

## Refinements on electron diffraction data of $\beta$ -glycine in MoPro: A quest for improved structure model

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### Supplementary Information

**Table S1.** Electron diffraction scattering factors  $f(s)$  as a function of  $s = \sin\theta/\lambda$ ,  
Coefficients A1 A2 A3 A4 B1 B2 B3 B4 published by Peng, (1999), in *Micron*. **30**, 625–648,  
where  $f(s) = \sum_{i=1,4} A_i * \exp(-B_i*s^2)$

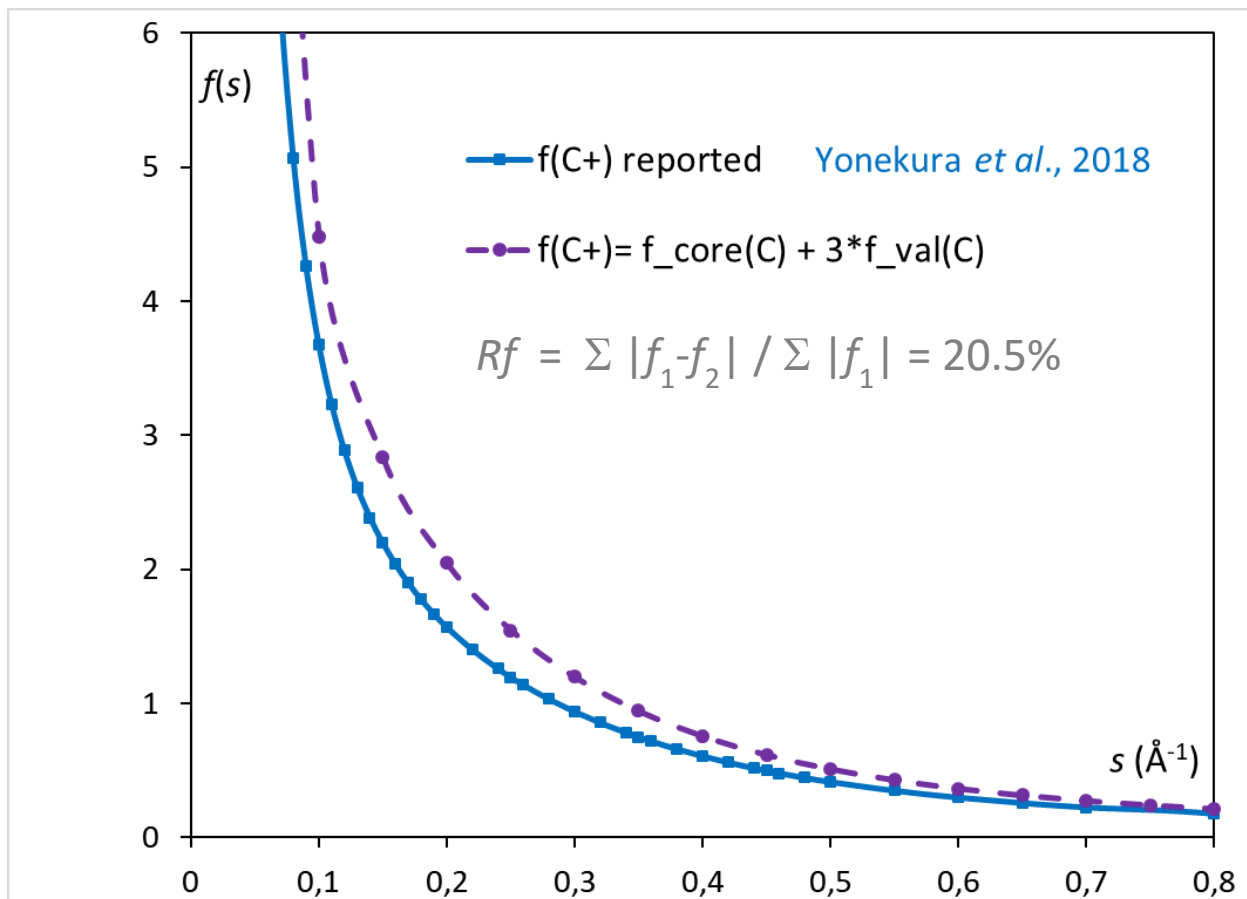
CHEM H										
CORE GAUS										
	.0367	.1269	.2360	.1290	.5608	3.7913	13.5557	37.7229		
CHEM C										
CORE GAUS										
	.1361	.5482	1.2266	.5971	.3731	3.2814	13.0456	41.0202		
CHEM N										
CORE GAUS										
	.1372	.5344	1.0862	.4547	.3287	2.6733	10.3165	32.7631		
CHEM O										
CORE GAUS										
	.1433	.5103	.9370	.3923	.3055	2.2683	8.2625	25.6645		

