

SUPPLEMENTARY MATERIAL

Experimental and Theoretical Characterization of the Zn-Zn Bond in [Zn₂(η⁵-C₅Me₅)₂].

Juan F. Van der Maelen, Enrique Gutiérrez-Puebla,* Ángeles Monge, Santiago García-Granda, Irene Resa, Ernesto Carmona, María Teresa Fernández-Díaz, Garry J. McIntyre, Philip Pattison, and Hans-Peter Weber*

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Table S1. Bond distances (\AA) and (selected) angles ($^{\circ}$) involving the Zn atoms from the neutron diffraction experiment.

Atom-Atom	Distance(esd)	Atom-Atom-Atom	Angle(esd)
Zn(1)-Zn(2)	2.292(1)	Zn(2)-Zn(1)-C(1)	145.2(4)
Zn(1)-C(1)	2.306(5)	Zn(2)-Zn(1)-C(2)	146.3(3)
Zn(1)-C(2)	2.304(5)	Zn(2)-Zn(1)-C(3)	149.4(3)
Zn(1)-C(3)	2.272(4)	Zn(2)-Zn(1)-C(4)	149.9(5)
Zn(1)-C(4)	2.272(5)	Zn(2)-Zn(1)-C(5)	147.8(2)
Zn(1)-C(5)	2.284(3)	Zn(1)-Zn(2)-C(6)	146.4(3)
Zn(2)-C(6)	2.326(3)	Zn(1)-Zn(2)-C(7)	147.2(3)
Zn(2)-C(7)	2.308(4)	Zn(1)-Zn(2)-C(8)	150.2(4)
Zn(2)-C(8)	2.308(3)	Zn(1)-Zn(2)-C(9)	149.3(5)
Zn(2)-C(9)	2.304(3)	Zn(1)-Zn(2)-C(10)	147.2(3)
Zn(2)-C(10)	2.305(5)		

Table S2. Bond distances (\AA) and (selected) angles ($^{\circ}$) involving the Zn atoms from the synchrotron X-ray diffraction experiment.

Atom-Atom	Distance(esd)	Atom-Atom-Atom	Angle(esd)
Zn(1)-Zn(2)	2.3186(3)	Zn(2)-Zn(1)-C(1)	148.91(5)
Zn(1)-C(1)	2.2909(12)	Zn(2)-Zn(1)-C(2)	150.65(4)
Zn(1)-C(2)	2.2756(12)	Zn(2)-Zn(1)-C(3)	148.36(4)
Zn(1)-C(3)	2.2794(11)	Zn(2)-Zn(1)-C(4)	145.12(6)
Zn(1)-C(4)	2.3124(14)	Zn(2)-Zn(1)-C(5)	145.72(3)
Zn(1)-C(5)	2.3132(9)	Zn(1)-Zn(2)-C(6)	147.77(5)
Zn(2)-C(6)	2.2892(13)	Zn(1)-Zn(2)-C(7)	146.45(5)
Zn(2)-C(7)	2.2811(11)	Zn(1)-Zn(2)-C(8)	147.02(4)
Zn(2)-C(8)	2.2859(10)	Zn(1)-Zn(2)-C(9)	148.52(6)
Zn(2)-C(9)	2.3027(14)	Zn(1)-Zn(2)-C(10)	149.12(3)
Zn(2)-C(10)	2.3060(10)		

Table S3. Monopole populations, radial parameters and net atomic charges from the multipole refinement.

