P, q \text{ in } Q} \{f_p f_q^* \cos[2\pi\mathbf{H}^T(\mathbf{X}_p - \mathbf{X}_q)] + f_q f_p^* \cos[2\pi\mathbf{H}^T(\mathbf{X}_q - \mathbf{X}_p)]\}$$

$$= C \sum_{p \text{ in } P, q \text{ in } Q} \{(f_p f_q^* + f_q f_p^*) \cos[2\pi\mathbf{H}^T(\mathbf{X}_p - \mathbf{X}_q)]\} \quad (4)$$

where C is a constant.  $I_{PQ}(\mathbf{H})$  is the term of intensity generated by atom pairs with one from union P and the other from union Q.

Suppose that there are two symmetry operations **A** and **B** which belong to the point group of the crystal's reciprocal lattice. The position of **A** and **B** are  $\mathbf{X}_A$  and  $\mathbf{X}_B$ . P remains unchanged under **A** and

Q remains unchanged under **B**. R will be changed under **A** or **B**, but **AR** and **BR** are identical. Union **AR=BR=R'**, consider a structure containing P, Q and R',  $I_{R'R'}(\mathbf{H})$  has the following form:

$$\begin{aligned}
 I_{R'R'}(\mathbf{H}) &= 2C \sum_{r_1', r_2' \text{ in } R'} f_{r_1'} f_{r_2'}^* \cos[2\pi \mathbf{H}^T (\mathbf{X}_{r_1'} - \mathbf{X}_{r_2'})] \\
 &= 2C \sum_{r_1, r_2 \text{ in } R} f_{r_1} f_{r_2}^* \cos[2\pi \mathbf{H}^T (\mathbf{A}(\mathbf{X}_{r_1} - \mathbf{X}_A) + \mathbf{X}_A - \mathbf{A}(\mathbf{X}_{r_2} - \mathbf{X}_A) - \mathbf{X}_A)] \\
 &= 2C \sum_{r_1, r_2 \text{ in } R} f_{r_1} f_{r_2}^* \cos[2\pi \mathbf{H}^T \mathbf{A}(\mathbf{X}_{r_1} - \mathbf{X}_{r_2})] \\
 &= 2C \sum_{r_1, r_2 \text{ in } R} f_{r_1} f_{r_2}^* \cos[2\pi (\mathbf{A}^T \mathbf{H})^T (\mathbf{X}_{r_1} - \mathbf{X}_{r_2})] \\
 &= I_{RR}(\mathbf{A}^T \mathbf{H})
 \end{aligned} \tag{5}$$

$I_{PR'}(\mathbf{H})$  can be written as:

$$I_{PR'}(\mathbf{H}) = C \sum_{p \text{ in } P, r' \text{ in } R'} \{(f_p f_{r'}^* + f_{r'} f_p^*) \cos[2\pi \mathbf{H}^T (\mathbf{X}_p - \mathbf{X}_{r'})]\} \tag{6}$$

$$= C \sum_{p \text{ in } P, r \text{ in } R} \{(f_p f_{r'}^* + f_{r'} f_p^*) \cos[2\pi \mathbf{H}^T (\mathbf{X}_p - \mathbf{A}(\mathbf{X}_r - \mathbf{X}_A) - \mathbf{X}_A)]\} \tag{7}$$

As  $\mathbf{AP} = P$ , for all  $p$  in P there is one and only one  $p'$  that makes  $\mathbf{X}_{p'} = \mathbf{A}(\mathbf{X}_p - \mathbf{X}_A) + \mathbf{X}_A$ . So we can reformulate Eq. 7 as follows:

$$\begin{aligned}
 I_{PR'}(\mathbf{H}) &= C \sum_{p \text{ in } P, r \text{ in } R} \{(f_p f_{r'}^* + f_{r'} f_p^*) \cos[2\pi \mathbf{H}^T (\mathbf{A}(\mathbf{X}_p - \mathbf{X}_A) + \mathbf{X}_A - \mathbf{A}(\mathbf{X}_r - \mathbf{X}_A) - \mathbf{X}_A)]\} \\
 &= I_{PR}(\mathbf{A}^T \mathbf{H})
 \end{aligned} \tag{8}$$

Similarly, considering  $\mathbf{BQ} = Q$ ,  $I_{QR'}(\mathbf{H})$  has

$$I_{QR'}(\mathbf{H}) = I_{QR}(\mathbf{B}^T \mathbf{H}) \tag{9}$$

As **A** and **B** are point group operations,  $\mathbf{A}^T \mathbf{H}$ ,  $\mathbf{B}^T \mathbf{H}$  and **H** have the same distance to the origin. Thus, they will contribute to the same peak in PXRD, and the summation of all contributions for certain reflection will be the same according to Eq. 3. So, the original structure containing unions P, Q, R and the structure containing unions P, Q, R' are identical in PXRD.

