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Supporting information for article:

**Low-temperature phase transition and high-pressure phase stability
of 1*H*-pyrazole-1-carboxamide nitrate**

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Supporting information

Table S1 Experimental frequencies (cm^{-1}) taken from Raman and IR spectra collected at various temperature conditions and assignment of the bands.

$\nu_{\text{exp, R 300K}}$	$\nu_{\text{exp, IR 300K}}$	$\nu_{\text{exp, IR 112K}}$	$\nu_{\text{exp, IR 12K}}$	Assignment
	3322	3318	3315	νNH
3145	3143	3146	3149	νCH
3125		3127	3129	νCH
3105				νCH
3065				$\nu_{\text{as}}\text{NH}_2$
	1707	1712	1713	$\nu_{\text{as}}\text{CN}_{(\text{gua})}$
1656	1658	1662	1665	$\nu_{\text{as}}\text{CN}_{(\text{gua})}$
1563	1573	1581	1584	$\delta_s\text{NH}_2$
1537		1562	1561	$\delta_s\text{NH}_2$
			1419	$\nu_3\text{NO}_3$
1409	1409	1409	1407	$\nu_3\text{NO}_3$
1389	1386	1388	1387	$\nu_3\text{NO}_3$
			1316	νCN
1309	1309	1310	1310	νCN
		1299	1298	νCN
1228	1231	1234	1232	$\nu\text{CN} + \nu\text{CC}$
1212	1216	1217	1217	$\nu\text{NN} + \delta\text{CH}$

			1122	ρNH_2
1109	1099	1102	1102	$\delta\text{CH} + \rho\text{NH}_2$
1087				$\delta\text{CH} + \rho\text{NH}_2$
1065				$\nu_1\text{NO}_3$
	1064	1071	1074	δCH
1044	1041	1041	1040	δCH
946	940	943	945	$\delta\text{CH} + \delta\text{CNN}$
	919			ring breathing
908	910	912	912	ring breathing
881	886	888	888	γCH
	824	823	824	$\nu_2\text{NO}_3$
		789	792	$\nu_4\text{NO}_3$
	780	780	780	$\nu_4\text{NO}_3$
722	722	721	723	$\tau\text{NH}_2 + \gamma\text{NH}_2$
		706	708	$\gamma\text{CN}_{(\text{gua})}$
687	668	655	657	$\gamma\text{CN}_{(\text{gua})}$
634		597	600	γCCN
584		551	558	γCCN

Table S2 Hydrogen-bond geometry (\AA , $^\circ$) for (HPyCA) NO_3 at 300 and 100 K.

$D\text{---}H\cdots A$	$D\text{---}H$ (\AA)	$H\cdots A$ (\AA)	$D\cdots A$ (\AA)	$D\text{---}H\cdots A$ ($^\circ$)
300 K				
$\text{N3---H3A}\cdots\text{O1}^i$	0.86	1.92	2.7726 (17)	174.4
$\text{N3---H3B}\cdots\text{N2}^{ii}$	0.86	2.25	2.9887 (17)	144.5
$\text{N4---H4A}\cdots\text{O3}^i$	0.86	2.10	2.9506 (19)	170.6
$\text{N4---H4B}\cdots\text{O3}^{iii}$	0.86	2.42	3.2752 (18)	170.4
100 K				
$\text{N13---H13A}\cdots\text{O12}$	0.86	1.91	2.7628 (14)	172.3
$\text{N13---H13B}\cdots\text{N12}^{iv}$	0.86	2.23	2.9607 (17)	143.4
$\text{N14---H14A}\cdots\text{O11}$	0.86	2.06	2.9188 (15)	172.3
$\text{N14---H14B}\cdots\text{O22}$	0.86	2.71	3.5663 (15)	177.4
$\text{N14---H14B}\cdots\text{O23}$	0.86	2.19	2.7864 (16)	126.4
$\text{N23---H23A}\cdots\text{O21}$	0.86	1.93	2.7834 (13)	174.4

N23—H23B···N22 ^v	0.86	2.21	2.9493 (18)	143.8
N24—H24A···O22	0.86	2.07	2.9191 (13)	168.1
N24—H24B···O11 ^{vi}	0.86	2.19	3.0253 (15)	163.6

Symmetry codes: (i) $-x+1, y-1/2, -z+1/2$; (ii) $-x+2, -y+1, -z+1$; (iii) $x-1, y, z$; (iv) $-x-1, -y+1, -z$; (v) $-x+3, -y, -z+1$; (vi) $x, y-1, z$.

