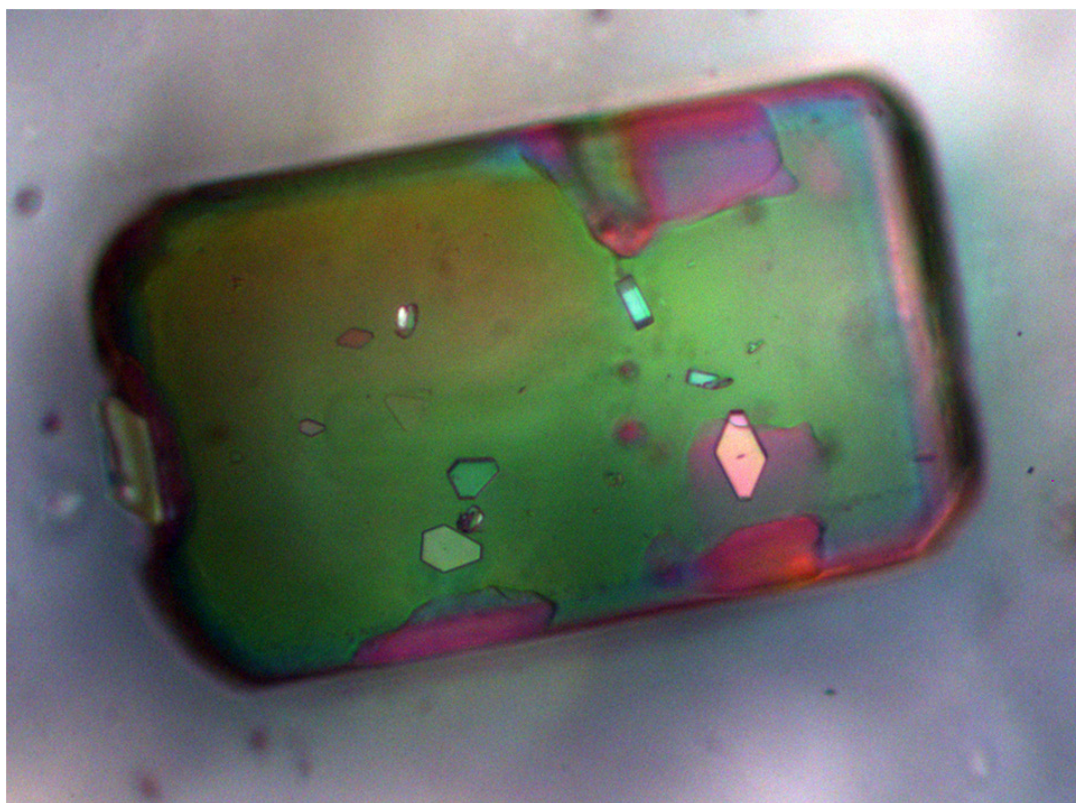
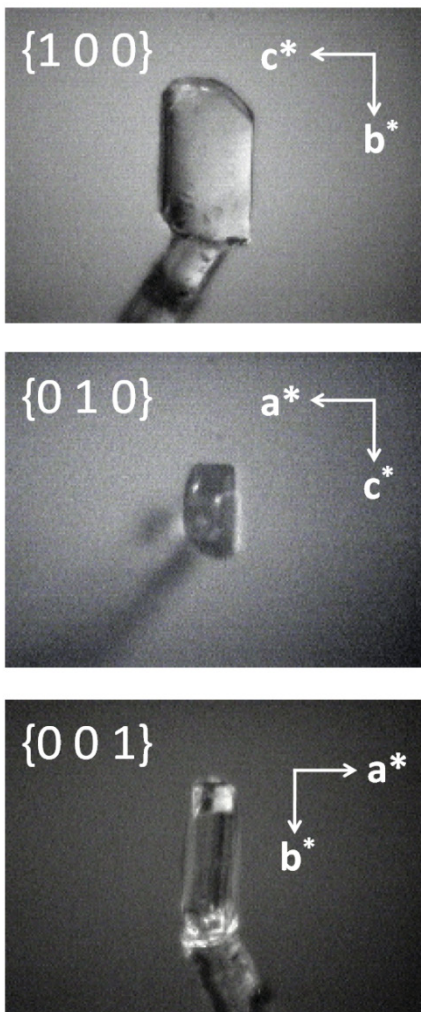


