

**An Improved experimental databank of transferable
multipolar atom models - ELMAM2.
Construction details and applications**

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SUPPORTING INFORMATION

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Supplementary Figures 1S

Supplementary Tables

Table 1S. List of crystal structures used for building the ELMAM2 database.

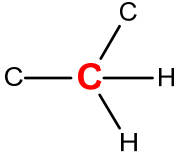
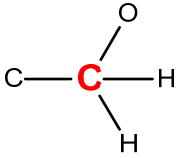
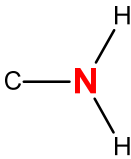
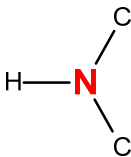
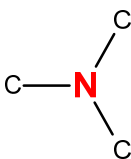
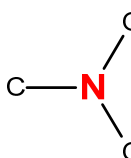
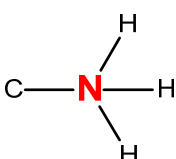
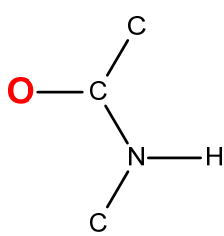
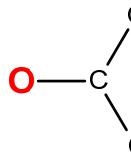
Nr	Abbreviation	Name, formula	Data type, temp	Reference
1	ala	L-Alanine, C ₃ H ₇ NO ₂	X-Ray, Mo, 23K	Destro <i>et al.</i> , 1988
2	enkephalin	Leu-Enkephalin Trihydrate, C ₂₈ H ₃₇ N ₅ O ₇ 3H ₂ O	X-Ray, Mo, 100K	Wiest <i>et al.</i> , 1994
3	Argphos	L-Arginine Phosphate Monohydrate, C ₆ H ₁₅ N ₄ O ₂ ⁺ H ₂ PO ₄ ⁻ H ₂ O	X-Ray, Mo, 130K	Espinosa <i>et al.</i> , 1996
4	oxalic	Oxalic acid, C ₂ H ₂ O ₄	X-Ray, Mo, 100K	Martin <i>et al.</i> , 1998
5	his	DL-Histidine, C ₆ H ₉ N ₃ O ₂	X-Ray, Mo, 110K	Coppens <i>et al.</i> , 1999
6	actyr	N-acetyl-L-tyrosine ethyl ester monohydrate, C ₁₃ H ₁₉ NO ₅	X-Ray, Mo, 110K	Dahaoui <i>et al.</i> , 1999
7	glythr	Glycyl-L-threonine dihydrate, C ₆ H ₁₂ N ₂ O ₄ 2H ₂ O	X-Ray, Mo, 110K	Benabicha <i>et al.</i> , 2000
8	GlyAsp	glycyl-aspartic acid dihydrate, C ₆ H ₁₀ N ₂ O ₅ 2H ₂ O	X-Ray, Mo, 123K	Pichon-Pesme <i>et al.</i> , 2000
9	TyrGlyGly	Tyrosyl-glycyl-glycine monohydrate, C ₁₃ H ₁₇ N ₃ O ₅ H ₂ O	X-Ray, Mo, 123K	Pichon-Pesme <i>et al.</i> , 2000
10	panb	<i>p</i> -amino- <i>p</i> '-nitrobiphenyl, C ₁₂ H ₁₀ N ₂ O ₂	X-Ray, Mo, 20K	Volkov <i>et al.</i> , 2000
11	ureaphos	urea-phosphoric acid, CH ₇ N ₂ O ₅ P	X-Ray, Mo, 100K	Rodrigues <i>et al.</i> , 2001
12	NTO	5-nitro-2,4-dihydro-3H-1,2,4- triazol-3-one, β -NTO, C ₂ H ₂ N ₄ O ₃	X-Ray, Ag, 100K	Zhurova <i>et al.</i> , 2001
13	cycloproala	cyclo-(D,L-Pro) ₂ -(L-Ala) ₄ monohydrate, C ₂₂ H ₃₄ N ₆ O ₆ H ₂ O	Synchrotron, 0.5583 Å, 100K	Dittrich <i>et al.</i> , 2002
14	Thiocoumarin 2	2H-chromene-2-thione, (2- thiocoumarin) C ₉ H ₆ OS	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2002
15	HPPM	hydrogen-[(2,4-diaminopyrimidin- 1-yl)methyl]-phosphonate monohydrate, C ₅ H ₉ N ₄ O ₃ P H ₂ O	X-Ray, Mo, 105K	Slouf <i>et al.</i> , 2002
16	BIGH	Biguanidinium Dinitramide, C ₂ H ₈ O ₄ N ₈	X-Ray, Mo, 90K	Zhurova <i>et al.</i> , 2002
17	BIGH2	Biguanidinium Bis-Dinitramide, C ₂ H ₉ O ₈ N ₁₁	X-Ray, Mo, 90K	Zhurova <i>et al.</i> , 2002
18	schiff1	dianil of 2-hydroxy-5-methyl- isophth-aldehyde, C ₂₁ H ₁₈ O ₁ N ₂	X-Ray, Mo, 100K	Dominiak <i>et al.</i> , 2003
19	schiff2	3,5-dinitro- <i>N</i> - salicylidenoethylamine, C ₉ H ₉ O ₅ N ₃	X-Ray, Mo, 100K	Dominiak <i>et al.</i> , 2003
20	schiff3	3-nitro- <i>N</i> - salicylidencyclohexylamine, C ₁₃ H ₁₆ O ₃ N ₂	X-Ray, Mo, 100K	Dominiak <i>et al.</i> , 2003
21	nad	NAD ⁺ , β -Nicotinamide Adenine Dinucleotide, C ₂₁ H ₂₇ N ₇ O ₁₄ P ₂ 4H ₂ O	X-Ray, Mo, 100K	Guillot <i>et al.</i> , 2003

