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Supporting information for article:

Structural flexibility of halogen bonds showed in a single-crystal-tosingle-crystal [2+2] photodimerization

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Supplementary information (SI)

Single-crystal-to-single-crystal [2+2] photodimerization mediated by the N \cdots I halogen bond

- S1. Syntheses
- S2. ¹H NMR Spectral Data
- S3. Powder and Single-Crystal X-ray Diffraction Data
- S4. Halogen Bond Metrics
- S5. CSD Data and IsoStar Video Files
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S1. Syntheses and Photoreactions

I4F16Cb was synthesized according to the literature. **bpe** (SIGMA) and solvents were commercially available and used without further purification (FISHER). Cocrystals of (**I4F16Cb**)·2(**bpe**) as colorless plates were obtained *via* the addition of a solution of **I4F16Cb** (23 mg, 0.02 mmol) in 1.5 mL of acetonitrile into a solution of **bpe** (11 mg, 0.04 mmol, 1:2 molar ratio) in 1.5 mL of acetonitrile. Single crystals suitable for single-crystal X-ray diffraction studies were obtained after one day. Photoreactions of (**I4F16Cb**)·2(**bpe**) were conducted on a glass plate using UV-radiation from a 450W medium-pressure mercury lamp inside an ACE Glass photochemistry cabinet. ¹H NMR spectra were recorded using a Bruker AVANCE-300 NMR spectrometer operating at 300 MHz using DMSO-*d*₆ as NMR solvent. All data were processed with MestReNova suite of software programs.

S2. ¹H NMR Spectral Data.

¹H-NMR spectra were recorded using a Bruker AVANCE-300 NMR spectrometer operating at 300 MHz using DMSO-d₆ as NMR solvent. All data were processed with MestReNova 10.0 suite of software programs.

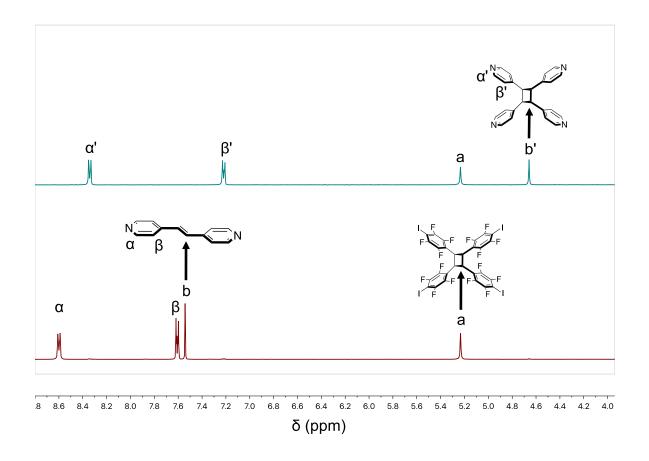


Figure S1. ¹H NMR spectra of (**I**₄**F**₁₆**cb**)·2(**bpe**) before UV irradiation (0 h) and (top) after UV irradiation (30 h) (DMSO-d₆).

S3. X-ray Diffraction Data

Powder X-ray Diffraction

PXRD data were collected from samples mounted on glass slides by a Siemens D5000 X-ray diffractometer using CuK α 1 radiation (λ = 1.54056 Å) (scan type: locked coupled; scan mode: continuous; step size: 0.02°).

Single-crystal X-ray Diffraction

Single crystal data for ($\mathbf{I_4F_{16}cb}$)·2(\mathbf{bpe}) were collected with a Bruker APEX II Kappa diffractometer equipped with an Oxford Cryostream low temperature device using MoKa radiation ($\mathbf{l} = 0.71073 \text{ Å}$) after having been secured to Mitegen magnetic mounts using Paratone oil. Data collection strategies to ensure maximum data redundancy and completeness were calculated using Apex2.² All calculation dealing with data collection, initial indexing, frame integration, Lorentz- polarization corrections and final cell parameter were again carried out by Apex2.² Single-crystal diffraction data for ($\mathbf{I_4F_{16}cb}$)·(\mathbf{tpcb}) were collected on a Nonius Kappa CCD single-crystal X-ray diffractometer at room temperatures using MoK α radiation (λ = 0.71073 Å). Data collection, cell refinement, and data reduction were performed using Collect³ and HKL Scalepack/Denzo,⁴ respectively. All structures were solved via direct methods using ShelXT⁵ and refined using ShelXL⁵ in the Olex2⁶ graphical user interface. All nonhydrogen atoms were identified from the difference Fourier map within several refinement steps. Hydrogen atoms associated with carbon atoms were refined in geometrically constrained positions with isotropic thermal parameter $U_{iso}(\mathbf{H}) = 1.2 \ U_{eq}(\mathbf{C}_{CH})$.