Table S3 Hydrogen-bond geometry (\AA , $^\circ$) for (HPyCA)NO₃ at 0.67 GPa and 0.89 GPa.

$D-H\cdots A$	$D-H$ (\AA)	$H\cdots A$ (\AA)	$D\cdots A$ (\AA)	$D-H\cdots A$ ($^\circ$)
0.67 GPa				
N3—H3A···O1 ⁱⁱⁱ	0.86	1.91	2.757 (5)	169.7
N3—H3B···N2 ^{iv}	0.86	2.22	2.956 (7)	142.8
N4—H4A···O3 ⁱⁱⁱ	0.86	2.11	2.966 (6)	173.2
N4—H4B···O3 ⁱⁱ	0.86	2.37	3.214 (6)	169.2
0.89 GPa				
N3—H3A···O1 ⁱⁱⁱ	0.86	1.92	2.769 (6)	168.4
N3—H3B···N2 ^{iv}	0.86	2.22	2.942 (8)	142.0
N4—H4A···O3 ⁱⁱⁱ	0.86	2.11	2.968 (6)	173.2
N4—H4B···O3 ⁱⁱ	0.86	2.36	3.209 (7)	169.0

Symmetry code(s): (i) $-x+1, y+1/2, -z+3/2$; (ii) $x+1, y, z$; (iii) $-x+1, y-1/2, -z+3/2$; (iv) $-x, -y+1, -z+1$.

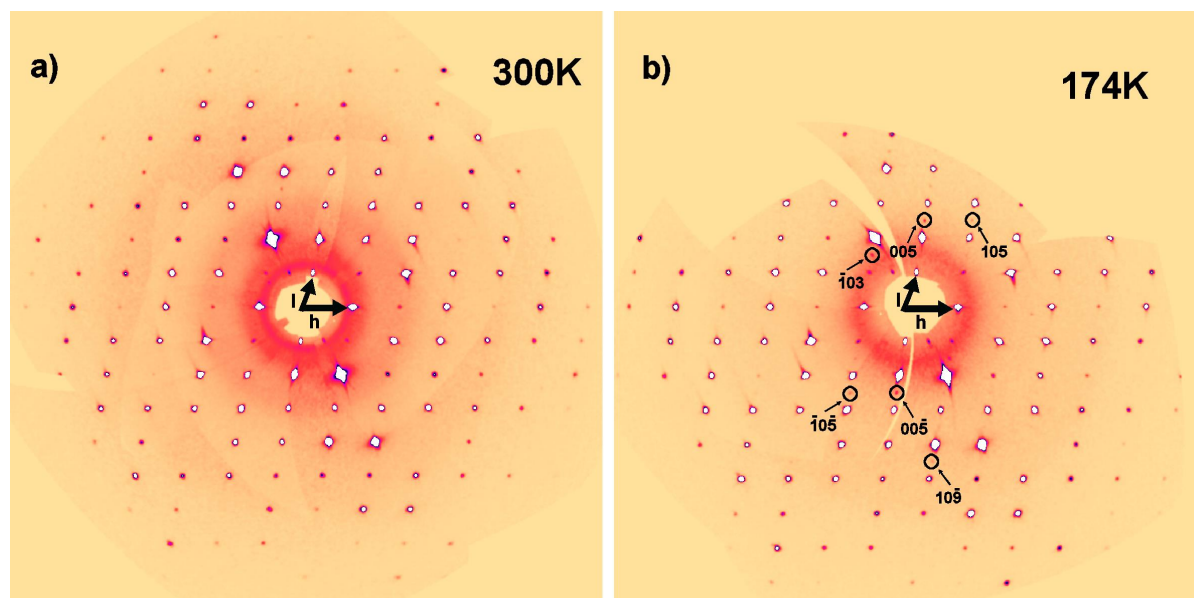


Figure S1 A comparison of diffraction pattern on the (h0l) plane obtained (a) at 300 K and (b) at 174 K. Additional l odd reflections indicate a disappearance of the *c* glide plane.