Fig. S-1



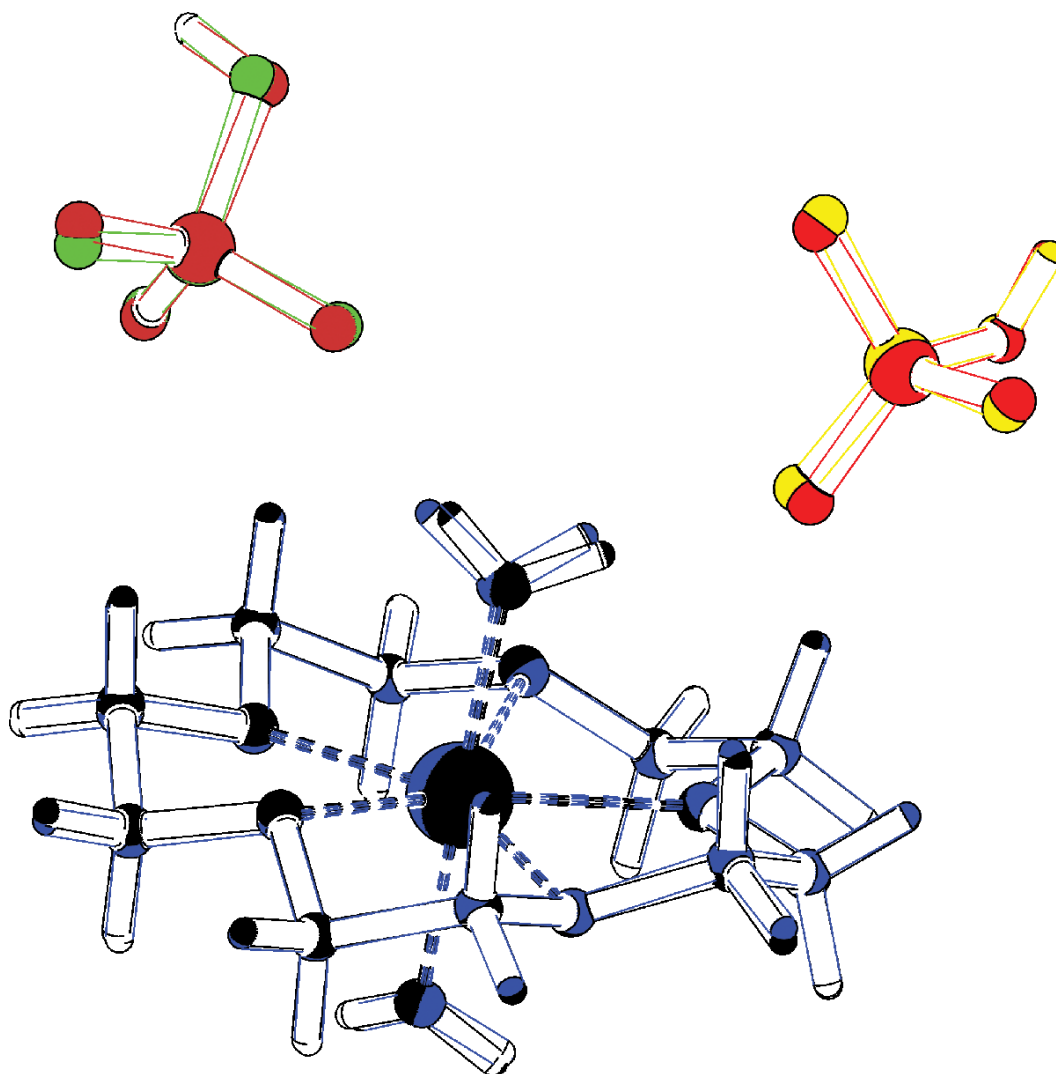
Crystal of $[\text{Ni}(\text{H}_2\text{O})_2(15\text{C}5)](\text{HSO}_4)_2$ removed from the mother liquor. Near room temperature, decomposition occurred after an hour, and growth of new green crystallites took place. These crystallites were those of the tetragonal polymorph of $\text{Ni}(\text{SO}_4)\cdot 6\text{H}_2\text{O}$.

