

# IUCrJ

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**Supporting information for article:**

**Crystal engineering of ionic cocrystals comprising Na/K salts of hesperetin with hesperetin molecules and solubility modulation**

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**Table S1** K/Na–O bond lengths observed in the structures of HESNAO and HESKOH.

Solid forms	Coordination number	Coordination Network	Distance Å
<b>Na–O bond lengths in HESNAO</b>			
<b>HESNA·H<sub>2</sub>O</b>	5	2D	2.3320(17)
			2.3792(15)
			2.6371(15)
			2.2610(16)
			2.3222(16)
<b>HESNAH</b>	6	1D	2.3361(15)
			2.4399(15)
			2.4039(13)
<b>HESNAH·2EtOH</b>	7	1D	2.4669(21)
			2.4557(19)
			2.5960(17)
			2.5267(24)
			2.2885(22)
			2.3117(22)
<b>K–O bond lengths in HESKOH</b>			
<b>HESK·3H<sub>2</sub>O</b>	6	2D	2.6450(19)
			3.0171(17)
			2.7970(17)
			2.7653(16)
			2.7911(17)
<b>HESKHE·2H<sub>2</sub>O</b>	8	1D	2.800(17)
			2.6541(22)
			2.8166(20)
<b>HESKHE·xEtOH</b>	6, 8	2D	2.8050(19)
			2.8218(23)
			2.8237(85)
			2.8647(37)
			2.7121(36)
<b>HESKHE·xMeOH</b>	6, 8	2D	2.6676(30)
			2.6712(26)
			2.6637(34)
			2.8430(27)
<b>HESKHE·xMeOH</b>	6, 8	2D	2.8548(19)
			2.6593(22)

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2.6587(16)  
2.6813(22)  
2.7434(21)  
2.8714(24)

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\*one K–O bond length in **HESKHE**·**xMeOH** is not included due to oxygen atom disorder.

**Table S2** Frequency of  $\text{PhOH}\cdots\text{PhO}^-$ ,  $\text{PhOH}\cdots\text{PhOH}$ , and  $\text{PhOH}\cdots\text{O}=\text{C}$  supramolecular synthons appearing in related structures deposited in CSD. (ConQuest version 2020.3.0 with Nov. 2022 update, search parameters: 3D coordinates present; only organics; R factor  $\leq 0.05$ , no disorder and single-crystal structure only).

**REFCODEs of structures deposited in CSD that contain  $\text{PhOH}$ ,  $\text{PhO}^-$ , and  $\text{O}=\text{C}$  moieties with no  $\text{COOH}$  or  $\text{COO}^-$  moieties**

COLHER	DAVLER	DUCWUV	FILGUF	HEGHEH	HIVBET
JICKIS	JYXEX	LUZRAZ02	NICCEJ	SAFMOF	UNIHUU
UYOYIR	WIHFAW	IWUGEP	XAZHIT	XAZHOZ	

$\text{PhOH}\cdots\text{PhO}^-$ ,  $\text{PhOH}\cdots\text{PhOH}$ ,  $\text{PhOH}\cdots\text{O}=\text{C}$ , and other supramolecular synthons occur in 58.8%, 11.8%, 23.5%, and 29.4% of structures, respectively.