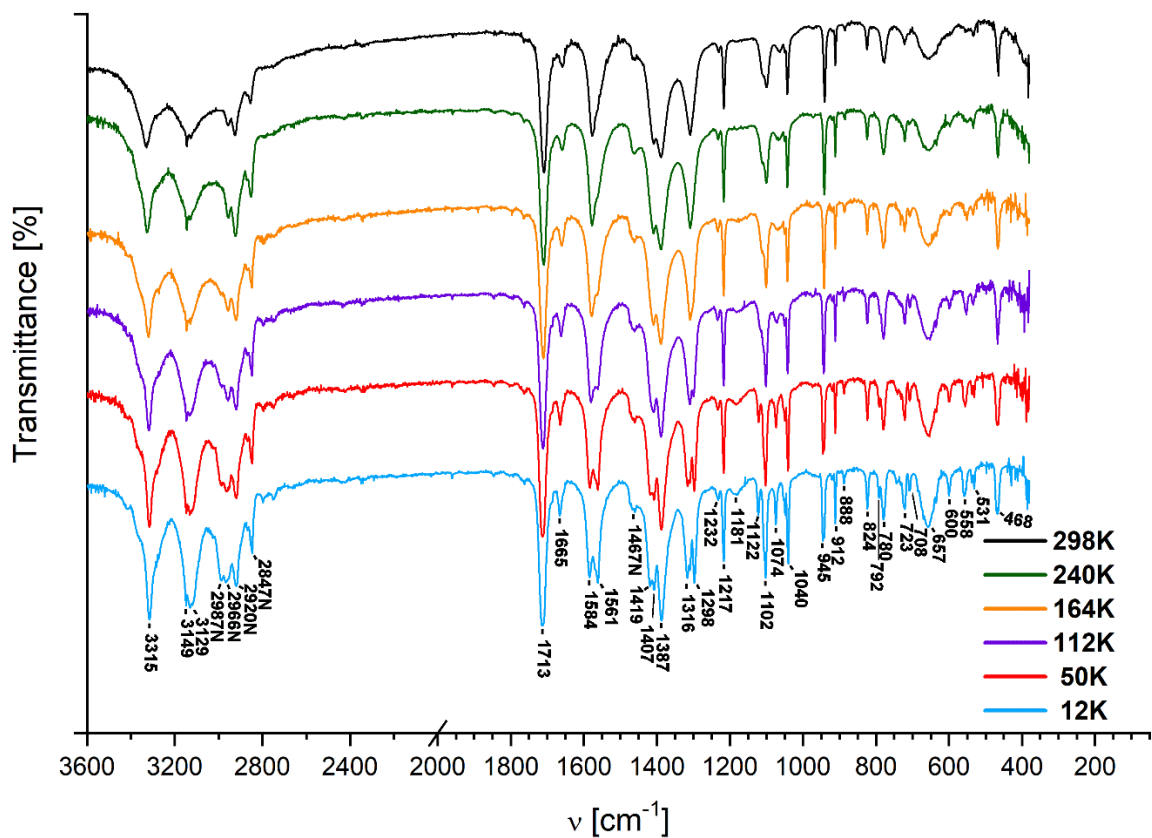


Figure S2 IR spectra collected for (HPyCA)NO₃ in the range of 300–12 K upon cooling the sample.