Fig. S-2



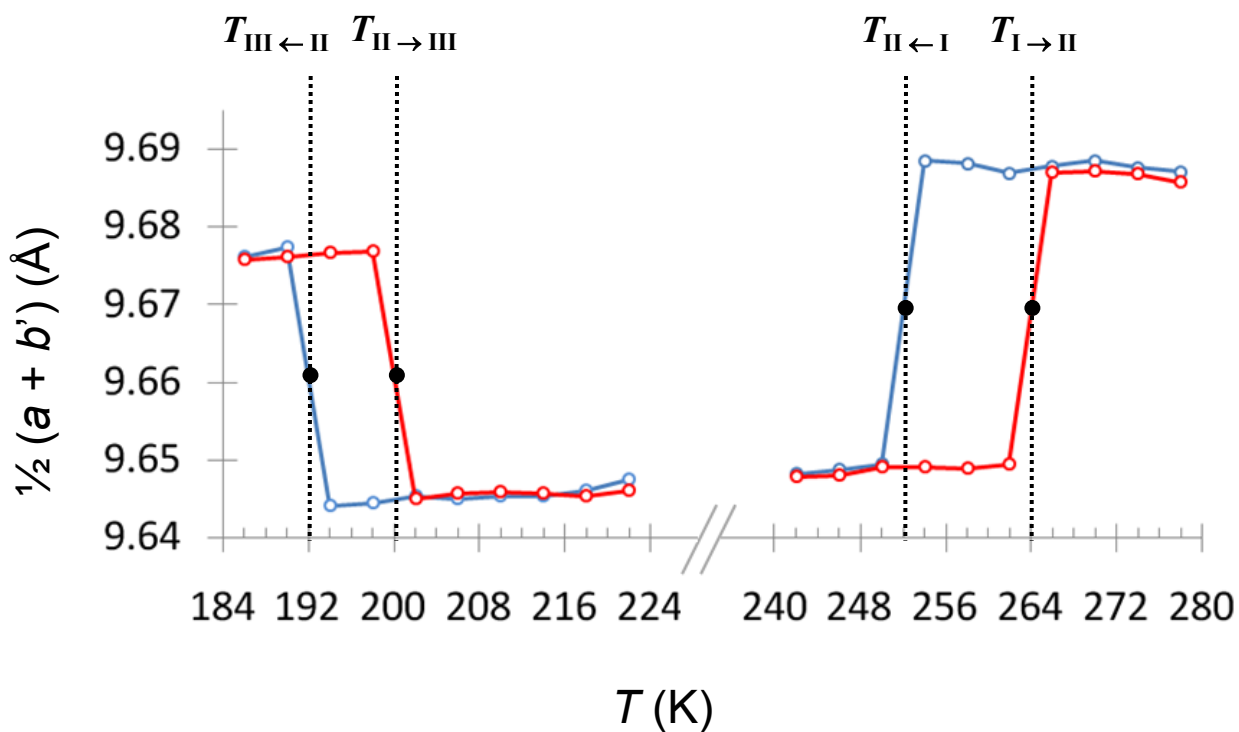
Crystal of $[\text{Ni}(\text{H}_2\text{O})_2(15\text{C}5)](\text{HSO}_4)_2$ mounted on the diffractometer. The indices are given in the $Cmc2_1$ cell of phase I. The most important crystal faces and directions of the reciprocal crystal axes are shown.