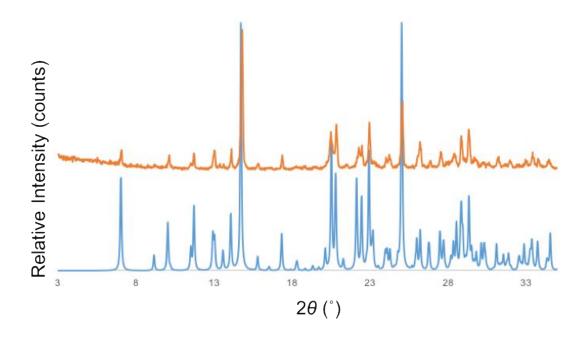


Figure S2. Powder X-ray diffraction of powder ($\mathbf{I_4F_{16}cb}$)·2(\mathbf{bpe}) (top) and simulated pattern from single-crystal X-ray diffraction data.

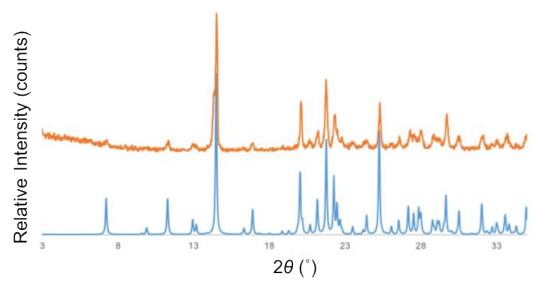


Figure S3. Powder X-ray diffraction of $(\mathbf{I_4F_{16}cb}) \cdot (\mathbf{tpcb})$ (top) and simulated pattern from single crystal X-ray diffraction data.

Table S1. Crystal data and structure refinement for $(\mathbf{I_4F_{16}cb})\cdot 2(\mathbf{bpe})$.

CCDC deposition number 1521899

Empirical formula $C_{52}H_{24}F_{16}I_4N_4$

Formula weight 1516.35 Temperature/K 298.15

Crystal system monoclinic

Space group $P2_1/c$

a/Å 13.0184(13)

b/Å 9.7153(10) c/Å 19.986(2)

α/° 90

 β /° 104.649(5)

γ/° 90

Volume/ $Å^3$ 2445.6(4)

Z 2

 $\rho_{calc} g/cm^3$ 2.059 μ/mm^{-1} 2.653

F(000) 1440.0

Crystal size/mm³ $0.4 \times 0.19 \times 0.04$

Radiation $MoK\alpha (\lambda = 0.71073)$

 2Θ range for data collection/° 5.926 to 52.87

Index ranges $-16 \le h \le 16, -12 \le k \le 11, -25 \le l \le 23$

Reflections collected 52897

Independent reflections 5012 [$R_{int} = 0.0305$, $R_{sigma} = 0.0162$]

Data/restraints/parameters 5012/704/639

Goodness-of-fit on F² 1.085

Final R indexes [I>= 2σ (I)] $R_1 = 0.0198$, $wR_2 = 0.0408$

Final R indexes [all data] $R_1 = 0.0310$, $wR_2 = 0.0460$

Largest diff. peak/hole / e Å⁻³ 0.45/-0.41

Table S2. Crystal data and structure refinement for (I₄F₁₆cb)·(tpcb).

CCDC deposition number 1521900

Empirical formula $C_{52}H_{24}F_{16}I_4N_4$

Formula weight 1516.35 Temperature/K 298.15

Crystal system monoclinic

Space group $P2_1/c$

a/Å 12.9647(13) b/Å 10.2219(10) c/Å 19.519(10)

 α /°

 β /° 109.387(5)

γ/° 90

Volume/Å³ 2440.0(13)

Z 2

 $\begin{array}{lll} \rho_{calc} g/cm^3 & 2.064 \\ \mu/mm^{-1} & 2.659 \\ F(000) & 1440.0 \end{array}$

 $Crystal~size/mm^3 \\ 0.14 \times 0.12 \times 0.05$

Radiation $MoK\alpha (\lambda = 0.71073)$

 2Θ range for data collection/° 5.956 to 50.786

Index ranges $-11 \le h \le 15, -12 \le k \le 12, -23 \le l \le 23$

Reflections collected 12227

Independent reflections 4464 [$R_{int} = 0.0670$, $R_{sigma} = 0.0580$]

Data/restraints/parameters 4464/473/548

Goodness-of-fit on F² 1.043

Final R indexes [I>=2 σ (I)] $R_1 = 0.0330$, $wR_2 = 0.0548$

Final R indexes [all data] $R_1 = 0.0626$, $wR_2 = 0.0627$

Largest diff. peak/hole / e Å⁻³ 0.50/-0.52

S4. Halogen Bond Metrics

Table S3. Halogen-bond bending metrics: (**I**₄**F**₁₆**cb**)·2(**bpe**), (**I**₄**F**₁₆**cb**)·(**tpcb**), and CSD ranges.

