# Supplementary Material for 

# More Examples of the $15-$ Crown- $5 \cdots \mathrm{H}_{2} \mathrm{O}-\mathrm{M}-\mathrm{OH}_{2} \cdots 15-\mathrm{Crown}-5$ Motif, $M=\mathbf{A l}{ }^{3+}, \mathbf{C r}^{3+}$, and $\mathbf{P d}^{2+}$ 

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#### Abstract

Five structures of co-crystals grown from aqueous solutions equimolar in 15-crown-5 (or, $15 \mathrm{C} 5)$ and $\left[M\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]\left(\mathrm{NO}_{3}\right)_{n}, M=\mathrm{Al}^{3+}, \mathrm{Cr}^{3+}$, and $\mathrm{Pd}^{2+}$, are reported. The H-bonding patterns in all are similar: metal complexes including the fragment trans $-\mathrm{H}_{2} \mathrm{O}-\mathrm{M}-\mathrm{OH}_{2}$ alternate with 15 C 5 molecules, to which they are hydrogen bonded, to form stacks. A literature survey shows that this H -bonding pattern is very common. In each of the two polymorphs of the compound $\left[\mathrm{Al}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]\left(\mathrm{NO}_{3}\right)_{3} \cdot 15 \mathrm{C} 5 \cdot 4 \mathrm{H}_{2} \mathrm{O}$ there are two independent cations; one forms H bonds directly to the 15 C 5 molecules adjacent in the stack while the other cation is H -bonded to two water molecules that act as spacers in the stack. These stacks are then crosslinked by H bonds formed by the three nitrate counterions and the three lattice water molecules. The H-bonded stacks in $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}\left(\mathrm{NO}_{3}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot(3 / 2)(15 \mathrm{C} 5) \cdot \mathrm{H}_{2} \mathrm{O}$ are discrete rather than infinite; each unit contains two $\mathrm{Cr}^{3+}$ complex cations and three 15 C 5 molecules. These units are again crosslinked by the uncoordinated nitrate ions and a lattice water molecule. In $\left[\mathrm{Pd}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\left(\mathrm{NO}_{3}\right)_{2}\right] \cdot 15 \mathrm{C} 5$ the infinite stacks are electrically neutral and are not crosslinked. In $\left[\mathrm{Pd}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\left(\mathrm{NO}_{3}\right)_{2}\right] \cdot 2(15 \mathrm{C} 5) \cdot 2 \mathrm{H}_{2} \mathrm{O} \cdot$ $2 \mathrm{HNO}_{3}$ a discrete, uncharged unit containing one Pd complex and two 15 C 5 molecules is "capped off" at either end by a lattice water molecule and an included nitric acid molecule. In all five structures the infinite stacks or discrete units form an array that is at least approximately hexagonal.


## Items Included

1. List of structures retrieved from the CSD (2 pp.)
2. Reciprocal lattice slices for Cr compound showing diffuse scattering ( 3 pp .)
3. Details of disorder of "inner" 15 -crown- 5 ring in the Cr compound ( 1 pg .)

| Refcode | R value | M | M <br> Oxidation <br> State | Charge of <br> Complex | $\boldsymbol{M}$-OH2 <br> distance | Motif <br> (if unusual) |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| DUCNEU | 0.050 | Zn | 2 | 0 | 2.00 | Comments |



Deleted Hits:

| CARYUP | 0.150 | Zn | 2 | 0 | 2.00 |
| :--- | :---: | :---: | :---: | :---: | :---: |
| CIGSAM | 0.117 | Cu | 2 | 0 | 1.99 |
| FUVFIL | 0.111 | U | 6 | $2+$ | 2.38 |
| GOHTEC | 0.109 | Nd | 3 | $1+$ | 2.44 |
| TUVYUE | 0.120 | Sc | 3 | $1+$ | $2.2 ?$ |

## Okl Slice of the Reciprocal Lattice for $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}\left(\mathrm{NO}_{3}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot 1.5\left(15\right.$-crown-5) $\cdot \mathrm{H}_{2} \mathrm{O}$ at 90 K



## hkO Slice of the Reciprocal Lattice for $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}\left(\mathrm{NO}_{3}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot 1.5\left(15\right.$-crown-5) $\cdot \mathrm{H}_{2} \mathrm{O}$ at 90 K



## hk3 Slice of the Reciprocal Lattice for $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}\left(\mathrm{NO}_{3}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot 1.5\left(15\right.$-crown-5) $\cdot \mathrm{H}_{2} \mathrm{O}$ at 90 K



## Inner 15-Crown-5 Rings in <br> $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}\left(\mathrm{NO}_{3}\right)\right]\left(\mathrm{NO}_{3}\right)_{2} \cdot 1.5\left(15\right.$-crown-5) $\cdot \mathrm{H}_{2} \mathrm{O}$ at 90 K