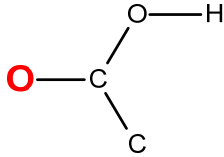
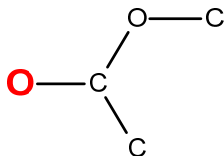
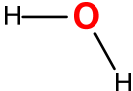
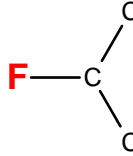
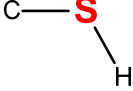
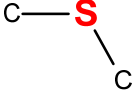
22	TFT	Tetrafluoroterephthalonitrile, C ₈ N ₂ F ₄	X-Ray, Mo, 122.4K	Sørensen <i>et al.</i> , 2003
23	urea	Urea, CH ₄ N ₂ O	Synchrotron, 0.5996 Å, 123K	Birkedal <i>et al.</i> , 2004
24	busulfan	1,4-butanediol-dimethylsulfonate, C ₆ H ₁₄ O ₆ S ₂	X-Ray, Mo, 100K	Ghermani <i>et al.</i> , 2004
25	barba	2,4,6,8-tetraphenylbarbaralane, C ₃₃ H ₂₆	Synchrotron, 0.56 Å, 110K	Luger <i>et al.</i> , 2004
26	xylitol	xylitol, C ₅ H ₁₂ O ₅	X-Ray, Mo, 122.4 K	Madsen <i>et al.</i> , 2004
27	famotidineA	Famotidine (polymorph A), C ₈ H ₁₅ N ₇ O ₂ S ₃	X-Ray, Mo, 100K	Overgaard <i>et al.</i> , 2004
28	famotidineB	Famotidine (polymorph B), C ₈ H ₁₅ N ₇ O ₂ S ₃	X-Ray, Mo, 100K	Overgaard <i>et al.</i> , 2004
29	trp	L-tryptophan formic acid, C ₁₁ H ₁₂ N ₂ O ₂ CH ₂ O ₂	Synchrotron, 0.54 Å, 100K	Scheins <i>et al.</i> , 2004
30	coumarin	2H-chromene-2-one, C ₉ H ₆ O ₂	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2005
31	Thiocoumarin 1	1-thiocoumarin, C ₉ H ₆ OS	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2005
32	pentaerythritol	Pentaerythritol, C ₅ H ₁₂ O ₄	X-Ray, Mo, 15K	Zhurov <i>et al.</i> , 2005
33	accoumarinA	3-Acetylcoumarin, (polymorph A), C ₁₁ H ₈ O ₃	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2006a
34	accoumarinB	3-Acetylcoumarin, (polymorph B), C ₁₁ H ₈ O ₃	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2006a
35	cytosine_hyd	cytosine monohydrate, C ₄ H ₅ N ₃ O H ₂ O	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2006b
36	salicylic	salicylic acid, C ₇ H ₆ O ₃	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2006b
37	thiosemicarb	1-formyl-3-thiosemicarbamide, C ₂ H ₅ N ₃ OS	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2006c
38	thiouracil	2-thiouracil, C ₄ H ₄ N ₂ OS	X-Ray, Mo, 90K	Munshi <i>et al.</i> , 2006b
39	stilbene	Dimethylstilbene	X-Ray, 16K	Ogawa <i>et al.</i> , 2006
40	estradiol	17β-Estradiol Urea, C ₁₈ H ₂₄ O ₂ CH ₄ N ₂ O	X-Ray, Mo, 100K	Parrish <i>et al.</i> , 2006
41	estrone	3-hydroxy-1,3,5-(10)-estratrien-17-one, C ₁₈ H ₂₂ O ₂	X-Ray, Mo, 20K	Zhurova <i>et al.</i> , 2006
42	diazapentalene	1,3,4-trinitro-7,8-diazapentalene, C ₆ H ₃ N ₅ O ₆	X-Ray, Ag, 90K	Chen <i>et al.</i> , 2007
43	sarcosine	Sarcosine, C ₃ H ₇ NO ₂	X-Ray, Mo, 100K	Dittrich <i>et al.</i> , 2007
44	alaproala	L-alanyl-L-prolyl-L-alanine hydrate, C ₁₁ H ₁₉ N ₃ O ₄ H ₂ O	Synchrotron, 100K	Kalinowski <i>et al.</i> , 2007
45	ornithine	L-ornithine hydrochloride, C ₅ H ₁₃ N ₂ O ₂ Cl	X-Ray, Mo, 100K	Dittrich <i>et al.</i> , 2007
46	thymidine	Thymidine, C ₁₀ H ₁₄ N ₂ O ₅	X-Ray, Mo, 20K	Hübschle <i>et al.</i> , 2008
47	thiuronium	2-(indol-3-yl)-1,1,3,3-tetramethylthiuronium nitrate, C ₁₃ H ₁₈ N ₃ S ⁺ NO ₃ ⁻	X-Ray, Mo, 110K	Lutz <i>et al.</i> , 2008
48	FOX-7	1,1-diamino-2,2-dinitroethylene, C ₂ H ₄ N ₄ O ₄	Synchrotron, 0.503 Å, 100K	Meents <i>et al.</i> , 2008
49	RhPCI	[Rh(C ₇ H ₈)(PPh ₃)Cl], C ₂₅ H ₂₃ ClPRh	X-Ray, Mo, 100K	Sparkes <i>et al.</i> , 2008
50	paracetamol	paracetamol, C ₈ H ₉ NO ₂	X-Ray, Mo, 100K	Bouhaida <i>et al.</i> , 2009

