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Supporting information for article:

Synthesis, structure and NLO properties of a new isostructural β -d-fructopyranose alkaline halide MOFs: a theoretical and experimental study

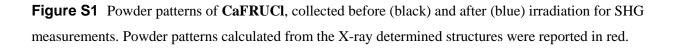
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Supporting information

Table S1Hydrogen bond distances for compounds CaFRUCI, CaFRUBr, SrFRUCI and SrFRUBr.X=Cl, Br.

D-H···A (Å)	CaFRUCI	CaFRUBr	SrFRUCI	SrFRUBr
$D \cdots A$ (Å)				
D-H···A (°)				
$H1WB{\cdots}X^i$	2.14	2.40	2.27	2.42
$O1W{\cdots}X^i$	3.0967(3)	3.241(3)	3.114(3)	3.251(4)
$O1W\text{-}H1WB\cdots X^i$	175	179	174	163
Н2…Х	2.30	2.45	2.26	2.46
02…X	3.1336(3)	3.293(4)	3.120(3)	3.277(5)
О2-Н2⋯Х	171	173	154	163
H1WA····O5 ⁱⁱ	1.97	2.03	2.00	2.09
O1W···O5 ⁱⁱ	2.8628(2)	2.880(6)	2.836(5)	2.828(7)
O1W-	160	178	167	156
$H1WA\cdots O5^{ii}$				
H2W···X	2.14	2.38	2.24	2.35
O2W···X	3.0650(3)	3.193(3)	3.052(3)	3.203(3)
O2W-H2W···X	168	163	154	180
H3····O4 ⁱⁱⁱ	1.97	2.00	2.06	2.01
O3····O4 ⁱⁱⁱ	2.7667(2)	2.780(5)	2.756(5)	2.777(7)
O3-H3····O4 ⁱⁱⁱ	161	152	131	152
$H4\cdots X^{iv}$	2.27	2.41	2.70	2.83
$O4\cdots X^{iv}$	3.0774(3)	3.225(3)	3.101(3)	3.243(4)
$O4\text{-}H4\cdots X^{\text{iii}}$	175	170	112	113
H5···O2W ^v	1.88	1.88	2.29	1.88
$O5 \cdots O2W^{v}$	2.6934(2)	2.695(4)	2.714(5)	2.692(5)
$O5-H5\cdots O2W^{v}$	169	172	111	171
Нб⋯Х	2.70	2.89	2.59	2.76
06…X	3.4686(3)	3.673(4)	3.392(4)	3.551(5)
Об-Н6⋯Х	162	161	158	162
Н6…О1	2.29	2.26	2.30	2.27
06…0	2.6942(2)	2.691(5)	2.690(4)	2.697(6)
O6-H6···O1	112	113	108	113
H6B····O4 ^{vi}	2.48	2.48	2.57	2.59
C6···O4 ^{vi}	3.3344(3)	3.314(6)	3.428(7)	3.448(8)
C6-H6B····O4 ^{vi}	148	145	147	147

Symmetry operations: i) x,-1+y,z; ii) -0.5+x,-0.5+y,0.5+z; iii) 0.5-x,-0.5+y,0.5-z; iv) -x,y,-z; v) 0.5+x, 0.5+y, 0.5+z; vi) 0.5-x,0.5+y,0.5-z



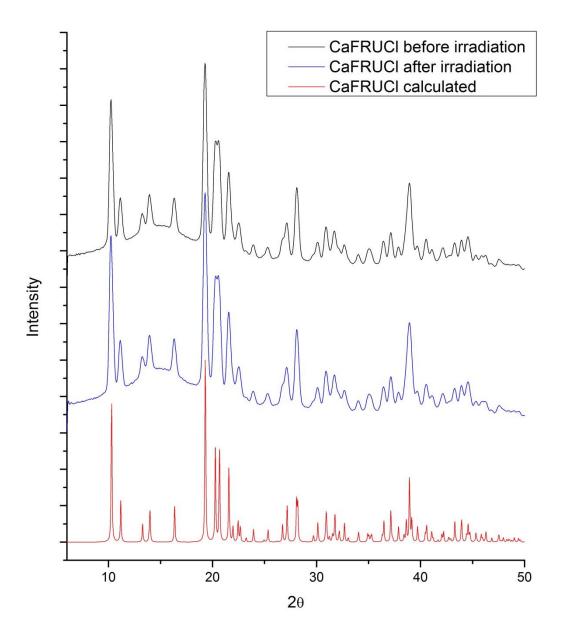


Figure S2 Powder patterns of **CaFRUBr** collected before (black) and after (blue) irradiation for SHG measurements. Powder patterns calculated from the X-ray determined structures were reported in red.