Fig. S-3



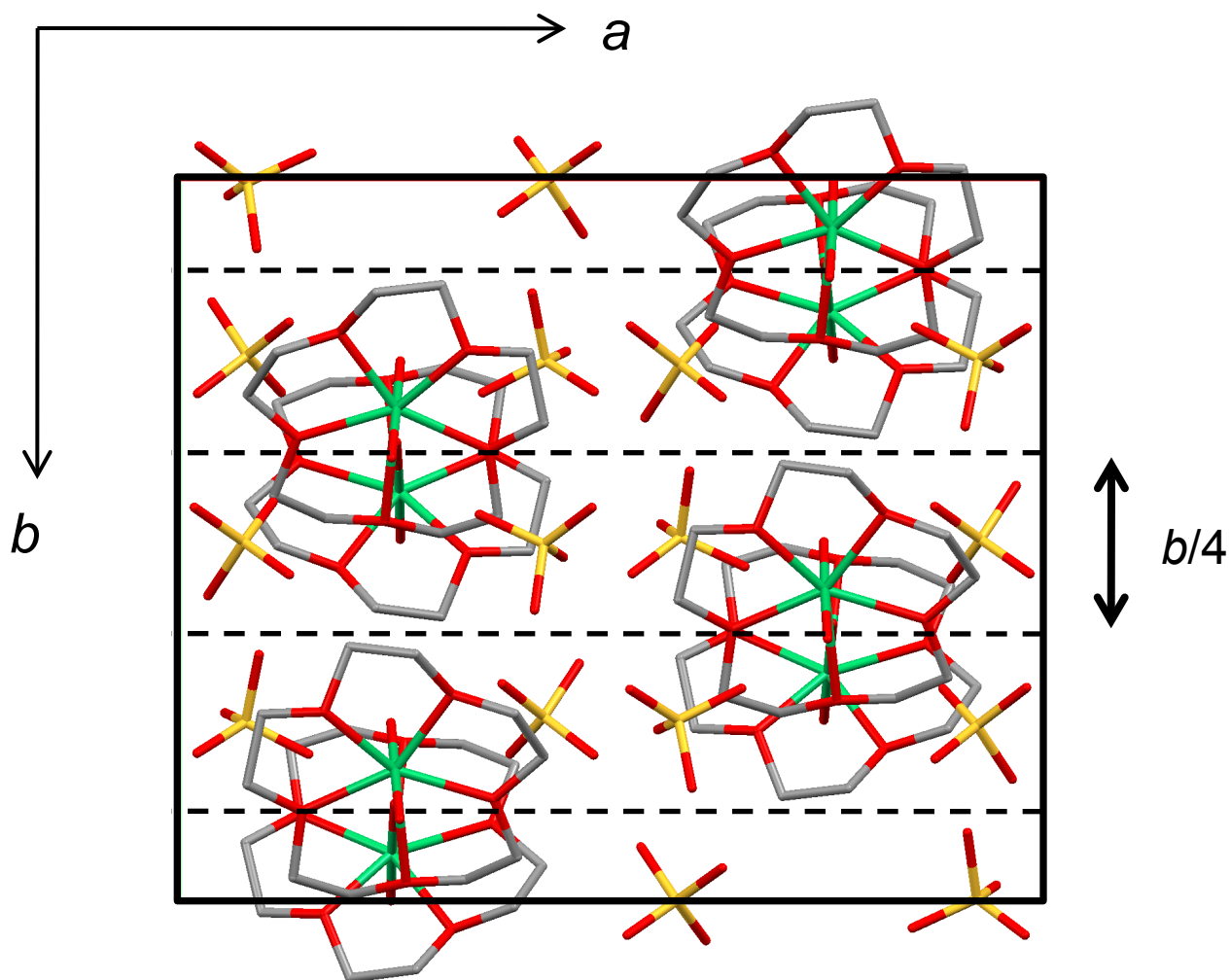
Overlay of the two crystallography independent formula units of $[\text{Ni}(\text{H}_2\text{O})_2(15\text{C}5)](\text{HSO}_4)_2$ (phase III, $Z' = 2$). The two cations nearly superimposed (even the H atoms nearly superimposed except for those of the water ligands), whereas differences in orientations are more noticeable for the counterions.

