

**Additional material to deposit for the paper of F. Liebau & X. Wang,
“A bond-valence investigation of two series of isostructural lanthanide compounds”.**

Table 3. Values of calculated bond-valence sums (BVS_{Ln}) and bond valences $s_{ij}(\text{LnO}_{\text{N}})$, both measured in v.u., for the $[\text{LnO}_8\text{N}_3]$ coordination polyhedra in the lanthanide chelates $[\text{Ln}_3(\text{C}_{36}\text{H}_{32}\text{N}_7\text{O}_{10}) \cdot \text{H}_2\text{O}]$ described by Seitz *et al.* (2007a, b).

| Ln | $Z_{\text{Ln}}(BVS)_{\text{Ln}}$ | O(1) | O(2) | O(3) | O(4) | O(5) | O(6) | O(7) | O(10) | N(3) | N(5) | N(7) | |
|----|----------------------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| La | 57 | 3.643 | 0.481 | 0.476 | 0.454 | 0.442 | 0.436 | 0.433 | 0.395 | 0.331 | 0.067 | 0.066 | 0.063 |
| Ce | 58 | 3.660 | 0.477 | 0.468 | 0.452 | 0.452 | 0.446 | 0.425 | 0.413 | 0.320 | 0.071 | 0.069 | 0.068 |
| Pr | 59 | 3.656 | 0.486 | 0.479 | 0.457 | 0.449 | 0.443 | 0.425 | 0.401 | 0.319 | 0.068 | 0.066 | 0.063 |
| Nd | 60 | 3.650 | 0.474 | 0.473 | 0.469 | 0.447 | 0.443 | 0.425 | 0.394 | 0.323 | 0.068 | 0.068 | 0.066 |
| Sm | 62 | 3.594 | 0.462 | 0.458 | 0.458 | 0.454 | 0.435 | 0.435 | 0.393 | 0.313 | 0.063 | 0.063 | 0.061 |
| Eu | 63 | 3.615 | 0.469 | 0.463 | 0.455 | 0.454 | 0.434 | 0.434 | 0.395 | 0.317 | 0.066 | 0.065 | 0.064 |
| Gd | 64 | 3.626 | 0.472 | 0.465 | 0.459 | 0.453 | 0.435 | 0.431 | 0.401 | 0.316 | 0.066 | 0.064 | 0.063 |
| Tb | 65 | 3.641 | 0.465 | 0.464 | 0.462 | 0.458 | 0.436 | 0.430 | 0.408 | 0.328 | 0.066 | 0.063 | 0.061 |
| Dy | 66 | 3.583 | 0.462 | 0.457 | 0.453 | 0.446 | 0.440 | 0.422 | 0.400 | 0.319 | 0.062 | 0.062 | 0.061 |
| Ho | 67 | 3.607 | 0.458 | 0.458 | 0.457 | 0.451 | 0.440 | 0.433 | 0.410 | 0.311 | 0.064 | 0.063 | 0.063 |
| Er | 68 | 3.538 | 0.448 | 0.447 | 0.446 | 0.441 | 0.440 | 0.421 | 0.396 | 0.311 | 0.063 | 0.063 | 0.062 |
| Tm | 69 | 3.533 | 0.452 | 0.447 | 0.444 | 0.440 | 0.435 | 0.415 | 0.401 | 0.319 | 0.061 | 0.060 | 0.059 |
| Yb | 70 | 3.529 | 0.468 | 0.453 | 0.448 | 0.443 | 0.427 | 0.426 | 0.390 | 0.295 | 0.061 | 0.059 | 0.059 |
| Lu | 71 | 3.366 | 0.431 | 0.426 | 0.425 | 0.420 | 0.418 | 0.390 | 0.380 | 0.298 | 0.058 | 0.057 | 0.057 |

The bonds $\text{LnO}(1-6)$ are strained chelate bonds to oxygen atoms of the organic anion forming the complex molecule. $\text{LnO}(7)$ is a bond to an oxygen atom of the organic anion of a neighbouring complex. O(10) is part of a H_2O molecule complementing the coordination sphere of the Ln cation. The weak $\text{LnN}(3, 5, 7)$ bonds to nitrogen atoms of the organic anion complete the coordination sphere of Ln to $[\text{LnO}_8\text{N}_3]$.