51	16Ni	[3,11-Bis(methoxycarbonyl)-1,5,9,13-tetraazacyclohexadeca-1,3,9,11-tetraenato-(2-)- κ^4 N]nickel(II), C ₁₆ H ₂₂ N ₄ O ₄ Ni	X-Ray, Mo, 100K	Domagała <i>et al.</i> , 2009
52	fidarestat	Fidarestat, C ₁₂ H ₁₀ N ₃ O ₄ F	Synchrotron, 0.450 Å, 100K	Fournier <i>et al.</i> , 2009
53	coumcarb	coumarin-3-carboxylic acid, C ₁₀ H ₆ O ₄	X-Ray, Mo, 100K	Howard <i>et al.</i> , 2009
54	transcin	trans-cinnamic acid, C ₉ H ₈ O ₂	X-Ray, Mo, 100K	Howard <i>et al.</i> , 2009

Table 2S. Schematic representation of the selected, available atom-types in the ELMAM2.

Symbol	Chemical connectivity	Schematic representation
H101	Hc[cc]	
H103	Hc[chh]	
H104	Hc[cch]	
H105	Hc[ccc]	
H115	Ho[c]	
C301	C1.5c1.5c1h	
C307	C1.5n1.5c1.5c	
C401	C1c1h1h1h	

C402	C1c1c1h1h	
C410	C1o1c1h1h	
N301	Nchh	
N302	N1.5c1.5c1h	
N304	Nccc	
N305	Nooc	
N401	Nchhh	
O102	O1.5c[1.5n1c]	
O104	O1.5c[1.5o1c]	

O105	O2c[1o(1h)1c]	
O106	O2c[1o(1c)1c]	
O201T	Ohh	
F101T	Fc[cc]	
S201T	Sch	
S202	Scs	

The connectivity is shown without the bond orders. In total 68 atom-types are available; 16 H; 31C; 7N; 11O; 1F; 2S

Table 3S. Benchmarked crystal structures.