### S1.2. Extended discussion of diffraction equality of NaNbO<sub>3</sub>(Str. 1a&b)

The correlation of the reflections between the two structures can be explained by dividing each structure into two substructures. One substructure which is the same in two structures contains Na and O in the space group of  $I2_12_12_1$ ; the other contains Nb in  $C222_1$ , which can be related by an inversion center at (0,0,0.125) from Str. 1a to Str. 1b. As the Na and O substructure and the Nb substructure have

different symmetries, the coordination environment of Nb atoms may change after the inversion center operation. For the reflections  $h k 2n$  which corresponds to the reflections from the averaged structure with half  $c$ -axis, the intensities and phases are equal since the average structure of Str. 1a and Str. 1b are the same. For the reflections  $h k 2n+1$ , as the Na and O substructure are body-centered, it only contributes the intensity to those reflections with  $h+k=2n+1$  where  $h+k+l=2n$ . The Nb substructure has a C-center which only contributes the intensity to those reflections with  $h+k=2n$ . So when  $l=2n+1$ , the Na and O substructure and Nb substructure contribute to different reflections and will not affect each other. Moreover, as Nb substructure only contains one type of element, only phases will change during the inversion center operation even with considering of resonant scattering. Consequently, the intensity of reflections with  $l=2n+1$  of Str. 1a and Str. 1b are the same.

## S2. Supporting tables

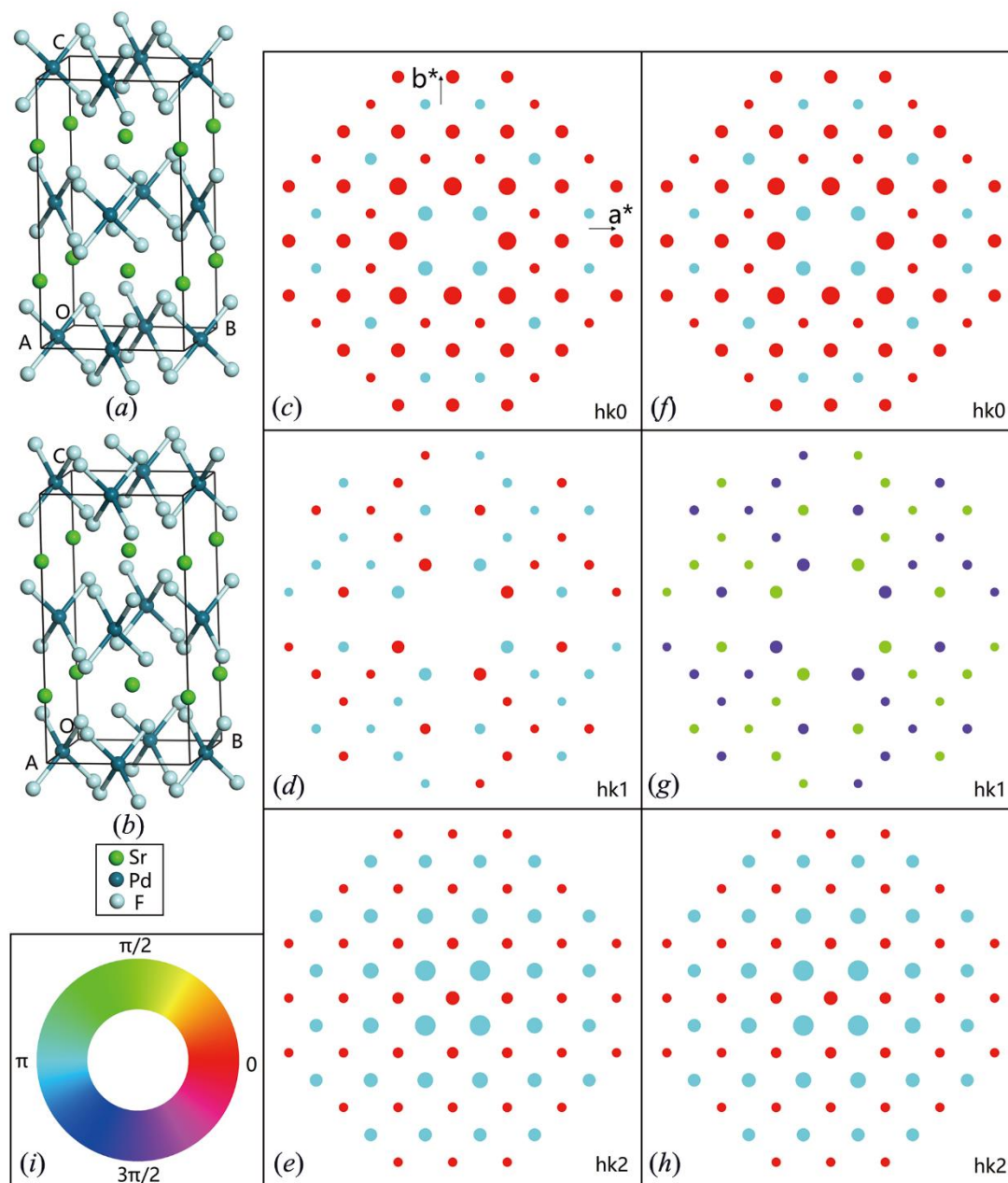
**Table S1** List of homometric structures found by the program mentioned in the text.