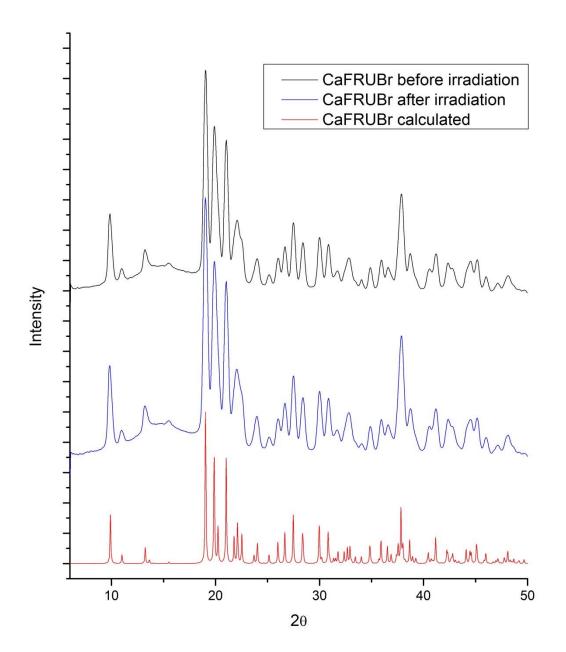


Figure S3 Powder patterns of **SrFRUCI**, collected before (black) and after (blue) irradiation for SHG measurements. Powder patterns calculated from the X-ray determined structures were reported in red.

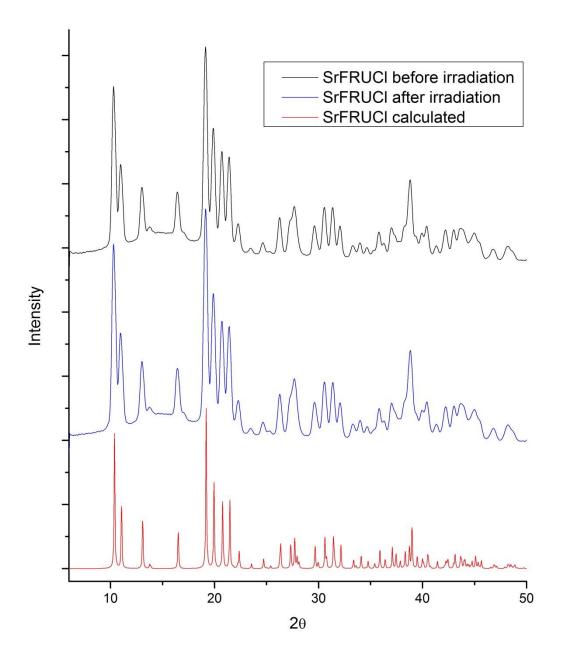


Figure S4 Powder patterns of **SrFRUBr** collected before (black) and after (blue) irradiation for SHG measurements. Powder patterns calculated from the X-ray determined structures were reported in red.

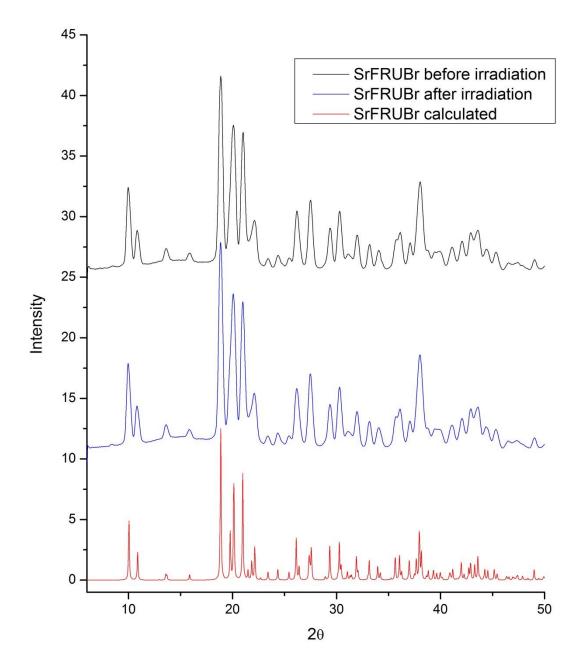


Figure S5 FT-infrared spectra of **CaFruCl** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.).

