

Supplementary materials

Within stereoatomic model¹² an atom or a molecule are characterized by such descriptors as:

${}^0d_{\min}$ and ${}^0d_{\max}$ – the shortest and the longest distances between the atoms of neighboring molecules.

${}^0D_{AZ}$ and ${}^1D_{AZ}$ – Relative contributions (in per cents) of a given type of non-valence interactions A/Z to the total area of molecular surface (0S) or the area of VDP faces internal to the molecular surface (1S).

${}^1f_{AZ}$ – The module of difference between the number of faces of a given A/Z type and RC value for two molecular VDPs. 1F – the sum of ${}^1\phi_{AZ}$ for all theoretically possible types of intramolecular A/Z interactions for one molecule.

G_3 – dimensionless second moment of inertia of a VDP (either atomic and molecular).

0k and 1k – as the number of inter- and intramolecular contacts per one molecule; calculated as the number of pyramids with a vertex situated in the atomic position and the base in a VDP face with corresponding RC (0 and >1).

MMVDP – the method of representation of a crystal space by the molecular Voronoi-Dirichlet polyhedra.

RC – Rank of interatomic contact, the number of chemical bonds in the shortest chain connecting the A and B atoms in the crystal structure. Rank = 1 for valence bonds, Rank > 1 for intramolecular non-valence interactions and the Rank of intermolecular contacts is set to zero.

0S and 1S – Total areas of VDP faces with RC = 0 and >1. The former denotes the area of a molecular surface and the latter is the area of intramolecular contacts.

v_{ij} – and a volume of a pyramid with a vertex in a position of A_i or A_j atom and the base in the VDP face common for the A_i and A_j , atoms. Upon the latter descriptor, a total volume of pyramids based on the faces with RC = 0 and with the vertexes in the atomic positions interior to the molecular surface (0V) can be calculated.

VDP – Voronoi-Dirichlet polyhedron of a point represents a convex polyhedron of minimum volume, containing this point, and bounded by perpendicular planes, which pass through middle points of segments, connecting this point with all other points.

Full set of molecular descriptors for **1 – 5** is listed in Table 1S.

Table 1S. Some characteristics of non-valence interactions in the crystals of polymorphs 1 – 5 obtained within the MMVDP^a

Polymorph number	Reference code Allen, 2002	Reference	Space group	V _{mol} , Å ³	T _c , K	Intermolecular interactions						Intramolecular interactions			
						⁰ k	⁰ d _{min} , Å	⁰ d _{max} , Å	⁰ S, Å ²	⁰ V, Å ³	^{>1} k	^{>1} d _{min} , Å	^{>1} d _{max} , Å	^{>1} S, Å ²	^{>1} V, Å ³
Polymorphs of C ₁₃ H ₁₁ NO (1)															
1-1A	SALCAN01	Arod <i>et al.</i> , 2005	P2 ₁ 2 ₁	255.8	120	268	2.55	4.11	281.5	146.6	49	1.72	3.35	100.3	38.4
1-2B	SALCAN03	Arod <i>et al.</i> , 2007	P bc2 ₁	256.8	120	264	2.57	4.36	282.2	148.3	41	1.79	2.92	97.8	36.7
1-3C ^b	SALCAN04	Arod <i>et al.</i> , 2007	P 1	271.3	120	258	2.56	4.60	293.6	157.6	53	1.83	3.46	103.8	40.8
Polymorphs of C ₁₃ H ₁₁ N ₃ O ₄ (2)															
2-1A	RIGHUK01	Scherl <i>et al.</i> , 1996	P 2 ₁ /c	320.3	296	282	2.43	4.49	324.3	170.9	72	1.55	3.83	181.5	69.3
2-2A	RIGHUK05	Naumov & Sakurai, 2005	P 2 ₁ /c	322.9	316	284	2.49	4.59	325.0	172.9	71	1.56	3.85	185.4	70.0
2-3A	RIGHUK06	Naumov & Sakurai, 2005	P 2 ₁ /c	322.8	319.7	282	2.49	4.59	325.3	172.7	72	1.45	3.90	183.8	69.6
2-4A	RIGHUK07	Naumov & Sakurai, 2005	P 2 ₁ /c	323.2	324	282	2.50	4.59	324.9	173.0	72	1.56	3.90	185.4	70.1
2-5B	RIGHUK08	Naumov & Sakurai, 2005	P 2 ₁ /c	325.7	328.9	284	2.52	4.59	327.2	176.2	71	1.56	3.80	179.1	67.8
2-6B	RIGHUK09	Naumov & Sakurai, 2005	P 2 ₁ /c	329.4	337.5	284	2.53	4.60	329.5	178.2	72	1.56	3.81	180.9	68.6
2-7B	RIGHUK	Eichen <i>et al.</i> , 1997	P 2 ₁ /c	327.6	343	286	2.53	4.54	327.6	176.2	78	1.31	4.44	177.3	67.4
2-8C	RIGHUK04	Schmidt <i>et al.</i> , 1999	C c	313.8	293	283	2.26	4.47	322.5	169.8	75	1.34	3.77	169.4	63.4
Polymorphs of C ₁₇ H ₁₂ N ₄ O ₄ (3)															