**Table S2.** Electron diffraction scattering factors  $f(s)$  as a function of  $s = \sin \theta / \lambda$   
Coefficients A1 A2 A3 A4 A5 B1 B2 B3 B4 B5 as published in the International Tables of  
Crystallography (Volume C ,Tables 4.2.6.8 and 6.1.1.4). Here,  $f(s) = \sum_{i=1,5} A_i * \exp(-B_i*s^2)$

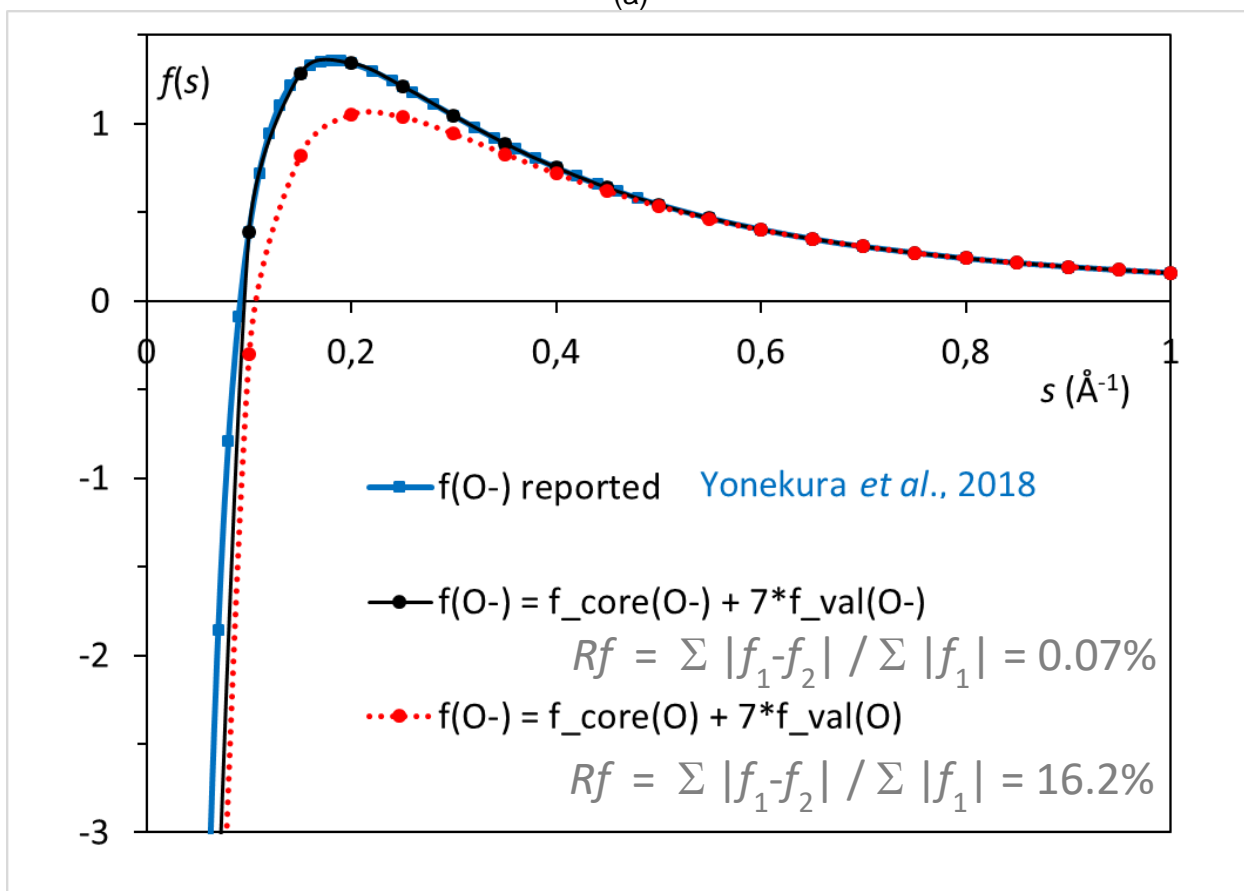
CHEM H											
CORE GAUS											
	0.0088	0.0449	0.1481	0.2356	0.0914	0.1152	1.0867	4.9755	16.5591	43.2743	
CHEM C											
CORE GAUS											
	0.0489	0.2091	0.7537	1.142	0.3555	0.114	1.0825	5.4281	17.8811	51.1341	
CHEM N											
CORE GAUS											
	0.0267	0.1328	0.5301	1.102	0.4215	0.0541	0.5165	2.8207	10.6297	34.3764	
CHEM O											
CORE GAUS											
	0.0365	0.1729	0.5805	0.8814	0.3121	0.0652	0.6184	2.9449	9.6298	28.2194	