Compound	Structure details	Theoretical calculations	Database transfer details	Reference
Glycine	high resolution X-Ray, 23K	Single Point	AMBER (a mixture of C-terminal glycine, N-terminal glycine and bulk glycine used for CA, HA1 and HA2 point charges) ELMAM ELMAM2	Destro, R.; Roversi, P.; Barzaghi, M.; Marsh, R. E. <i>J. Phys. Chem. A</i> , 2000 , <i>104</i> , 1047-1054.
AlaGlyAla anhydrous	X-Ray, 293K	Single Point, X-H distances extended to average neutron values	AMBER ELMAM ELMAM2	Sreekanta Padiyar, G.; Parthasarathy Seshadri, T. <i>Acta Crystallogr., Sect. C</i> , 1996 , <i>52</i> , 1693-1695.
AlaGlyAla monohydrate	X-Ray, 20K	Single Point, X-H distances extended to average neutron values	AMBER ELMAM ELMAM2	Förster, D.; Messerschmidt, M.; Luger, P. <i>Acta Crystallogr., Sect. C</i> , 2005 , <i>61</i> , o420-o421.
AlaProAla monohydrate	high resolution, X-Ray, 100K	Single Point	AMBER ELMAM ELMAM2	Kalinowski, R.; Dittrich, B.; Hübschle, C. B.; Paulmann, C.; Luger, P. <i>Acta Crystallogr., Sect. B</i> , 2007 , <i>63</i> , 753-767.
AlaTyrAla ethanol	X-Ray, 20K	Single Point, X-H distances extended to average neutron values	AMBER ELMAM ELMAM2	Chęcińska, L.; Förster, D.; Morgenroth, W.; Luger, P. <i>Crystallogr., Sect. C</i> , 2006 , <i>62</i> , o454-o457.
benzene	Neutron, 138K	Single Point	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Bacon, G. E.; Curry, N. A.; Wilson, S. A. <i>Proc. R. Soc. London Ser. A</i> , 1964, <i>279</i> , 98-110.
catechol	X-Ray, 100K	Full Optimization	ELMAM (tyrosine aromatic ring used to mimic atom-types)	Fronczek, F. R.; Kim, K. K.; Strongin, R. M.; <i>Priv. Comm.</i> , 2002 .

			ELMAM2	(CATCOL13)
resorcinol	Neutron, 298K	Single Point	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Bacon, G. E.; Jude R. J. <i>Z.Kristallogr., Kristallgeom., Kristallphys., Kristallchem.</i> 1973 , <i>138</i> , 19-40.
p-nitrophenol α-form	high resolution X-Ray, 110K	Single Point	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Kulkarni, G. U.; Kumaradhas, P.; Rao, C. N. R. <i>Chem. Mater.</i> 1998 , <i>10</i> , 3498-3505.
p-nitrophenol β-form	high resolution X-Ray, 110K	Single Point	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Kulkarni, G. U.; Kumaradhas, P.; Rao, C. N. R. <i>Chem. Mater.</i> 1998 , <i>10</i> , 3498-3505.
p-nitroaniline	X-Ray, 123K	hydrogen atoms positions optimized	ELMAM2	Nieger, M. <i>Priv. Comm.</i> 2007 . (NANILI23)
p-nitrobenzoic acid	X-Ray, 123K	hydrogen atoms positions optimized	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Groth, P. <i>Acta Chem. Scand. A</i> 1980 , <i>334</i> , 229-230.
2,5-dihydroxybenzoic acid	X-Ray, 130K	hydrogen atoms positions optimized	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Cohen, D. E.; Benedict, J. B.; Morlan, B.; Chiu, D. T.; Kahr, B. <i>Cryst. Growth Des.</i> 2007 , <i>7</i> , 492-495.
p-dinitrobenzene	Neutron, 120K	Single Point	ELMAM (tyrosine aromatic ring used to mimic atom-types) ELMAM2	Tonogaki, M.; Kawata, T.; Ohba, S.; Iwata, Y.; Shibuya, I. <i>Acta Crystallogr., Sect. B</i> 1993 , <i>49</i> , 1031-1039.
quercetin monohydrate	X-Ray, 100K	Full Optimization	ELMAM2	Domagała, S., Munshi, P., Ahmed, M., Guillot, B. & Jelsch, C. (2011). <i>Acta Crystallogr., Sect. B</i> 2011 , <i>67</i> , 63-78.