ICSD No.	Composition	Note	ICSD No.	Composition	Note
1381	Cu <sub>16</sub> O <sub>14.15</sub>	<i>I4<sub>1</sub>/amd</i>	52289	Cr <sub>7</sub> C <sub>3</sub>	<i>P6<sub>3</sub>mc</i> to <i>Cmc2<sub>1</sub></i>
4141	Eu <sub>2</sub> Sn <sub>2</sub> O <sub>7</sub>	<i>Fd<math>\bar{3}m</math></i>	53191	CrNb <sub>4</sub> Se <sub>8</sub>	<i>P6<sub>3</sub>/mmc</i>
9965	V <sub>2</sub> C	<i>Pbcn</i>	76432	NaNbO <sub>3</sub>	<i>P222<sub>1</sub></i> to <i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>
15958	Hg(NH <sub>3</sub> ) <sub>2</sub> Cl <sub>2</sub>	weakly; <i>Cmmm</i> to <i>P4/mmm</i> ; H not determined	77675	Cu <sub>4</sub> O <sub>3</sub>	<i>I4<sub>1</sub>/amd</i>
18126	HgI <sub>2</sub>	<i>I4<sub>1</sub>/amd</i>	85824	FeH	<i>P6<sub>3</sub>/mmc</i>
22073	PtCl <sub>4</sub>	<i>Pa<math>\bar{3}</math></i>	100566	Cu <sub>4</sub> O <sub>3</sub>	<i>I4<sub>1</sub>/amd</i>
22177	V <sub>8</sub> C <sub>7</sub>	weakly; <i>P4<sub>3</sub>32</i> to <i>C2/m</i>	108192	V <sub>2</sub> C	<i>Pbcn</i>
22399	GeI <sub>4</sub>	<i>Pa<math>\bar{3}</math></i>	108594	Mn <sub>0.25</sub> TaS <sub>2</sub>	<i>P6<sub>3</sub>/mmc</i>
24314	TaSe <sub>2</sub>	<i>P6<sub>3</sub>/mmc</i>	108990	SrPdF <sub>4</sub>	<i>I4/mcm</i>
24317	TaSe <sub>2</sub>	<i>P6<sub>3</sub>/mmc</i>	109091	Li <sub>5</sub> GaO <sub>4</sub>	<i>P4<sub>1</sub>22</i>
26888	Bi <sub>1.5</sub> Cd <sub>0.5</sub> O <sub>2.75</sub>	<i>Ia<math>\bar{3}d</math></i>	153979	Ge <sub>1.00009</sub> V <sub>0.5495</sub>	weakly; <i>I4/mmm</i> to <i>Cmcm</i> ; V not determined
27035	D <sub>2</sub> S	<i>Pa<math>\bar{3}</math></i>	171908	Cu <sub>2</sub> FeS <sub>2</sub>	weakly; <i>F<math>\bar{4}3m</math></i> to <i>R3m</i>
27680	CaUO <sub>3</sub>	<i>Ia<math>\bar{3}</math></i>	171910	Cu <sub>2</sub> FeS <sub>2</sub>	weakly; <i>F<math>\bar{4}3m</math></i> to <i>R3m</i>
28387	Ta <sub>2</sub> O	weakly; <i>I<math>\bar{4}3m</math></i> to <i>Cm</i>	607870	AlGdGe	weakly; <i>I4<sub>1</sub>/amd</i> to <i>C2/m</i>
29311	FeNiS <sub>2</sub>	weakly; <i>Fm<math>\bar{3}m</math></i> to <i>R<math>\bar{3}m</math></i>	645370	NbSe <sub>2</sub>	<i>P6<sub>3</sub>/mmc</i>
31625	UCl <sub>6</sub>	<i>P<math>\bar{3}m1</math></i>	651092	TaS <sub>2</sub>	<i>P6<sub>3</sub>/mmc</i>
37226	TeI <sub>4</sub>	<i>I4<sub>1</sub>/amd</i> to <i>Imma</i>	655978	Mn <sub>2</sub> Y	weakly; <i>I4<sub>1</sub>/amd</i> to <i>C2/m</i>