**Table S3:** Bond distances (Å) derived from IAM and TAAM refinement of  $\beta$ -Glycine

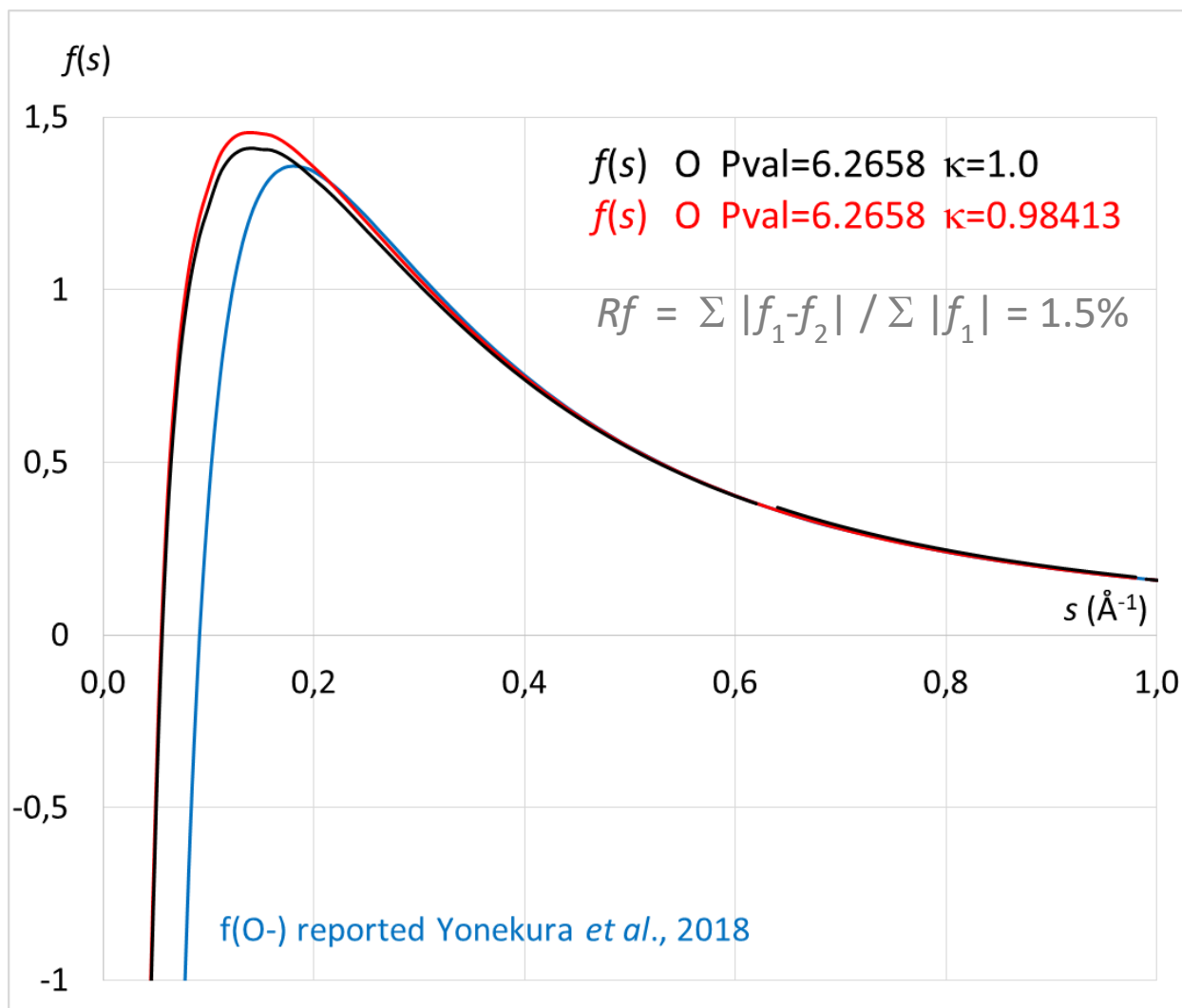
Method	Bond	IAM	IAM_Peng	IAM_IT	TAAM_ELMAM2	TAAM_UBDB
Program	Type	SHELX	MoPro	MoPro	MoPro	MoPro
1	C1-C2	1.58(6)	1.58(6)	1.58(6)	1.58(6)	1.58(9)
2	C2-O1	1.32(3)	1.32(6)	1.32(6)	1.32(8)	1.32(1)
3	C2-O2	1.32(2)	1.32(1)	1.32(1)	1.32(3)	1.32(2)
4	N1-C1	1.54(4)	1.54(7)	1.53(7)	1.54(6)	1.54(5)
5	N1-H1A	1.21(4)	1.22(4)	1.22(4)	1.19(2)	1.19(2)
6	N1-H1B	1.11(7)	1.09(6)	1.10(6)	1.07(7)	1.10(6)
7	N1-H1C	1.23(5)	1.22(3)	1.22(3)	1.18(3)	1.17(3)
8	C1-H1D	1.16(5)	1.17(4)	1.17(3)	1.14(3)	1.14(3)
9	C1-H1E	1.32(6)	1.31(5)	1.31(5)	1.20(5)	1.17(5)