Table 4S. Cross-validation statistics for the L-SerVal structure. The average values over the 20 working and test sets are given. All the *R*-factor values are in %.

Model	$wR^2(F)$	$wR^2(F)_{\text{free}}$	$S(F)$	$S(F)_{\text{free}}$
IAM_U _{iso}	4.15	5.20	2.55	2.92
IAM_U _{anis}	5.10	5.33	2.62	2.99
ELMAM_U _{iso}	2.47	2.95	1.52	1.65
ELMAM_U _{anis}	2.52	3.00	1.54	1.67
ELMAM2_U _{iso}	2.53	3.07	1.55	1.71
ELMAM2_U _{anis}	2.40	2.87	1.47	1.60

Electrostatic Potential quantities computed at the van der Waals surface for tripeptides and aromatic molecules.

All the notations used here to describe the descriptors are the same as in the original papers (Murray & Politzer, 1998; Murray *et al.*, 2000).

The quantities: $V_{S,\max}$, $V_{S,\min}$, \bar{V}_S^+ , \bar{V}_S^- , \bar{V}_S and Π are given in kcal/mol; σ_+^2 , σ_-^2 and σ_{tot}^2 are in (kcal/mol)²; ν is dimensionless. V_S^+ and V_S^- designate the regions of positive and negative potential respectively. The following definitions were used to calculate surface quantities:

$$\begin{aligned}\bar{V}_S^+ &= \frac{1}{p} \sum_{i=1}^p V_S^+(\mathbf{r}_i), & \bar{V}_S^- &= \frac{1}{n} \sum_{i=1}^n V_S^-(\mathbf{r}_i), & \bar{V}_S &= \frac{1}{p+n} \sum_{i=1}^{p+n} V_S(\mathbf{r}_i), \\ \Pi &= \frac{1}{p+n} \sum_{i=1}^{p+n} |V_S(\mathbf{r}_i) - \bar{V}_S|, & \sigma_+^2 &= \frac{1}{p} \sum_{i=1}^p [V_S^+(\mathbf{r}_i) - \bar{V}_S^+]^2, & \sigma_-^2 &= \frac{1}{n} \sum_{i=1}^n [V_S^-(\mathbf{r}_i) - \bar{V}_S^-]^2, \\ \sigma_{\text{tot}}^2 &= \sigma_+^2 + \sigma_-^2, & \nu &= \frac{\sigma_+^2 \sigma_-^2}{(\sigma_{\text{tot}}^2)^2}.\end{aligned}$$

Table 5S. Electrostatic Potential quantities computed at the van der Waals surface for glycine and tri-peptides.

Surface quantity	AMBER	ELMAM	ELMAM2	THEO	AMBER	ELMAM	ELMAM2	THEO
glycine	AGA							
$V_{S,max}$	125.1	102.8	86.3	91.4	153.2	119.8	103.0	118.5
$V_{S,min}$	-122.3	-102.0	-77.9	-78.9	-159.2	-138.9	-105.8	-110.0
\bar{V}_S^+	59.7	52.8	44.2	45.0	61.3	49.7	40.4	49.8
\bar{V}_S^-	-63.5	-57.5	-44.7	-46.3	-63.2	-62.9	-49.3	-49.9
\bar{V}_S	1.9	2.1	4.7	5.1	-2.9	-6.6	-0.9	-0.6
Π	61.4	54.8	44.0	45.0	62.2	56.5	44.6	49.8
σ_+^2	1055.0	566.7	467.5	477.8	1360.4	810.7	510.8	697.1
σ_-^2	872.9	635.8	377.7	383.2	1611.9	1555.3	948.2	1032.7
σ_{tot}^2	1927.9	1202.5	845.2	861.0	2972.3	2366.0	1459.1	1729.8
V	0.248	0.249	0.247	0.247	0.248	0.225	0.228	0.241
RMS	68.9	60.2	49.0	50.1	73.3	66.3	52.1	57.9
RMSD	20.7	12.9	2.8		20.4	15.4	8.9	
r	0.992	0.992	0.999		0.979	0.982	0.992	
$\Delta R/R$	0.352	0.234	0.058		0.314	0.248	0.163	