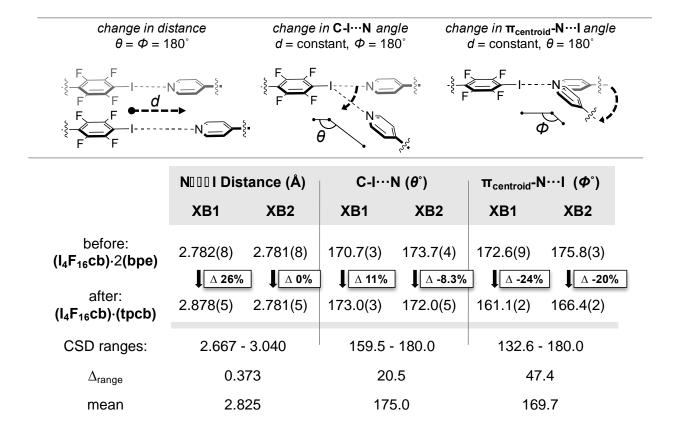


Table S4. Halogen Bond Metrics

	halogen bond	N····I Distance (Å)	C-I···N (°)	π _{centroid} -N···I (°)
	$C1A\text{-}I1A\cdots N1A\text{-}\pi_{centroid}$	2.782(8)	170.7(3)	172.6(9)
$(I_4F_{16}cb)\cdot 2(bpe)$	$C1A\text{-}I1A\cdots N1B\text{-}\pi_{centroid}$	2.80(3)	167(1)	167.1(2)
	$C1B\text{-}I1B\cdots N1A\text{-}\pi_{centroid}$	2.821(9)	164.6(4)	173.0(8)
	$C1B\text{-}I1B\cdots N1B\text{-}\pi_{centroid}$	2.84(3)	161(1)	161.9(2)
	C12A-I2A · · · N2A- $\pi_{centroid}$	2.781(8)	173.7(4)	175.8(3)
	$\text{C12A-I2A} \cdot \cdot \cdot \text{N2B-} \pi_{\text{centroid}}$	2.80(2)	178.3(9)	162.6(1)
	$\text{C12B-I2B} \cdot \cdot \cdot \text{N2A-} \pi_{\text{centroid}}$	2.908(9)	171.3(5)	176.7(5)
$(I_4F_{16}cb)\cdot(tpcb)$	C12B-I2B···N2B-π _{centroid}	2.94(2)	166.3(7)	160.1(1)
	$C1A\text{-}I1A\cdots N1\text{-}\pi_{centroid}$	2.878(5)	173.0(3)	161.1(2)
	$C1B\text{-}I1B\cdots N1\text{-}\pi_{centroid}$	2.84(1)	177(1)	158.9(4)
	$C12A\text{-}I2A\cdots N2\text{-}\pi_{centroid}$	2.781(5)	172.0(5)	166.4(2)
	C12B-I2B····N2-π _{centroid}	2.80(1)	166(1)	167.4(4)

S5. CSD Data and IsoStar Video Files

IsoStar Video Files

Video S1. IsoStar scatterplot of structures from the CSD (wireframe) and XB1 from (I₄F₁₆cb)·2(bpe) (gray) and (I4F16cb)·(tpcb) (black). Highest occupied XB1 in one quadrant.

Video S2. IsoStar scatterplot of structures from the CSD (wireframe) and XB2 from (I₄F₁₆cb)·2(bpe) (gray) and $(I_4F_{16}cb)\cdot(tpcb)$ (black). Highest occupied XB2 in one quadrant.

CSD data

Search Overview

Database(s):

CSD version 5.38 updates (Nov 2016) CSD version 5.38 (November 2016) CSD version 5.38 (November 2016)

Restriction Info: No refcode restrictions applied

Filters: 3D coordinates determined R factor <= 0.075 No errors

Not polymeric No powder structures No ions

Only Organics

Percentage Completed: 100%

Single query used. Search found structures that:

match Query 1 **Query 1**

S6. References

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