Atom	P_{val}	κ	P_{∞}	κ'_0	Net charge
ZN(1)	11.49(13)	0.998(7)	0.000	1.050(9)	+0.508
ZN(2)	11.55(13)	0.998(7)	0.000	1.050(9)	+0.450
C(1)	4.0(3)	0.900(7)	0.000	0.776(9)	-0.034
C(2)	4.6(3)	0.900(7)	0.000	0.776(9)	-0.553
C(3)	4.3(3)	0.900(7)	0.000	0.776(9)	-0.312
C(4)	4.6(3)	0.900(7)	0.000	0.776(9)	-0.605
C(5)	4.2(3)	0.900(7)	0.000	0.776(9)	-0.200
C(6)	3.6(3)	0.900(7)	0.000	0.776(9)	+0.430
C(7)	4.8(3)	0.900(7)	0.000	0.776(9)	-0.835
C(8)	3.7(3)	0.900(7)	0.000	0.776(9)	+0.286
C(9)	3.3(3)	0.900(7)	0.000	0.776(9)	+0.740
C(10)	6.2(3)	0.900(7)	0.000	0.776(9)	-2.236
C(11)	5.3(2)	0.900(7)	0.000	0.776(9)	-1.324
C(12)	5.3(2)	0.900(7)	0.000	0.776(9)	-1.264
C(13)	4.9(2)	0.900(7)	0.000	0.776(9)	-0.909
C(14)	5.2(2)	0.900(7)	0.000	0.776(9)	-1.199
C(15)	4.6(2)	0.900(7)	0.000	0.776(9)	-0.623
C(16)	4.8(2)	0.900(7)	0.000	0.776(9)	-0.781
C(17)	4.9(2)	0.900(7)	0.000	0.776(9)	-0.890
C(18)	5.0(2)	0.900(7)	0.000	0.776(9)	-1.006
C(19)	5.3(2)	0.900(7)	0.000	0.776(9)	-1.305
C(20)	5.6(2)	0.900(7)	0.000	0.776(9)	-1.571
H(11A) [#]	1.000	1.200	-0.579(51)	1.200	+0.579
H(12A)	1.000	1.200	-0.271(55)	1.200	+0.271
H(13A)	1.000	1.200	-0.364(55)	1.200	+0.364
H(14A) [*]	0.12(4)	1.200	-0.331(569)	1.200	+0.040
H(14D)	0.88(4)	1.200	-0.467(76)	1.200	+0.411
H(15A)	0.67(4)	1.200	-0.679(97)	1.200	+0.455
H(15D)	0.33(4)	1.200	0.083(194)	1.200	-0.027
H(16A)	0.28(4)	1.200	-0.802(241)	1.200	+0.225
H(16D)	0.72(4)	1.200	-0.187(95)	1.200	+0.135
H(17A)	1.000	1.200	-0.362(54)	1.200	+0.362
H(18A)	1.000	1.200	-0.604(55)	1.200	+0.604
H(19A)	0.30(4)	1.200	-0.770(254)	1.200	+0.231
H(19D)	0.70(4)	1.200	-0.248(109)	1.200	+0.174
H(20A)	1.000	1.200	-0.589(53)	1.200	+0.589

[#] Only one hydrogen atom is included in the table for each methyl group.^{*} Some of the methyl groups were split into two components for treating the disorder problems.

Table S4. Dipole population parameters.

Atom	D ₁₁₊	D ₁₁₋	D ₁₀	κ_1'
ZN(1)	-0.062(21)	0.249(39)	-0.020(22)	1.283(71)
ZN(2)	0.040(21)	0.339(43)	-0.115(22)	1.283(71)
C(1)	0.088(81)	0.199(137)	-0.034(168)	0.776
C(2)	0.149(83)	0.220(133)	0.205(177)	0.776
C(3)	0.100(85)	-0.043(133)	-0.345(174)	0.776
C(4)	-0.233(84)	0.134(163)	0.284(159)	0.776
C(5)	0.097(78)	-0.184(134)	-0.038(158)	0.776
C(6)	0.173(82)	-0.457(184)	-0.165(138)	0.776
C(7)	-0.075(82)	0.426(141)	0.290(181)	0.776
C(8)	0.194(83)	0.091(138)	-0.229(176)	0.776
C(9)	0.131(87)	0.246(168)	-0.485(170)	0.776
C(10)	0.074(81)	-0.121(155)	0.464(153)	0.776
C(11)	0.303(92)	0.197(71)	0.214(93)	0.776
C(12)	0.054(105)	0.167(76)	0.054(67)	0.776
C(13)	0.330(76)	0.180(61)	0.010(94)	0.776
C(14)	0.239(116)	0.255(98)	0.173(76)	0.776
C(15)	0.133(70)	0.038(74)	0.142(61)	0.776
C(16)	-0.263(102)	0.114(94)	0.077(83)	0.776
C(17)	-0.230(87)	0.050(100)	0.098(64)	0.776
C(18)	0.345(75)	0.264(66)	0.017(95)	0.776
C(19)	-0.339(122)	0.016(88)	0.021(87)	0.776
C(20)	-0.351(71)	0.187(82)	-0.018(62)	0.776
H(11A) [#]	0.000	0.000	0.005(25)	1.200
H(12A)	0.000	0.000	0.013(30)	1.200
H(13A)	0.000	0.000	0.035(30)	1.200
H(14A)	0.000	0.000	-0.078(223)	1.200
H(14D)	0.000	0.000	-0.027(32)	1.200
H(15A)	0.000	0.000	0.030(43)	1.200
H(15D)	0.000	0.000	0.264(81)	1.200
H(16A)	0.000	0.000	-0.061(97)	1.200
H(16D)	0.000	0.000	-0.010(38)	1.200
H(17A)	0.000	0.000	0.039(29)	1.200
H(18A)	0.000	0.000	0.039(30)	1.200
H(19A)	0.000	0.000	-0.068(99)	1.200
H(19D)	0.000	0.000	0.028(42)	1.200
H(20A)	0.000	0.000	-0.022(28)	1.200