	AGA	H ₂ O			APA	H ₂ O		
$V_{S,max}$	164.1	125.9	109.5	103.9	169.4	121.7	102.9	107.9
$V_{S,min}$	-164.4	-130.4	-101.2	-94.0	-164.4	-130.8	-102.7	-99.8
\bar{V}_S^+	68.8	57.8	48.8	46.6	62.2	48.3	40.9	40.0
\bar{V}_S^-	-62.3	-53.4	-39.3	-36.0	-61.8	-55.7	-42.0	-41.0
\bar{V}_S	0.9	-2.0	2.8	4.0	0.0	-4.2	1.5	2.0
Π	65.5	55.3	44.0	41.3	62.0	52.1	41.4	40.4
σ_+^2	1704.8	953.1	574.4	513.0	1796.3	822.8	556.3	549.3
σ_-^2	1556.8	1038.3	746.6	632.7	1553.1	963.8	667.2	586.5
σ_{tot}^2	3261.7	1991.4	1321.0	1145.7	3349.4	1786.6	1223.5	1135.9
V	0.249	0.250	0.246	0.247	0.249	0.248	0.248	0.250
RMS	76.9	63.8	51.1	47.7	74.3	60.0	48.2	46.9

RMSD	32.2	20.4	6.7	31.0	17.9	4.4
r	0.976	0.980	0.993	0.971	0.981	0.996
$\Delta R/R$	0.531	0.369	0.136	0.525	0.337	0.092

AYA	EtOH			
$V_{S,max}$	165	126.6	105.6	120.6
$V_{S,min}$	-159.7	-124.1	-97.1	-91.5
\bar{V}_S^+	56.1	46.9	35.5	44.6
\bar{V}_S^-	-47.8	-44	-33.8	-32
\bar{V}_S	-1.1	-2.9	1.7	-0.7
Π	51.4	45.1	34.6	37
σ_+^2	1840.6	1143.2	693.7	1013.7
σ_-^2	1419.1	884.8	605.6	540.5
σ_{tot}^2	3259.7	2028	1299.3	1554.2
V	0.246	0.246	0.249	0.227
RMS	65.4	55.3	43.0	46.4
RMSD	24.3	14.8	9.7	
r	0.963	0.974	0.981	
$\Delta R/R$	0.441	0.292	0.216	
all peptides				
<RMS>	71.8	61.1	48.7	49.8
<RMSD>	25.7	16.3	6.5	
<r>	0.976	0.982	0.992	
<$\Delta R/R$>	0.433	0.296	0.133	

Table 6S. Computed Electrostatic Potential surface quantities for the aromatic molecules.

Surface quantity	ELMAM	ELMAM2	THEO	ELMAM	ELMAM2	THEO
benzene	catechol					
$V_{S,max}$	39.3	20.5	18.9	62.7	62.9	60.2
$V_{S,min}$	-56.2	-36.5	-27.2	-59.6	-45.0	-36.3
\bar{V}_S^+	16.2	10.9	9.4	23.5	18.7	16.1
\bar{V}_S^-	-24.9	-15.1	-13.3	-21.9	-18.3	-15.1
\bar{V}_S	-5.2	0.2	0.3	-2.2	0.7	1.0
Π	20.8	12.6	10.9	22.4	18.5	15.6
σ_+^2	87.5	21.3	15.6	209.5	122.7	103.8
σ_-^2	212.2	94.6	66.9	156.6	108.2	68.1
σ_{tot}^2	299.7	115.9	82.5	366.0	230.9	171.9
V	0.207	0.150	0.153	0.245	0.249	0.239
RMS	24.5	14.7	12.6	26.3	21.4	18.2
RMSD	13.5	2.4		11.7	6.0	
r	0.961	0.995		0.935	0.967	
$\Delta R/R$	0.767	0.178		0.536	0.305	