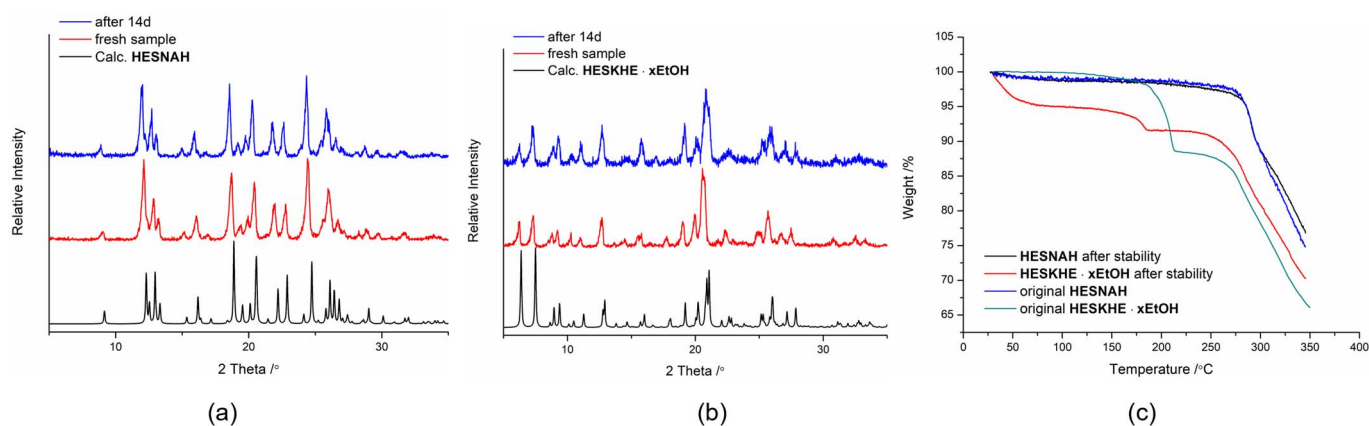
**Table S3** HES entries in the CSD where infinite chains of HES molecules form via either  $\text{PhOH}\cdots\text{PhOH}$  or  $\text{PhOH}\cdots\text{O}=\text{C}$  supramolecular synthons.

HES Chains	
$\text{PhOH}\cdots\text{PhOH}$	$\text{PhOH}\cdots\text{O}=\text{C}$
FOYTOC	IJIWAC
LAVLEC	LAVLAY
RUWHEX	LAVLIG
RUWHIB	YEHROS
CANPOA	

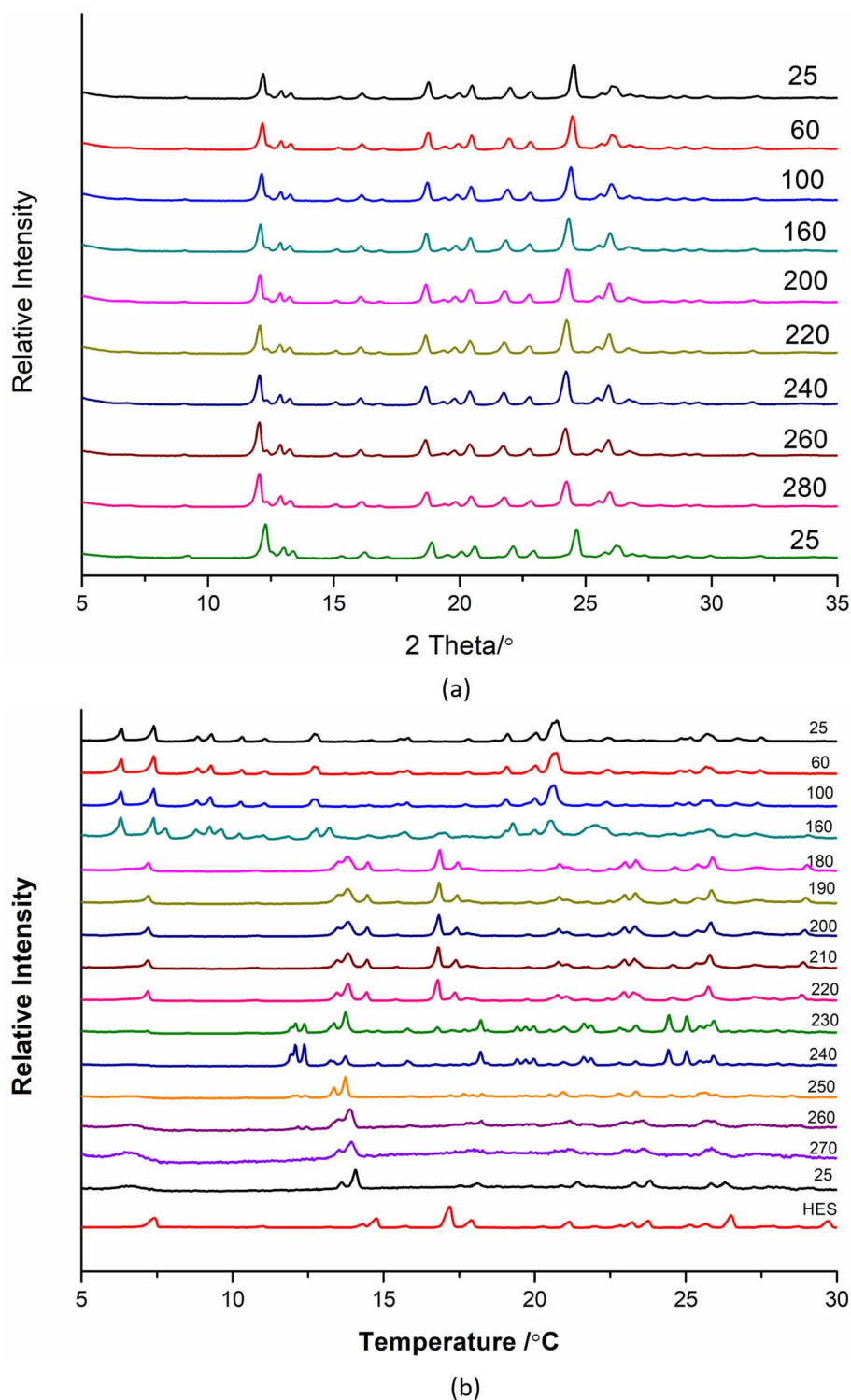
**Table S4** Dihedral angles between benzopyrone rings and phenolic rings of HES moieties extracted from structures of HES reported herein.