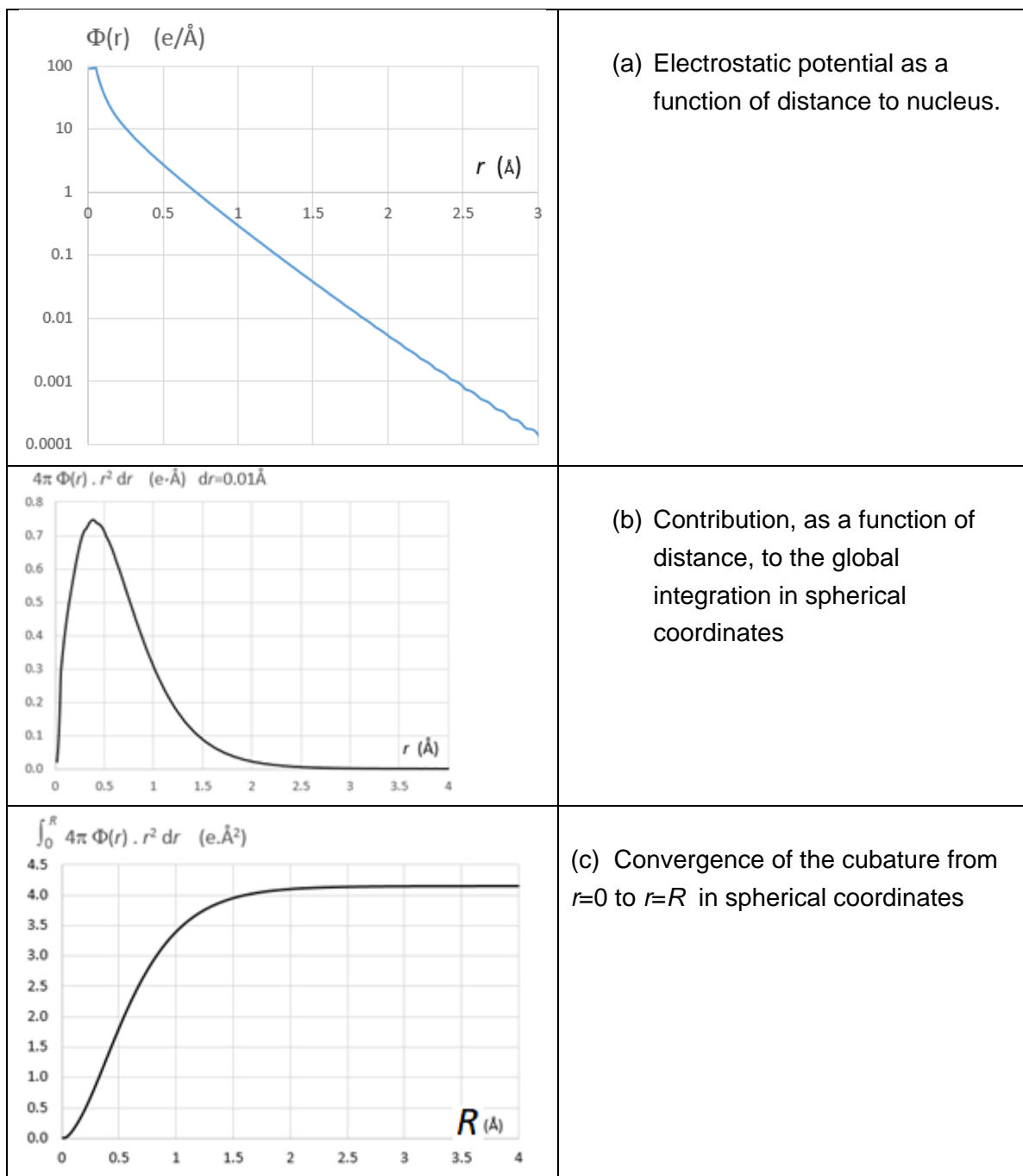
(a)