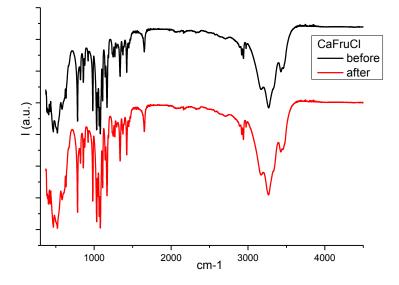
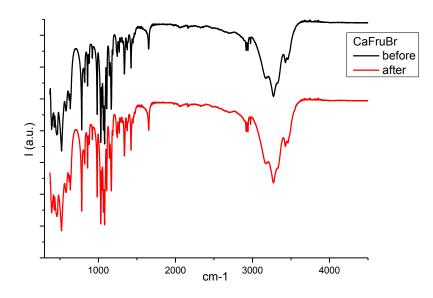
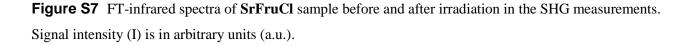


Figure S6 FT-infrared spectra of **CaFruBr** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.).





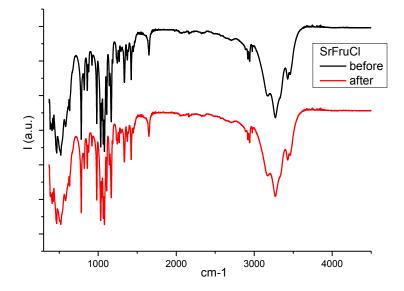


Figure S8 FT-infrared spectra of **SrFruBr** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.).

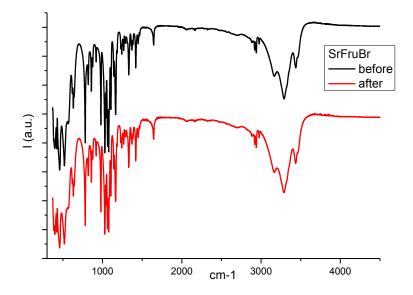


Figure S9 FT-Raman spectra of **CaFruCl** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.).

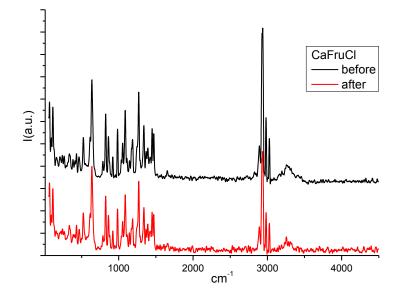


Figure S10FT-Raman spectra of **CaFruBr** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.)

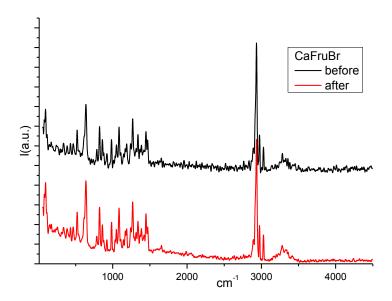


Figure S11FT-Raman spectra of **SrFruCl** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.).

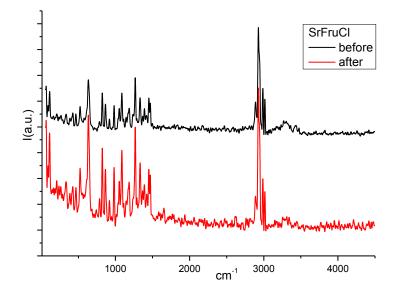


Figure S12FT-Raman spectra of **SrFruBr** sample before and after irradiation in the SHG measurements. Signal intensity (I) is in arbitrary units (a.u.).