Solid forms	Dihedral angles/ $^{\circ}$
<b>HESNA·H<sub>2</sub>O</b>	75.11
<b>HESNAH·2EtOH</b>	86.94, 82.01
<b>HESK·3H<sub>2</sub>O</b>	55.83
<b>HESKHE·xEtOH</b>	81.47, 89.13
<b>HESKHE·xMeOH</b>	88.67
<b>HESNAH</b>	89.89*
<b>HESKHE·2H<sub>2</sub>O</b>	86.58*

\*folded conformation

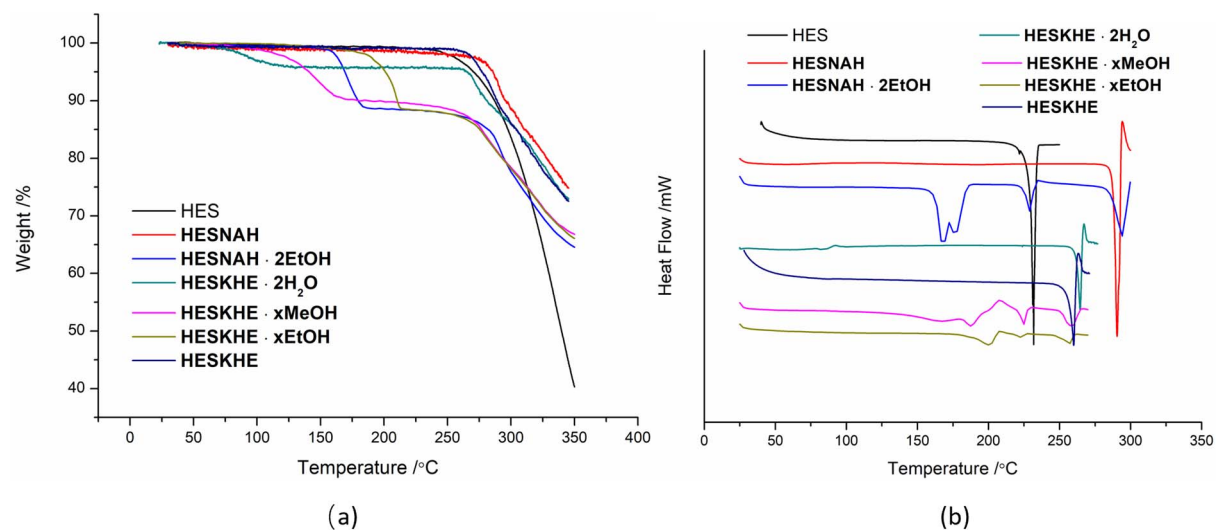


**Figure S1** (a) (b) PXR D patterns and (c) TGA plots of **HESNAH** and **HESKHE·xEtOH** after 14 days in accelerated stability test (there is peak shift due to thermal expansion as ICCs were collected at low temperature).