(b)

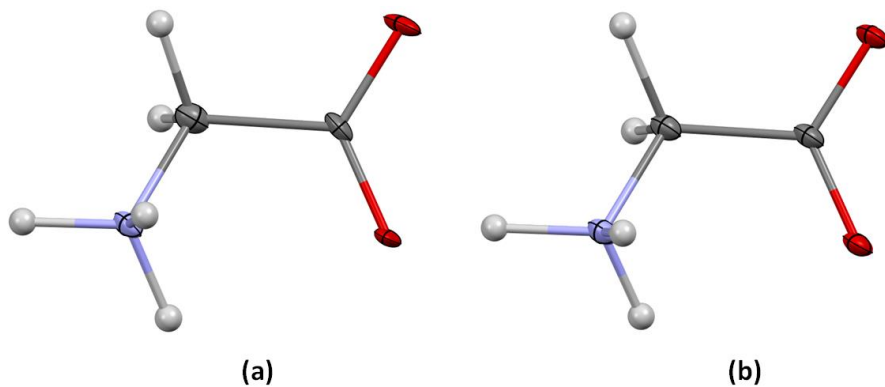


**Figure S1.** Comparison of  $f(s)$  electron scattering factors ( $\text{\AA}$ ) as a function of reciprocal resolution  $s = \sin \theta / \lambda$  ( $\text{\AA}^{-1}$ ) for (a) a carbon atom with charge  $+1e$ ,  $f(\text{C}^+)$  and (b) an oxygen atom with charge  $-1e$ ,  $f(\text{O}^-)$  from current study and earlier reported by Yonekura *et al.*, 2018. Current  $f(s)$  were obtained by the Mott-Bethe equation on X-ray scattering factors for neutral atoms (C, O) and for ions ( $\text{O}^-$ ) with the valence part rescaled to reproduce desired charges of carbon and oxygen using the  $f = f_{\text{core}} + P_{\text{val}} f_{\text{val}}$  approach. The reported  $f(s)$  values were computed on isolated carbon and oxygen ions (Yonekura *et al.*, 2018). The  $f(s)$  derived from neutral atoms differ from the reported ones by  $R$ -factor  $Rf = 20.5$  and  $16.2\%$  for (a)  $f(\text{C}^+)$  and (b)  $f(\text{O}^-)$ , respectively. The  $f(s)$  derived from oxygen anion,  $f(\text{O}^-) = f_{\text{core}}(\text{O}^-) + 7f_{\text{val}}(\text{O}^-)$ , differ from the reported one by  $Rf = 0.07\%$ . (c) Electron scattering factor of the spherical part of carboxylate oxygen atom after ELMAM2 database transfer (charge  $-0.2658 e$ ). For comparison,  $f(s)$  was also computed for the atom with same  $P_{\text{val}}$  valence population, but  $\kappa$  expansion contraction coefficient set to unity.