42433	$V_2D_{0.85}$	weakly; $I4_1/amd$ to $C2/m$
51540	$Cr_{0.35}Mo_{0.65}$	$I4/mmm$ to $I4_1/amd$
51542	$Cr_{0.16}Mo_{0.84}$	$I4/mmm$ to $I4_1/amd$

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## S3. Supporting figures

**Figure S1**

Structures and diffraction equality of SrPdF<sub>4</sub> (Str. 2a&b). (a) Original structure of SrPdF<sub>4</sub> (Str. 2a); (b) The homometric structure of Str. 2a (Str. 2b). (c)(d)(e) Structure factors of Str. 2a. (f)(g)(h) Structure factors of Str. 2b. (i) Correspondence of phase angle and color in figure (c)~(h). For the reflections with  $l=2n$ , Str. 2a and Str. 2b are equal in both intensities and phases. For the reflections with  $l=2n+1$ , Str. 2a and Str. 2b are equal in intensities, but different in phases.

The difference between two structures is that the F-substructure is shifted by one quarter of the unit cell along the *c*-axis, so similar to Str. 1a and Str. 1b, the structure factor correlation between two structures can be explained by dividing each structure into two substructures. One substructure contains Sr and Pd atoms in  $P4/mmm$  symmetry with  $\frac{1}{\sqrt{2}}$  of the *a*-axis and half of the *c*-axis and the other contains F in  $I4/mcm$  symmetry.

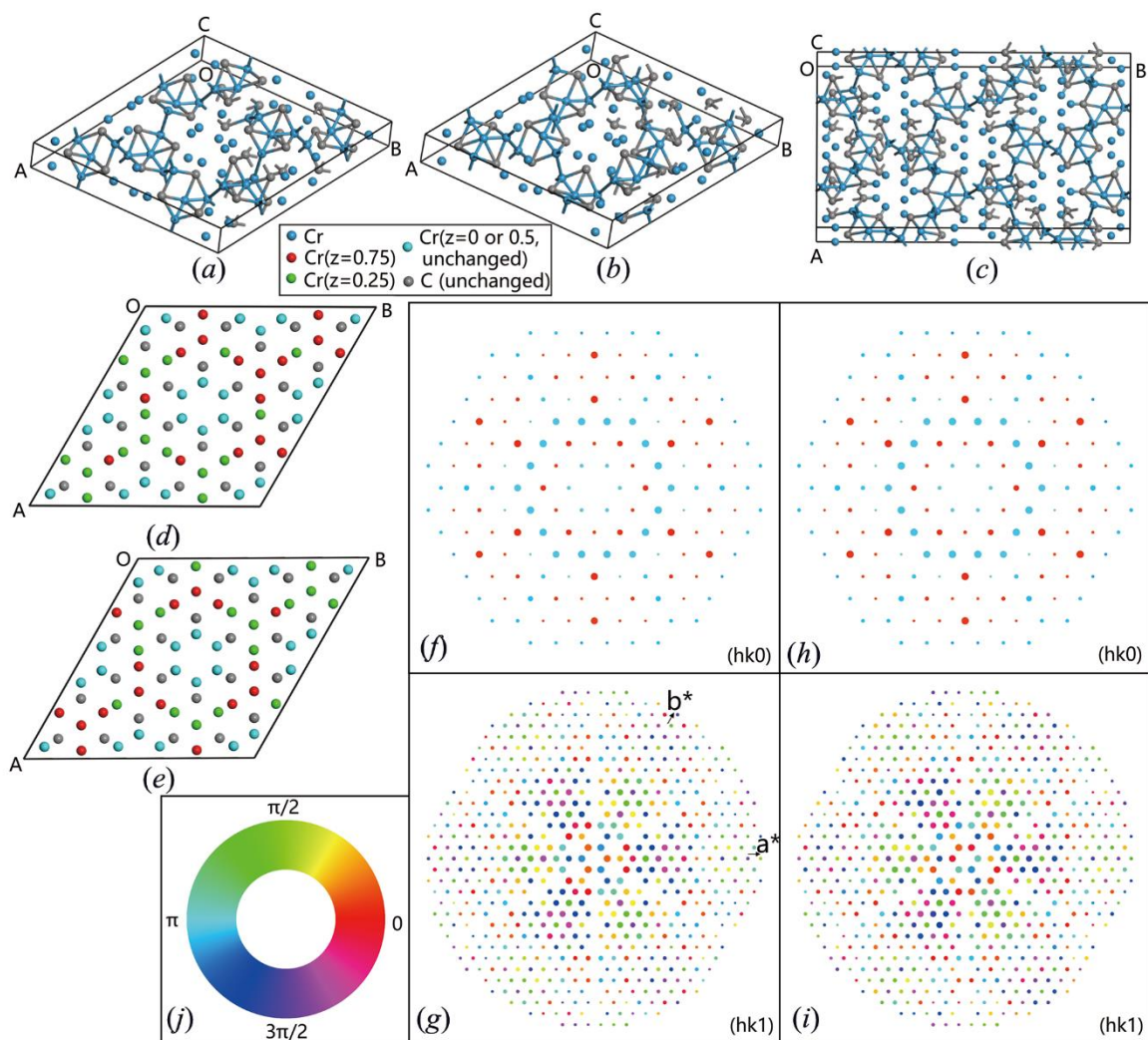
For the reflections  $h k 2n+1$ , as the *c*-axis of the Sr/Pd substructure is half of original cell, it doesn't contribute any intensities to these reflections, i.e. only the F-substructure contributes to  $h k 2n+1$  reflections, so these reflections have equal intensities but different phases between two structures due to the shift of one quarter of the unit cell for the F-substructure.