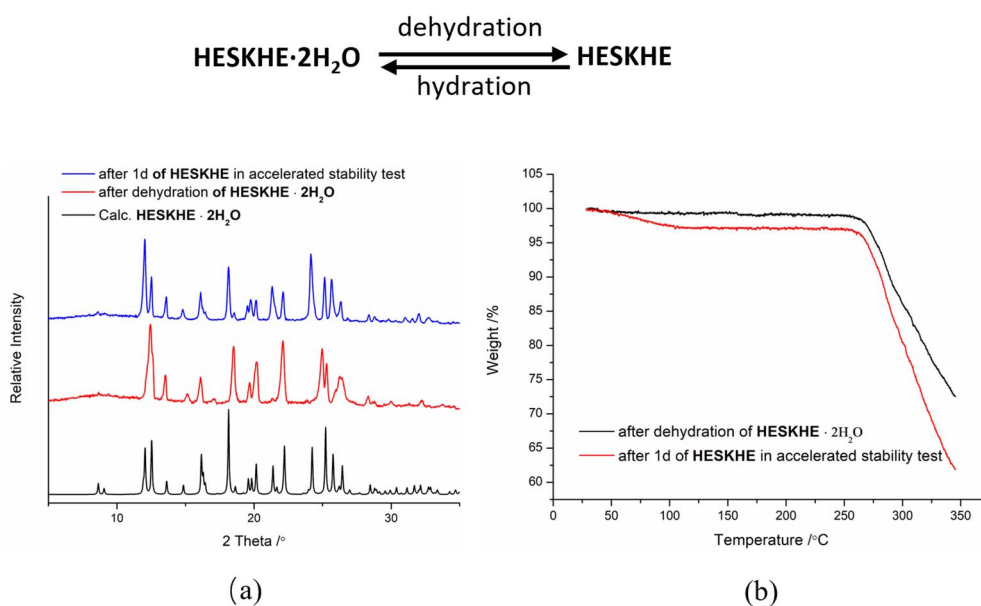


**Figure S2** Variable temperature PXRD of (a) **HESNAH** from 25 °C to 280 °C and back to 25 °C which retained the structure until 280 °C; and (b) **HESKHE·xEtOH** from 25 °C to 270 °C and back to 25 °C. **HESKHE·xEtOH** remained unchanged before desolvation at *ca.* 160 °C confirmed by TGA in Figure S3, and then changed to a phase displaying similar PXRD to HES until 220 °C (likely caused by dissociation to HES which is supported by DSC displaying HES melting peak at similar temperature) followed by another phase change during 230 - 240 °C and then decomposition after 250

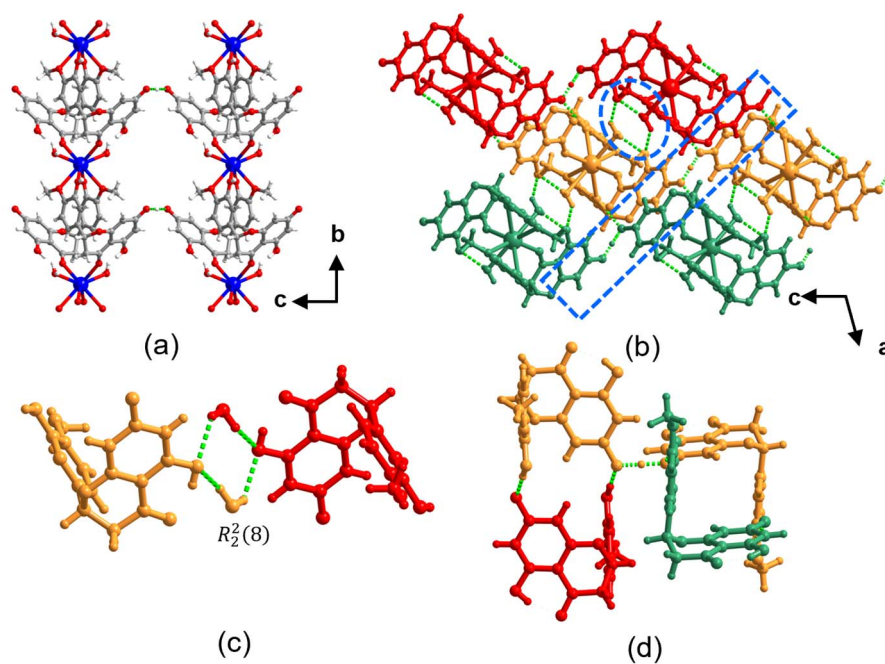
°C as black decomposed compound was observed after test (there is peak shift due to thermal expansion as ICC samples were measured at different temperature).



