Di-p-bromophenyl ether, a redetermined crystal structure derived from low quality diffraction data.

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## Appendix 1: Details of MOPAC calculation.

The map in Fig. S3 was calculated on a 10-degree grid with values for each torsion angle ranging from $-180^{\circ}$ to $180^{\circ}$, a total of 37 steps in each direction. This sums up to a total of 1369 energy optimizations where only the two torsion angles are locked at each grid point. The molecular conformation at each grid point could thus be said to have been relaxed. The end values $\left(-180^{\circ}\right.$ and $\left.180^{\circ}\right)$ designate the same conformation and were included only to check the periodicity of the calculations. As each of the rings have C 2 symmetry along the $\mathrm{O}-\mathrm{C} 1 \ldots \mathrm{C} 4-\mathrm{Br}$ direction the conformational map will have translation symmetry along both torsion angles with a translation period of $180^{\circ}$. The total computation time on a standard 1 GHz Pentium III processor was of the order of 10 hours.

## Appendix 2: Details of the scaling and merging process

Details of the merging and scaling procedure are given below.
All data sets were collected and integrated with the diffractometer software using the same unit cell setting. A check for having the correct absolute configuration was done on all individual datasets. None of the refinements on the individual datasets gave indications of having the opposite absolute configuration. One of the six componnet datasets gave a Flack parameter in the order of 0.5 , however with a large s.u. After that the complete structure model was derived from the individual data sets and checked that it fitted with acceptable residual R-factors we determined individual batch scale factor using the BASF and MERG 0 instructions in SHELXL97. One data set together with an additional BASF was added at a time. Approximately 30 least square cycles were necessary in order for the refinements to be completely convergent. When the refinements had converged for all added individual data sets, totally six individual data sets, the refined BASF were used to scale each of the individual intensity data sets. The scale factor applied to each, except the first data set, was computed as 1.0/BASF for each of the individual data sets. The scaling was applied both to $\mathrm{F}^{2}$ and $\operatorname{sig}\left(\mathrm{F}^{2}\right)$ using a locally written computer program REFLEX available from one of the authors. A final check that the computed scale factors were correct was done such as setting all BASF parameters in the instruction file for the SHELXL97 runs to unity and doing refinements with the scaled data set. All BASF parameters refined to unity within the estimated standard uncertainty derived by SHELXL97. In all these calculation no merging i.e. MERG 0 was used in SHELXL97. Thereafter the MERG 2 instruction was used in SHELXL97 to perform the merging of the whole data set producing the final dataset of 1166 unique reflections.

## References

Smith, G. S. \& Snyder, R. L. (1979). J. Appl. Cryst. 12, 60-65.
deWolff, P.M. de (1968). J. Appl. Cryst. 1, 108-113.

## Table S1.

Unitcell data for the six component data sets and their corresponding average values.

| crystal | $\mathbf{a}(\AA)$ | $\mathbf{b}(\AA)$ | $\mathbf{c}(\AA)$ | volume $\left(\AA^{3}\right)$ |
| :--- | :--- | :--- | :--- | :--- |
| I | $26.98(5)$ | $7.692(3)$ | $5.725(5)$ | $1188(4)$ |
| II | $26.74(8)$ | $7.729(10)$ | $5.753(8)$ | $1189(7)$ |
| III | $26.75(3)$ | $7.759(6)$ | $5.731(4)$ | $1189(3)$ |
| IV | $26.65(1)$ | $7.577(2)$ | $5.715(1)$ | $1154(1)$ |
| V | $26.14(6)$ | $7.696(10)$ | $5.807(15)$ | $1168(7)$ |
| VI | $26.76(3)$ | $7.741(8)$ | $5.699(5)$ | $1180(4)$ |
| Average | 26.67 | 7.699 | 5.738 | 1178 |

## Table S2

Indexed powder diffraction data for (I). Data recorded with $\mathrm{CuK} \alpha 1$-radiation $(\lambda \approx 1.5405981 \AA)$

The refined cell parameters are: $a=26.645(6) \AA, b=7.6902(14) \AA, c=5.7223(8) \AA$ $\mathrm{M}(19)=48.2($ deWolff, 1968) and $\mathrm{F}(19)=83.2(0.006522,35)$ (Smith \& Snyder, 1979)

| $\mathbf{N}$ | $\mathbf{H}$ | $\mathbf{K}$ | $\mathbf{L}$ | $\mathbf{2 \theta}_{\text {obs }}$ | $\mathbf{2} \boldsymbol{\theta}_{\text {calc }}$ | $\mathbf{\Delta} \mathbf{2 \boldsymbol { \theta }}$ |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 2 | 0 | 0 | 6.6294 | 6.6294 | 0.0000 |
| 2 | 1 | 1 | 0 | 11.9728 | 11.9685 | 0.0044 |
| 3 | 1 | 1 | 1 | 19.6052 | 19.6086 | -0.0034 |
| 4 | 3 | 1 | 1 | 21.7783 | 21.7771 | 0.0012 |
| 5 | 0 | 2 | 0 | 23.1220 | 23.1129 | 0.0092 |
| 6 | 2 | 2 | 0 | 24.0766 | 24.0701 | 0.0065 |
| 7 | 5 | 1 | 1 | 25.5964 | 25.5925 | 0.0038 |
| 8 | 8 | 0 | 0 | 26.7568 | 26.7450 | 0.0118 |
|  | 4 | 2 | 0 |  | 26.7488 | 0.0080 |
| 10 | 2 | 2 | 1 | 28.7437 | 28.7422 | 0.0015 |
| 11 | 7 | 1 | 1 | 30.4893 | 30.4840 | 0.0053 |
| 12 | 0 | 0 | 2 | 31.2270 | 31.2423 | -0.0153 |
| 13 | 2 | 0 | 2 | 31.9604 | 31.9733 | -0.0129 |
| 14 | 1 | 1 | 2 | 33.5806 | 33.5667 | 0.0139 |
|  | 10 | 0 | 0 |  | 33.6083 | -0.0277 |
| 15 | 4 | 0 | 2 | 34.0872 | 34.0820 | 0.0051 |
| 16 | 3 | 1 | 2 | 34.9463 | 34.9282 | 0.0182 |
| 17 | 1 | 3 | 0 | 35.1385 | 35.1419 | -0.0034 |
| 18 | 8 | 2 | 0 | 35.6228 | 35.6343 | -0.0115 |
| 19 | 9 | 1 | 1 | 36.0745 | 36.0718 | 0.0027 |

Figure S1
Electron density map caclulated from the combined data set, clearly showing the bromine and the six resolved carbon atoms of the benzene ring. Figure drawn with PLATON (Spek, 2003).


## Figure S2

Heat of formation map for different values of the two torsion angles $\left(\varphi_{1}, \varphi_{2}\right)$ describing the molecular conformation. The $\Delta \mathrm{H}^{\circ}$ f was calculated at each grid point from $-180^{\circ}$ to $180^{\circ}$ in $10^{\circ}$ steps. The contours are drawn at $1 \mathrm{kcal} / \mathrm{mole}$. All calculations done with MOPAC6 (Stewart, 1990) and figure drawn with Easyplot (Karon, 1999). The conformation derived from the crystal structure $\left(\varphi_{1} \approx 150^{\circ}, \varphi_{2} \approx-34^{\circ}\right)$, marked with a dot, resides in an energy minimum.