Fig. S-4



Plots of $\frac{1}{2}(a + b')$ (Å) versus T (K). Values of $\frac{1}{2}(a + b')$ are measured for the transitions I \leftrightarrow II and II \leftrightarrow III at ± 4 K for both cooling (blue) and heating (red) regimens. Discontinuities in these values are found for both transitions. The positions of the vertical dotted lines give the estimated transition temperatures.

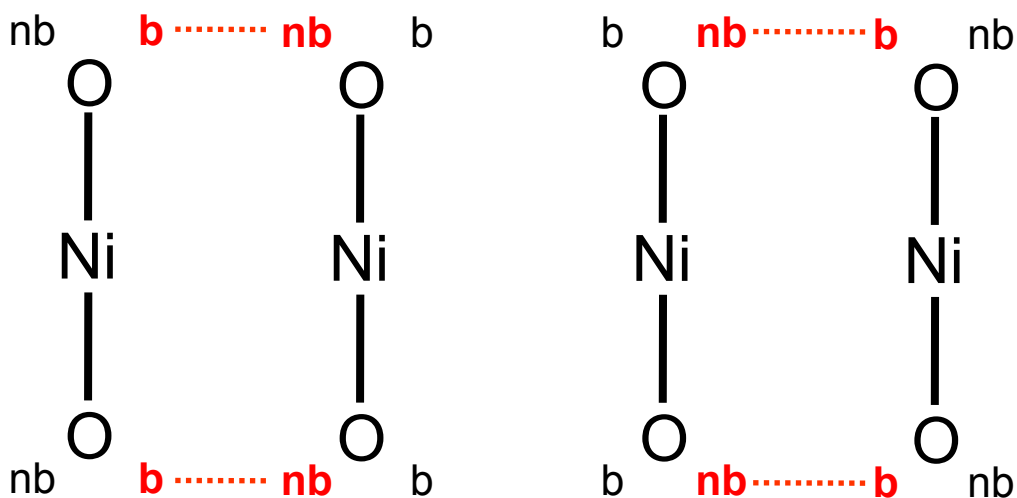
Fig. S-5



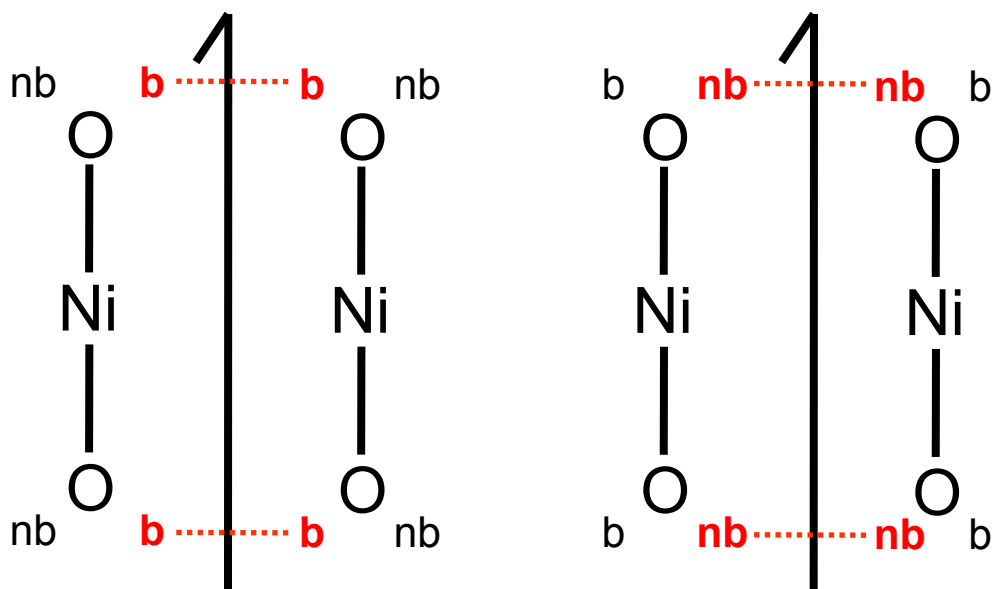
Drawing showing the packing in the unit cell of the $Pbn2_1$, $Z' = 2$ structure down the c direction. The structure is pseudosymmetric, and there are four non-crystallographic pseudo c glide planes (\perp to b) per unit cell separated by $b/4$. The pseudo glide planes are shown as black dashed lines. H atoms are omitted for clarity.