[#] Same values for the other hydrogen atoms of the same methyl group.

Table S5. Quadrupole population parameters.

Atom	Q_{20}	Q_{21+}	Q_{21-}	Q_{22+}	Q_{22-}	κ_2'
ZN(1)	1.696(22)	0.029(21)	0.425(43)	-2.710(9)	1.270(7)	0.888(28)
ZN(2)	-0.146(68)	-2.512(75)	-1.579(88)	-0.251(89)	-1.157(76)	0.888(28)
C(1)	0.175(21)	0.028(15)	0.537(21)	-0.060(26)	-0.229(21)	0.776
C(2)	0.243(50)	0.217(32)	-0.252(60)	0.062(54)	-0.149(45)	0.776
C(3)	0.426(9)	0.248(9)	-0.139(9)	0.037(7)	0.031(7)	0.776
C(4)	-0.002(21)	-0.172(9)	0.336(22)	-0.216(7)	0.032(43)	0.776
C(5)	0.026(75)	0.020(89)	-0.238(68)	0.108(8)	0.145(88)	0.776
C(6)	-0.043(15)	0.344(26)	0.426(21)	-0.070(22)	-0.020(21)	0.776
C(7)	0.273(32)	0.180(54)	0.296(50)	-0.099(58)	-0.205(60)	0.776
C(8)	0.040(9)	0.087(9)	-0.023(9)	0.079(8)	0.282(8)	0.776
C(9)	-0.285(9)	-0.334(9)	0.232(21)	-0.310(7)	0.185(22)	0.776
C(10)	0.315(89)	-0.220(9)	-0.767(75)	-0.216(9)	-0.183(68)	0.776
C(11)	0.025(26)	-0.002(22)	0.044(15)	0.046(9)	-0.236(21)	0.776
C(12)	0.142(54)	-0.009(58)	-0.063(32)	0.289(61)	-0.082(50)	0.776
C(13)	0.152(9)	-0.048(9)	0.108(9)	-0.284(7)	-0.139(8)	0.776
C(14)	0.062(9)	-0.272(9)	0.259(9)	0.052(7)	-0.245(21)	0.776
C(15)	0.100(9)	-0.033(9)	-0.032(89)	-0.398(7)	-0.022(75)	0.776
C(16)	0.117(22)	0.241(9)	-0.142(26)	-0.090(8)	0.075(15)	0.776
C(17)	-0.025(58)	0.238(61)	-0.015(54)	-0.132(7)	0.002(32)	0.776
C(18)	0.207(9)	0.100(7)	0.116(9)	-0.300(9)	-0.283(9)	0.776
C(19)	0.142(9)	0.084(9)	-0.022(9)	0.012(9)	-0.066(9)	0.776
C(20)	0.342(9)	0.108(9)	-0.019(9)	-0.206(9)	0.077(89)	0.776

Table S6. Octupole population parameters.