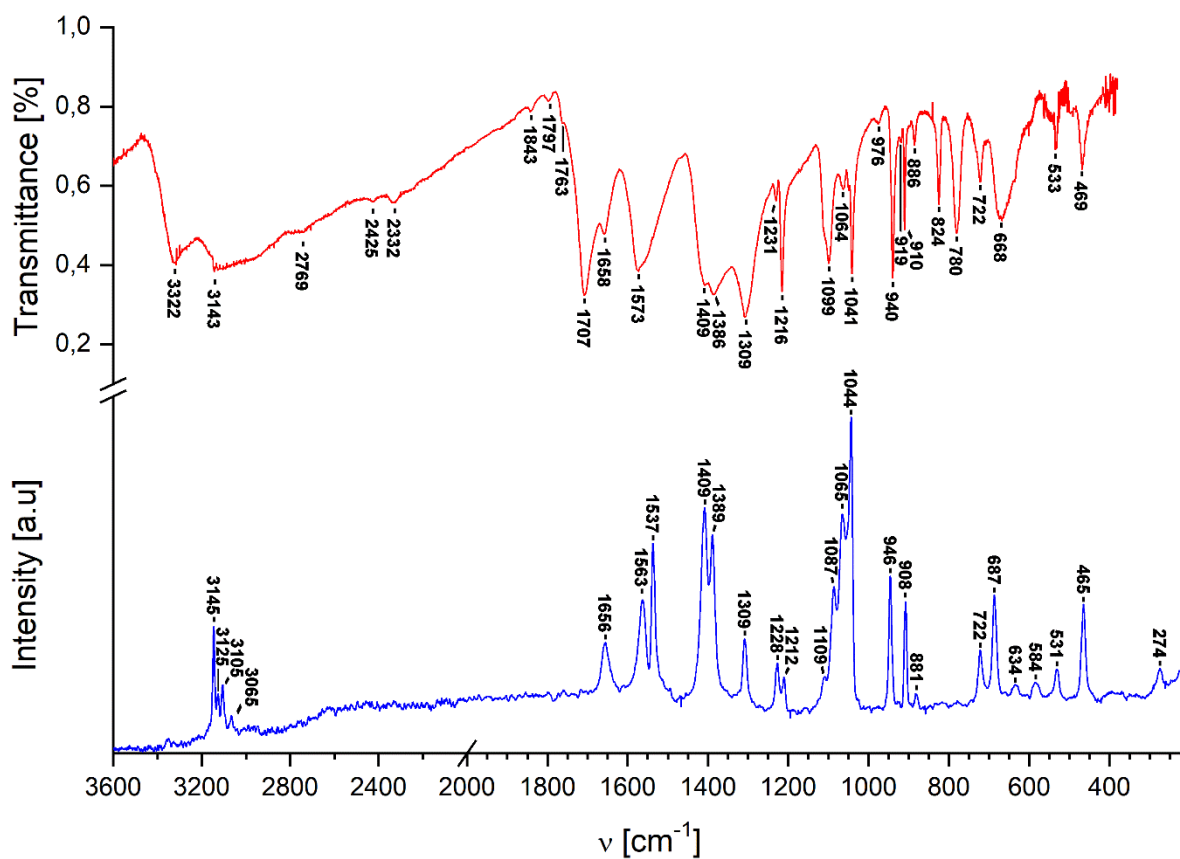
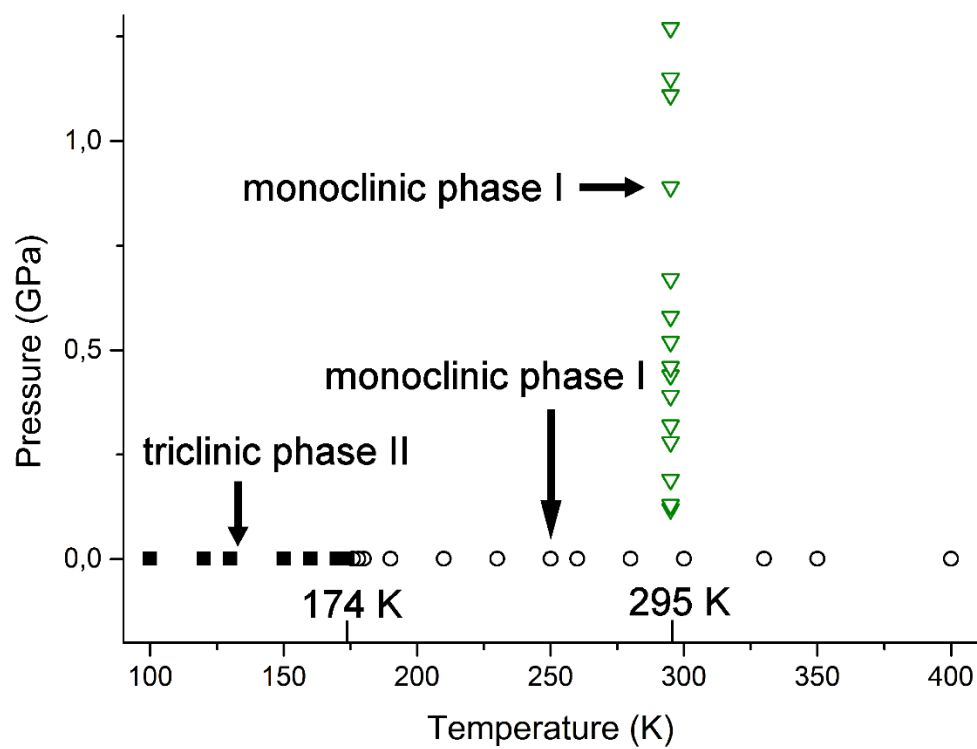


Figure S3 IR and Raman spectra collected for (HPyCA)NO₃ at ambient conditions.**Figure S4** Phase diagram for (HPyCA)NO₃