resorcinol	dihydroxybenzoic acid					
$V_{S,max}$	58.7	61.0	55.4	67.9	68.0	63.8
$V_{S,min}$	-58.1	-42.3	-37.4	-55.1	-38.4	-35.9
\bar{V}_S^+	26.3	22.8	21.0	23.6	16.4	18.2
\bar{V}_S^-	-23.9	-19.3	-19.0	-17.6	-12.2	-14.0
\bar{V}_S	-1.5	1.3	0.4	0.7	2.8	2.3
Π	24.9	21.1	20.0	20.4	14.4	16.2
σ_+^2	223.9	123.4	112.7	254.8	158.9	157.9
σ_-^2	174.5	104.9	87.9	134.5	76.5	75.5
σ_{tot}^2	398.4	228.3	200.6	389.3	235.4	233.3
V	0.246	0.248	0.246	0.226	0.219	0.219
RMS	28.7	23.7	22.3	24.7	18.2	19.6
RMSD	10.0	5.1		9.2	5.7	

<i>r</i>	0.955	0.978	0.944	0.957
$\Delta R/R$	0.400	0.222	0.417	0.302

quercetin	nitroaniline			
$V_{S,max}$	72.8	76.9	80.2	68.4
$V_{S,min}$	-60.4	-44.9	-50.5	-51.2
\bar{V}_S^+	18.4	19.6	32.3	24.7
\bar{V}_S^-	-14.2	-13.7	-25.9	-22.1
\bar{V}_S	3.4	2.8	1.3	3.3
Π	16.5	16.8	29.0	23.3
σ_+^2	197.1	208.7	434.4	222.9
σ_-^2	158.2	113.0	157.9	230.2
σ_{tot}^2	355.2	321.7	592.4	453.0
<i>V</i>	0.247	0.228	0.196	0.250
<i>RMS</i>	21.3	21.1	33.7	27.9
<i>RMSD</i>	6.7		10.3	
<i>r</i>	0.950		0.964	
$\Delta R/R$	0.316		0.335	

nitrophenol alpha	Nitrophenol beta					
$V_{S,max}$	97.5	76.4	71.7	101.8	78.4	77.5
$V_{S,min}$	-58.1	-32.2	-42.0	-57.6	-32.1	-41.2
\bar{V}_S^+	29.3	20.3	18.2	29.1	20.5	17.8
\bar{V}_S^-	-26.5	-12.8	-18.6	-26.7	-12.8	-17.9
\bar{V}_S	-1.3	3.5	4.5	-1.3	3.4	4.4
Π	27.7	16.8	17.7	27.7	16.8	17.2
σ_+^2	477.3	209.1	178.1	493.3	214.0	196.2
σ_-^2	183.8	69.8	156.1	181.8	69.2	154.2
σ_{tot}^2	661.1	279.0	334.2	675.1	283.2	350.4
<i>V</i>	0.201	0.188	0.249	0.197	0.185	0.246
<i>RMS</i>	33.0	20.6	22.5	33.1	20.7	22.3
<i>RMSD</i>	21.6	6.0		21	5.8	
<i>r</i>	0.786	0.964		0.806	0.966	

$\Delta R/R$	0.791	0.280	0.772	0.271
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	<i>p</i> -nitrobenzoic acid		dinitrobenzene			
$V_{S,max}$	56.6	48.6	47.7	59.2	48.1	49.5
$V_{S,min}$	-36.2	-32.0	-37.4	-37.2	-22.4	-30.7
\bar{V}_S^+	23.2	15.3	16.3	23.0	16.5	22.2
\bar{V}_S^-	-13.2	-13.4	-18.2	-12.0	-10.9	-16.1
\bar{V}_S	0.6	4.2	5.4	0.9	5.9	7.7
Π	17.1	14.1	15.3	16.3	13.4	18.4
σ_+^2	207.1	117.6	99.4	258.5	103.8	91.0
σ_-^2	57.2	72.3	84.2	90.8	30.0	56.1
σ_{tot}^2	264.2	189.9	183.6	349.3	133.9	147.1
V	0.17	0.236	0.248	0.192	0.174	0.236
RMS	20.6	17.7	19.5	20.9	15.9	21.9
RMSD	17.2	8.2		24.1	6.7	
r	0.652	0.900		0.376	0.969	
$\Delta R/R$	0.857	0.444		1.126	0.358	

All aromatic			
$\langle RMS \rangle$	26.5	20.8	20.8
$\langle RMSD \rangle$	16.0	6.3	
$\langle r \rangle$	0.802	0.961	
$\langle \Delta R/R \rangle$	0.708	0.301	