Fig. S-6

(i) non alternations RR or SS



(ii) alternations RS or SR



The different geometries of the $S-O-H\cdots O-S$ bridges for non-alternations RR (or SS) and alternations RS (or SR) along the $\mathbf{a} \pm \mathbf{b}/2$ directions (phase III). For alternations RS (or SR), the two bridges $b\cdots b$ and $nb\cdots nb$ are related by 2_1 axes.

Table S-1. Measurement temperature (T), number of reflections used for cell refinement, cell constants (a , b , c), and cell volume V for the transitions I \rightarrow II and II \rightarrow III in the cooling mode (blue), and for the transitions III \rightarrow II and II \rightarrow I in the heating mode (red). The cell constants are given in the cells $Cmc2_1$ (I), $Pbc2_1$ (II) and $Pbn2_1$ (III)

T (K)	phase	ref#	a (Å)	b (Å)	c (Å)	v(Å ³)
270	I	807	17.8604(9)	7.4836(3)	14.7905(5)	1976.89(18)
260	I	820	17.8628(9)	7.4818(3)	14.7761(6)	1974.77(19)
250	II	834	17.7546(6)	7.5264(3)	14.7896(6)	1976.32(20)
240	II	868	17.7533(10)	7.5222(2)	14.7754(4)	1973.15(13)
230	II	872	17.7541(8)	7.5253(3)	14.7627(6)	1972.38(18)
220	II	880	17.7525(11)	7.5240(3)	14.7471(6)	1969.78(17)
210	II	908	17.7507(07)	7.5245(4)	14.7321(5)	1967.69(18)
200	III	1060	17.8507(07)	14.9346(4)	14.7071(5)	3920.81(26)
190	III	1177	17.8480(6)	14.9298(7)	14.6955(5)	3915.85(37)
180	III	1243	17.8434(8)	14.9240(6)	14.6850(4)	3910.52(31)
170	III	1321	17.8424(7)	14.9226(7)	14.6749(5)	3907.29(39)
160	III	1339	17.8416(4)	14.9203(4)	14.6648(4)	3903.81(26)
150	III	1388	17.8410(6)	14.9171(5)	14.6533(4)	3899.75(33)
140	III	1416	17.8386(6)	14.9170(6)	14.6421(5)	3896.23(26)
130	III	1461	17.8396(8)	14.9132(6)	14.6331(6)	3893.06(32)
120	III	1485	17.8392(8)	14.9111(7)	14.6228(6)	3889.69(36)
110	III	1479	17.8377(9)	14.9086(8)	14.6118(7)	3885.78(39)
110	III	1519	17.8395(8)	14.9060(6)	14.6111(7)	3885.34(42)
120	III	1495	17.8406(10)	14.9100(8)	14.6227(6)	3889.70(50)
130	III	1451	17.8396(10)	14.9124(7)	14.6338(5)	3893.02(39)
140	III	1419	17.8404(8)	14.9152(6)	14.6428(6)	3896.34(39)
150	III	1398	17.8410(7)	14.9188(6)	14.6536(6)	3900.30(31)
160	III	1347	17.8416(8)	14.9213(6)	14.6642(6)	3903.89(27)
170	III	1311	17.8437(10)	14.9237(6)	14.6754(5)	3907.99(33)
180	III	1268	17.8474(8)	14.9272(5)	14.6852(4)	3912.29(29)
190	III	1207	17.8444(7)	14.9288(6)	14.6985(4)	3915.62(32)
200	III	1131	17.8500(6)	14.9305(6)	14.7088(4)	3920.03(34)
210	II	902	17.7525(12)	7.5236(4)	14.7342(6)	1967.95(21)
220	II	912	17.7534(8)	7.5241(3)	14.7479(5)	1970.00(18)
230	II	884	17.7528(8)	7.5253(3)	14.7633(5)	1972.30(15)
240	II	872	17.7546(11)	7.5261(4)	14.7761(6)	1974.44(20)
250	II	861	17.7556(9)	7.5255(3)	14.7893(6)	1976.13(19)
260	II	850	17.7549(9)	7.5274(4)	14.8023(6)	1978.31(18)
270	I	807	17.8605(10)	7.4850(4)	14.792(10)	1977.49(18)