Atom	O_{30}	O_{31+}	O_{31-}	O_{32+}	O_{32-}	O_{33+}	O_{33-}	κ_3'
ZN(1)	0.127(22)	-0.010(15)	-0.047(22)	0.180(25)	-0.055(21)	-0.103(24)	-0.116(24)	1.050(9)
ZN(2)	0.174(21)	-0.019(15)	-0.047(21)	0.267(26)	0.067(21)	0.133(22)	-0.147(26)	1.050(9)
C(1)	0.285(141)	0.041(86)	0.010(143)	0.139(101)	-0.029(89)	0.107(64)	-0.049(71)	0.776
C(2)	0.452(161)	0.416(95)	-0.370(143)	0.274(101)	0.226(90)	0.065(65)	-0.029(70)	0.776
C(3)	0.410(169)	0.227(93)	-0.017(145)	0.261(106)	-0.021(92)	0.208(68)	0.035(75)	0.776
C(4)	0.358(128)	-0.031(79)	-0.115(146)	0.316(120)	-0.111(96)	0.122(73)	-0.053(87)	0.776
C(5)	0.392(151)	0.072(85)	-0.241(138)	0.368(105)	-0.075(91)	-0.103(63)	-0.083(74)	0.776
C(6)	0.269(125)	0.207(79)	-0.216(138)	-0.019(111)	-0.108(88)	0.002(82)	-0.025(10)	0.776
C(7)	0.438(163)	0.082(92)	-0.252(145)	0.373(102)	0.072(90)	0.050(64)	0.048(72)	0.776
C(8)	0.306(171)	0.156(94)	-0.521(150)	0.269(108)	-0.039(92)	-0.037(66)	-0.047(74)	0.776
C(9)	-0.139(132)	0.009(84)	0.409(151)	0.137(119)	-0.053(97)	-0.070(75)	-0.046(87)	0.776
C(10)	1.074(145)	0.063(81)	-0.468(148)	0.648(121)	0.121(10)	-0.011(71)	-0.324(91)	0.776
C(11)	-0.101(91)	0.339(81)	-0.443(65)	-0.473(73)	-0.410(73)	0.178(54)	-0.465(71)	0.776
C(12)	0.233(81)	-0.071(71)	0.013(56)	-0.159(75)	0.394(69)	-0.007(76)	-0.296(90)	0.776
C(13)	-0.010(100)	0.125(77)	-0.186(69)	-0.136(64)	-0.208(66)	0.125(52)	-0.305(67)	0.776
C(14)	0.069(105)	0.046(70)	0.301(63)	-0.046(82)	-0.397(104)	-0.178(82)	-0.165(10)	0.776
C(15)	0.045(97)	0.376(62)	-0.317(56)	-0.566(96)	0.141(69)	0.099(70)	-0.266(68)	0.776
C(16)	0.104(106)	0.003(72)	0.352(66)	0.140(86)	0.285(97)	0.250(72)	-0.186(89)	0.776
C(17)	0.022(78)	-0.018(55)	0.628(63)	-0.149(67)	0.476(74)	0.117(79)	-0.029(91)	0.776
C(18)	0.134(104)	-0.109(76)	0.354(71)	-0.004(63)	-0.420(68)	-0.173(54)	-0.367(68)	0.776
C(19)	0.094(118)	-0.183(86)	-0.338(60)	-0.500(11)	-0.068(87)	0.286(99)	-0.320(87)	0.776
C(20)	0.041(78)	-0.012(54)	0.500(62)	-0.035(65)	0.541(63)	0.148(68)	-0.205(81)	0.776

Table S7. Hexadecapole population parameters.

