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Supporting information for article:

Octamolybdate isomers concomitance in metastable crystal structures isolated using homoleptic Co(II)/(III)-complexes as structuring directing templates

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**Figure S1** Photo of the solid obtained using 1:6 Co:Mo molar ratio at 160 °C, in which phases **1-5** are identified by SC-XRD takin into account different morphology and color.

**Figure S2** Graphical representation for octamolybdate isomers found in the crystal structures of **1**. Ball-and-stick model of the  $\alpha$ - (a) and  $\beta$ -isomer (c). Polyhedral models that display an evident difference between the geometry forms of the  $\alpha$ - (b) and  $\beta$ -isomer (d).



**Figure S3** (a) Ball-and-stick and (b) polyhedral representation for the *y*-isomer found in the crystal structures of **2**.



**Figure S4** *Pseudo*-tetrahedral site where NH₄<sup>+</sup> is hosted in the crystal structure of **2**. Formed hydrogen bonds agree well with the local site. N1A—H1AA····O22: 1.83 Å and 142.6°. N1A—H1AB···O4: 2.13 Å and 142.6°. N1A—H1AC···O101: 2.01 Å and 155.2°. N1A—H1AD···O2: 1.91 Å and 150.4°.



**Figure S5** Graphical representation of the crystal structure of **3**, showing channels parallel to the *a*-direction where most of the crystallization water molecules are accommodated.



**Figure S6** (a) Building block observed in the crystal structure of **4**, including labeling. (b) 1D-polymer running parallel to the *a*-axis found in **4**.



**Figure S7** Comparison between P-XRD pattern from solid II-T140 (obtained at 140 °C and Co:Mo ratio: 1.8) and those simulated from SC-XRD of  $\alpha$ - $\beta$  (1),  $\gamma$ - $\beta$  (2) and  $\beta_{cs}$ - $\beta_{gp}$  (3), Polymer 4, and Polymer 5 (VAFYOQ).



**Figure S8** Comparison between P-XRD pattern from solid II-T160 (obtained at 160 °C and Co:Mo ratio: 1.8) and II-T180 (obtained at 160 °C and Co:Mo ratio: 1.8) and those simulated from SC-XRD of **1-5**.



<sup>\*</sup>  $\beta_{cs}$  and  $\beta_{gp}$  correspond to the  $\beta$ -isomers located on a center of symmetry and on a general position in the crystal structure of **3**, respectively.

**Figure S9** Relative contributions of each intermolecular contact as calculated from HS analysis of each octamolybdate found in  $\alpha$ - $\beta$  (1),  $\gamma$ - $\beta$  (2), and  $\beta_{cp}$ - $\beta_{gp}$  (3).



**Figure S10** Relative contributions of intermolecular contacts as calculated from HS analysis of cations  $[Co(bpy)_3]^n$  (n= 2 or 3) found in the crystal structures of **1-3**.  $\Delta$  or  $\Lambda$  correspond to the configuration of the isomer observed in its respective asymmetry units for each crystal structure.



Figure S11 FT-IR for compound 1.



Figure S12 FT-IR for compound 2.



Figure S13 FT-IR for compound 3.



Figure S14 Thermal Gravimetric curves for compounds  $\{[NH_4^+][Co(bpy)_3][(\gamma-Mo_8O_{26})_{0.5}(\beta-Mo_8O_{26})_{0.5}]\} \cdot 4H_2O$  (2) and  $\{[Co(bpy)_3]_2[(\beta-Mo_8O_{26})_{0.5}(\beta-Mo_8O_{26})]\} \cdot 12H_2O$  (3).



Figure S15 Room temperature EPR spectra for 1 and 3.



**Figure S16** Comparison of P-XRD simulated from SC-XRD of **1** (black) *vs.* P-XRD of selected crystals of **1** (red).



**Figure S17** Comparison of P-XRD simulated from SC-XRD of **2** (black) *vs.* P-XRD of selected crystals of **2** (red).



**Figure S18** Comparison of P-XRD simulated from SC-XRD of **3** (black) vs. P-XRD of selected crystals of **3** (red).