3-1A	ZOGQAN04	Naumov & Ohashi, 2004	P n a 21	377.8	78	358	2.39	4.74	387.1	202.4	87.5	1.60	3.75	171.8	68.9	
3-2A	ZOGQAN02	Scherl <i>et al.</i> , 1996	P n a 21	380.7	173	353	2.36	4.79	386.5	204.0	91.5	1.57	3.76	180.2	71.6	
3-3B	ZOGQAN03	Scherl <i>et al.</i> , 1996	P 21/c	382.5	173	374	2.40	4.59	387.9	204.2	80	1.72	3.79	173.8	70.2	
3-4C	ZOGQAN05	Naumov & Ohashi, 2004	P 21/c	375.1	78	346	2.51	4.61	379.8	198.8	90	1.50	3.93	176.1	70.9	
Polymorphs of C ₃₉ H ₄₆ N ₂ O ₂ (4)																
4-1A	ATIWOP	Taneda <i>et al.</i> , 2004	P 21/n	878.3	296	596	2.20	5.00	772.2	409.8	300	1.36	4.60	750.0	281.2	
4-2B	ATIWOP01	Taneda <i>et al.</i> , 2004	C 2/c	856.8	296	628	2.20	4.91	796.6	408.2	293	1.47	4.30	621.6	241.4	
Polymorphs of C ₂₂ H ₂₇ NO ₃ (5)																
5-1A	MOPNAH	Johmoto <i>et al.</i> , 2009	P 2 ₁ /n	<u>497.1</u>	295	416	1.85	5.15	471.8	246.1	161	1.57	3.91	389.7	140.6	
5-2B	MOPNAH01	Johmoto <i>et al.</i> , 2009	P 2 ₁ /c	<u>511.0</u>	293	406	1.82	5.05	488.4	257.2	166	1.57	3.59	405.3	145.2	
5-3C	MOPNAH02	Johmoto <i>et al.</i> , 2009	C 2/c	<u>502.5</u>	295	412	2.42	5.01	459.4	244.7	161.5	1.57	3.57	402.9	143.2	

a T_e – the temperature of the X-ray diffraction experiment. b The Characteristics of non-valence interactions are given for a major orientation of a molecule in a twinning crystal.

Table 2S. Relative contributions of inter- ($^0\Delta$) and intramolecular ($^{>1}\Delta$) non-valence contacts A/Z into the area of molecular surface of polymorphs **1, 3, 4, 5**^a

The polymorph number	Relative contribution, %	The type of a non-valence interaction A/Z									
		O/O	N/O	C/O	H/O	N/N	C/N	H/N	C/C	H/C	H/H
Polymorphs of C ₁₃ H ₁₁ NO (1)											
1-1A	$^0\Delta$	-	-	0.8	9.5	-	0.2	3.4	1.2	38.0	46.9
	$^{>1}\Delta$	-	-	-	5.7	-	-	19.5	0.6	15.8	58.3
1-2B	$^0\Delta$	-	-	0.5	9.3	-	0.2	4.5	1.1	41.6	42.9
	$^{>1}\Delta$	-	-	-	4.3	-	-	19.8	0.5	14.0	61.4
1-3C	$^0\Delta$	-	-	0.6	8.6	-	-	3.8	0.5	40.2	46.3
	$^{>1}\Delta$	-	-	-	4.7	-	-	21.3	0.5	17.0	56.5
Polymorphs of C ₁₇ H ₁₂ N ₄ O ₄ (3)											
3-1A	$^0\Delta$	4.4	1.5	5.1	26.4	-	1.9	8.9	4.0	21.8	25.9
	$^{>1}\Delta$	3.2	1.1	2.4	23.4	2.3	4.0	13.3	2.8	14.5	33.0
3-2A	$^0\Delta$	4.7	1.6	5.4	25.8	-	2.0	8.9	4.0	21.0	26.6
	$^{>1}\Delta$	2.9	1.3	2.6	23.3	2.0	3.9	13.3	3.1	17.3	30.4
3-3B	$^0\Delta$	2.6	2.1	2.4	32.2	1.2	2.3	6.8	6.0	18.9	25.5
	$^{>1}\Delta$	2.8	-	6.0	22.5	-	2.4	17.8	2.6	12.4	33.4
3-4C	$^0\Delta$	5.0	2.1	5.3	25.3	-	1.5	8.8	3.7	22.9	25.3
	$^{>1}\Delta$	1.8	1.3	4.2	23.5	2.2	3.9	13.3	2.1	12.9	34.8
Polymorphs of C ₃₉ H ₄₆ N ₂ O ₂ (4)											
4-1A	$^0\Delta$	-	-	0.1	7.2	-	-	0.4	0.1	24.0	68.2
	$^{>1}\Delta$	-	-	<0.1	1.3	-	0.1	6.7	0.6	24.6	66.6
4-2B	$^0\Delta$	-	-	0.4	6.6	-	-	0.4	0.1	25.1	67.4
	$^{>1}\Delta$	-	-	<0.1	1.2	-	<0.1	3.9	1.2	26.0	67.8
Polymorphs of C ₂₂ H ₂₇ NO ₃ (5)											
5-1A	$^0\Delta$	0.2	-	0.9	14.4	-	<0.1	2.2	0.6	22.2	59.6
	$^{>1}\Delta$	-	-	0.2	9.5	-	-	5.3	1.1	20.5	63.4
5-2B	$^0\Delta$	0.6	-	0.2	15.5	-	<0.1	2.1	0.8	22.1	58.6
	$^{>1}\Delta$	-	-	0.1	10.0	-	-	4.6	0.9	18.4	65.9
5-3C	$^0\Delta$	-	-	<0.1	12.2	-	0.6	1.8	4.2	16.7	64.5
	$^{>1}\Delta$	-	-	0.1	8.4	-	-	5.3	0.8	17.9	67.4

a The values of $^0\Delta$ and $^{>1}\Delta$ are round to one decimal place and calculated in respect to 0S and $^{>1}S$, correspondingly. The absence of some type of intermolecular interactions in the crystal is denoted with hephen.

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