Table S-2. Measurement temperature (T), number of reflections used for cell refinement, cell constants (a , b , c), and cell volume V for the transitions I \rightarrow II (cooling, blue) and II \rightarrow I (heating, red). The cell constants are given in the cells $Cmc2_1$ (I) and $Pbc2_1$ (II)

T (K)	phase	refl. #	a (Å)	b (Å)	c (Å)	V(Å ³)
278	I	833	17.8575(8)	7.4881(3)	14.8071(5)	1979.99(17)
274	I	723	17.8589(7)	7.4876(3)	14.8025(5)	1979.40(16)
270	I	852	17.8610(8)	7.4875(3)	14.7970(5)	1978.86(18)
266	I	834	17.8600(8)	7.4859(3)	14.7911(6)	1977.53(19)
262	I	857	17.8581(9)	7.4858(4)	14.7853(6)	1976.53(20)
258	I	847	17.8619(8)	7.4838(3)	14.7792(5)	1975.60(18)
254	I	860	17.8627(9)	7.4833(3)	14.7748(7)	1974.98(20)
250	II	888	17.7593(7)	7.5275(3)	14.7928(6)	1977.56(20)
246	II	894	17.7576(10)	7.5286(3)	14.7869(6)	1976.86(18)
242	II	899	17.7567(5)	7.5276(3)	14.7815(5)	1975.77(15)
242	II	899	17.7564(8)	7.5267(3)	14.7828(4)	1975.68(15)
246	II	913	17.7568(8)	7.5270(3)	14.7875(6)	1976.43(18)
250	II	896	17.7582(9)	7.5288(4)	14.7926(5)	1977.75(19)
254	II	878	17.7586(7)	7.5279(3)	14.7978(5)	1978.24(14)
258	II	883	17.7584(10)	7.5277(4)	14.8032(7)	1978.90(21)
262	II	878	17.7593(7)	7.5279(3)	14.8090(5)	1979.82(16)
266	I	840	17.8586(7)	7.4858(2)	14.7914(5)	1977.41(15)
270	I	843	17.8585(7)	7.4869(3)	14.7955(4)	1978.23(18)
274	I	840	17.8574(9)	7.4876(3)	14.8017(6)	1979.11(21)
278	I	828	17.8546(6)	7.4886(3)	14.8081(4)	1979.94(14)

Table S-3. Measurement temperature (T), number of reflections used for cell refinement, cell constants (a , b , c), and cell volume V for the transitions II \rightarrow III (cooling, blue) and III \rightarrow II (heating, red). The cell constants are given in the cells $Pbc2_1$ (II) and $Pbn2_1$ (III)

