

# **Hydrogen Bond Computing Server (HBCS): An online web server to compute hydrogen bond interactions with their precision**

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## CALCULATION OF ATOMIC COORDINATE ERROR:

Diffraction precision index (DPI) was first introduced by Cruickshank (Cruickshank, 1999) as a quantitative descriptor to estimate the coordinate error  $\sigma(x, B_{avg})$  of an 'average atom' *i.e.* whose B factor is the average B factor of the refined model. DPI, derived from experimental crystallographic parameters, is computationally very efficient unlike the full-matrix error estimate. Though it is an approximation, it has been shown that the DPI agrees very well with the full-matrix error estimate (Cruickshank, 1999). The coordinate error of each atom can be calculated using equation (1), which takes account of the B factor of an individual atom versus that of an average atom,

$$\text{Atomic coordinate error} = \text{DPI} (B_{atom}/B_{average})^{1/2} \quad (1)$$

The overall DPI itself can be calculated using equation 2

$$\sigma(x, B_{avg}) = (N_i/p)^{1/2} C^{-1/3} R_{work} d_{min} \quad (2a)$$

$$\sigma(x, B_{avg}) = (N_i/n_{obs})^{1/2} C^{-1/3} R_{free} d_{min} \quad (2b)$$

where  $N_i$  is the number of fully occupied atoms,  $p = n_{obs} - n_{params}$ ,  $n_{obs}$  is the number of reflections used in the refinement,  $n_{params}$  is the number of

parameters,  $C$  is the diffraction data completeness,  $d_{min}$  is the resolution of the diffraction data and the  $R$  (reliability) factor (either  $R_{work}$  or  $R_{free}$ ).

Finally, the precision of any interatomic distance ( $d$ ) and angle ( $\theta$ ) can be quantified by harnessing the atomic displacement parameters of any individual interacting atoms (labeled 'A', 'B' and 'C' in equation 3 and 4). A more detailed discussion on this subject has been presented by the authors elsewhere (Kumar *et al.*, 2015, Gurusaran *et al.*, 2014).

$$\sigma_d = (\sigma_A^2 + \sigma_B^2)^{1/2} \quad (3)$$

$$\sigma_\theta = \left( \frac{\sigma_A^2}{d_{AB}^2} + \frac{\sigma_B^2 d_{AC}^2}{d_{AB}^2 d_{BC}^2} + \frac{\sigma_C^2}{d_{BC}^2} \right)^{1/2} \quad (4)$$

## REFERENCES

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