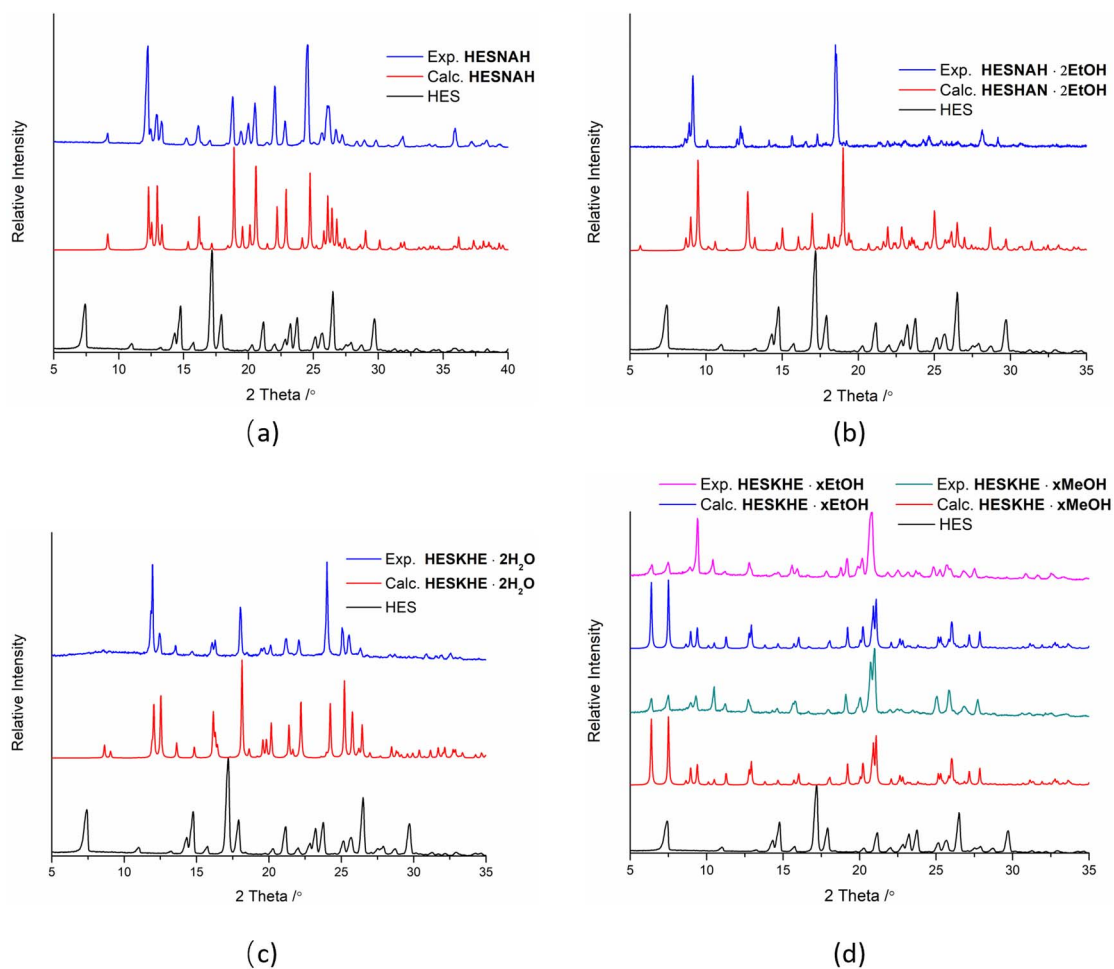
**Figure S3** TGA (a) and DSC (b) plots of HES and ICCs. The weight losses in **HESKHE·xMeOH** and **HESKHE·xEtOH** samples calculated from TGA curves are 9.8% and 11.6%, respectively, which indicates 1.75~2.2 solvent molecules in both structure formulas.