For the reflections  $h k 4n$ , as the translation vector of the F part is  $\frac{1}{4}c$ , the translation operation will not affect the reflections  $h k 4n$ , so these reflections will have same intensities and phases in both structures.

For the reflections  $h k 4n+2$ , the F-substructure doesn't contribute to these reflections, because the  $h k 4n+2$  reflections of the F-substructure are same as  $h k 2n+1$  reflections in a folded structure with half of the *c*-axis and the folded structure contains two same layers at  $z=0.25$  and  $z=0.75$ , causing the extinction of these reflections. As only Sr and Pd part contributes to  $h k 4n+2$  reflections, these reflections have same intensities and phases in both structures.

Although both structures are centrosymmetric, as the inversion center of Str. 2b is not at the origin of the cell, there are reflections with phases other than 0 and  $\pi$ .



**Figure S2**

Structures of  $\text{Cr}_7\text{C}_3$  (Str. 5a&b). (a) Original structure of  $\text{Cr}_7\text{C}_3$  (Str. 5a, No.52289 in ICSD); (b) The homometric structure of  $\text{Cr}_7\text{C}_3$  (Str. 5b). (c) The orthorhombic cell of Str. 5b. (d) Str. 5a along the  $c$ -axis; (e) Str. 5b along the  $c$ -axis. Blue and black spheres are Cr and C atoms respectively in (a), (b) and (c). Green and red spheres are Cr atoms with  $c=0.25$  and  $0.75$  respectively, and cyan spheres are Cr atoms with  $c=0$  or  $0.5$ , in (d) and (e). (f)(g) Structure factors of Str. 5a. (h)(i) Structure factors of Str. 5b. (j) Correspondence of phase angle and color in figure (f)~(i). The phases for the  $hk1$  reflections in Str. 5a have a 3-fold axis, while those of Str. 5b do not.

Note that if resonant scattering is considered, Str. 5a and Str. 5b can be distinguished by their slightly different intensities of Friedel's pairs. For example, the symmetry correlated reflections  $101$ ,  $011$ ,  $\bar{1}11$ ,  $\bar{1}01$ ,  $0\bar{1}1$ ,  $1\bar{1}1$  have higher intensities than their Friedel's pairs in Str. 5a, but in Str. 5b,  $10\bar{1}$ ,  $01\bar{1}$ ,  $\bar{1}11$ ,  $\bar{1}0\bar{1}$ ,  $0\bar{1}\bar{1}$ ,  $1\bar{1}\bar{1}$  reflections have higher intensities than others.