T (K)	phase	refl. #	a (Å)	b (Å)	c (Å)	v(Å ³)
222	II	900	17.7557(10)	7.5269(2)	14.7553(5)	1971.98(17)
218	II	904	17.7525(10)	7.5266(3)	14.7487(5)	1970.66(16)
214	II	909	17.7514(11)	7.5260(3)	14.7429(7)	1969.60(23)
210	II	906	17.7516(10)	7.5258(3)	14.7385(6)	1968.98(18)
206	II	923	17.7506(9)	7.5259(3)	14.7334(5)	1968.21(17)
202	II	915	17.7518(10)	7.5252(3)	14.7273(8)	1967.37(20)
198	II	923	17.7504(10)	7.5236(3)	14.7225(6)	1966.12(23)
194	II	918	17.7494(10)	7.5242(3)	14.7182(6)	1965.60(19)
190	III	1165	17.8453(9)	14.9355(7)	14.6997(6)	3917.88(40)
186	III	1156	17.8427(7)	14.9349(6)	14.6965(6)	3916.32(44)
186	III	1186	17.8417(8)	14.9352(7)	14.6951(5)	3915.80(38)
190	III	1147	17.8425(6)	14.9353(6)	14.7012(5)	3917.64(33)
194	III	1124	17.8434(8)	14.9375(8)	14.7050(6)	3919.43(42)
198	III	1105	17.8430(8)	14.9409(5)	14.7083(6)	3921.08(33)
202	II	921	17.7514(10)	7.5240(3)	14.7264(6)	1966.88(17)
206	II	926	17.7520(7)	7.5260(3)	14.7323(5)	1968.26(16)
210	II	908	17.7527(10)	7.5256(3)	14.7371(6)	1968.87(16)
214	II	925	17.7523(11)	7.5255(4)	14.7434(7)	1969.65(25)
218	II	909	17.7515(10)	7.5257(4)	14.7487(6)	1970.31(23)
222	II	901	17.7531(9)	7.5258(2)	14.7431(5)	1971.11(16)

Table S-4. Geometrical parameters for OH...O hydrogen bonds found in phases I, II and III of [Ni(H₂O)₂(15C5)](HSO₄)₂

	O-H...O (Å)	H...O (Å)	O...O (Å)	O-H-O (°)
Phase I				
O6-H...O8	0.767(19)	1.91(2)	2.667(3)	168(4)
O6-H...O8'	0.767(19)	2.02(2)	2.743(4)	157(4)
O7-H...O9	0.797(17)	1.889(18)	2.650(3)	159(3)
O7-H...O8'	0.797(17)	2.06(2)	2.767(5)	149(3)
O11-H...O10	0.82	1.82	2.581(8)	153
O11-H...O10'	0.82	1.82	2.619(14)	164
Phase II				
O6-H...O8	0.79(2)	1.88(2)	2.663(4)	168(4)
	0.81(2)	1.87(2)	2.676(3)	174(3)
O6-H...O8'	0.79(2)	2.00(3)	2.728(5)	154(4)
	0.81(2)	2.01(3)	2.757(16)	153(4)
O7-H...O9	0.79(2)	1.89(2)	2.645(4)	161(4)
	0.81(2)	1.87(2)	2.672(3)	170(4)
O7-H...O8'	0.79(2)	2.02(3)	2.724(5)	148(3)
	0.81(2)	2.15(4)	2.799(18)	137(3)
O11-H...O10	0.83	2.03	2.596(5)	125
	0.83	2.12	2.69(2)	126
O11-H...O10'	0.83	1.67	2.44(3)	153
	0.83	1.70	2.50(2)	160
Phase III				
O6-H...O8	0.85(2)	1.88(2)	2.712(3)	168(3)
	0.87(2)	1.91(2)	2.730(3)	158(3)
	0.78(2)	1.97(2)	2.738(3)	165(3)
	0.83(2)	1.89(2)	2.697(3)	162(3)
O7-H...O8	0.77(2)	1.99(2)	2.748(3)	164(3)
	0.86(2)	1.88(2)	2.729(3)	173(3)
O7-H...O9	0.84(2)	1.85(2)	2.686(3)	176(3)
	0.82(2)	1.87(2)	2.671(3)	167(3)
O11-H...O10	0.84(2)	1.76(2)	2.583(3)	166(3)
	0.87(2)	1.75(2)	2.579(3)	158(3)
	0.82(2)	1.77(2)	2.565(3)	163(3)
	0.85(2)	1.76(2)	2.583(3)	161(3)