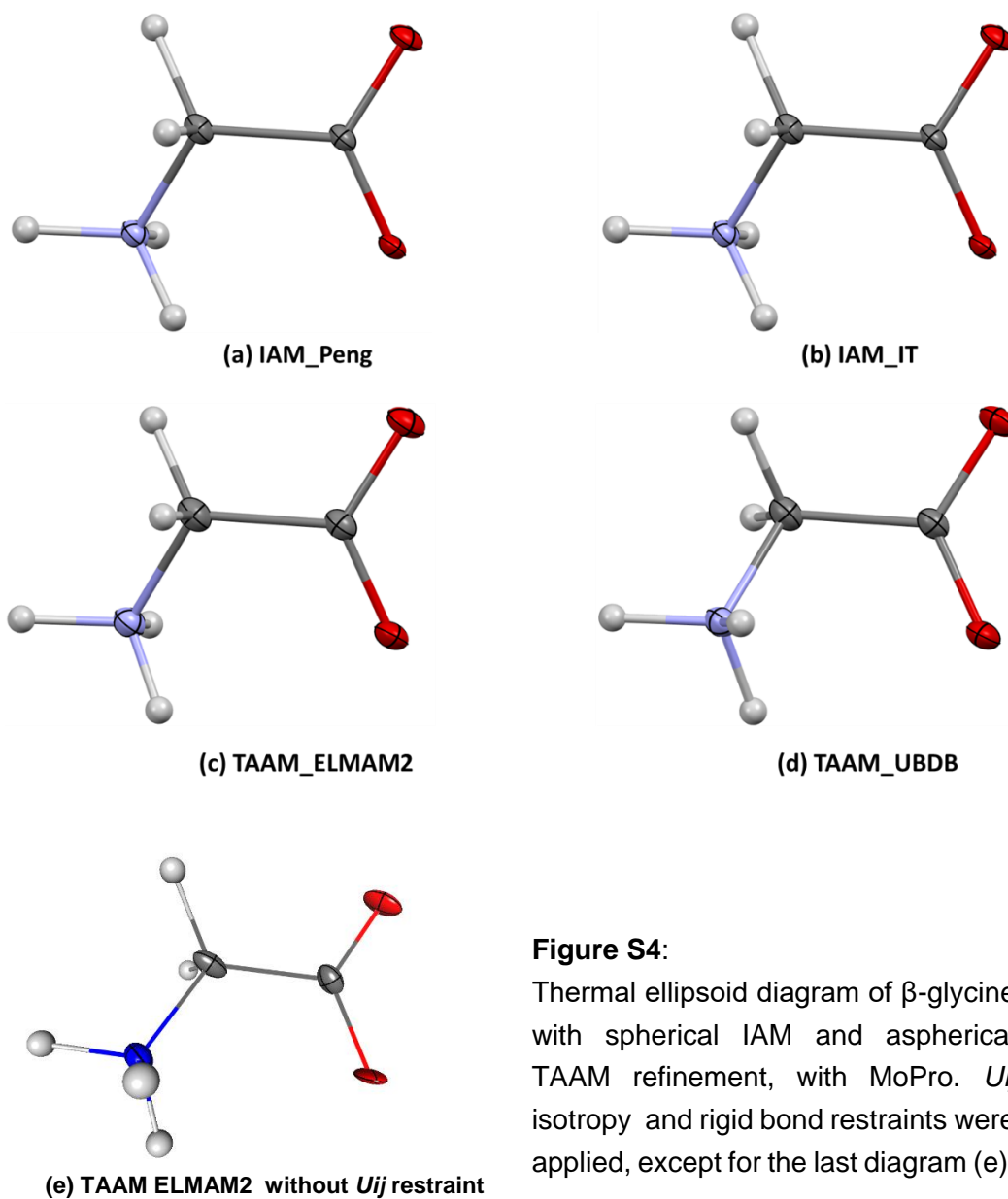


**Figure S2.** Cubature of the electrostatic potential generated by an IAM carbon atom (neutral and spherical). The integration is done in spherical coordinates:  $I_a = 4\pi \int \Phi_a(r) r^2 dr$

1 e/Å corresponds to 14,40 Volts.



**Figure S3:** Thermal ellipsoid diagram of  $\beta$ -glycine (a) reported structure (Broadhurst *et al.*, 2020) (b) re-refined structure (Table 2) with  $U_{ij}$  isotropy and rigid bond restraints in SHELX.



**Figure S4:** Thermal ellipsoid diagram of  $\beta$ -glycine with spherical IAM and aspherical TAAM refinement, with MoPro.  $U_{ij}$  isotropy and rigid bond restraints were applied, except for the last diagram (e).