Atom	H _{4o}	H ₄₁₊	H ₄₁₋	H ₄₂₊	H ₄₂₋	H ₄₃₊	H ₄₃₋	H ₄₄₊	H ₄₄₋	κ_4'
ZN(1)	0.005(46)	-0.630(37)	0.524(68)	0.185(49)	0.320(43)	0.770(59)	-0.177(60)	0.327(59)	0.092(63)	0.983(38)
ZN(2)	0.032(50)	0.308(32)	0.726(60)	-0.179(54)	0.250(45)	-0.728(58)	-0.162(67)	-0.148(61)	-0.343(53)	0.983(38)
C(1)	-0.008(167)	-0.099(90)	-0.193(170)	-0.066(130)	0.071(124)	-0.014(90)	-0.063(104)	-0.046(78)	0.091(69)	0.776
C(2)	0.186(183)	0.142(97)	0.064(172)	0.040(131)	-0.032(129)	-0.127(88)	0.041(103)	0.016(79)	0.020(73)	0.776
C(3)	-0.170(184)	0.014(96)	0.223(167)	-0.306(136)	0.253(134)	0.092(87)	-0.050(107)	-0.054(81)	0.021(75)	0.776
C(4)	0.465(158)	0.106(85)	0.041(176)	0.074(142)	0.200(127)	0.140(99)	-0.146(127)	0.252(91)	0.101(79)	0.776
C(5)	0.174(166)	0.071(86)	-0.390(164)	0.013(132)	0.101(127)	0.041(88)	0.124(103)	-0.079(79)	-0.240(70)	0.776
C(6)	-0.214(155)	0.106(85)	-0.178(176)	-0.167(141)	0.039(125)	-0.010(91)	-0.288(121)	0.132(106)	0.070(87)	0.776
C(7)	0.185(186)	-0.108(96)	-0.141(181)	-0.037(136)	0.261(128)	-0.070(86)	0.016(106)	-0.053(80)	0.087(73)	0.776
C(8)	-0.035(183)	0.062(96)	0.025(179)	-0.052(138)	0.121(134)	0.221(88)	-0.079(108)	-0.022(81)	-0.101(74)	0.776
C(9)	-0.193(167)	0.153(88)	0.312(183)	-0.105(148)	-0.142(138)	0.490(100)	0.075(127)	0.233(91)	-0.140(80)	0.776
C(10)	0.708(157)	0.055(86)	0.263(174)	0.374(143)	0.128(132)	0.146(99)	0.368(120)	0.206(90)	-0.125(79)	0.776
C(11)	-0.229(91)	0.072(86)	-0.231(72)	0.059(79)	-0.152(78)	-0.125(68)	-0.148(85)	0.185(56)	0.323(65)	0.776
C(12)	-0.222(83)	-0.018(80)	-0.025(65)	0.094(80)	-0.138(70)	-0.119(77)	0.127(90)	-0.172(86)	-0.200(86)	0.776
C(13)	-0.046(110)	-0.191(93)	-0.090(87)	0.165(78)	-0.061(79)	0.053(67)	-0.008(80)	0.059(58)	-0.012(61)	0.776
C(14)	-0.017(90)	-0.052(81)	0.333(72)	-0.265(71)	0.011(73)	-0.049(81)	-0.154(109)	-0.062(84)	0.101(93)	0.776
C(15)	-0.169(96)	-0.006(69)	-0.130(72)	0.116(68)	-0.398(69)	-0.352(92)	-0.012(81)	0.057(74)	-0.074(78)	0.776
C(16)	-0.173(99)	0.264(82)	0.148(75)	-0.487(76)	0.056(79)	-0.158(85)	-0.136(105)	0.068(80)	0.145(86)	0.776
C(17)	-0.049(75)	0.207(67)	0.171(68)	-0.251(74)	-0.041(74)	0.159(77)	-0.109(88)	-0.244(80)	0.082(89)	0.776
C(18)	-0.022(109)	-0.152(91)	-0.033(82)	-0.028(75)	0.271(84)	0.166(65)	0.085(79)	-0.065(60)	-0.211(63)	0.776
C(19)	-0.120(107)	-0.089(94)	0.069(71)	-0.046(81)	0.191(76)	-0.039(106)	0.302(94)	-0.096(90)	0.014(86)	0.776
C(20)	-0.196(75)	0.320(63)	0.164(76)	-0.038(72)	-0.143(69)	0.059(73)	0.008(90)	-0.203(74)	0.049(78)	0.776