Supplementary Figures

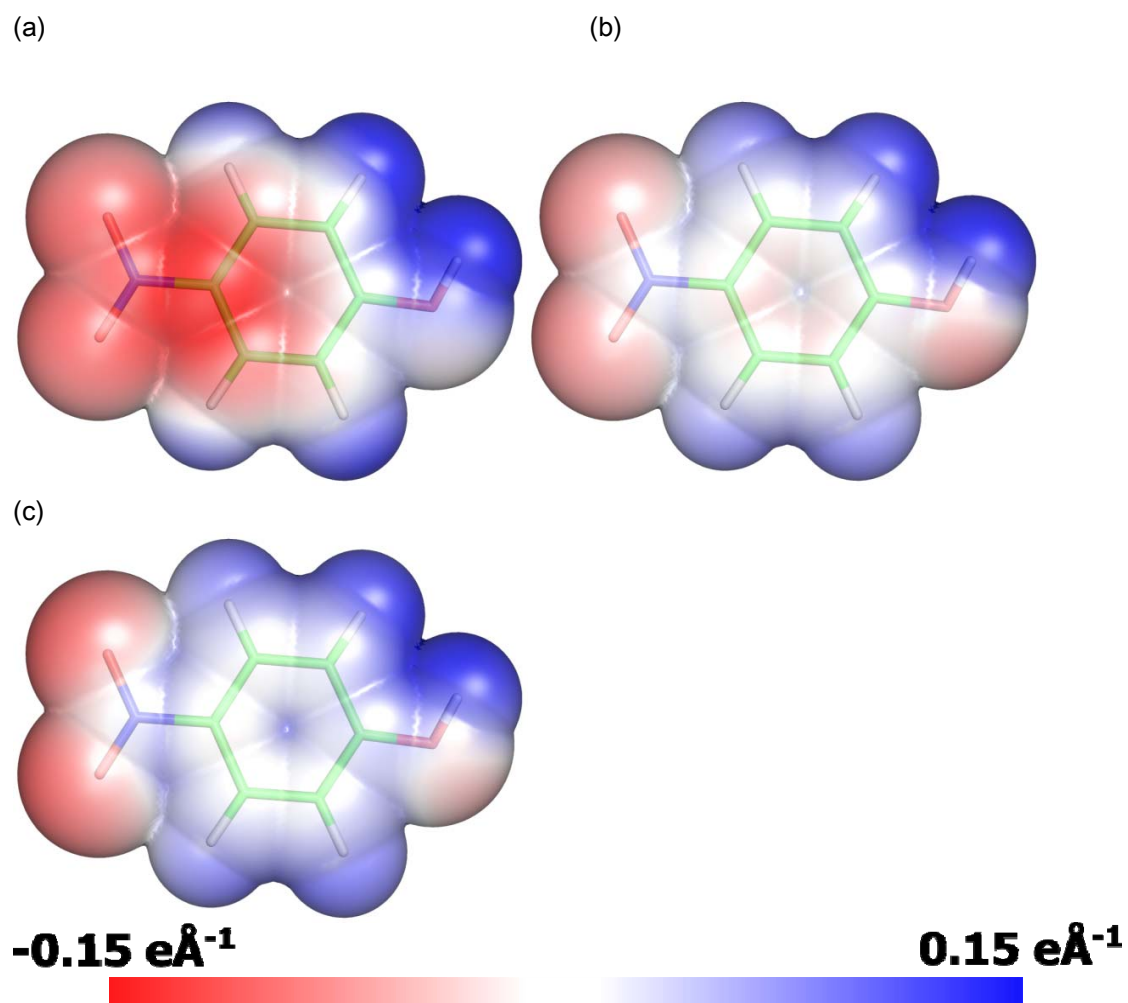


Figure 1S. The electrostatic potential mapped the van der Waals surface in the nitrophenol (alpha) molecule for (a) ELMAM (b) ELMAM2 and (c) THEO_CRYS models. The maximum negative (blue) and positive (red) values of the ESP correspond to -0.15 and $0.15 \text{ e}\text{\AA}^{-1}$ values, respectively. The $\Delta V/V$ values are 0.791 and 0.280 for ELMAM and ELMAM2, respectively.

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