**Figure S4** (a) PXR patterns and (b) TGA plot of dehydration of **HESKHE·2H<sub>2</sub>O** and hydration of **HESKHE** exposed to 75% RH for one day.

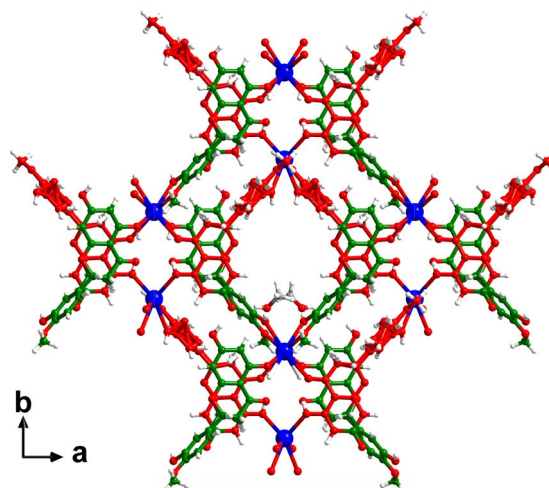


**Figure S5** Crystal structure of HESKHE·2H<sub>2</sub>O. (a) Coordination polymeric chains give rise to 2D network via [PhO···H···PhO<sup>-</sup>] H-bonds. (b) Packing of layers (represented in different colors) causing the formation of (c)  $R_2^2(8)$  H-bonded motif between HES moieties and H<sub>2</sub>O molecules, and (d) a chain of cyclic dimers of HES moieties (Motif I).

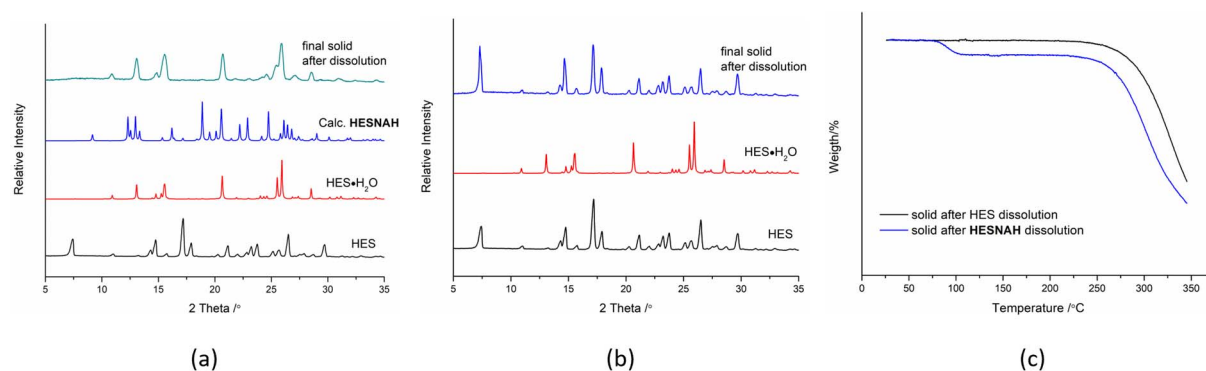


**Figure S6** Calculated PXRD generated from single crystal structures of (a) **HESNAH**, (b) **HESNAH·2EtOH**, (c) **HESKHE·2H<sub>2</sub>O** and (d) **HESKHE·xMeOH** and **HESKHE·xEtOH** compared with experimental PXRD (there is peak shift due to thermal expansion as ICCs were collected at low temperature).





**Figure S7** ‘Double-wall’ square grid filled with MeOH molecules in the crystal structure of **HESKHE·xMeOH**. HES molecules and HES<sup>-</sup> anions are colored green and red, respectively.



**Figure S8** (a) (b) PXR patterns and (c) TGA plots of final solid forms after dissolution test of **HESNAH** and **HES** at PBS buffer at pH6.8.