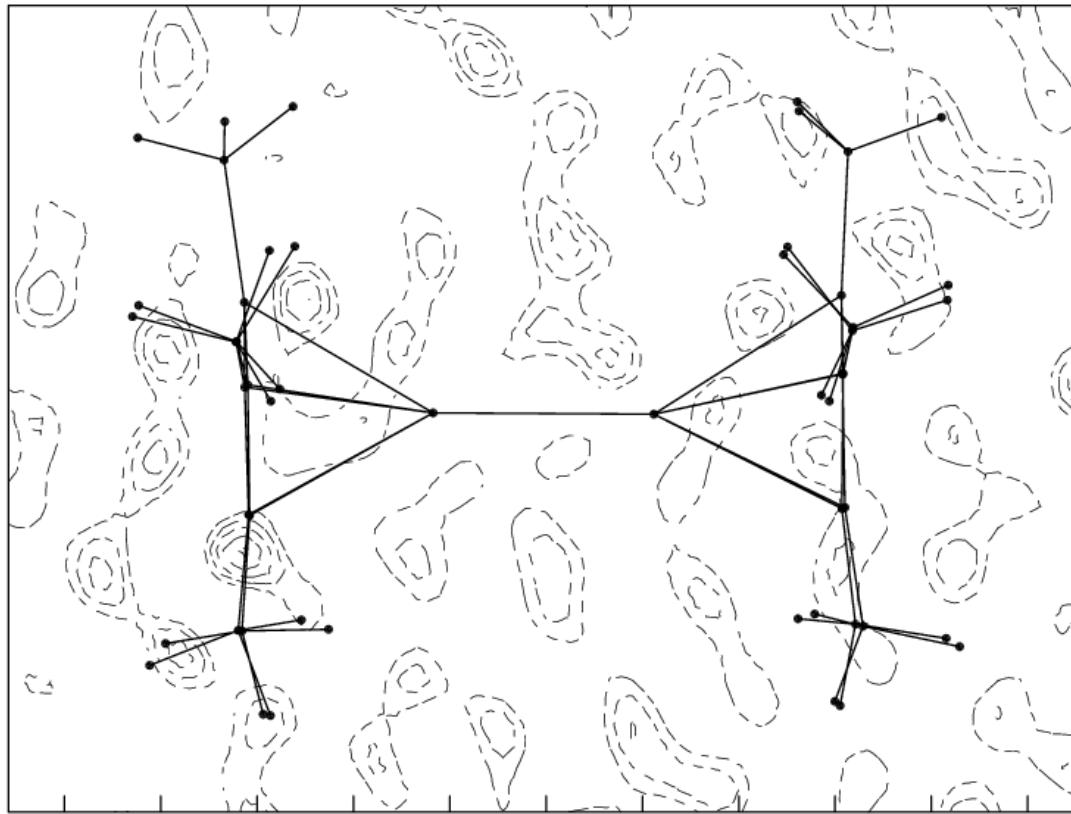


Figure S1. Final Difference Fourier map from neutron diffraction experiment in the Zn(2)-Zn(1)-C(1) plane. Contour values at 0.1 e Å⁻³.

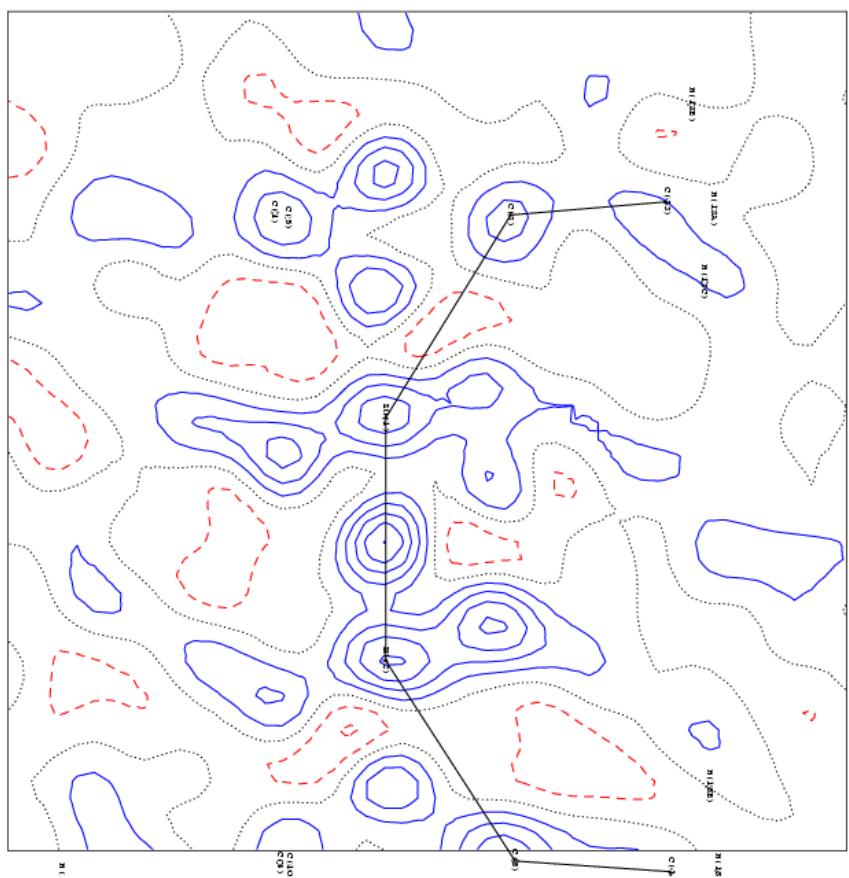


Figure S2. Final Difference Fourier map from X-ray synchrotron radiation experiment in the Zn(2)-Zn(1)-C(1) plane. Contour values at $0.1 \text{ e } \text{\AA}^{-3}$. Positive values plotted as solid blue lines, negative values as dashed red lines, and zero value as dotted black lines.