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**Supporting information for article:**

**Structure of a double-domain phosphagen kinase reveals an  
asymmetric arrangement of the tandem domains**

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**Table S1** The melting temperature ( $T_m$ ) of ddAK from *Anthopleura japonicas* in different buffers.

Buffer <sup>a</sup>	pH	$T_m$ (°C)
Sodium acetate	4.5	37.5
Sodium acetate	5.0	42.7
MES	5.5	44.9
BIS-TRIS	5.5	45.0
MES	6.0	46.2
MES	6.5	47.3
BIS-TRIS	6.5	47.4
HEPES	7.0	48.3
Tris-HCl	7.5	48.0
Tris-HCl	8.0	48.3
Tris-HCl	8.5	47.8
CHES	9.0	47.4
CHES	9.5	44.0

<sup>a</sup> Buffers were prepared at a concentration of 50 mM in the presence of 20 mM sodium chloride.

**Table S2** The melting temperature ( $T_m$ ) of ddAK from *Anthopleura japonicas* in the presence of different ligands.

Ligands <sup>a</sup>	Concentration	$T_m$ (°C)
Control	N/A	48.4
ADP	10	49.7
ATP $\gamma$ S	1	49.5
AMPPNP	1.5	50.2
Arginine	10	53.0
ADP/Arginine	5 / 5	53.6
ATP $\gamma$ S/Arginine	1/1	50.2
AMPPNP/Arginine	1.5 / 5	53.4

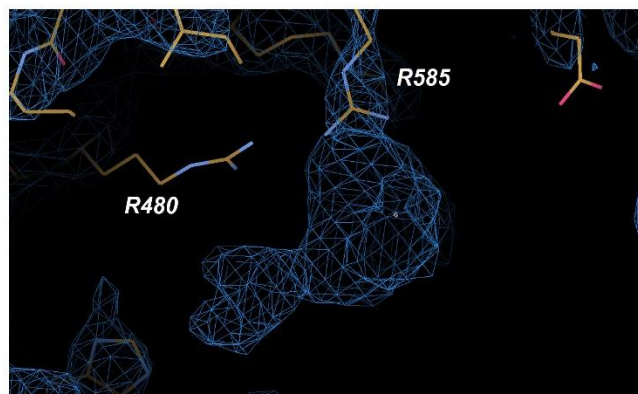
<sup>a</sup> Ligands were solved in a buffer containing 50 mM HEPES (pH 7.0) and 20 mM sodium chloride.

**Table S3** The specific activity of ddAK from *Anthopleura japonicas*

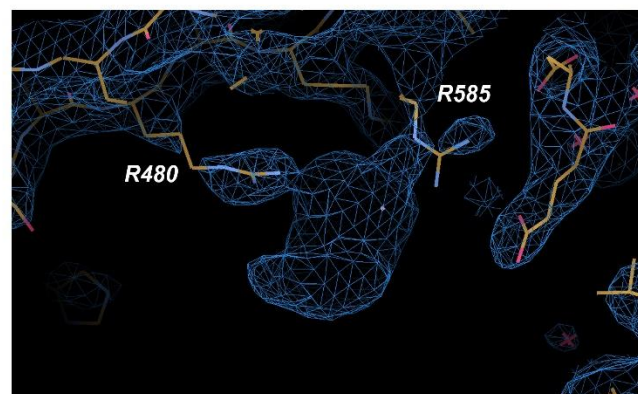
Mutants	Specific activity ( $\mu\text{M}/\text{min}/\text{mg}$ ) <sup>a</sup>	Relative activity (%) <sup>b</sup>
Wild type	279.87 $\pm$ 10.90	100
R122A	166.49 $\pm$ 25.70	59.49 $\pm$ 9.18
R122K	193.14 $\pm$ 8.37	69.01 $\pm$ 2.99
D408A	200.30 $\pm$ 32.29	71.57 $\pm$ 11.54
D408E	169.21 $\pm$ 8.38	60.46 $\pm$ 2.30
R501A	159.02 $\pm$ 16.96	56.82 $\pm$ 6.06
R122D/D408R	167.66 $\pm$ 18.78	59.91 $\pm$ 6.71
R122Y/D408H	286.54 $\pm$ 19.03	102.38 $\pm$ 6.80
$\Delta\text{TGE1}$	180.74 $\pm$ 19.26	64.58 $\pm$ 6.88
$\Delta\text{TGE2}$	156.00 $\pm$ 8.73	55.74 $\pm$ 3.12
$\Delta\text{TGE1/2}$	92.65 $\pm$ 8.91	33.10 $\pm$ 3.18

<sup>a</sup>The specific activities are the mean  $\pm$  SD of triplicate determinations.

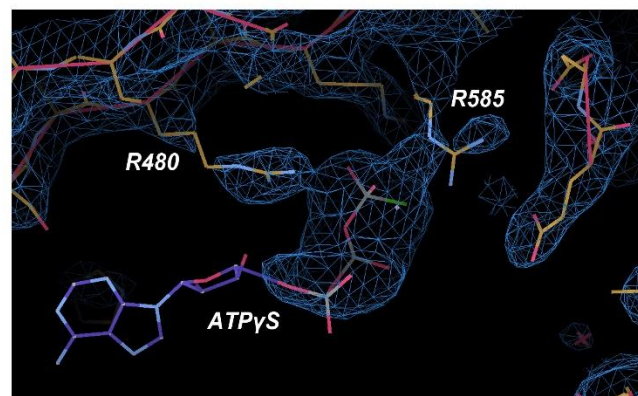
<sup>b</sup>The relative activity of the wild type is defined as 100.00%.



(a)



(b)



(c)

**Figure S1** Uninterpretable electron density maps around the active site of ddAK-D2 in the arginine/ATP $\gamma$ S-bound complex structure. The omit map (at 1.0 contour level) depicted in blue meshes was calculated by using PHENIX.CalculateMaps (Afonine, Grosse-Kunstleve et al. 2012) with a protein model without any ligands to provide phases. Figures were prepared using COOT (Emsley and Cowtan 2004). (a) The uninterpretable map around the active site of ddAK-D2 in chain A. (b) The uninterpretable map around the active site of ddAK-D2 in chain B. (c) The same map in (b) overlapped to a refined model with ATP $\gamma$ S at the active site. A

reasonable explanation to this bulk of electron density map could be the negatively charged phosphoryl groups of ATP $\gamma$ S, of which the nucleoside group cannot be observed due to its extremely high flexibility. In the crystallization solution of this complex, there is no more small molecules than ATP $\gamma$ S that can provide such a strong electron density, but other possibilities still remain. Therefore, we decided to leave these uninterpretable maps empty with no small molecules to fit in.